

ANTHRAQUINONES FROM *GALIUM SINAICUM*

ALI A. EL-GAMAL, KOICHI TAKEYA, HIDEJI ITOKAWA,\* AHMED F. HALIM,\*† MOHAMED M. AMER,†  
HASSAN-ELRADY A. SAAD† and SABRI A. AWAD†

Tokyo College of Pharmacy, 1432-1 Horinouchi, Hachioji, Tokyo 192-03, Japan; †Faculty of Pharmacy, University of Mansoura, El-Mansoura-35516, Egypt

(Received in revised form 18 January 1995)

**Key Word Index**—*Galium sinaicum*; Rubiaceae; roots; anthraquinones; bianthraquinone; cytotoxic activity.

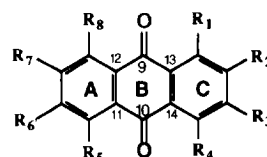
**Abstract**—The new anthraquinones, 7-methylanthragallol 1,3-dimethyl ether, 7-methyl-anthragallol 2-methyl ether, 6-methylanthragallol 3-methyl ether, 8-hydroxyanthragallol 2,3-dimethyl ether, 7-formylanthragallol 1,3-dimethyl ether, copareolatin 5,7-dimethyl ether, copareolatin 6,7-dimethyl ether, 6-methoxylucidin  $\omega$ -ethyl ether and 6-hydroxyxanthopurpurin, as well as a novel bianthraquinone, bisinaiquinone [10,2'-bi(9-hydroxy-3-methyl-1,4-anthraquinonyl)] were isolated from the roots of *Galium sinaicum*. Their structures were established by spectroscopic techniques including 2D-NMR. In addition, eight known anthraquinones were also isolated and fully characterized.

## INTRODUCTION

The genus *Galium* has long been known to contain substantial amounts of anthraquinones, with the roots being especially rich sources of these secondary metabolites. Alizarin-, anthragallol-, lucidin-, munjistin-, purpurin-, xanthopurpurin-, and rubiadin-derived anthraquinones and others are reported [1–3]. In a previous study [3], the new anthraquinones, 6-hydroxyanthragallol 1,3-dimethyl ether and 6-(or 7-)hydroxymethylanthragallol 1,3-dimethyl ether, were isolated from *G. sinaicum* roots. In addition, alizarin 1-methyl ether, and anthragallol 1,3-dimethyl ether were also isolated and fully characterized. We have now reinvestigated the roots of the title species for their anthraquinone content.

## RESULTS AND DISCUSSION

From the cytotoxic *n*-hexane extract (P388 cell line,  $IC_{50}$ ,  $3.8 \mu g ml^{-1}$ ) of *G. sinaicum* roots, 18 anthraquinones were isolated. In addition to the known ones, quinizarin (11) [2, 4], 2-methylquinizarin (12) [5], alizarin 2-methyl ether (13) [5], lucidin  $\omega$ -ethyl ether (14) [6], soranjidiol (15) [7], anthragallol 2-methyl ether (16) [8, 9], copareolatin 6-methyl ether (17) [10] and 7-methyl-8-hydroxyanthragallol 1,3-dimethyl ether (18) [11], 10 new compounds were isolated and spectroscopically characterized as 7-methylanthragallol 1,3-dimethyl ether 1, 7-methylanthragallol 2-methyl ether 2, 6-methyl-anthragallol 3-methyl ether (3), 8-hydroxyanthragallol 2,3-dimethyl ether (4), 7-formylanthragallol 1,3-dimethyl ether (5), copareolatin 5,7-dimethyl ether (6),



- (1)  $R_1 = R_3 = OMe$ ,  $R_2 = OH$ ,  $R_7 = Me$ , others = H
- (2)  $R_1 = R_3 = OH$ ,  $R_2 = OMe$ ,  $R_7 = Me$ , others = H
- (3)  $R_1 = R_2 = OH$ ,  $R_3 = OMe$ ,  $R_6 = Me$ , others = H
- (4)  $R_1 = R_8 = OH$ ,  $R_2 = R_3 = OMe$ , others = H
- (5)  $R_1 = R_3 = OMe$ ,  $R_2 = OH$ ,  $R_7 = CHO$ , others = H
- (6)  $R_1 = R_6 = OH$ ,  $R_2 = Me$ ,  $R_5 = R_7 = OMe$ , others = H
- (7)  $R_1 = R_5 = OH$ ,  $R_2 = R_6 = R_7 = OMe$ , others = H
- (8)  $R_1 = R_3 = OH$ ,  $R_2 = CH_2-O-CH_2-Me$ ,  $R_6 = OMe$ , others = H
- (9)  $R_1 = R_3 = R_6 = OH$ , others = H
- (10) Bianthraquinone, see Fig. 2.
- (11)  $R_1 = R_4 = OH$ , others = H
- (12)  $R_1 = R_4 = OH$ ,  $R_2 = Me$ , others = H
- (13)  $R_1 = OH$ ,  $R_2 = OMe$ , others = H
- (14)  $R_1 = R_3 = OH$ ,  $R_2 = CH_2-O-CH_2-Me$ , others = H
- (15)  $R_1 = R_6 = OH$ ,  $R_2 = Me$ , others = H
- (16)  $R_1 = R_3 = OH$ ,  $R_2 = OMe$ , others = H
- (17)  $R_1 = R_5 = R_7 = OH$ ,  $R_2 = Me$ ,  $R_6 = OMe$ , others = H
- (18)  $R_1 = R_3 = OMe$ ,  $R_2 = R_8 = OH$ ,  $R_7 = Me$ , others = H
- (19)  $R_1 = R_3 = OMe$ ,  $R_2 = OH$ ,  $R_7 = CH_2OH$ , others = H

copareolatin 6,7-dimethyl ether (7), 6-methoxylucidin  $\omega$ -ethyl ether (8), 6-hydroxyxanthopurpurin (9) and 10,2'-bi(9-hydroxy-3-methyl-1,4-anthraquinonyl), named bisinaiquinone (10) whose structure elucidation is described herein. On the other hand, the previously reported 6-(or 7-) hydroxymethylanthragallol 1,3-dimethyl

\*Authors to whom correspondence should be addressed.

ether [3] was confirmed to be the 7-hydroxymethyl compound **19** by the two or three bond correlations from H-4 to C-2, C-10, C-13, from H-5 to C-7, C-10, C-12, from the hydroxymethyl protons to C-6, C-7, C-8, and from H-8 to C-6, C-9, C-11 in the HMBC spectrum.

Compound **1** showed IR bands at 3310, 1167  $\text{cm}^{-1}$  of a free hydroxyl and unchelated carbonyl groups, respectively. The EI-mass spectrum displayed a very weak  $[\text{M}]^+$  at  $m/z$  298, confirmed by FAB, consistent with the molecular formula  $\text{C}_{17}\text{H}_{14}\text{O}_5$ . The  $^1\text{H}$ NMR spectrum (Table 1) showed signals for one methyl at  $\delta$ 2.51, two methoxyl at  $\delta$ 4.03 and 4.08 and a phenolic  $\beta$ -hydroxyl group at  $\delta$ 6.3. A trisubstituted ring C was indicated by the aromatic proton singlet at  $\delta$ 7.7 (H-4) and the  $\beta$ -monosubstitution pattern in ring A by the three coupled protons at  $\delta$ 8.04 ( $d$ ,  $J = 8$  Hz, H-5), 7.51 ( $dd$ ,  $J = 8.0$ , 1.6 Hz, H-6) and 7.97 ( $d$ ,  $J = 1.6$  Hz, H-8).  $^{13}\text{C}$ NMR data (Table 2) and DEPT experiments confirmed the presence of three Me, four CH and 10 quaternary carbons, of which two carbonyl carbons were observed at  $\delta$ 182.1 and 182.3. NOE measurements revealed the pres-

ence of a 1,3-dimethoxy-2-hydroxy pattern and a  $\beta$ -Me group. HMQC and HMBC confirmed the location of the methyl group at C-7 through two- and three-bond correlations (Fig. 1) from the methyl protons to C-7, C-6 and C-8, from H-6 to C-8 and C-11, from H-8 to C-6, C-9 and C-11 and from H-5 to C-7, C-10 and C-12. These correlations together with the above spectra support the structure **1** of a 7-methylantragallol 1,3-dimethyl ether.

Compound **2** showed IR bands at 3392 1633 and 1673  $\text{cm}^{-1}$  of a free hydroxyl group and two chelated and unchelated carbonyl absorptions, respectively. The EI-mass spectrum displayed a  $[\text{M}]^+$  at  $m/z$  284, 14 mu less than that of **1**, and calculated for  $\text{C}_{16}\text{H}_{12}\text{O}_5$ . The  $^1\text{H}$ NMR data (Table 1) showed, like **1**, the same splitting pattern of the aromatic protons but a *peri*-hydroxyl and only one methoxyl group were detected at  $\delta$ 13.14 and 4.13, respectively. NOE exhibited 5 and 7% enhancements for the protons at  $\delta$ 7.55 and 8.07, respectively, upon irradiation of the methyl resonance at  $\delta$ 2.52. No effect was observed upon irradiation of the methoxyl protons at  $\delta$ 4.13 and, thus, it must be at C-2. Methylation

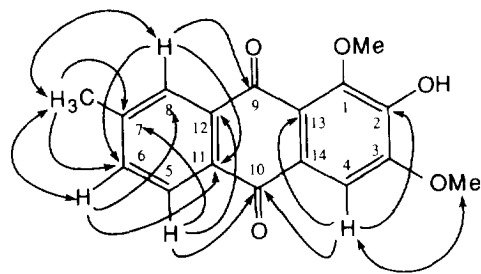
Table 1.  $^1\text{H}$ NMR spectral data of compounds **1–9** (400 MHz,  $\text{CDCl}_3$ )

H	1	2	3*	4	5	6	7	8	9*
OH-1	—	13.14 <i>b, s</i>	—	12.21 <i>b, s</i>	—	12.90 <i>s</i>	12.84 <i>s</i>	13.21 <i>s</i>	—
OMe-1	4.03 <i>s</i>	—	—	—	4.05 <i>s</i>	—	—	—	—
OH-2	6.30 <i>b, s</i>	—	—	—	6.41 <i>b, s</i>	—	—	—	—
Me-2	—	—	—	—	—	2.35 <i>s</i>	2.31 <i>s</i>	—	—
2	—	—	—	—	—	—	—	—	6.62 <i>d</i>
OMe-2	—	4.13 <i>s</i>	—	4.02 <i>s</i>	—	—	—	—	—
CH <sub>2</sub> -2 (a)	—	—	—	—	—	—	—	4.96 <i>s</i>	—
CH <sub>2</sub> -2 (b)	—	—	—	—	—	—	—	3.71 <i>q</i>	—
								(7.2)	—
Me-2 (c)	—	—	—	—	—	—	—	1.32 <i>t</i>	—
								(7.2)	—
OH-3	—	6.50 <i>b, s</i>	—	—	—	—	—	9.55 <i>b, s</i>	—
3	—	—	—	—	—	7.84 <i>d</i>	7.49 <i>d</i>	—	—
						(7.6)	(7.6)	—	—
OMe-3	4.08 <i>s</i>	—	3.99 <i>s</i>	4.05 <i>s</i>	4.11 <i>s</i>	—	—	—	—
4	7.70 <i>s</i>	7.47 <i>s</i>	7.62 <i>s</i>	7.49 <i>s</i>	7.73 <i>s</i>	7.67 <i>d</i>	7.65 <i>d</i>	7.29 <i>s</i>	7.22 <i>d</i>
						(7.6)	(7.6)	—	(2.3)
OH-5	—	—	—	—	—	—	12.89 <i>s</i>	—	—
OMe-5	—	—	—	—	—	4.02 <i>s</i>	—	—	—
5	8.04 <i>d</i>	8.14 <i>d</i>	7.59 <i>d</i>	7.80 <i>dd</i>	8.38 <i>d</i>	—	—	7.69 <i>d</i>	7.56 <i>d</i>
	(8.0)	(7.9)	(0.8)	(7.6, 1.2)	(7.8)	—	—	(2.6)	(2.6)
Me-6	—	—	2.28 <i>s</i>	—	—	—	—	—	—
OMe-6	—	—	—	—	—	—	3.96 <i>s</i>	3.97 <i>s</i>	—
6	7.51 <i>dd</i>	7.55 <i>dd</i>	—	7.63 <i>t</i>	8.22 <i>dd</i>	—	—	—	—
	(8.0, 1.6)	(7.9, 0.5)	—	(7.6)	(7.8, 1.6)	—	—	—	—
Me-7	2.51 <i>s</i>	2.52 <i>s</i>	—	—	—	—	—	—	—
CHO-7	—	—	—	—	10.21 <i>s</i>	—	—	—	—
OMe-7	—	—	—	—	—	4.08 <i>s</i>	3.98 <i>s</i>	—	—
7	—	—	7.40 <i>dd</i>	7.25 <i>dd</i>	—	—	—	7.23 <i>dd</i>	7.17 <i>dd</i>
			(7.7, 0.8)	(7.6, 1.2)	—	—	—	(8.6, 2.6)	(8.4, 2.4)
OH-8	—	—	—	12.09 <i>b, s</i>	—	—	—	—	—
8	7.97 <i>d</i>	8.07 <i>d</i>	7.61 <i>d</i>	—	8.74 <i>d</i>	7.71 <i>s</i>	7.41 <i>s</i>	8.20 <i>d</i>	8.14 <i>d</i>
	(1.6)	(0.5)	(7.6)	—	(1.6)	—	—	(8.6)	(8.4)

\*Measured in  $\text{CDCl}_3$ - $\text{CD}_3\text{OD}$ .

Table 2.  $^{13}\text{C}$  NMR spectral data of some isolated compounds and xanthopurpurin (100 MHz,  $\text{CDCl}_3$ )

C	1	3*	5†	6	8	14	9†	xanthopurpurin† [14]
1	147.4	160.3	148.3	160.5	161.7	161.9	165.0	164.8
2	144.7	152.1	147.2	114.7	114.9	114.7	107.8	107.7
3	151.5	152.8	152.7	137.1	163.7	164.2	164.5	165.3
4	106.3	108.8	106.1	119.0	109.7	109.8	108.2	108.3
5	126.9	118.9	126.9	147.6	110.4	127.4	112.6	126.7
6	134.2	134.4	132.7	145.3	164.4	134.1	163.2	134.6
7	145.1	136.6	139.4	151.4	121.0	134.1	121.4	134.5
8	127.4	112.4	127.9	106.0	129.2	126.7	129.4	126.3
9	182.1	188.4	180.0	188.0	186.4	186.9	185.3	185.9
10	182.3	182.0	181.0	181.1	182.4	182.3	182.1	181.8
11	134.6	131.5	134.8	121.2	134.1	134.1	135.1	133.0
12	130.5	128.4	135.7	127.3	126.9	133.6	124.6	133.0
13	120.8	115.1	120.7	133.9	109.4	109.7	109.0	108.4
14	127.8	127.7	125.6	133.0	135.8	133.6	135.0	135.0
OMe-1	61.8	—	60.6	—	—	—	—	—
Me-2	—	—	—	16.1	—	—	—	—
OMe-3	56.6	56.2	56.1	—	—	—	—	—
OMe-5	—	—	—	61.9	—	—	—	—
Me-6	—	15.8	—	—	—	—	—	—
OMe-6	—	—	—	—	56.0	—	—	—
Me-7	22.0	—	—	—	—	—	—	—
OMe-7	—	—	—	56.6	—	—	—	—
CHO-7	—	—	192.6	—	—	—	—	—
$\text{CH}_2\text{-O-CH}_2\text{-Me}$	—	—	—	—	67.6	67.6	—	—
$\text{CH}_2\text{-O-CH}_2\text{-Me}$	—	—	—	—	67.0	67.0	—	—
$\text{CH}_2\text{-O-CH}_2\text{-Me}$	—	—	—	—	15.0	15.0	—	—

\* $\text{CDCl}_3 + \text{CD}_3\text{OD}$ .† $(\text{CD}_3\text{O})_2\text{SO}_2$ .Fig. 1. HMBC ( $\rightarrow$ ) and NOESY ( $\leftrightarrow$ ) correlations of compound 1.

of both **1** and **2** with methyl iodide gave one and the same methylated product to confirm the identity of **2** as 7-methylantragallol 2-methyl ether.

Compound **3** showed IR bands at  $3353\text{ cm}^{-1}$  of a free hydroxyl group and two chelated and unchelated carbonyl absorptions at  $1632$  and  $1652\text{ cm}^{-1}$ , respectively. The EI-mass spectrum and  $^1\text{H}$  NMR data (Table 1) were very similar to those of **2**. However, the methoxyl protons exhibited a significant NOE upon H-4 confirming its disposition at C-3.  $^{13}\text{C}$  NMR data (Table 2) and DEPT experiments confirmed the presence of two  $\text{CH}_3$  four CH and 10 quaternary carbons, of which two carbonyl carbons were observed at  $\delta 182.0$  and  $188.4$ . The dimethyl ether of **3** was different from that of **2** and, therefore, the

methyl group must be at C-6 to support the proposed structure for **3** as 6-methylantragallol 3-methyl ether.

Compound **4** showed IR bands at  $1628$  and  $1673\text{ cm}^{-1}$  due to two strongly chelated and unchelated carbonyl absorptions, respectively. The EI-mass spectrum displayed  $[\text{M}]^+$  at  $m/z$  300, calculating for  $\text{C}_{16}\text{H}_{12}\text{O}_6$ . The trisubstitution pattern of ring C was deduced from the aromatic protons singlet at  $\delta 7.49$  (H-4) and the  $\alpha$ -mono-substitution pattern of ring A by the three coupled protons at  $7.25$  ( $dd$ ,  $J = 7.6, 1.2\text{ Hz}$ , H-7),  $7.63$  ( $t$ ,  $J = 7.6\text{ Hz}$ , H-6) and  $7.80$  ( $dd$ ,  $J = 7.6, 1.2\text{ Hz}$ , H-5). Two methoxyl and two *peri*-hydroxyl groups were also detected at  $\delta 4.02$ ,  $4.05$ ,  $12.09$  and  $12.21$ , respectively. A NOESY cross-peak correlation was observed between the methoxyl proton signals at  $\delta 4.05$  and H-4 at  $\delta 7.49$  and, hence, the two methoxyl groups are located at C-2 and C-3. From the two possible 1,5- and 1,8-dihydroxy-2,3-dimethoxy substitution patterns, the latter was chosen based upon the strong chelation of only one carbonyl absorption in its IR spectrum. Therefore, **4** is as 8-hydroxyanthragallol 2,3-dimethyl ether.

Compound **5** showed IR bands at  $3382$ ,  $1696$  and  $1668\text{ cm}^{-1}$  of a free hydroxyl and two unchelated carbonyl groups and the  $[\text{M}]^+$  at  $m/z$  312 calculated for  $\text{C}_{17}\text{H}_{12}\text{O}_6$ . The  $^1\text{H}$  NMR spectrum (Table 1) showed, like **1**, a trisubstituted ring C with two methoxyl and a  $\beta$ -hydroxyl groups and a monosubstituted ring A with

an aldehydic group ( $\delta$  10.21). However, the three coupled aromatic protons of ring A exhibited a significant downfield shift of *ca* 0.4–0.7 ppm indicating the disposition of the aldehydic group at C-6 or C-7.  $^{13}\text{C}$  NMR data (Table 2) and DEPT experiments displayed signals for two unchelated carbonyl carbons at  $\delta$  180.0 and 181.0 as well as a strongly downfield shifted one at  $\delta$  192.6 assigned to that of the aldehyde group. Moreover, reduction of **5** with  $\text{NaBH}_4$  [12] afforded the corresponding 7-hydroxymethylanthragallol 1,3-dimethyl ether **19**. Consequently, **5** is 7-formylanthragallol 1,3-dimethyl ether.

Compound **6** showed IR bands at 3372, 1627 and  $1664\text{ cm}^{-1}$  of a free hydroxyl group and two chelated and unchelated carbonyl absorptions, respectively, and the  $[\text{M}]^+$  at  $m/z$  314 calculated for  $\text{C}_{17}\text{H}_{14}\text{O}_6$ . The  $^1\text{H}$  NMR data (Table 1) revealed signals for one methyl and two methoxyl groups at  $\delta$  2.35, 4.02 and 4.08, respectively, and the aromatic proton signals appeared as a 1-proton singlet at  $\delta$  7.71 and two 1-proton doublets at  $\delta$  7.48 and 7.67 ( $J = 7.6\text{ Hz}$ , each) suggesting that it could be a copareolatin derivative [2]. The substitution pattern was verified by NOE measurements. Irradiation of the methyl protons at  $\delta$  2.35 produced a 9% enhancement of H-3 at  $\delta$  7.48. Similarly, irradiation of the methoxyl protons at  $\delta$  4.08 produced a 16% enhancement of H-8 at  $\delta$  7.71, while upon irradiation of that at  $\delta$  4.02 no effect was observed. Moreover, methylation of both **6** and **17** (copareolatin 6-methyl ether) with methyl iodide gave one and the same methylated product. The above findings together with the  $^{13}\text{C}$  NMR data (Table 2) confirmed the identity of **6** as copareolatin 5,7-dimethyl ether.

Compound **7** showed IR bands at 3436 and  $1625\text{ cm}^{-1}$  of an hydroxyl group and a chelated carbonyl absorption, respectively. The EI-mass spectrum displayed, like **6**, a  $[\text{M}]^+$  at  $m/z$  314, calculating for  $\text{C}_{17}\text{H}_{14}\text{O}_6$ . The close similarity between **7** and **6** was evidenced from the  $^1\text{H}$  NMR data (Table 1) by the same substituent groups and splitting pattern of the aromatic protons. NOESY experiments showed cross-peak correlation between the Me-2 at  $\delta$  2.31 and the aromatic proton at  $\delta$  7.49 (*d*, H-3) and between the MeO-7 at  $\delta$  3.98 and the aromatic proton at  $\delta$  7.41 (*s*, H-8). These spectral data, together with the presence of only one chelated carbonyl absorption at  $1625\text{ cm}^{-1}$  support the 1,5-dihydroxy-2-methyl substitution pattern [13] and confirmed the identity of **7** as copareolatin 6,7-dimethyl ether.

Compound **8** showed IR bands at 3431, 1627 and  $1672\text{ cm}^{-1}$  of a free hydroxyl and two chelated and unchelated carbonyl groups, respectively and the  $[\text{M}]^+$  at  $m/z$  328, calculated for  $\text{C}_{18}\text{H}_{16}\text{O}_6$ . The  $^1\text{H}$  NMR data (Table 1) established, a lucidin-derived structure [2] from the ethoxymethyl proton resonances at  $\delta$  1.32, 3.71 and 4.96, as well as the hydroxyl protons at  $\delta$  9.55 and 13.21 of a  $\beta$ -phenolic and a *peri*-hydroxyl group. Ring C was identical to ring C of lucidin  $\omega$ -ethyl ether **14**, but there is, in addition, a  $\beta$ -methoxyl substituent in ring A resonating at  $\delta$  3.97. Comparison of the  $^{13}\text{C}$  NMR data of **8** with those of lucidin  $\omega$ -ethyl ether **14** (Table 2) revealed an upfield shift of *ca* 0.5 ppm in the  $\delta$  value of the C-9

carbonyl carbon of **8** with respect to that of **14** as a result of transfer of electron density by cross-conjugation [3, 6, 14], while those at C-10 were observed at approximately the same  $\delta$  values. Accordingly, **8** is 6-methoxylucidin  $\omega$ -ethyl ether.

Compound **9** showed IR bands at 3413, 1665 and  $1631\text{ cm}^{-1}$  of a free hydroxyl, unchelated and chelated carbonyl groups and the  $[\text{M}]^+$  at  $m/z$  256 calculated for  $\text{C}_{14}\text{H}_8\text{O}_5$ . The  $^1\text{H}$  NMR (Table 1) indicated an  $\alpha,\beta$ -disubstituted ring C from two *meta*-coupled protons at  $\delta$  6.62 and 7.22 (*d*, 2,3) and a  $\beta$ -substituted ring A from three coupled protons at  $\delta$  7.56 (*d*, 2,6), 7.17 (*dd*, 8,4, 2,40) and 8.14 (*d*, 8,4). The  $^{13}\text{C}$  NMR data (Table 2) revealed a total of 14 carbon signals for five CH and nine quaternary, of which two carbonyl signals at  $\delta$  181.8 and 185.9 and three C-hydroxyl signals at  $\delta$  163.2, 164.5 and 165.0

Table 3.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of compound **10** (400 and 100 MHz,  $\text{CDCl}_3$ )

Position	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$^1\text{H}$ long-range coupled	
			$^3J_{\text{CH}}$	$^2J_{\text{CH}}$
1		189.5 s		
2	6.87 <i>d</i> (1.3)	136.2 <i>d</i>	4, 13, 15	
3		150.9 s		
4		184.9 s		
5	7.75 <i>d</i> (7.8)	127.0 <i>d</i>	7, 10, 12	
6	7.61–7.70 <i>m</i>	131.1 <i>d</i>		
7	7.61–7.70 <i>m</i>	129.5 <i>d</i>		
8	8.42 <i>d</i> (8.0)	124.8 <i>d</i>	6, 11	
9		163.1 s		
10		128.1 <sup>a</sup> s		
11		136.2 s		
12		127.8 <sup>a</sup> s		
13		109.1 s		
14		127.9 <sup>a</sup> s		
15	2.03 <i>d</i> (1.3)	17.1 <i>q</i>	2, 4	3
OH-9	14.23 <i>br, s</i>	—	12, 13	9
1'		187.8 s		
2'		147.6 s		
3'		145.2 s		
4'		184.0 s		
5'	7.96 <i>d</i> (8.0)	130.5 <i>d</i>	7', 10', 13'	
6'	7.61–7.70 <i>m</i>	132.1 <i>d</i>		
7'	7.61–7.70 <i>m</i>	128.9 <i>d</i>		
8'	8.57 <i>d</i> (7.0)	125.5 <i>d</i>	6', 14'	
9'		162.6 s		
10'	8.22 <i>s</i>	121.9 <i>d</i>	4', 5', 12', 13'	
11'		125.2 s		
12'		109.2 s		
13'		127.8 <sup>a</sup> s		
14'		134.1 s		
15'	1.73 <i>s</i>	14.5 <i>q</i>	2', 4'	3'
OH-9'	13.67 <i>br, s</i>		12', 13'	9'

Signal assignments are based on analysis of 1D ( $^1\text{H}$ ,  $^{13}\text{C}$ ) and 2D (COSY, NOESY, HMQC, HMBC) NMR spectra ( $\delta$ , in ppm from TMS,  $J$  in Hz).  $^3J_{\text{CH}}$  and  $^2J_{\text{CH}}$  indicate the protons long-range-coupled with carbons through three and two bonds, respectively observed in the HMBC spectrum.

<sup>a</sup>Signal assignments may be interchanged.

were observed. The choice between the 1,3,6- and 1,3,7-trihydroxylation patterns was deduced from comparison of the  $^{13}\text{C}$  NMR data of **9** with xanthopurpurin [14] (Table 2). The upfield shift of the C-9 carbonyl carbon by *ca* 0.6 ppm with respect to that of xanthopurpurin, as a result of cross-conjugation [3, 6, 14], confirmed its identity as 6-hydroxy-xanthopurpurin.

Compound **10** showed IR bands at 1661, 1642 and  $1626\text{ cm}^{-1}$  of unchelated and two chelated CO absorptions, respectively. The EI-mass spectrum displayed a  $[\text{M}]^+$  at  $m/z$  474, calculating for  $\text{C}_{30}\text{H}_{18}\text{O}_6$  and a significant fragment ion at  $m/z$  238, calculating for  $\text{C}_{15}\text{H}_{10}\text{O}_3$ . The  $^1\text{H}$  NMR data (Table 3) revealed signals for two methyl groups at  $\delta$  2.03 (d, 1.3) and 1.73 (s), two *peri*-hydroxyl protons at  $\delta$  13.67 and 14.23, as well as nine aromatic and one quinonoid proton. A total of 30 carbon signals were detected in  $^{13}\text{C}$  NMR data (Table 3) and DEPT experiments demonstrated two Me, 10 CH and 18 quaternary carbons of which four are carbonyl at  $\delta$  189.5, 187.8, 184.9 and 184.0 and two are C-hydroxyl at  $\delta$  163.1 and 162.6. These spectral findings together with the fragment ion at  $m/z$  238 suggested a bianthraquinone skeleton with one methyl and one *peri*-hydroxyl group in each monomer. The NOESY experiments demonstrated a few important cross-peak correlations (Fig. 2) between the methyl protons at  $\delta$  2.03 and the proton at  $\delta$  6.87, the methyl protons at  $\delta$  1.73 and the proton at  $\delta$  7.75, as well as the protons at  $\delta$  7.96 and 8.22. The structure **10** was finally established by detailed analysis of HMQC and HMBC spectra. The result of these analyses (Fig. 2, Table 3) revealed the key connectivity between the two monomers by the carbon signals at  $\delta$  128.1 and 147.6 ascribed to C-10 and C-2', respectively, through  $^3J_{\text{CH}}$  correlations with H-5 at  $\delta$  7.75 and the methyl protons at  $\delta$  1.73.  $^2J_{\text{CH}}$  and  $^3J_{\text{CH}}$  correlations presented in Fig. 2 and Table 3 support the structure of **10** as 10,2'-bi(9-hydroxy-3-methyl-1,4-anthraquinonyl), named bi-sinaquinone.

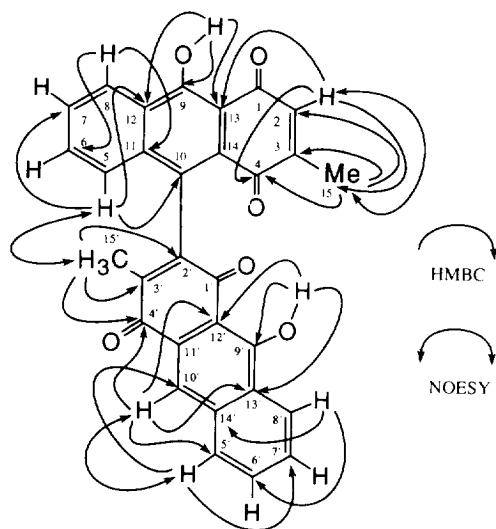


Fig. 2. HMBC and NOESY correlations of compound **10**.

Table 4.  $\text{IC}_{50}$  values of anthraquinones from *G. sinaicum* in cytotoxic activity against P388 cells

compound	$\text{IC}_{50}$ values ( $\mu\text{g ml}^{-1}$ )
<b>1</b>	10.1
<b>3</b>	21.0
<b>5</b>	25.0
<b>6</b>	12.5
<b>8</b>	11.2
<b>9</b>	13.0
<b>10</b>	10.9
<b>11</b>	5.8
<b>13</b>	80.0
<b>14</b>	18.0
<b>17</b>	5.5
<b>19</b>	32.0

Anthraquinones isolated from *G. sinaicum* were subjected to the cytotoxic bioassay against P388 leukemic cells. Compounds **11** and **17** showed stronger activity than the other anthraquinones (Table 4).

#### EXPERIMENTAL

**General.** Mps: uncorr.  $^1\text{H}$  and  $^{13}\text{C}$  NMR:  $\text{CDCl}_3$  or  $\text{DMSO}-d_6$  with TMS as int. standard, Bruker AM400.  $^{13}\text{C}$  NMR multiplicities were determined by DEPT pulse sequence. 2D NMR (COSY, NOESY, HMQC, HMBC), Bruker AM500. EI-MS: 70 eV. IR: KBr. UV: MeOH. Prep. HPLC: 10  $\mu\text{m}$  ODS column. MPLC: 20  $\mu\text{m}$  ODS column, Kusano Scientific. CC: Kieselgel 60. TLC: silica gel 60F<sub>254</sub> and RP-18 F<sub>254</sub> (0.25 mm) precoated plates. Spots detection: UV light at 254 nm, exposure to  $\text{NH}_3$  vapour and/or spraying with 10%  $\text{H}_2\text{SO}_4$  or 5% KOH in MeOH. Prep. TLC: silica gel 60F<sub>254</sub> (0.5 mm). Cytotoxicity bioassay: Tohso MPR A4i microplate reader.

**Plant material.** Roots of *Galium sinaicum* Boiss. were collected in May 1991 from the mountains of Saint Katherine district, Sinai Peninsula. Identification was verified by Professor Dr N. El-Hadidi (Faculty of Science, Cairo University). A voucher specimen, documenting this collection, is deposited at the Pharmacognosy Department, Faculty of Pharmacy, University of Mansoura.

**Extraction and isolation.** Air-dried and powdered roots (3 kg) were extracted by percolation with 95% EtOH. The alcoholic extract (96 g;  $\text{IC}_{50}$ ,  $14\text{ }\mu\text{g ml}^{-1}$  to P388 cell line) was diluted with  $\text{H}_2\text{O}$  and successively partitioned with *n*-hexane,  $\text{CH}_2\text{Cl}_2$  and *n*-BuOH. The *n*-hexane extract (15 g;  $\text{IC}_{50}$ ,  $3.8\text{ }\mu\text{g ml}^{-1}$  to P388 cell line) was then successively extracted with 1 N  $\text{Na}_2\text{CO}_3$  and 1 N NaOH to afford extracts A and B (1.5 g, each), respectively. Both extracts were fractionated separately using silica-gel-packed columns eluted with an *n*-hexane-EtOAc gradient; similar frs were pooled together. The frs obtained were subjected either to direct recrystallization or to further purification on RP-18 columns using various chromatographic techniques (HPLC,

MPLC, PTLC) to afford the anthraquinone compounds 1–18.

**7-Methylantragaolol 1,3-dimethyl ether (1).** Yellow needles (5 mg) from NaOH fr. Mp 197°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 204, 244, 250, 280. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3310, 2923, 1667, 1595, 1561, 1496, 1458, 1347, 1293, 1242, 1168, 1133, 1101, 1047, 905, 874, 790, 732, 692. EI-MS  $m/z$  (rel. int.): 298  $[M]^+$  (3) (calcd for  $\text{C}_{17}\text{H}_{14}\text{O}_5$  298.0841, found 298.0835), 284 (100), 268 (5), 255 (6), 213 (6.5), 128 (5.3). FABMS  $m/z$  (rel. int.): 299  $[M + 1]^+$  (100).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ): Table 2.

**7-Methylantragaolol 2-methyl ether (2).** Yellow crystals (2 mg) from  $\text{Na}_2\text{CO}_3$  fr. Mp 174–176°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 205, 250, 280, 407. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3392, 2923, 2853, 1673, 1633, 1599, 1580, 1492, 1398, 1363, 1342, 1248, 1213, 1163, 1097, 799, 759, 734, 671. EI-MS  $m/z$  (rel. int.): 284  $[M]^+$  (100) (calcd for  $\text{C}_{16}\text{H}_{12}\text{O}_5$  284.0685, found 284.0681), 266 (65), 255 (9), 241 (47), 210 (21), 184 (6), 139 (11), 128 (14), 89 (5), 74 (9).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.

**6-Methylantragaolol 3-methyl ether (3).** Yellow needles (5 mg) from  $\text{Na}_2\text{CO}_3$  fr. Mp 270–273°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 212, 240 sh, 280, 408. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3353, 2926, 2852, 1652, 1632, 1594, 1570, 1523, 1438, 1371, 1293, 1325, 1311, 1256, 1210, 1188, 1121, 1041, 1005, 892, 815, 782, 752, 731, 672, 623. EI-MS  $m/z$  (rel. int.): 284  $[M]^+$  (100), (calcd for  $\text{C}_{16}\text{H}_{12}\text{O}_5$  284.0685, found 284.0668), 269 (8), 241 (17), 213 (12), 185 (10), 139 (9), 127 (8), 111 (8), 97 (15), 82 (17), 69 (23), 56 (23).  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{OD} + \text{CDCl}_3$ ): Table 1.  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{OD} + \text{CDCl}_3$ ): Table 2.

**8-Hydroxyanthragallol 2,3-dimethyl ether (4).** Yellow crystals (2 mg) from NaOH fr. Mp 226–227°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 206, 238 sh, 244 sh, 274, 417. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3436, 2924, 2853, 1673, 1628, 1599, 1505, 1459, 1420, 1371, 1321, 1273, 1241, 1220, 1196, 1163, 1092, 1048, 905, 990, 981, 920, 864, 749, 721. EI-MS  $m/z$  (rel. int.): 300  $[M]^+$  (38) (calcd for  $\text{C}_{16}\text{H}_{12}\text{O}_6$  300.0634, found 300.0654), 282 (13), 257 (20), 223 (11), 205 (7), 186 (15), 167 (6), 149 (100), 135 (8), 121 (13), 97 (19), 86 (25), 69 (31).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.

**7-Formylantragaolol 1,3-dimethyl ether (5).** Yellow powder (5 mg) from NaOH fr. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 206, 244 sh, 280. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3382, 2960, 2925, 2854, 1696, 1668, 1597, 1560, 1495, 1455, 1381, 1347, 1325, 1289, 1237, 1211, 1161, 1126, 1089, 1043, 991, 882, 819, 800, 790, 735, 726. EI-MS  $m/z$  (rel. int.): 312  $[M]^+$  (100) (calcd for  $\text{C}_{17}\text{H}_{12}\text{O}_6$  312.0634, found 312.0639), 294 (97), 265 (73), 253 (7), 239 (11), 211 (10), 198 (34), 181 (11), 153 (7), 141 (13), 126 (8), 113 (10), 97 (6), 69 (8), 56 (11).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ ): Table 2.

**Copareolatin 5,7-dimethyl ether (6).** Yellow powder (5 mg) from  $\text{Na}_2\text{CO}_3$  fr. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 213, 278, 410. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3372, 2925, 2854, 1664, 1627, 1594, 1561, 1496, 1470, 1457, 1426, 1366, 1323, 1289, 1258, 1199, 1173, 1130, 1100, 1072, 1027, 1001, 904, 872, 847, 811, 792, 738, 717, 688. EI-MS  $m/z$  (rel. int.): 314  $[M]^+$  (100) (Calcd for  $\text{C}_{17}\text{H}_{14}\text{O}_6$  314.0790, found 314.0808), 296 (58), 281 (26), 267 (43), 255 (6), 241 (10), 213 (9), 200 (27), 172 (7),

144 (10), 128 (12), 115 (23), 89 (7), 77 (15).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ): Table 2.

**Copareolatin 6,7-dimethyl ether (7).** Yellow needles (2 mg) from NaOH fr. Mp 205–207°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 220, 272, 308 sh, 432. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3436, 2925, 2853, 1625, 1578, 1500, 1445, 1418, 1368, 1303, 1245, 1195, 1180, 1136, 1100, 1076, 1035, 990, 956, 855, 801, 742, 719, 692. EI-MS  $m/z$  (rel. int.): 314  $[M]^+$  (18) (calcd for  $\text{C}_{17}\text{H}_{14}\text{O}_6$  314.0790, found 314.0812), 299 (11), 270 (18), 256 (28), 236 (15), 221 (7), 199 (13), 185 (15), 167 (18), 149 (76), 135 (12), 123 (17), 111 (28), 97 (49), 85 (36), 71 (60), 70 (100), 56 (98).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.

**6-Methoxylucidin  $\omega$ -ethyl ether (8).** Yellow crystals (3 mg) from NaOH fr. Mp 143–145°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 213, 273, 336 sh, 422. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3431, 3235, 2924, 2852, 1672, 1627, 1597, 1482, 1444, 1400, 1376, 1334, 1309, 1293, 1279, 1230, 1185, 1163, 1139, 1090, 1057, 1025, 977, 888, 833, 759, 732. EI-MS  $m/z$  (rel. int.): 328  $[M]^+$  (13) (calcd for  $\text{C}_{18}\text{H}_{16}\text{O}_6$  328.0947, found 328.0955), 299 (9), 268 (5), 282 (100), 254 (15), 226 (8), 149 (20), 129 (5), 97 (8), 71 (10), 70 (17), 57 (20).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ): Table 1.  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ): Table 2.

**6-Hydroxyxanthopurpurin (9).** Orange red crystals (2 mg) from NaOH fr. Mp 247–250°. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 210, 278, 300 sh, 340 sh, 428. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3413, 2926, 2852, 1665, 1631, 1597, 945, 798, 754, 664. EI-MS  $m/z$  (rel. int.): 256  $[M]^+$  (30) (calcd for  $\text{C}_{14}\text{H}_8\text{O}_5$  256.0372, found 256.0399), 284 (100), 236 (15), 228 (7), 220 (25), 205 (7), 185 (12), 167 (12), 157 (9), 149 (72), 137 (19), 129 (25), 111 (25), 97 (40), 80 (52), 69 (100).  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{OD} + \text{CDCl}_3$ ): Table 1.  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ ): Table 2.

**Bisinaquinone (10,2'-bi(9-hydroxy-3-methyl-1,4-anthraquinonyl)) (10).** Red crystals (5 mg) from NaOH fr. Mp 245–248°.  $[\alpha]_D^{20} \pm 0^\circ$   $\text{CHCl}_3$ ;  $c$  0.08. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 206, 226, 252 sh, 278 sh, 400 sh, 434 sh. IR  $\nu^{\text{KBr}}$   $\text{cm}^{-1}$ : 3436, 2921, 2851, 1737, 1661, 1642, 1626, 1597, 1577, 1499, 1459, 1427, 1399, 1376, 1341, 1303, 1249, 1218, 1103, 986, 954, 903, 857, 834, 806, 786, 759, 750. EI-MS  $m/z$  (rel. int.): 474  $[M]^+$  (100) (calcd for  $\text{C}_{30}\text{H}_{18}\text{O}_6$  474.1103, found 474.1132), 431 (35), 401 (8), 368 (11), 227 (9), 238 (18), 211 (6), 183 (22), 142 (9), 114 (12), 98 (15), 83 (20), 57 (42).  $^1\text{H}$  and  $^{13}\text{C}$  NMR (40–100 MHz,  $\text{CDCl}_3$ ): Table 3.

**Cytotoxicity.** The MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide) colorimetric assay was performed in a 96-well plate [15, 16]. The assay is dependent on the reduction of MTT by the mitochondrial dehydrogenase of viable cells to a blue formazan product which can be measured spectrophotometrically. Mouse P388 leukemia cells ( $2 \times 10^4$  cells  $\text{ml}^{-1}$ ) were inoculated in each well with 100  $\mu\text{l}$   $\text{ml}^{-1}$  of RPMI-1640 medium (Nissui Pharm. Co., Ltd) supplemented with 5% fetal calf serum (Mitsubishi Chemical Industry Co., Ltd) and kanamycin (100  $\mu\text{g}$   $\text{ml}^{-1}$ ) at 37° in a humidified atmosphere of 5%  $\text{CO}_2$ . Various drug concs (10  $\mu\text{l}$ ) were added to the cultures at day 1 after transplantation. At day 3, 20  $\mu\text{l}$  of MTT soln (5 mg  $\text{ml}^{-1}$ ) per well was added to each cultured medium. After a further 4 hr of incubation, 100  $\mu\text{l}$  of 10% SDS–0.01 M HCl soln was added to

each well and the formazan crystals in each well were dissolved by stirring with a pipette. Absorbance measurements were made using a microplate reader (Tohso MPR-A4i) with a two wavelength system (550 and 700 nm). In all expts, 3 replicate wells were used to determine each point.

**Acknowledgements**—The authors are grateful to Dr K. Tagahara (Kobe College of Pharmacy, Japan) for supplying authentic 2,8-dihydroxy-1,3-dimethoxy-7-methyl-9,10-anthraquinone and also to the Ministry of Higher Education in Egypt for financial support.

#### REFERENCES

1. Thomson, R. H. (1971) *Naturally Occurring Quinones*, 2nd Ed. Academic Press, London.
2. Wijnsma, R. and Verpoorte, R. (1986) in *Progress in the Chemistry of Organic Natural Products* Vol. 49 (W. Herz, H. Grisebach, G. W. Kirby and C. H. Tamm, eds), p. 79. Springer, Wein.
3. Halim, A. F., Abd El-Fattah, H., El-Gamal, A. A. and Thomson, R. H. (1992) *Phytochemistry* **31**, 355.
4. Berg, W., Hesse, A., Kraft, R. and Hermann, M. (1974) *Pharmazie* **29**, 478.
5. Dosseh, Ch., Tessier, A. M. and Delaveau, P. (1981) *Planta Med.* **43**, 141, 360.
6. Itokawa, H., Mihara, K. and Takeya, K. (1983) *Chem. Pharm. Bull.* **31**, 2353.
7. Demagos, G. P., Baltus, W. and Hofle, G. (1981) *Z. Naturforsch* **36b**, 1180.
8. Brigg, L. H., Beachen, J. F., Cambie, R. C., Dudman, N. P. B., Steggles, A. W. and Rutledge, P. S. (1976) *J. Chem. Soc. Perk. Trans 1*, 1789.
9. Roberge, G. and Brassard, P. (1981) *J. Org. Chem.* **46**, 1461.
10. Park, Y. H. (1977) PhD thesis. University of Mississippi.
11. Koyama, J., Ogura, T. and Tagahara, K. (1993) *Phytochemistry* **33**, 1540.
12. Zhou, Z., Jiang, S. H., Zhu, D. Y., Lin, L. Z. and Cordell, G. A. (1994) *Phytochemistry* **36**, 765.
13. Wijnsma, R. and Verpoorte, R., Mulder-Krieger, T. H. and Svendsen, A. B. (1984) *Phytochemistry* **23**, 2307.
14. Berger, Y. and Castonguay, A. (1978) *Org. Magn. Res.* **11**, 375.
15. Twentyman, P. R. and Luscombe, M. (1987) *Br. J. Cancer* **56**, 279 Carmichael, J., DeGraff, W. G., Gazdar, A. F., Minna, J. D. and Mitchell, B. (1987) *Cancer Res.* **47**, 936.