



TRITERPENOIDS FROM *ANTRODIA CINNAMOMEA*

I-HWA CHERNG, DE-PENG WU and HUNG-CHEH CHIANG*

Institute of Chemistry, National Taiwan Normal University, Taipei 117, Taiwan

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Abstract—Four novel ergostane-type triterpenoids (antcins E and F and methyl antcinate G and H) were isolated from the fruiting body of the fungus *Antrodia cinnamomea*, a newly identified species of *Antrodia*, Polyporaceae in Taiwan. Their structures were elucidated by spectroscopic methods.

INTRODUCTION

Several highly oxygenated triterpenoids from *Ganoderma lucidum* were isolated previously [1, 2] whose biological activities included histamine release-inhibition [3], inhibition or stimulation of platelet aggregation [4, 5], cytotoxicity to hepatoma cells [6], inhibition of angiotensin-converting enzyme [7], hypolipidemic activity and inhibition of cholesterol synthesis [8].

Antrodia cinnamomea Chang & Chou, sp. nov. growing rarely on the inner cavity wall of *Cinnamomum kanehirai* Hay was identified as a new species of *Antrodia* in 1994 [9]. It was initially identified as a new *Ganoderma* species, *Ganoderma camphoratum* in 1990 [10] because there were several similar characteristics. However, it tastes more bitter than *G. lucidum* and the percentage of methanolic extract from *Antrodia cinnamomea* (30%) is 10 times higher than that of *G. lucidum* (3%) [11]. *Antrodia cinnamomea* is used as an antidote, anticancer and antitumescenic (anti-itching) drug, but no phytochemical or biological studies have been published. This study attempts to explore the differences between *Antrodia cinnamomea* and *G. lucidum* in their constituents and biological activities by constituent analysis. Previously, three triterpenoids were isolated and determined from *Antrodia cinnamomea* which were designated as antcin A (1, 4 α -methylergost-8,4(8)-diene-3,11-dione-26-oic acid), antcin B (2, 4 α -methylergost-8,4(8)-diene-3,7,11-trione-26-oic acid) and antcin C (3, 7 β -hydroxy-4 α -methylergost-8,4(8)-diene-3,11-dione-26-oic acid) [12]. In this paper we report five additional triterpenoids isolated from the methanolic extract of *Antrodia cinnamomea*. They are antcin D (4, 14-hydroxy-4 α -methyl-3,7,11-trioxo-ergost-8,24(28)-dien-26-oic acid), antcin E (5, 3,11-dioxo-4 α -methylergost-8,14,24(28)-trien-26-oic acid), antcin F (6, 3,11-dioxo-7 β -hydroxy-4 α -methylergost-8,14,24(28)-trien-26-oic acid), methyl antcinate G (7a, 7 α -acetoxy-

3,11-dioxo-4 α -methylergost-8,24(28)-dien-26-oate), and methyl antcinate H (8a, 3 α , 12 α -dihydroxy-7,11-dioxo-4 α -methylergost-8,24(28)-dien-26-oate). With the exception of antcin D, the other four triterpenoids reported in this study are new.

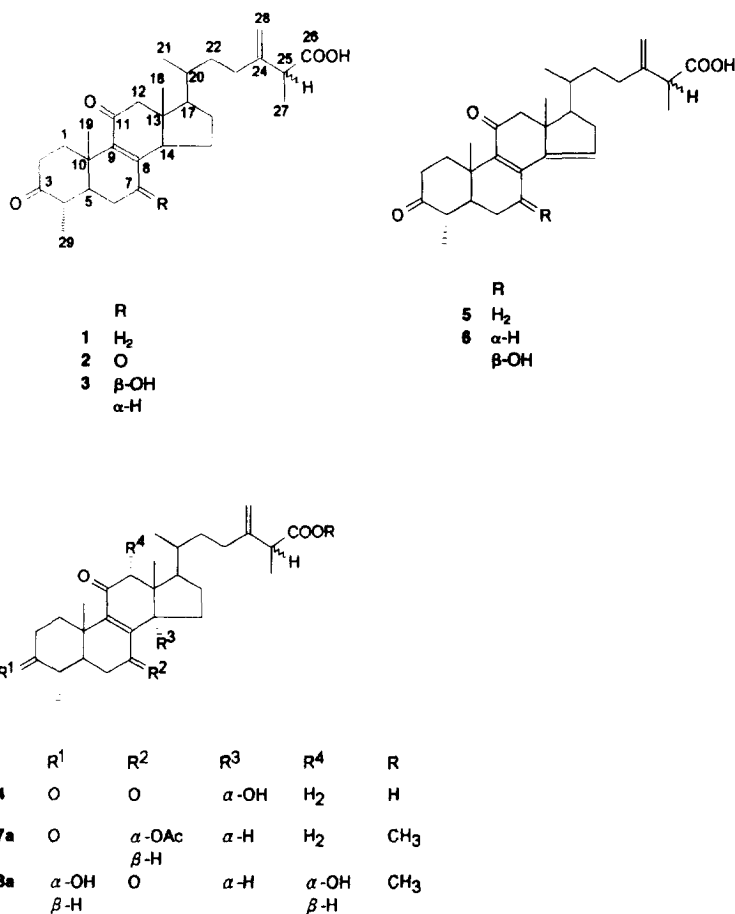
RESULTS AND DISCUSSION

In this study, five compounds from *Antrodia cinnamomea* were isolated from the MeOH extraction and are designated as compounds 4, 5, 6, 7a and 8a.

Compound 4, antcin D, analysed for C₂₉H₄₀O₆ by HRMS, had an absorption at 256 nm (log ϵ 3.92), in the UV spectrum characteristic of an α,β -unsaturated carbonyl group [13]. Its IR absorption bands were observed at 3400–2500 (br., carboxylic acid), 1710 (carbonyl), 1680 (α,β -unsaturated carbonyl), and 875 cm⁻¹ (terminal methylene). Compound 4 gave a M⁺ at m/z 484 [M]⁺, and a fragment ion peak by the elimination of water at m/z 466 [M – H₂O]⁺. The fragment ion peaks at m/z 440 [M – CO₂]⁺, 371 [M – C₆H₉O₂]⁺, and m/z 329 [M – C₉H₁₅O₂]⁺ suggested that the side chain was of the 24(28)-ene-26-carboxylic type [14]. The latter fragment ion peak results from cleavage of the side chain, and the peak at m/z 371 results from the allylic cleavage between C-22 and C-23. In the ¹³C NMR spectrum of 4, four carbonyl carbon signals (δ 210.7, 202.9, 201.3 and 179.7), three quaternary olefinic carbon signals (δ 152.0, 148.1 and 144.2) and a secondary olefinic carbon signal (δ 111.4) suggested the presence of a conjugated 8-ene-7, 11-dione system, a 24(28)-ene-26-carboxylic side chain and a saturated, six-membered cyclic ketone at C-3 [13]. In addition, a quaternary carbon signal bearing a hydroxyl group was observed at δ 81.0.

The ¹H NMR spectrum of 4 showed two methyl singlets at δ 1.53 and 0.81, three methyl doublets at δ 1.30 (d , J = 7.3), δ 1.06 (d , J = 6.4) and δ 0.94 (d , J = 6.4) and two singlets at δ 4.97 and 4.92. Its ¹H NMR spectral pattern resembled that of antcin B (2), but the signal of the

*Author to whom correspondence should be addressed



18-methyl was more downfield than that in **2**. The ^{13}C NMR spectrum of **4** also resembled that of **2**, except for the presence of a quaternary carbon signal at $\delta 81.0$. The ^1H - ^1H and ^1H - ^{13}C shift correlation NMR spectra were analyzed to examine the proton and carbon sequences in **4**. Comparison of its ^{13}C NMR data with those of **2** indicated that the chemical shifts of the carbons in rings A and B, and in the side chain, were practically the same as those of the corresponding carbons in **2** [12]. Furthermore, from a consideration of the ^1H splitting pattern and the coupling constants, together with a comparison of its ^1H NMR and ^{13}C NMR data with those of **2**, it could be deduced that the skeleton of **2** and **4** were similar. The signal due to the quaternary carbon with a hydroxy group at $\delta 81.0$ and the ^{13}C NMR spectrum of **4**, which showed a more upfield shift for the 18-methyl signal at $\delta 17.2$ than that in **2**. The γ -effect of 14 α -hydroxy group, coupled with a blue shift due to the substitution of a 14 α -hydroxy in the UV spectrum (265 nm, $\log \epsilon$ 3.27 for **2**), led to the formulation of **4** as 14-hydroxy-4 α -methyl-3,7,11-trioxo-ergost-8,24(28)-dien-26-oic acid [15].

Compound **5**, antcin E ($[\text{M}]^+$, m/z 452), was assigned the molecular formula $\text{C}_{29}\text{H}_{40}\text{O}_6$ by HRMS. The IR spectrum of **5** showed the presence of a carboxylic acid group (3400–2400, 1710 cm^{-1}), a saturated carbonyl group (1705 cm^{-1}), an α,β -unsaturated carbonyl group

(1658 cm^{-1}) and a terminal methylene group (891 cm^{-1}). The side chain could be identified by its fragment ion peaks at m/z 408 $[\text{M} - \text{CO}_2]^+$ and 297 $[\text{M} - \text{C}_9\text{H}_{15}\text{O}_2]^+$ as of the 24(28)-ene-26-carboxylic type. The ^{13}C NMR spectrum showed signals due to three carbonyl carbons at $\delta 213.1$, 200.6, and 179.9, four olefinic quaternary carbons at $\delta 148.4$, 148.0, 143.0, and 138.0, one olefinic methine at $\delta 126.0$, and one olefinic methylene at $\delta 111.3$, which had one more extending olefinic double bond and one less carbonyl group than **4**. The ^{13}C NMR and UV absorption at 295 nm ($\log \epsilon$ 3.95) suggested that the conjugated system contained an 8,14-dien-11-one, a carbonyl group at C-3 and a 24(28)-ene-26-carboxylic side chain. The ^1H NMR spectrum of **5** exhibited three methyl doublets similar to those of **4**, a more downfield signal of H_3 -18, and broad singlet of H-15 at $\delta 5.87$. Thus, the structure of **5** can be assigned as 3,11-dioxo-4 α -methylergost-8,14,24(28)-trien-26-oic acid.

The HRMS of compound **6**, antcin F, revealed the molecular formula to be $\text{C}_{29}\text{H}_{40}\text{O}_5$, i.e. one additional oxygen atom than **5**. The mass spectrum of compound **6** revealed fragment ion peaks at m/z 450 $[\text{M} - \text{H}_2\text{O}]^+$, 313 $[\text{M} - \text{C}_9\text{H}_{15}\text{O}_2]^+$ and 295 $[\text{M} - \text{C}_9\text{H}_{15}\text{O}_2 - \text{H}_2\text{O}]^+$, showing that **6** has the same side chain and an additional hydroxy group in comparison with **5**. The UV, IR and ^1H NMR spectra of **6** closely resembled those of **5**, but a signal due to a carbinol methine proton at $\delta 4.66$

(*dd*, $J = 7.3, 9.3$), and assigned to H-7 was not observed in the ^1H NMR spectrum of **5**. The ^{13}C NMR spectrum of compound **6** was also similar to that of compound **5**, except for the tertiary carbon at $\delta 67.0$ (C-7) and the chemical shifts of several proximate carbons (C-8, C-9, C-14, C-15). Thus, the structure of compound **6** can be formulated as 3,11-dioxo-7 β -hydroxy-4 α -methylergost-8,14,24(28)trien-26-oic acid.

Compound **7a**, methyl antcinate G, analysed for $\text{C}_{32}\text{H}_{46}\text{O}_6$ by HRMS, and showed in the UV spectrum an absorption at 249.5 nm ($\log \epsilon$ 4.05). The IR spectrum showed a carboxylate group (1735 cm^{-1}), two keto groups ($1706, 1665\text{ cm}^{-1}$), and a terminal methylene (890 cm^{-1}). The MS spectrum of **7a** showed a base peak at m/z 466 $[\text{M} - \text{HOAc}]^+$ and fragment ion peaks at m/z 357 $[\text{M} - \text{C}_{10}\text{H}_{17}\text{O}_2]^+$, m/z 325 $[\text{M} - \text{C}_8\text{H}_{13}\text{O}_2 - \text{HOAc}]^+$, m/z 339 $[\text{M} - \text{C}_7\text{H}_{11}\text{O}_2 - \text{HOAc}]^+$, and m/z 297 $[\text{M} - \text{C}_8\text{H}_{13}\text{O}_2 - \text{HOAc}]^+$, indicating the presence of a methyl 24(28)-ene-26-carboxylate side chain and an acetoxy group. The ^{13}C NMR spectrum showed signals due to the carbonyl carbons at $\delta 212.5$ (C-3) and $\delta 200.3$ (C-11), two carboxylate carbons at $\delta 175.0$ (C-26) and $\delta 170.0$ (OAc), three quaternary olefinic carbons at $\delta 150.0$ (C-9), $\delta 148.4$ (C-24) and $\delta 142.8$ (C-8), a secondary olefinic carbon at $\delta 110.9$ (C-28), and two carbons bearing oxygens at $\delta 68.4$ (C-7) and $\delta 51.9$ (OCH₃). The ^1H NMR contained four methyl singlets at $\delta 3.68$ (COOCH₃), $\delta 2.08$ (acetoxy), $\delta 1.32$ (19-methyl), $\delta 0.71$ (18-methyl), three methyl doublets, an acetoxymethine proton signal at $\delta 5.36$ (*br s*, 7 β -H), and two olefinic protons slightly shifted upfield to $\delta 4.93, 4.88$ (28-CH₂), in comparison with **4**. In the ^1H - ^1H COSY, the H-7 was coupled with the 6-CH₂ protons at $\delta 1.82, 1.61$, while the 6-CH₂ protons were coupled with H-5 proton at $\delta 1.66$, which was assigned by the correlation between H-4 and the relation between H-4 and H₃-29. The configuration of H-7 was determined by its coupling constant. The clear doublet of doublets for the H-7 α proton signal ($J = 8.5, 8.5$) of the acetylated derivative of ganodermic acid E [12] indicated that H-7 of compound **7a** should be a 7 β -configuration proton which had a broad singlet at $\delta 5.36$. Thus, compound **7a** can be assigned as methyl 7 α -acetoxy-3,11-dioxo-4 α -methylergost-8,24(28)-dien-26-oate.

Compound **8a**, methyl antcinate H, has the molecular formula $\text{C}_{30}\text{H}_{44}\text{O}_6$ as confirmed by the HRMS data. The UV spectrum of **8a** showed an absorption band at 274 nm ($\log \epsilon$ 3.79) and its IR spectrum showed the presence of a hydroxy group (3435 cm^{-1} , *br*), an ester group (1728 cm^{-1}), an α,β -unsaturated carbonyl group (1676 cm^{-1}) and a terminal methylene (879 cm^{-1}). The MS spectrum of **8a** showed a molecular ion peak at m/z 500 $[\text{M}]^+$, and other fragment ion peaks at m/z 482 $[\text{M} - \text{H}_2\text{O}]^+$, m/z 469 $[\text{M} - \text{OCH}_3]^+$, m/z 441 $[\text{M} - \text{COOCH}_3]^+$, m/z 359 $[\text{M} - \text{C}_8\text{H}_{13}\text{O}_2]^+$, and m/z 331 $[\text{M} - \text{C}_{10}\text{H}_{17}\text{O}_2]^+$, indicating the presence of a hydroxy group and a side chain, the same as in **7a**. The ^{13}C NMR spectrum showed signals at $\delta 202.6$ (C-11), $\delta 201.7$ (C-7), $\delta 152.2$ (C-9) and $\delta 144.6$ (C-8) similar to that of **4**. Also, in the ^{13}C NMR spectrum are signals due to an 8-ene-7,11-dione system, similar to that of **7a**, and two

tertiary carbon signals at $\delta 80.7$ and $\delta 70.4$. Comparing the ^{13}C NMR spectrum with those of **4** and **7a**, it can be observed that the 3-carbonyl carbon signal (*ca* $\delta 210$) is absent and that the signals due to ring A are changed. The ^1H NMR spectrum of compound **8a** exhibited more upfield tertmethyl signals ($\delta 0.96$, *d*, $J = 7.3$, 29-methyl; $\delta 0.97$, *d*, $J = 6.8$, 21-methyl; and $\delta 1.27$, *d*, $J = 6.8$, 27-methyl), a broad singlet at $\delta 3.79$ and a singlet at $\delta 4.05$ due to two carbinol methine protons. Only the 12-carbinol methine proton resulted in the singlet at $\delta 4.05$ and in the ^1H - ^1H COSY, the absence of the cross peak by the W conformation of 18-methyl and H-12 α and the NOE of H-12 β and H₃-18 suggested that the hydroxy group is attached at 12- α site. The other hydroxy group was assigned at 3- α site by the chemical shift difference and coupling constant of 3 β -carbinol proton ($\delta 3.79$, *brs*) from that of 3 α -carbinol proton (*ca* $\delta 3.24$, $J = \text{ca } 11$ and 5) [14]. Thus, the compound **8a** can be formulated methyl 3 $\alpha, 12\alpha$ -dihydroxy-7,11-dioxo-4 α -methylergost-8,24(28)-dien-26-oate.

Finally, in comparison with constituents from *G. ludicum* and *A. cinnamomea*, the former are triterpenoids of the lanostane type, which have two methyl groups at C-4 [6, 7, 16, 17] and a side-chain with a trisubstituted allylic group at C-24 (25) [6, 7, 17], whereas *A. cinnamomea* has a terminal allyic group at C-2 (28) and only one methyl group at C-4.

EXPERIMENTAL

General. Mps (Yamato MP-21 MEL-TEMP apparatus) are uncorr.; IR: CHCl_3 soln. in CaF_2 cell or KBr disc (only compound **7a**). ^1H and ^{13}C NMR spectra were recorded in CDCl_3 at 400 and 100 MHz, respectively, with TMS as int. standard. EIMS: 30 eV. HPLC was performed with *n*-Hexane-EtOAc or EtOAc- CHCl_3 on a Si60 (Waters, 6 μm , $7.8 \times 300\text{ nm}$) column employing a refractive index detector at a flow rate of 2.0 ml min^{-1} .

Extraction and separation. *Antrodia cinnamomea*, growing in Pin-Tung, Taiwan, was collected and identified by Prof. Chiu-Yuan Chien, Institute of Biological Sciences, National Taiwan Normal University. A Voucher specimen was deposited by Dr T. T. Chang to the Division of Forest Protection, Taiwan Forestry Research Institute. The dry fruit bodies of *Antrodia cinnamomea* (300 g) were cut into small pieces and refluxed $\times 6$ with MeOH (2 l) for 5 hr. The concd MeOH ext was partitioned between H_2O and CHCl_3 , and the CHCl_3 fr. (70 g) was CC on Si gel (800 g) by stepwise elution with *n*-Hexane-EtOAc (7:3), *n*-Hexane-EtOAc (1:1), and *n*-Hexane-EtOAc (2:3). The *n*-Hexane-EtOAc (7:3) elution was cc on HPLC using *n*-Hexane-EtOAc (3:1) to give **7a** (10 mg). The *n*-Hexane-EtOAc (1:1) elution was chromatographed on Si gel repeatedly (EtOAc- CHCl_3 , 2:3) and then separated by HPLC (*n*-Hexane-EtOAc, 55:45) to afford **5** (14 mg), **4** (24 mg) and **8a** (21 mg). The 60% EtOAc-*n*-Hexane elution was separated by HPLC (*n*-Hexane-EtOAc, 2:3) to give **6** (7 mg).

Table 1. ^1H NMR spectral data of compounds **4**, **5**, **6**, **7a** and **8a***

Atom	4	5	6	7a	8a
1	1.56, 3.02	1.29, 2.91	2.83, 1.26	1.42, 3.10	2.34, 1.43
2	2.60, 2.44	2.53, 2.18	2.64, 2.33	2.40, 2.55	1.79, 1.79
3	—	—	—	—	3.79
4	2.50	2.40	2.40	2.35	1.72
5	2.01	1.45	1.40	1.66	2.14
6	2.46, 2.53	1.84, 1.48	2.30, 1.53	1.61, 1.82	2.40, 2.22
7	—	2.49, 2.46	4.66 <i>dd</i> ($J = 7.3, 9.3$)	5.36 <i>br s</i>	—
12 β	3.18 <i>d</i> ($J = 13.7$)	2.66 <i>d</i> ($J = 13.2$)	2.74 <i>d</i> ($J = 13.7$)	2.86 <i>d</i> ($J = 14.6$)	4.05 <i>s</i>
12 α	2.50 <i>d</i> ($J = 13.7$)	2.53 <i>d</i> ($J = 13.2$)	2.54 <i>d</i> ($J = 13.7$)	2.38 <i>d</i> ($J = 14.6$)	—
14	—	—	—	2.69 <i>dd</i>	3.02 <i>dd</i>
15	1.88, 2.34	5.87 <i>br s</i>	6.04 <i>br s</i>	1.45, 1.61	2.53, 1.49
16	1.38, 2.07	2.52, 2.50	2.40, 2.40	1.36, 1.96	1.95, 1.26
17	2.04	1.75	1.66	1.48	1.83
18	0.81 <i>s</i>	0.92 <i>s</i>	0.98 <i>s</i>	0.71 <i>s</i>	0.64 <i>s</i>
19	1.53 <i>s</i>	1.41 <i>s</i>	1.51 <i>s</i>	1.32 <i>s</i>	1.30 <i>s</i>
20	1.42	1.60	1.64	1.45	1.40
21	0.94 <i>d</i> ($J = 6.4$)	0.92 <i>d</i> ($J = 5.4$)	0.95 <i>d</i> ($J = 6.4$)	0.93 <i>d</i> ($J = 5.8$)	0.97 <i>d</i> ($J = 6.8$)
22	1.25, 1.62	1.62, 1.26	1.59, 1.27	1.20, 1.52	1.57, 1.17
23	1.98, 2.17	2.20, 2.05	2.20, 2.05	1.93, 2.13	2.17, 1.89
25	3.15 <i>q</i> ($J = 7.3$)	3.14 <i>q</i> ($J = 6.8$)	3.17 <i>q</i> ($J = 6.8$)	3.13 <i>q</i> ($J = 6.8$)	3.14 <i>q</i> ($J = 6.8$)
27	1.30 <i>d</i> ($J = 7.3$)	1.30 <i>d</i> ($J = 6.8$)	1.31 <i>d</i> ($J = 6.8$)	1.28 <i>d</i> ($J = 6.8$)	1.27 <i>d</i> ($J = 6.8$)
28	4.97, 4.92	4.98, 4.93	4.99, 4.95	4.88, 4.93	4.92, 4.89
29	1.06 <i>d</i> ($J = 6.4$)	1.05 <i>d</i> ($J = 6.8$)	1.08 <i>d</i> ($J = 6.8$)	1.00 <i>d</i> ($J = 6.8$)	0.96 <i>d</i> ($J = 7.3$)
OAc	—	—	—	2.08 <i>s</i>	—
OCH ₃	—	—	—	3.68 <i>s</i>	3.68 <i>s</i>

*Values in parentheses are coupling constants in Hz (CDCl_3 , TMS as int. standard, 400 MHz).

Antcin D (4). Compound **4** was crystallized from *n*-Hexane- CHCl_3 as yellow needles (24 mg), mp 173–176°, $[\alpha]^{24} + 109^\circ$ (CHCl_3 ; c 0.55). IR: (CaF_2 , CHCl_3) $\nu_{\max} \text{ cm}^{-1}$: 3400, 3020, 2978, 1720, 1709, 1680, 1460, 1421, 1377, 1230, 1174, 1107, 875; UV $\lambda^{\text{MeOH}} \text{ nm}$ ($\log \epsilon$): 256 (3.92). HRMS m/z 484.2831 $[\text{M}]^+$ $\text{C}_{29}\text{H}_{40}\text{O}_6$ requires: 484.2826. EIMS (30 eV) m/z (rel. int.): 484 $[\text{M}]^+$ (100), 466 (33), 456 (65), 440 (20), 404 (18), 371 (18), 369 (19), 358 (38), 356 (30), 329 (32), 312 (30), 274 (50), 262 (50), 162 (32), 109 (43). ^1H NMR (CDCl_3): (Table 1). ^{13}C NMR (CDCl_3): (Table 2).

Antcin E (5). Compound **5** was obtained as a yellow syrup (14 mg). $[\alpha]^{24} + 76.7^\circ$ (CHCl_3 ; c 1.2). IR: (CaF_2 , CHCl_3) $\nu_{\max} \text{ cm}^{-1}$: 3400, 3024, 2970, 2934, 2881, 1720, 1705, 1658, 1616, 1570, 1458, 1423, 1377, 1236, 1195, 906, 891. UV $\lambda^{\text{MeOH}} \text{ nm}$ ($\log \epsilon$): 295 nm (3.95). HRMS m/z 452.2930 $[\text{M}]^+$ $\text{C}_{29}\text{H}_{40}\text{O}_4$ requires: 452.2928. EIMS (30 eV) m/z (rel. int.): 452 (46), 438 (9), 408 (9), 378 (100), 379 (35), 370 (5), 363 (9), 323 (35), 297 (11), 281 (8), 199 (5), 159 (9), 121 (15), 109 (8), 95 (5). ^1H NMR (CDCl_3): (Table 1). ^{13}C NMR (CDCl_3): (Table 2).

Antcin F (6). Compound **6** was crystallized from EtOAc- CHCl_3 as needles (7 mg). $[\alpha]^{24} + 120^\circ$ (CHCl_3 ;

c 0.15). IR: (CaF_2 , CHCl_3) $\nu_{\max} \text{ cm}^{-1}$: 3400, 3038, 3020, 2934, 1720, 1707, 1666, 1608, 1458, 1377, 1234, 1087, 898, 873, 854. UV $\lambda^{\text{MeOH}} \text{ nm}$ ($\log \epsilon$): 294 (3.87), 230 (3.49). HRMS m/z 468.2878 $[\text{M}]^+$ $\text{C}_{29}\text{H}_{40}\text{O}_5$ requires: 468.2877. EIMS (30 eV) m/z (rel. int.): 468 (30) $[\text{M}]^+$, 450 (88), 404 (3), 394 (100), 395 (33), 377 (40), 339 (6), 323 (10), 313 (5), 295 (15), 253 ((8), 225 (5), 172 (10), 171 (12), 154 (35), 137 (8), 109 (16), 91 (3). ^1H NMR (CDCl_3): (Table 1). ^{13}C NMR (CDCl_3): (Table 2).

Methyl antcinatate G (7a). Compound **7a** was a colourless liquid. (10 mg). $[\alpha]^{24} + 114.5^\circ$ (CHCl_3 ; c 1.1). IR: (KBr , CHCl_3) $\nu_{\max} \text{ cm}^{-1}$: 1735, 1707, 1666, 1373, 1226, 1192, 890; UV $\lambda^{\text{CHCl}_3} \text{ nm}$ ($\log \epsilon$): 249.50 (4.05). HRMS m/z 526.3285 $[\text{M}]^+$ $\text{C}_{32}\text{H}_{46}\text{O}_6$ requires: 526.3294. EIMS (30 eV) m/z (rel. int.): 526 (2) $[\text{M}]^+$, 466 (100) $[\text{M} - \text{HOAc}]^+$, 484 (12), 467 (25), 419 (20), 357 (6), 339 (10), 325 (52), 297 (20), 245 (22), 171 (10), 135 (15), 109 (22), 95 (8). ^1H NMR (CDCl_3): (Table 1). ^{13}C NMR (CDCl_3): (Table 2).

Methyl antcinatate H (8a). Compound **8a** was crystallized from *n*-Hex- CHCl_3 as yellow needles (21 mg), mp 170–173°. $[\alpha]^{24} + 102^\circ$ (CHCl_3 ; c 0.45). IR: (CaF_2 , CHCl_3) $\nu_{\max} \text{ cm}^{-1}$: 3437, 3634, 3018, 2935, 1730, 1676,

Table 2. ^{13}C NMR spectral data of compounds **4**, **5**, **6**, **7a** and **8a**

Atom	4	5	6	7a	8a
1	35.6	35.0	35.2	34.7	27.8
2	37.4	36.1	36.5	37.6	28.9
3	210.7	213.1	212.1	212.5	70.4
4	44.0	44.2	43.8	43.5	34.5
5	48.8	50.3	47.1	45.3	40.7
6	38.9	20.7	30.6	27.8	38.1
7	201.3	27.8	67.0	68.4	201.7
8	144.2	138.0	145.0*	142.8	144.6
9	152.0	143.0	144.0*	150.0	152.2
10	38.4	36.7	37.1	37.1	38.3
11	202.9	200.6	200.9	200.3	202.6
12	50.1	55.8	56.1	57.6	80.7
13	49.3	48.3	49.5	47.4	49.5
14	81.0	148.0	140.0*	51.0	41.8
15	33.3	126.0	121.6	22.9	23.9
16	26.6	37.8	37.7	27.4	26.9
17	48.3	56.5	56.3	55.0	45.5
18	17.2	17.4	17.3	11.9	11.4
19	16.3	18.0	17.8	16.3	16.1
20	35.4	33.4	33.4	35.7	35.4
21	18.5	18.3	18.5	18.4	18.0
22	34.1	33.6	33.7	33.8	33.8
23	31.4	31.2	31.1	31.1	31.3
24	148.1	148.4	148.1	148.4	148.5
25	45.4	45.4	45.5	45.6	45.6
26	179.7	179.9	179.0	175.0	175.0
27	16.2	16.2	16.3	16.3	16.3
28	111.4	111.3	111.4	110.9	110.9
29	11.5	11.6	11.6	51.9	15.6
OCH ₃	—	—	—	51.9	51.9
OCOMe	—	—	—	170.0	—
OCOMe	—	—	—	21.2	—

*Assignments may be reversed (CDCl_3 , TMS as int. standard, 100 MHz).

1458, 1437, 1377, 1238, 1170, 1055, 896, 877. UV λ^{MeOH} nm (log ϵ) 274 (3.79). HRMS m/z 500.3140 $[\text{M}]^+$ $\text{C}_{30}\text{H}_{44}\text{O}_6$ requires: 500.3139. EIMS (30 eV) m/z (rel. int.): 500 (100) $[\text{M}]^+$, 482 (5), 441 (7), 423 (6), 413 (5), 359 (3), 341 (11), 331 (6), 329 (14), 313 (13), 277 (12), 275 (24), 215 (21), 201 (22), 75 (25), 121 (24), 109 (39), 95 (13). ^1H NMR (CDCl_3): (Table 1). ^{13}C NMR (CDCl_3): (Table 2).

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REFERENCES

1. Hirotsani, M., Furuya, T. and Shiro, M. (1984) *Phytochemistry* **24**, 2055.
2. Nishitoba, T., Oda, K. and Sato, H. (1988) *Agric. Biol. Chem.* **52**, 367.
3. Kohda, H., Tokumoto, W., Sakamoto, K., Fugli, M., Hirai, Y., Yamasaki, K., Komoda, Y., Nakamura, H., Ishihara, S. and Uchida, M. (1985) *Chem. Pharm. Bull.* **33**, 1367.
4. Wang, T.-N., Chen, J.-C., Shiao, M.-S. and Wang, C.-T. (1989) *Biochim. Biophys. Acta* **986**, 151.
5. Wang, C.-N., Chen, J.-C. Shiao, M.-S. and Wang, C.-T. (1991) *Biochem. J.* **277**, 189.
6. Toth, J. O., Luu, B., Beck, J.-P. and Ourisson, G. (1983) *J. Chem. Res. (S)*, 299.
7. Morigiwa, A., Kitabatake, K., Fujimoto, Y. and Ikekawa, H. (1986) *Chem. Pharm. Bull.* **34**, 3025.
8. Komoda, Y., Shimizu, M., Sonoda, Y. and Sato, Y. (1989) *Chem. Pharm. Bull.* **37**, 531.
9. Chang, T. T. and Chou, W. N. (1995) *Mycol. Res.* **99**, 756.
10. Mu, Z. and Su, Q. (1990) *Acta Bot. Yunnan.* **12**, 395.
11. Arisawa, M., Fujita, A., Saga, M., Fukumura, H., Hayashi, T., Shimizu, M. and Morita, N. (1986) *J. Nat. Prod.* **49**, 621.
12. Cherng, I.-H. and Chiang, H.-C. (1995) *J. Nat. Prod.* **58**, 365.
13. Kikuchi, T., Kanomi, S., Murai, Y., Kadota, S., Tsubono, K. and Ogita, Z.-I. (1986) *Chem. Pharm. Bull.* **34**, 4030.
14. Thappa, P. K., Agarwal, S. G. and Atal, C. K. (1981) *Phytochemistry* **20**, 1746.
15. Young, S.-W. (1989) Masters Thesis, The Institute of Pharmacy, National Taiwan University, Taiwan.
16. Hirotsani, M., Furuya, T. and Shiro, M. (1985) *Phytochemistry* **24**, 2055.
17. Lin, L. J., Shiao, M. S. and Yeh, S. F. (1988) *J. Nat. Prod.* **51**, 918.