



BIOTRANSFORMATION OF (–)-NOPOL BY *GLOMERELLA CINGULATA*

MITSUO MIYAZAWA, YASUHIRO SUZUKI and HIROMU KAMEOKA

Department of Applied Chemistry, Faculty of Science and Engineering, Kinki University, Kowakae, Higashiosaka-shi, Osaka 577, Japan

(Received in revised form 14 September 1994)

Key Word Index—*Glomerella cingulata*; biotransformation; (–)-nopol; (4*R*)-(–)-4-hydroxynopol; 4-oxonopol; 5-hydroxynopol.

Abstract—The biotransformation of (–)-nopol to (4*R*)-(–)-4-hydroxynopol, 4-oxonopol and 5-hydroxynopol by *Glomerella cingulata* has been demonstrated. The structures of the biotransformation products were determined by spectral methods.

INTRODUCTION

As part of a programme concerned with the use of specific microorganisms for the production of fine chemicals, we have been studying the biotransformation of terpenoids by the plant pathogenic microorganisms, *Glomerella cingulata*, *Rhizoctonia solani* and *Botrytis allii*. In our previous publications we reported that, (+)-cedrol [1], (–)- α -bisabolol [2], 1,8-cineole [3], (–)-globulol [4] and (+)-ledol [4] are transformed to novel terpenes by *G. cingulata*. (–)-Nopol (1) has been isolated from the essential oil of carrot root [5], and is easily synthesized from β -pinene [6]. This paper deals with the microbial oxidation of (–)-nopol (1) to (4*R*)-(–)-4-hydroxynopol (2), 4-oxonopol (3) and 5-hydroxynopol (4) by *G. cingulata*. The results show that 1 is predominantly oxidized at the 4-position.

RESULTS AND DISCUSSION

The time course of metabolite production following the addition of a small amount of (–)-nopol (1) to a culture of *G. cingulata* was monitored by TLC and quantitatively measured by GC (Fig. 1). The results suggested that the metabolic route leading to compounds 2 and 3 was the major one by which compound 1 was metabolized by *G. cingulata* and that the route leading to compound 4 was of minor importance in quantitative terms. The major products, 2 and 3, each accounted for some 40% of the total monoterpenoid content of the cultures after 8 days whereas compound 4 accounted for only ca 5% after the same period of time. In order to isolate these metabolic products (2–4), 1 (3.60 g) was incubated with *G. cingulata* for 8 days. At the end of this time, the culture media and mycelia were extracted with CH_2Cl_2 and the extract (3.92 g) worked up to give 2 (326 mg), 3 (349 mg) and 4 (88 mg).

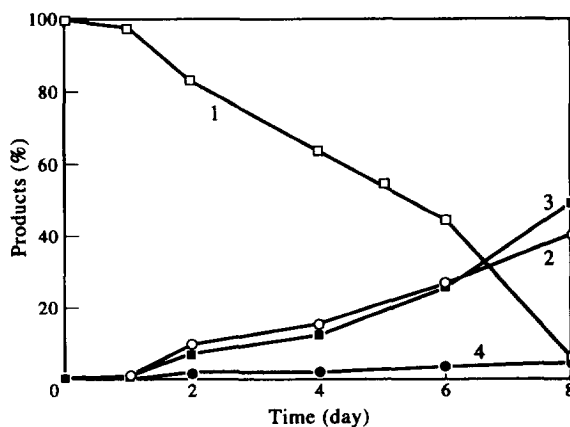
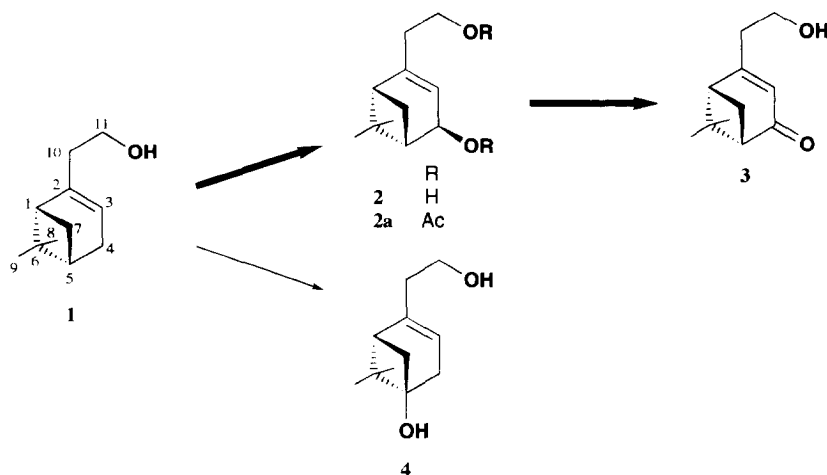


Fig. 1. Time course of metabolic products of (–)-nopol (1) by *G. cingulata*. □: (–)-nopol (1); ○, (4*R*)-(–)-4-hydroxynopol (2); ■: 4-oxonopol (3); ●: 5-hydroxynopol (4).

Compound 2 showed a specific ion peak at m/z 164 $[M - \text{H}_2\text{O}]^+$ in its GC-mass spectrum. The IR spectrum contained a C–O absorption band (1006 cm^{-1}) due to the presence of a secondary alcohol group as well as the primary alcohol absorption band (1049 cm^{-1}) of the parent compound. The ^1H NMR spectral data contained the signals (overlapped) for two hydroxyl groups at δ 1.73 and a methine proton adjacent to the secondary hydroxyl group at δ 4.32. The ^{13}C NMR spectral data showed the signals for a new methine carbon in place of the 4-methylene carbon present in 1. Furthermore, the signals for C-3 and C-5 of 2 were shifted to lower field from those of 1. These spectral data indicated that the new hydroxyl group was located at C-4. The structure of 2 was confirmed by the assignment of the NMR spectral data of its 4, 11-diacetate (2a). The absolute configuration at C-4 was established by comparison of the NMR spectral data with those of the known related compounds,



Scheme 1. The metabolism of (–)-nopol (**1**) by *G. cingulata*.

(–)-*cis*-verbenol and (–)-*trans*-verbenol [7, 8]. The ^1H NMR data showed that the signals for the proton at C-4 of **2**, (–)-*cis*-verbenol and (–)-*trans*-verbenol were at δ 4.32, 4.45 and 4.25, and the signals for the proton at C-8 were at δ 0.88, 1.07 and 0.87, respectively. The ^{13}C NMR data showed that the signals for C-4 of **2**, (–)-*cis*-verbenol and (–)-*trans*-verbenol were at δ 70.2, 73.3 and 70.3, and those for C-6 were at δ 46.3, 39.0 and 46.1, respectively. Consideration of the above data suggested that the chemical shifts of **2** were very similar to those of (–)-*trans*-verbenol. In addition, the other signals for C-2, C-5, C-7, C-8 and C-9 of **2** were almost identical with those of (–)-*trans*-verbenol. These results allowed us to determine the absolute configuration of C-4 as *R*. The optical rotation of **2** had a large laevorotatory value (–121.4°). Therefore, compound **2** was determined to be (4*R*)-(–)-4-hydroxynopol.

The metabolic product **3** contained specific ion peaks at m/z 180 $[\text{M}]^+$, 165 $[\text{M} - \text{Me}]^+$ and 149 $[\text{M} - \text{CH}_2\text{OH}]^+$ in the GC-mass spectrum. The IR spectrum showed a strong absorption band at 1665 cm^{-1} (C=O) due to the presence of a carbonyl group. The ^{13}C NMR spectral data also indicated the presence of a newly introduced carbonyl group in place of the 4-methylene group of **1**. In addition, the signals for H-3, H-5 and H-7 in the ^1H NMR spectrum and those for C-2, C-3, C-5, C-6, and C-7 in the ^{13}C NMR spectrum were shifted to lower field than the corresponding signals of **1**. These spectral data established that compound **3** was 4-oxonopol.

The mass spectral data of the minor metabolic product **4** showed specific ion peaks at m/z 182 $[\text{M}]^+$, 167 $[\text{M} - \text{Me}]^+$ and 137 $[\text{M} - \text{CH}_2\text{CH}_2\text{OH}]^+$. The IR spectrum of **4** contained the novel absorption at 1134 cm^{-1} (C–O) due to the presence of a tertiary hydroxy group. The ^1H NMR spectral data contained the signals (overlapped) for two hydroxyl groups at δ 1.68, however, there were no signals for a proton adjacent to a newly introduced hydroxy group. This indicated the

presence of a tertiary alcohol group in **4**. The ^{13}C NMR data showed the presence of a quaternary carbon instead of the 4-methine C present in **1**. In addition, movement of the chemical shifts between **1** and **4** was observed for the 4-position but not the 2-position. Therefore, compound **4** was shown to be 5-hydroxynopol.

This study established that compound (–)-**1** is oxidatively biotransformed to compounds **2–4** by the fungus *G. cingulata* (Scheme 1). The oxidations were confined mainly to the 4-position to give (4*R*)-(–)-4-hydroxy- and 4-oxo-nopol. This enantioselective hydroxylation might be due to steric hindrance by the geminal dimethyl group at C-6 dictating that the hydroxyl group is incorporated from the opposite side of the molecule to the dimethyl group.

EXPERIMENTAL

General. (–)-Nopol was purchased from Fluka Chem. ^1H and ^{13}C NMR: 270.05 and 67.80 MHz, respectively. GC-MS: 20 eV (ion voltage), 250° (ion source), OV-1 (0.25 mm \times 30 m) capillary column; TLC: silica gel 60 F₂₅₄ pre-coated (layer thickness 0.25 mm, Merck); CC: silica gel with *n*-hexane–EtOAc gradients.

Cultivation of *G. cingulata*. Spores of *G. cingulata* (provided by Dr M. Hyakumachi, Gifu University) which had been preserved at low temp. were inoculated into sterilized culture media in an Erlenmeyer flask and shaken at 27° for 2 days. The components of the culture medium (g/250 ml) were: sucrose 3.75, glucose 3.75, polypeptone 1.25, $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ 0.125, KCl 0.125, K_2HPO_4 0.25, $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ 0.0025. The mycelia were then transplanted in to petri dishes which contained 15 ml of the same sterilized culture media (pH 7.2) and incubated at 27° without shaking for 3 days. After growth of *G. cingulata*, **1** (3.60 g) was added directly to the medium (15 mg/15 ml) and the cultures further incubated under the same conditions for 8 days.

Table 1. ¹H NMR spectral data for (–)-nopol (1) and its metabolic products (2–4) and derivative 2a (270.05 MHz, CDCl₃, TMS as int. standard)

H	1	2	2a	3	4
1	2.04 ddd (1.5, 5.8, 5.8)	2.12 ddd (1.5, 5.3, 5.5)	2.15 ddd (1.2, 5.3, 5.5)	2.52 ddd (1.3, 5.8, 5.8)	2.00 dd (2.0, 6.3)
3	5.34 m	5.46 m	5.42 m	5.80 dd (1.3, 1.5)	5.48 m
4	2.24 m	4.32 dd (2.8, 3.0)	5.37 dd (2.8, 3.0)	—	2.24–2.31 m
5	2.15 m	2.20 dddd (1.5, 3.0, 5.5, 5.7)	2.24 dddd (1.5, 3.0, 5.5, 5.5)	2.68 ddd (1.5, 5.8, 5.8)	—
7 _a	1.15 d (8.5)	1.32 d (7.8)	1.45 d (8.5)	2.10 d (9.5)	1.59 d (8.2)
7 _b	2.39 ddd (5.8, 5.8, 8.5)	2.30 ddd (5.3, 5.7, 7.8)	2.33 ddd (5.3, 5.5, 8.5)	2.84 ddd (5.8, 5.8, 9.5)	2.34 ddd (0.8, 6.3, 8.2)
8	0.85 s	0.88 s	0.91 s	1.02 s	0.87 s
9	1.28 s	1.35 s	1.35 s	1.51 s	1.23 s
10	2.24 dt (1.0, 6.0)	2.28 t (6.5)	2.37 ddd (1.0, 6.5, 7.0)	2.55 t (6.5)	2.26 dt (1.0, 6.5)
11	3.61 dt (1.0, 6.0)	3.67 t (6.5)	4.08 ddd (6.5, 7.0, 11.5)	3.82 t (6.5)	3.64 dt (1.0, 6.5)
4-OH	—	1.73 br s	—	—	—
5-OH	—	—	—	—	1.68 br s
11-OH	1.64 br s	1.73 br s	—	1.73 br s	1.68 br s
4-COMe	—	—	2.03 ^a s	—	—
11-COMe	—	—	2.04 ^a s	—	—

Coupling constants in Hz.

^a Values are interchangeable within each column.Table 2. ¹³C NMR spectral data for (–)-nopol (1) and its metabolic products (2–4) and derivative 2a (67.80 MHz, CDCl₃, CHCl₃, as int. standard)

C	1	2	2a	3	4
1	45.6 d	46.8 d	46.2 ^b d	48.7 d	40.8 d
2	144.7 s	148.7 s	149.9 s	170.5 s	143.7 s
3	119.3 d	121.1 d	117.4 d	121.5 d	121.8 d
4	31.3 t	70.2 d	73.3 d	204.2 s	38.0 ^f t
5	40.7 d	47.1 ^a d	44.4 ^b d	57.8 d	75.0 s
6	37.9 s	46.3 s	46.2 s	54.2 s	44.7 s
7	31.7 t	28.9 t	29.6 t	41.2 ^e t	39.4 ^f t
8	21.1 q	20.8 q	21.4 q	22.2 q	18.6 q
9	26.2 q	26.6 q	26.4 q	26.5 q	20.7 q
10	40.2 t	39.7 t	35.4 t	40.0 ^e t	41.2 ^f t
11	59.9 t	60.0 t	62.0 t	59.5 t	60.1 t
4-COMe	—	—	20.9 ^c q	—	—
4-COMe	—	—	170.9 ^d s	—	—
11-COMe	—	—	20.8 ^c q	—	—
11-COMe	—	—	170.9 ^d s	—	—

^{a–f} Values are interchangeable within each column.

Purification of the metabolic products (2–4). After incubation, the culture media were collected, acidified to pH 2 with HCl, saturated with NaCl and extracted with CH₂Cl₂ for 3 days. The mycelia were also collected and extracted with CH₂Cl₂ for 3 days. Both CH₂Cl₂ extracts were mixed, and the solvent evapd under red. press. The extract (3.92 g) was dissolved in CH₂Cl₂ and separated into neutral and acid fractions in the usual manner. The neutral fraction (3.55 g) was chromatographed over silica gel with a hexane–EtOAc gradient repeatedly to give the metabolic products **2** (326 mg), **3** (349 mg) and **4** (88 mg).

(4R)-(–)-*Hydroxynopol* (**2**). Crystal; mp. 111.5–112.4°; $[\alpha]_D^{20}$ –121.4° (MeOH; *c* 1.0); EIMS *m/z* (rel. int.): 164 [M – H₂O]⁺ (17), 149 (25), 133 (30), 131 (36), 119 (46), 104 (45), 91 (100), 69 (36), 59 (24), 43 (47), 40 (67); IR ν_{\max} cm^{–1}: 3255, 2929, 1440, 1049, 1006, 929, 858, 766, 652; ¹H and ¹³C NMR: Tables 1 and 2.

(4R)-(–)-4,11-*Diacetoxy-4-hydroxynopol* (**2a**). Compound **2** (30 mg) was acetylated in the usual manner to yield **2a** (43 mg). Oil; $[\alpha]_D^{20}$ –125.9° (CHCl₃; *c* 1.0); EIMS *m/z* (rel. int.): 224 (0.8), 223 [M – Ac]⁺ (0.4), 181 (0.7), 180 (0.2), 163 (17), 147 (47), 131 (75), 121 (52), 105 (91), 91 (56), 43 (100); IR ν_{\max} cm^{–1}: 2938, 1741, 1472, 1370, 1240, 1087, 1019, 972; ¹H and ¹³C NMR: Tables 1 and 2.

4-*Oxonopol* (**3**). Oil; $[\alpha]_D^{20}$ –78.2° (CHCl₃; *c* 0.5); EIMS *m/z* (rel. int.): 180 [M]⁺ (42), 165 (58), 149 (33), 147 (50), 135 (48), 121 (65), 91 (71), 79 (80), 55 (82), 41 (100); IR ν_{\max} cm^{–1}: 3411, 2950, 1665, 1610, 1042, 986, 865, 748; ¹H and ¹³C NMR: Tables 1 and 2.

5-*Hydroxynopol* (**4**). Oil; $[\alpha]_D^{20}$ –45.0° (CHCl₃; *c* 0.30); EIMS *m/z* (rel. int.): 182 [M]⁺ (2), 167 (2), 164 (2), 149 (10), 137 (11), 126 (34), 121 (34), 95 (48), 91 (48), 79 (45), 43 (100), 41 (85); IR ν_{\max} cm^{–1}: 3326, 2943, 1723, 1667, 1134, 1042, 755, 659; ¹H and ¹³C NMR: Tables 1 and 2.

REFERENCES

1. Miyazawa, M., Nankai, H. and Kameoka, H. (1993) *Chem. Express* **8**, 573. [*Chem. Abs.* (1993), **119**, 177384e].
2. Miyazawa, M., Nankai, H. and Kameoka, H. (1993) *Chem. Express* **8**, 401. [*Chem. Abs.* (1993), **119**, 137467c].
3. Miyazawa, M., Nakaoka, H. and Kameoka, H. (1991) *Chem. Express* **6**, 667. [*Chem. Abs.* (1993), **116**, 251829k].
4. Miyazawa, M., Uemura, T. and Kameoka, H. (1994) *Phytochemistry* **37**, 1027.
5. Alabran, D. M., Moskowitz, H. R. and Mabrouk, A. F. (1975) *J. Agric. Food Chem.* **23**, 229.
6. Bain, J. P., Best, A. H., Clark, C. K. and Hampton, B. L. (1946) *J. Am. Chem. Soc.* **68**, 638.
7. Abraham, R. J., Cooper, M. A., and Salmon a taker, J. R. (1972) *Org. Magn. Res.* **4**, 489.
8. Coxon, J. M., Hydes, G. J. and Steel, P. J. (1984) *J. Chem. Soc. Perkin Trans. II* 1351.