



AN ARTEMISININIC ACID ANALOGUE FROM *TITHONIA DIVERSIFOLIA*

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Key Word Index—*Tithonia diversifolia*; compositae; stem, artemisinic acid analogue.

Abstract—A new artemisinic acid analogue compound has been isolated from mature stems of *Tithonia diversifolia* and its structure determined.

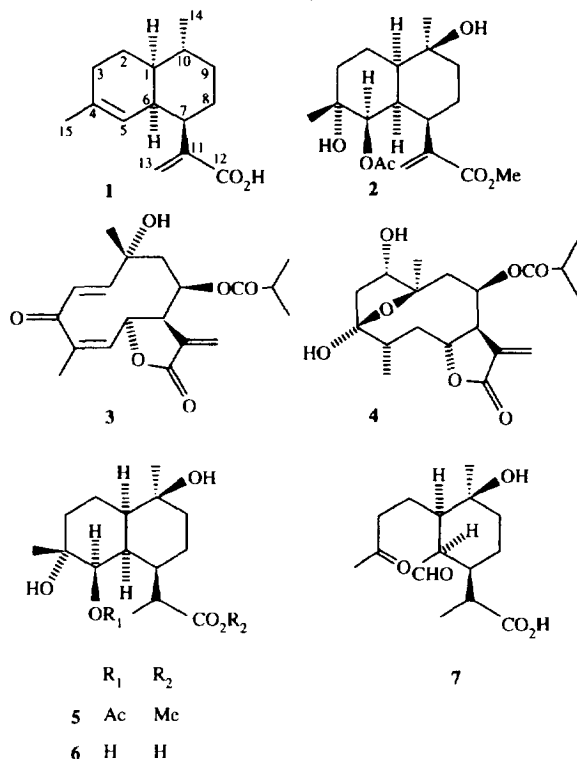
INTRODUCTION

The discovery of artemisinin, a potent antimalarial drug isolated from the Chinese traditional medicinal plant, *Artemisia annua*, and the co-occurrence of artemisinic acid **1**, the biogenetic precursor of artemisinin [1, 2], has intensified the search for other analogues of artemisinin, as well as artemisinic acid, in the Compositae in general and in other *Artemisia* species in particular. In continuation of our earlier work on *Tithonia diversifolia*, from which we isolated several antifeedant sesquiterpene lactones [3, 4], we investigated mature stems of this species and isolated an ester **2**, related to artemisinic acid **1**. The structure of this ester has been established through extensive use of ^1H and ^{13}C NMR spectrometry and chemical transformations. Compound **2** has functionalities suitable for further conversion to 1,2,4-trioxane-type artemisinin analogues [5-7].

RESULTS AND DISCUSSION

The chloroform extract of air-dried mature stems after separation of **3** and **4**, gave by further thin layer chromatographic separation on silica gel, a crystalline compound A, $\text{C}_{18}\text{H}_{28}\text{O}_6$ by HR-mass spectrometry with a $[\text{M}]^+$ at m/z 340. Other important peaks were observed at m/z 323, 322, 307, 291, 280, 262, etc. The IR spectrum showed characteristic bands at 3500 (—OH stretching), 1725 (—OCOCH₃ stretching), 1685 and 1625 cm^{-1} (α,β -unsaturated acid). In the ^1H NMR spectrum, four singlets at δ 0.8, 1.15, 2.0 and 3.6, each integrating for three protons revealed the presence of two tertiary methyls, one acetoxyl methyl and one ester methyl, respectively. A broad one-proton singlet at δ 4.6 was attributed to the proton of the acetoxyl group. Two broad one-proton singlets at δ 5.4 and 6.0 were assigned to an exocyclic methylene of an α,β -unsaturated acid. The broad-band decoupled ^{13}C NMR spectrum showed the presence of 18 carbons, three of which were quaternary, three methines,

five methylenes and four methyl carbons as revealed by the off-resonance ^{13}C NMR spectrum.



Hydrogenation of A under atmospheric pressure using platinum oxide catalyst, gave a dihydro compound B, as revealed by its mass spectrum. The IR spectra showed characteristic bands at 3500, 2950, 2850 and 1725 cm^{-1} . The ^1H NMR spectrum showed the absence of the two broad singlets at δ 5.4 and 6.0 present in the ^1H NMR spectrum of A and the appearance of a three proton doublet at δ 1.2 with $J = 7$ Hz. Four three proton singlets at δ 0.8, 1.1, 2.05, and 3.65, were assigned to two tertiary methyls, one acetoxyl and one ester methyl group, respectively. The dihydro derivative was hydrolysed by sodium methoxide to give a trihydroxy acid $\text{C}_{15}\text{H}_{26}\text{O}_5$ whose IR spectrum revealed the presence of hydroxyl

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groups (3550 cm^{-1}) and a carboxylic acid group (1725 cm^{-1}). The $^1\text{H NMR}$ spectrum contained a three-proton doublet at $\delta 1.2$ ($J = 7\text{ Hz}$), two three-proton singlets at $\delta 0.8$ and 1.05 and two broad one-proton singlets at $\delta 3.6$ and 11.5 . The hydrolysed compound was easily cleaved to a ketoaldehyde carboxylic acid ($\text{C}_{15}\text{H}_{24}\text{O}_5$) by activated manganese dioxide in dichloromethane, thus showing the presence of a tertiary–secondary vicinal diol system. The IR spectrum revealed the presence of hydroxyl groups (3550 cm^{-1}), overlapping strong bands of a carboxylic acid group and an aldehyde group (1725 cm^{-1}), and a ketone (1710 cm^{-1}). The $^1\text{H NMR}$ spectrum showed the presence of one methyl doublet at $\delta 1.1$ ($J = 7\text{ Hz}$), two methyl singlets at $\delta 0.95$ and 2.2 , one broad two-proton singlet at $\delta 3.5$, a broad one-proton singlet at $\delta 11.5$ and a one-proton doublet at $\delta 9.6$. These results led to structures **5**, **6** and **7** for compound B, trihydroxy acid and the ketoaldehyde carboxylic acid, respectively. Therefore, the gross structure of the parent compound A could be expressed as **2**. The $^{13}\text{C NMR}$ data of A also fully supported structure **2** for this compound.

The relative stereochemistry of **2** was assigned on the basis of coupling constants obtained by decoupling experiments. Since a *cis*-decalin can exist in two conformation, it is clear from the coupling constants $J_{5,6} = 2.0\text{ Hz}$ and $J_{1,6} = 4\text{ Hz}$ that H-6 is equatorial and H-1 is axial. In the presence of trifluoroacetyl isocyanide, the signals of H-5 and H-6 were shifted to $\delta 5.30$ and 3.10 , respectively, indicating that the hydroxyl and the acetoxy group at C-4 and C-5 are α and β , respectively, and also that H-6 is α . Since the H-1 signal was not appreciably shifted, it was inferred that the hydroxyl at C-10 is β .

EXPERIMENTAL

General. Mps: uncorr. IR: KBr or neat film. UV: EtOH. $[\alpha]_D^{25}$: CHCl_3 . $^1\text{H NMR}$: 270 MHz ; $^{13}\text{C NMR}$: 75 MHz ; CDCl_3 with TMS as int. standard. EIMS at 70 eV . Petrol refers to bp $60\text{--}80^\circ\text{ fr}$.

Isolation of methyl 5-acetoxy-4,5-dihydro-4,10-dihydroxy-5-acetoxyartemisininate 2. Air-dried mature stems of *T. diversifolia* Hemsl. A. Grey (2 kg) collected in the Bokakhat area of the Golaghat District, Assam, India, were extracted with CHCl_3 in a Soxhlet apparatus until the extract was colourless (*ca* 8 h). The crude material, after evapn of CHCl_3 under red. pres., was dissolved in MeOH (400 ml) containing 10% H_2O and left overnight. The ppt. was filtered and the filtrate washed with petrol ($8 \times 200\text{ ml}$). The MeOH layer was concd under red. pres. and then extracted with CHCl_3 . The washed and dried extract after evapn yielded 10 g of a crude gum which was chromatographed over 200 g of silica gel (60–120 mesh) and 200 ml frs were collected in the following order: frs 1–5 (petrol), 6–10 (petrol–EtOAc, 9:1), 11–16 (petrol–EtOAc, 4:1), 17–21 (petrol–EtOAc, 2:1), 22–28 (petrol–EtOAc, 1:1), 19–32 (petrol–EtOAc, 1:2), 33–38 (petrol–EtOAc, 1:4), 39–41 (EtOAc). Frs 11–16 showed several very minor compounds on TLC which were not investigated further. Frs 28–30 showed a minor spot on

TLC which was identified by spectroscopic means as tagitinin C **3** reorted by us earlier from the leaves of this species. Similarly frs 35–40 contained another minor compound which was identified by spectroscopic means as tagitinin A **4**, another compound reported by us earlier from *T. diversifolia* [2].

Frs 17–25 showed a major spot on TLC (petrol–EtOAc, 1:9) which were combined to give *ca* 900 mg of crystalline methyl 5-acetoxy-4,5-dihydro-4,10-dihydroxy-5-acetoxyartemisininate **2** (0.45%). Mp 170° (petrol–EtOAc, 1:1). UV λ_{max} 220 nm ($\epsilon = 10,000$). IR $\nu_{\text{max}}\text{ cm}^{-1}$: 3500, 2925, 2850, 1710, 1625, 1440. $^1\text{H NMR}$: $\delta 0.8$ (3H, s, H-14), 1.15 (3H, s, H-15), 2.0 (3H, s, CH_3COO), 2.3 (1H, *ddd*, $J = 2.5, 4$ and 4.5 Hz , H-1), 2.6 (1H, *ddd*, $J = 2, 4$ and 4 Hz , H-6), 3.6 (3H, s, COOMe), 4.6 (1H, *dq*, $J = 2$ and 1 Hz , H-5), 5.4 (1H, *br s*, H-13a), 6.0 (1H, *br s*, H-13b). $^{13}\text{C NMR}$: $\delta 173.1$ (s, acetoxy CO), 171.7 (s, C-12), 130.8 (s, C-11), 128.8 (t, C-13), 119.4 (d, C-5), 77.5 (s, C-4), 60.1 (s, C-10), 48.9 (d, C-6), 43.8 (d, C-1), 41.4 (d, C-7), 38.9 (t, C-3), 34.4 (t, C-9), 31.8 (t, C-8), 30.5 (t, C-2), 29.6 (q, acetoxy methyl), 28.9 (q, ester methyl), 26.9 (q, C-15), 25.8 (q, C-14). MS m/z : 340 $[\text{M}]^+$, 323, 322, 307, 291, 280, 262, 249, 246, 223, 191, 187, 181, 145, 135, 121, 119, 107. (Found: C, 63.52; H, 8.63; $\text{C}_{18}\text{H}_{28}\text{O}_6$ requires C, 63.53; H, 8.64%).

5-Acetoxy-4,5-dihydroxy-4,5,11,13-tetrahydroartemisinic acid methyl ester 5. A soln of 50 mg (0.15 mmol) of Me 6-acetoxy-4,5-dihydro-4,10-dihydroxy-5-acetoxyartemisininate **2** in 20 ml of MeOH was hydrogenated at room temp. and atm. pres. using platinum oxide (50 mg) as catalyst. After 6 hr, the catalyst was filtered off and washed with MeOH (5 ml $\times$ 3). The filtrate was evapd under red. pres. to obtain a crude mass which on purification by prep. TLC on silica gel afforded 40 mg of **5**. IR $\nu_{\text{max}}\text{ cm}^{-1}$: 3500, 2950, 2850, 1725, 1450, 1380. $^1\text{H NMR}$: $\delta 0.8$ (3H, s, H-14), 1.1 (3H, s, H-15), 1.2 (3H, *d*, $J = 7\text{ Hz}$, H-13), 2.05 (3H, s, CH_3COO), 3.65 (3H, s, COOMe), 4.75 (1H, *br s*, H-5). MS m/z : 342 $[\text{M}]^+$, 324, 309, 293, 282, 264, 249, 223, 193, 161, 123, 119, 109, 107, 105. (Found: C, 63.11; H, 8.74; $\text{C}_{18}\text{H}_{30}\text{O}_6$ requires C, 63.15; H, 8.77%).

4,5,6-Trihydroxy-4,5,11,13-tetrahydroartemisinic acid 6. A soln of 100 mg (0.35 mmol) of 6-acetoxy-4,5-dihydroxy-4,5,11,13-tetrahydroartemisinic acid Me ester **5** in 5 ml of MeOH was treated with freshly prep NaOMe (from 150 mg (6.50 mmol) of Na in 3 ml dry MeOH) and the reaction mixt. stirred at 20° monitoring the reaction by TLC. When no starting material remained, the reaction mixt. was dild with 200 ml of H_2O , neutralized with HOAc and then extracted with EtOAc (100 ml $\times$ 5). The dried extract was evapd under red. pres. and HOAc removed by co-distilling with toluene under vacuum. Purification by prep. TLC (petrol–EtOAc, 1:1) furnished 70 mg of **6**. Mp 198° (EtOAc). IR $\nu_{\text{max}}\text{ cm}^{-1}$: 3550, 2950, 2850, 1725, 1650, 1450, 1375. $^1\text{H NMR}$: $\delta 0.8$ (3H, s, H-14), 1.05 (3H, s, H-15), 1.2 (3H, *d*, $J = 7\text{ Hz}$, H-13), 3.6 (1H, *br s*, H-5), 11.5 (1H, *br s*, COOH). MS m/z : 286 $[\text{M}]^+$, 269, 252, 235, 224, 207, 193, 183, 177, 175, 163, 133 and 107. (Found: C, 62.91; H, 9.05; $\text{C}_{15}\text{H}_{26}\text{O}_5$ requires C, 62.93; H, 9.09%).

Ketoaldehydcarboxylic acid 7. To a soln of 70 mg (0.23 mmol) of 4,5,6-trihydroxy-4,5,11,13-tetrahydroartemisinic acid **6** in dry CH_2Cl_2 , was added 100 mg

(1.15 mmol) of activated MnO_2 (freshly prepd [8]) and the reaction mixt. stirred continuously until no starting material could be detected by TLC. The reaction mixt. was then filtered and the residue washed with CH_2Cl_2 (5 ml $\times$ 4). The combined organic phases were evapd under red. pres. to give a crude product, which on purification by prep. TLC gave a gummy mass, yield 57 mg. IR ν_{max} cm^{-1} : 2900, 2850, 1725, 1710, 1450, 1375, 1350. ^1H NMR: δ 0.95 (3H, s, H-14), 1.1 (3H, d, $J = 7$ Hz, H-13), 2.2 (3H, s, H-15), 3.5 (overlapping signals H-3 and H-6), 9.8 (1H, d, $J = 7$ Hz, CHO at H-5), 11.5 (1H, br s, COOH). MS m/z : 296 $[\text{M}]^+$, 282, 280, 265, 223, 206, 177, 133, 107, 95, 88 and 81. (Found: C, 63.36; H, 8.43; $\text{C}_{15}\text{H}_{24}\text{O}_5$ requires C, 63.38; H, 8.45%).

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