



## A FLAVONE FROM LEAVES OF *ARRABIDAEA CHICA* f. *CUPREA*

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**Key Word Index**—*Arrabidaea chica* f. *cuprea*; Bignoniaceae; Carajurú; 5-methoxylated flavone; carajuflavone.

**Abstract**—A new flavone, 6,7,3',4'-tetrahydroxy-5-methoxyflavone, named carajuflavone, was isolated from the leaves of the Brazilian plant *Arrabidaea chica* f. *cuprea*. The structure was established by spectroscopic analysis.

### INTRODUCTION

The family Bignoniaceae comprises about 120 genera and 650 species that are distributed mainly in tropical America and Africa [1, 2]. The genus *Arrabidaea* occurs in tropical America, from south of Mexico to central Brazil. Although *A. chica* f. *cuprea* has a limited occurrence in the south of Brazil, Paraguay and northeast of Argentina, *A. chica* is very common in the Amazon region and has been widely used as an anti-inflammatory and astringent agent as well as a remedy for intestinal colic, sanguine diarrhoea, leucorrhoea, anaemia and leukaemia [3]. In the past, the chemical constituents of the leaves, which supply a precious red dye, were studied by Chapman *et al.* who identified the 3-desoxyanthocyanidin named carajurin [4]. Later, Harborne [5] and Scogin [1] proposed that the occurrence of this rare pigment in Bignoniaceae is probably restricted to the species *A. chica*. In a previous study of the leaves of *A. chica* (H&B) Verlot f. *cuprea* (Cham.) Sandw. locally called 'carajurú', the presence of anthocyanins, flavonoids, tannins and phytosterols in addition to 7,4'-dihydroxy-5-methoxyflavone (thevetiaflavone) [6] has been reported. In the present paper, the isolation and structural determination of a new flavone from the leaves is described.

### RESULTS AND DISCUSSION

An ethyl acetate-soluble layer of 95% methanol extract of the leaves of *Arrabidaea chica* f. *cuprea* was subjected to column chromatography on silica gel, polyamide and polyclar, successively, to yield **1** and **2**. Compound **1**, obtained as a pale yellow amorphous powder, showed  $[M]^+$  at *m/z* 316.0594 in the high resolution EI-mass spectrum, which corresponds to the molecular formula

$C_{16}H_{12}O_7$  (calcd 316.0583). The UV absorption bands at 275 and 336 nm revealed that **1** was a flavone, and the bathochromic shifts observed by addition of shift reagents such as  $AlCl_3/HCl$  and  $NaOAc/H_3BO_3$  indicated that **1** had free hydroxyl groups at C-7, and C-6 or C-8 on the A ring and an *ortho*-dihydroxyl group on the B ring in a flavone skeleton [7]. The  $^1H$  NMR spectrum showed the presence of four hydroxyls [ $\delta$  8.28, 8.37, 8.63 and 8.92 (1H each, *br s*)] and a methoxyl group [ $\delta$  3.88 (3H, *s*)]. Among them, two hydroxyls and the methoxyl group were located on the A ring, and the remaining two hydroxyl groups were on the B ring, which was also supported by the EI-mass spectrum fragment ions caused by retro-Diels-Alder (RDA) cleavage (*m/z* 181 derived from the A ring and *m/z* 134 from the B ring). The  $^1H$  NMR spectrum further exhibited three protons in a ABX system [ $\delta$  6.98 (1H, *d*, *J* = 8 Hz), 7.41 (1H, *dd*, *J* = 8, 2 Hz) and 7.46 (1H, *d*, *J* = 2 Hz)] and two one-proton singlets [ $\delta$  6.45 and 6.88]. The above data suggested the partial structure of the B ring moiety is a 3',4'-dihydroxyl substituent, which was confirmed by the HMBC experiment (Fig. 1). On the other hand, the methoxyl group appeared in a lower field at  $\delta$  62.3 in the  $^{13}C$  NMR spectrum, indicating that both *ortho*-positions of the methoxyl group were occupied by substituents. The aromatic carbons bearing *O*-function were observed at  $\delta$  137.3, 145.9, 152.2 and 152.7 in the  $^{13}C$  NMR spectrum, indicating the A ring moiety to be a 1,3,4,5-tetraoxogenated benzene. As no chelated hydroxyl group was observed in the  $^1H$  NMR and the UV spectrum, the structure of **1** was either 6,7,3',4'-tetrahydroxy-5-methoxy- or 7,8,3',4'-tetrahydroxy-5-methoxyflavone. In the HMBC spectrum, the methoxyl group ( $\delta$  3.88) was correlated to the aromatic carbon at  $\delta$  145.9 assigned to C-5. The aromatic proton ( $\delta$  6.88) assignable to H-8 was correlated to three aromatic carbons with *O*-function ( $\delta$  137.3, 152.2 and 152.7) in addition to a quaternary

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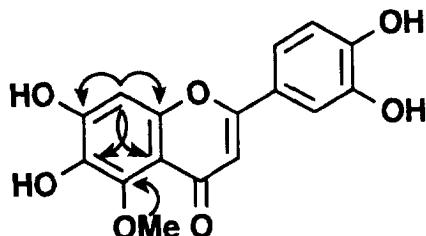


Fig. 1. HMBC spectrum of 1 (correlations observed in the B and C rings were omitted).

aromatic carbon ( $\delta$  112.6, C-10) through  $^2J$  and  $^3J$ . Therefore the methoxyl group was located at the *para*-position (C-5) to the aromatic proton ( $\delta$  6.88). The structure of carajuflavone was, then, concluded to be 6,7,3',4'-tetrahydroxy-5-methoxyflavone (1).

In addition to 1, luteolin (2) was isolated and the structure was elucidated by means of spectral analysis. Biological assays, using the isolated compounds, revealed that 1 had a weak activity on the inhibition of enzyme release and superoxide production by rabbit neutrophils stimulated by formyl-methionyl-leucyl-phenylalanine (FMLP), and 2 showed strong inhibitory effects on these responses of neutrophils (Takemura *et al.*, unpublished data).

## EXPERIMENTAL

**Plant material.** Leaves of *Arrabidaea chica* f. *cuprea* were collected at Alto da Glória district, Curitiba City, Paraná, Brazil in April 1989. The specimen was identified by Dr Gerdt Hatschbach, Botanical Garden of Curitiba City. A voucher specimen is deposited in the Herbarium of Department of Botany, Setor de Ciências Biológicas, Universidade Federal do Paraná, Curitiba City, under registration number n° 20.549.

**Extraction and isolation.** Dried leaves (2.0 kg) were exhaustively extracted with 95% MeOH. The concd extract was partitioned with petrol, toluene,  $\text{CHCl}_3$ , and  $\text{EtOAc}$ , successively. Upon concentration, the  $\text{EtOAc}$  layer (17.8 g) was subjected to CC on silica gel using  $\text{CHCl}_3$ ,  $\text{EtOAc}$  and MeOH as eluent. The  $\text{EtOAc}$  elution (7.29 g) was rechromatographed on silica gel using  $\text{CHCl}_3$ –MeOH– $\text{HCO}_2\text{H}$  (40:10:1). The initial frs I and II were further transferred to another chromatograph on polyamide 200 eluted with  $\text{CH}_2\text{Cl}_2$ –MeOH– $\text{H}_2\text{O}$

(40:10:1). Fr. I was finally submitted to chromatography on polyclar eluted with  $\text{CHCl}_3$ –EtOH–MeCOEt– $\text{Me}_2\text{CO}$  (40:20:5:1) to give 2 (18 mg). The fr. II was also subjected to chromatography over polyclar eluted with  $\text{CHCl}_3$ –MeOH–MeCOEt– $\text{Me}_2\text{CO}$  (20:10:5:1) with gradual increase in MeOH in  $\text{CHCl}_3$  to give 1 (26 mg).

**Compound 1, carajuflavone, 6,7,3',4'-tetrahydroxy-5-methoxyflavone.** A pale yellow amorphous powder [MeOH]. HR-EIMS  $m/z$  316.0594 (calcd 316.0583 for  $\text{C}_{16}\text{H}_{12}\text{O}_7$ ). EIMS  $m/z$  (rel. int.): 316 ( $\text{M}^+$ , 91), 301 (26), 298 (100), 273 (23), 270 (10), 242 (10), 181 (5), 167 (6), 153 (14), 139 (7), 137 (13), 135 (10), 134 (26), 69 (27), 44 (21). UV  $\lambda_{\text{max}}$  (nm, MeOH): 275, 336; + NaOH: 265sh, 296, 403; +  $\text{AlCl}_3$ : 260sh, 315, 370sh, 473; +  $\text{AlCl}_3$ /HCl: 275, 338, 413; + NaOAc: 270, 318sh, 364; + NaOAc/ $\text{H}_3\text{BO}_3$ : 252sh, 283, 356.  $^1\text{H}$  NMR (400 MHz, acetone- $d_6$ )  $\delta$ : 3.88 (3H, s, OMe), 6.45 (1H, s, H-3), 6.88 (1H, s, H-8), 6.98 (1H, d,  $J$  = 8 Hz, H-5'), 7.41 (1H, dd,  $J$  = 8, 2 Hz, H-6'), 7.46 (1H, d,  $J$  = 2 Hz, H-2'), 8.28, 8.37, 8.63 and 8.92 (1H each, br s, OH  $\times$  4).  $^{13}\text{C}$  NMR (100 MHz, acetone- $d_6$ ):  $\delta$  62.3 (MeO), 100.3 (C-8), 106.6 (C-3), 112.6 (C-10), 113.9 (C-2'), 116.6 (C-5'), 119.5 (C-6'), 124.5 (C-1'), 137.3 (C-6), 145.9 (C-5), 146.4 (C-3'), 149.4 (C-4'), 152.2 (C-7\*), 152.7 (C-9\*), 162.1 (C-2), 176.9 (C-4), (\*: interchangeable). All carbons were assigned with the aid of HMBC spectrum.

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