



STRUCTURE REVISION OF A FURANOCOUMARIN FROM *DORSTENIA CONTRAJERVA*

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Abstract—Two furanocoumarins have been isolated from *Dorstenia contrajerva* L.: bergapten and 4-[[3-(4,5-dihydro-5,5-dimethyl-4-oxo-2-furanyl)-butyl]oxy]-7H-furo[3,2-g][1]benzopyran-7-one. The structures were established by UV, NMR and mass spectrometry. The structure of the second compound, an unusual furanocoumarin, was confirmed by X-ray analysis. This substance has already been isolated from *D. contrajerva*, but the structure was incorrectly elucidated.

INTRODUCTION

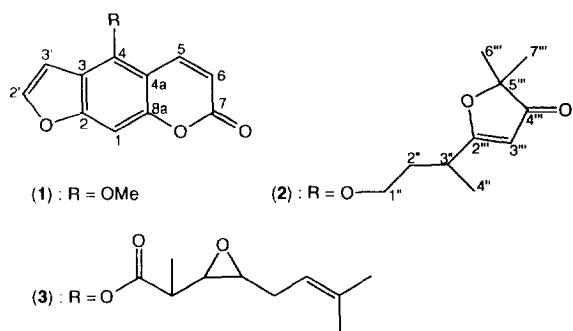
During the course of our systematic search for new compounds of natural origin, Panamanian medicinal plants have been screened. *Dorstenia contrajerva* L. is a herbaceous plant reported to be used by the Kuna Indians of Panama as a cold remedy [1] and for the treatment of snakebites and muscle aches [2]. In the screening for biological activity, the dichloromethane extract did not show antifungal properties but exhibited a slight larvicidal activity against *Aedes aegypti*, the vector of yellow fever, [3] and antimicrobial activity against *Bacillus subtilis* [4]. But in fact, the interest devoted to this plant was due to the HPLC-UV analysis of the extract. The chromatogram of the dichloromethane extract showed a simple feature with only two main peaks having typical UV spectra of furanocoumarins. These two compounds were isolated and their structures determined by spectral methods as bergapten (1) and 4-[[3-(4,5-dihydro-5,5-dimethyl-4-oxo-2-furanyl)-butyl]oxy]-7H-furo[3,2-g][1]benzopyran-7-one (2). X-Ray analysis of 2 confirmed its structure and also showed the presence of a racemic mixture. Another furanocoumarin with a monoterpenoid side chain (3) has previously been described as a major constituent of *D. contrajerva* [5]. However, this product was not detected in our plant sample, but comparison of the data of both compounds

indicated a full correlation. Thus, the structure of furanocoumarin 3 was revised to that of 2.

RESULTS AND DISCUSSION

Separation of the dichloromethane extract of *D. contrajerva* on a silica gel open column yielded 1 and 2 after crystallization. The compounds were purified by recrystallization to give needles and small prisms, respectively. Compound 1 was rapidly identified from ¹H and ¹³C NMR data and by comparison with an authentic sample as bergapten.

The structure of the side chain of the other furanocoumarin (2) was elucidated by ¹H and ¹³C NMR, 2D-COSY NMR, mass spectrometry and confirmed by X-ray crystallography, which fixed the conformational



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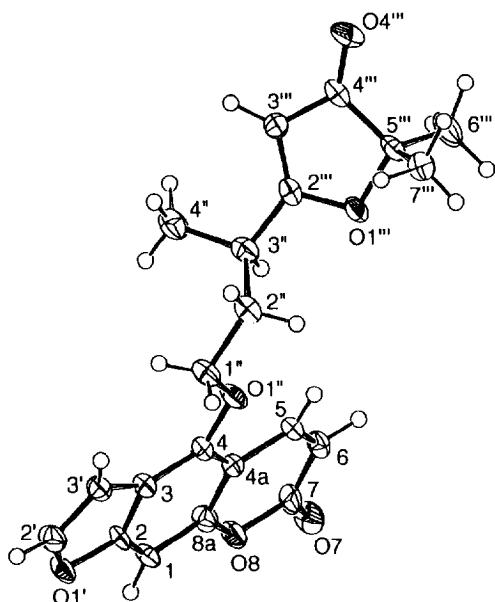


Fig. 1. Molecular formula of **2** as determined by X-ray analysis.

form of **2** (Fig. 1 gives the ORTEP plot of **2**). It should be noted that the X-ray analysis of **2** showed the presence of a racemic mixture and this was confirmed by the determination of the $[\alpha]_D$ value as -3.3° , a value quite near to 0.

This compound was already known from *Dorstenia cayapia* [6] and all the NMR data were in complete agreement with those indicated in this reference. Recently, **2** was also isolated from the rhizomes of another species of *Dorstenia*: *D. brasiliensis* [7], but an $[\alpha]_D$ value of +1.4° was reported.

In the EI mass spectrum of **2**, the molecule (m/z 368) is fragmented into two parts: m/z 202 and 167. The ions at m/z 202, 174 and 145 can be attributed unequivocally to the furanocoumarin moiety [8]. The other peaks at m/z 139 and 111, together with the signal at m/z 167, belong to the side chain of the molecule. These values are explained by the loss of two C_2H_4 groups successively leaving the 4,5-dihydro-5,5-dimethyl-4-oxo-2-furanyl cation [7]. The next step of the fragmentation is probably the loss of an isopropyl group from the position 5'', giving a unit with m/z 69 containing the two oxygen atoms.

A previous study of *D. contrajerva* [5], collected at the Fairchild Tropical Garden in Miami, FL, yielded another furanocoumarin (**3**), with an epoxide-containing monoterpenoid side chain (5-EDOP, 5-[3,4-epoxy-2,7-dimethyl-6,7-octenoyl]psoralen). This compound, an isomer (*m/z* 368) of **2**, was isolated by preparative TLC from the chloroform-soluble fraction of the methanol extract. Surprisingly, even using the most modern analytical techniques (including LC-TSP-MS), this furanocoumarin was not detected in the dichloromethane extract or the methanol extract of our plant sample. By comparing the physical and spectral data of

3 (mp, UV, MS, ^1H NMR) mentioned in the literature with the original data of **2**, we found that everything was identical except one difference in the ^1H NMR spectrum. Consultation of the original ^1H and ^{13}C NMR spectra of **3** showed that they were strictly the same as those of **2**. As a result, we ascertain that the structure of the furanocoumarin described by Swain *et al.* [5] is not correct and should be replaced by the structure of **2**. The mistake was due to an error in interpreting the ^1H NMR spectrum and to the low sensitivity of the ^{13}C NMR spectrum.

Methanol and dichloromethane extracts were tested in our different benchtop screening assays. The dichloromethane extract exhibited a slight larvicidal activity and antibacterial properties. The larvicidal effect was tested against larvae of *Aedes aegypti* in a routine test [3] and the dichloromethane extract was active (100% of larvae killed in 24 hr) at a concentration of 125 ppm. All the fractions were also tested. Three of them were active, but less than the total extract. Both **1** and **2** were inactive. The antibacterial test against *Bacillus subtilis* was performed by direct bioautography [4]. The dichloromethane extract was active with a 30 μ g sample deposited on the TLC plate. However, the active constituents exhibited tailing on TLC and were not identified in this work.

Furthermore, geiparvarin, a closely related 7-substituted coumarin with the same monoterpenoid side chain, was demonstrated to inhibit *in vitro* human carcinoma of the nasopharynx [9]. Consequently, **2** was tested at the NCI for its potential antitumour properties. But our compound did not show significant activity on the different strains of cancer cells tested.

EXPERIMENTAL

General. Mps: uncorr., measured on a Mettler FP 80/82 apparatus. EI-mass spectra were recorded on a Finnigan MAT TSQ 700 triple-stage quadrupole MS instrument. ^1H and ^{13}C NMR spectra were recorded at 200.03 and 50.30 MHz, respectively, in CDCl_3 with TMS as int. standard.

Plant material. The whole plant of *Dorstenia contrajerva* L. was collected in July 1992 in Parque Soberania, Canal Area, Panama and identified by Prof. Mireya D. Correa (Smithsonian Tropical Research Institute, Panama). A voucher specimen is deposited at the Herbarium of the University of Panama.

Extraction and isolation. The powder of the whole plant (230 g) was extracted successively with CH_2Cl_2 and MeOH (3×1 l of each). The CH_2Cl_2 extract was fractionated on a silica gel Si 60 open column (60×5 cm) using a step-gradient of petrol-EtOAc (4:1–1:2), EtOAc and finally MeOH to give 11 fractions (A1–A11). Bergapten (**1**) crystallized from fr. A7 and crystals of **2** were obtained from fr. A10. Both compounds were purified by recrystallization in a mixture of *n*-hexane and EtOAc.

Bergapten (**1**). Mp 190–191°. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 225 (4.03), 251 (4.06), 259 (4.06), 267 (4.05), 308 (4.05). ^1H and

¹³C NMR data in accord with ref. [7]. EI-MS: *m/z* 216 [M]⁺.

4-[3-(4,5-dihydro-5,5-dimethyl-4-oxo-2-furanyl)-butyl]oxy]-7H-furo[3,2-g][1]benzopyran-7-one (2). Mp 141–142°. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 222 (4.03), 249 (4.18), 260 (4.17), 267 (4.17), 309 (4.07). [α]_D – 3.3° (CHCl₃; *c* 0.5). ¹H NMR (200.03 MHz, CDCl₃): δ 1.33 (3H, *d*, *J* = 7 Hz, H-4''), 1.31 (3H, *s*, H-6''), 1.33 (3H, *s*, H-7''), 2.14 (2H, *m*, H-2''), 2.98 (1H, *q*, *J* = 7 Hz, H-3''), 4.46 (2H, *m*, H-1''), 5.38 (1H, *s*, H-3'''), 6.23 (1H, *d*, *J* = 10 Hz, H-6), 6.86 (1H, *d*, *J* = 2.4 Hz, H-3'), 7.08 (1H, *s*, H-1), 7.56 (1H, *d*, *J* = 2.4 Hz, H-2'), 8.08 (1H, *d*, *J* = 10 Hz, H-5). ¹³C NMR (50.30 MHz, CDCl₃): δ 17.8 (C-4''), 22.8 (C-6'', C-7''), 32.3 (C-3''), 34.2 (C-2''), 70.1 (C-1''), 88.5 (C-5''), 94.1 (C-1), 100.1 (C-3''), 104.7 (C-3'), 106.6 (C-4a), 112.7 (C-6), 113.3 (C-3), 138.9 (C-5), 145.0 (C-2'), 148.4 (C-8a), 152.5 (C-4), 158.1 (C-2), 160.9 (C-7), 194.0 (C-2''), 207.1 (C-4''). EI-MS: *m/z* 368 [M]⁺ (100), 202 (17), 174 (14), 167 (33), 139 (98), 111 (8), 83 (12), 69 (13).

X-Ray analysis. Suitable crystals of 2 were grown from hexane–EtOAc as prisms. Intensity data were collected at –80° on a Stoe AED2 4-circle diffractometer using MoK_α graphite monochromated radiation (λ = 0.7073 Å) with $\omega/2\Theta$ scans in the 2Θ range 3–50°. The structure was solved by Direct Methods using the NRCVAX system [10], which was used for all further calculations. The H-atoms were refined isotropically and the non-hydrogen atoms anisotropically using weighted full-matrix least-squares. Crystal data for 2: C₂₁H₂₀O₆, triclinic, space group *P*1, *a* = 9.015 (3), *b* = 9.960 (3), *c* = 10.934 (3) Å, α = 76.88 (2), β = 68.99 (2), γ = 81.13 (2)°, *V* = 889.7 (5) Å³, *Z* = 2. 1990 observed reflections [*I* > 3σ(*I*)], final *R* = 0.061, *R*_W = 0.087. Further details of the crystal structure investigation can be obtained from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ (U.K.) on quoting the full journal citation.

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REFERENCES

1. Morton, J. F. (1981) *Atlas of Medicinal Plants of Middle America*. Charles C. Thomas, Springfield, IL.
2. Gupta, M. P. (1995) in *Phytochemistry of Plants Used in Traditional Medicine* (Hostettmann, K., Marston, A., Maillard, M. and Hamburger, M., eds). Oxford University Press, Oxford (in press).
3. Cepleanu, F. (1993) Ph.D. Thesis. University of Lausanne.
4. Rahalison, L., Hamburger, M., Hostettmann, K., Monod, M. and Frenck, E. (1991) *Phytochem. Anal.* **2**, 199.
5. Swain, L. A., Quirke, J. M. E., Winkle, S. A. and Downum, K. R. (1991) *Phytochemistry* **30**, 4196.
6. Llabres, G., Baiwir, M., Vilegas, W., Pozetti, G. L. and Vilegas, J. H. Y. (1992) *Spectrochim. Acta* **48A**, 1347.
7. Kuster, R. M., Bernardo, R. R., Da Silva, A. J. R., Parente, J. P. and Mors, W. B. (1994) *Phytochemistry* **36**, 221.
8. Murray, R. D. H., Mendez, J. and Brown, S. A. (1992) *The Natural Coumarins*. Wiley, New York.
9. Jerris, P. J. and Smith, A. B. (1981) *J. Org. Chem.* **46**, 577.
10. Gabe, E. J., Le Page, Y., Charland, J.-P., Lee, F. L., White, P. S. (1989) *J. Appl. Crystallogr.* **22**, 384.