



# SCROKOELZISIDE A, A TRITERPENE GLYCOSIDE FROM SCROPHULARIA KOELZII\*

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**Abstract**—Scrokoelziside A, isolated from the aerial parts of *Scrophularia koelzii* was shown to be 3-O- $\{[\alpha-L-rhamnopyranosyl-(1 \rightarrow 3), \beta-D-glucopyranosyl-(1 \rightarrow 2)]-\beta-D-fucopyranosyl-(1 \rightarrow 4)-\beta-D-glucopyranosyl}-13<math>\beta$ ,28-epoxyolean-11-en-23-ol, on the basis one- and two-dimensional NMR homo- and hetero-nuclear spectroscopic evidence.

### INTRODUCTION

In continuation of our chemical studies on *Scrophularia* koelzii [1], we now report the isolation and structural elucidation of a triterpene glycoside, scrokoelziside A.

#### RESULTS AND DISCUSSION

The chloroform extract of the dried aerial parts on chromatographic purification on a silica gel column and preparative TLC yielded one glycoside, scrokoelziside A (1). The glycosidic nature of compound 1 was indicated by the broad absorption bands at 3400 and 1070 cm<sup>-1</sup> for hydroxyl groups in its infrared spectrum and many resonances in the region 61–78 ppm in its  $^{13}$ C NMR spectrum. The mass spectrum of this compound was successfully determined by negative-ion FAB mass spectrometry in which the [M]<sup>+</sup> peak appeared at m/z 1071, which is compatible with the molecular formula  $C_{54}H_{88}O_{21}$ .

The <sup>1</sup>H NMR spectrum (Table 1) displayed a pattern characteristic of a pentacyclic triterpene (six tertiary methyl singlets at  $\delta 0.80$ , 0.90, 0.93, 0.94, 1.06 and 1.31), two AB quartets (J=11.2 and 11.6 Hz) at  $\delta 3.32$  and 3.73, and at  $\delta 3.72$  and 4.38, respectively, and an oxymethine at  $\delta 4.15$ . There were two olefinic protons ( $\delta 5.53$ , 1H, d, J=10.4 Hz; 5.53 dd, J=10.4, 1.5 Hz) of an endocyclic disubstituted olefinic bond which indicated a  $\Delta^{11}$ -oleanene skeleton [2, 3]. The sugar moiety showed in the <sup>1</sup>H NMR spectrum two doublets (J=6.2 Hz) for secondary methyl groups at  $\delta 1.38$  and 1.73, three doublets for the anomeric protons ( $\delta 5.56$ , J=7.2 Hz, Glc-1; 4.89, d, J=7.5 Hz, Fuc-1; 5.24, d, J=7.5 Hz, Glc-1) indicative

of  $\beta$ -anomeric configuration [4] of three monosaccharides and a broad singlet ( $W_{1/2} = 1.8 \text{ Hz}$ ) at  $\delta 5.82$  attributed to H-1 of a monosaccharide (Rha) with  $\alpha$ anomeric configuration [4]. The assignment of the methyl resonances at  $\delta 1.38$  and 1.73 was based upon the fact that these exhibited their correlation with the methine resonance at  $\delta$ 3.59 and 4.94 in the COSY spectrum. The cross-peak connectivity was further elucidated, which led to the assignment of the upfield methyl resonance ( $\delta$ 1.38) to a fucopyranosyl and the resonance at lower field ( $\delta$ 1.73) to a rhamnopyranosyl residue. Many overlapping multiplets were observed between  $\delta$ 3.5-5.0, the assignments of which were performed by means of two dimensional <sup>1</sup>H-<sup>1</sup>H HOMOCOSY and <sup>1</sup>H-<sup>13</sup>C HETCOR spectra. The analysis of these spectral data confirmed the presence of a tetrasaccharide residue composed of two hexopyranose (2 x glucopyranose) and two 6-deoxyhexopyranose (fuc + rha) moieties in 1.

Starting with H-3 at  $\delta$ 4.15, the resonances at  $\delta$ 2.02 and 1.95 could be assigned to CH<sub>2</sub>-2 which, in turn, exhibited their connectivity to CH<sub>2</sub>-1 at  $\delta$ 0.97 and 1.78. Among the olefinic methine resonances, the high-field methine resonance at  $\delta$ 5.53 corresponded to H-11 as it displayed a cross-peak with H-9 at  $\delta$ 2.03. The H-23 and H-28 methylene protons were assigned to the resonances at  $\delta$ 3.72 and 4.38, and  $\delta$ 3.32 and 3.73, respectively, related by the geminal couplings.

The methyl signals at  $\delta 0.80$ , 0.90, 0.93, 0.94, 1.06 and 1.31 were correlated with the  $^{13}$ C NMR signals at  $\delta 23.62$ , 33.66, 18.70, 19.63, 12.75 and 19.90, respectively. The chemical shift values of the rest of the methylene and methine resonances could readily be established from the COSY and HETCOR spectra.

The <sup>13</sup>C NMR spectrum exhibited 54 carbon signals and of these, 30 carbons accounted for the aglycone moiety while the remaining 24 carbon resonances were

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Table 1. 13C NMR and 1H NMR spectral data for compound 1\*

Atom No.	Aglycone residue†			Sugar residue†	
	13C	¹H	Atom No.	<sup>13</sup> C	¹H
1	38.66 (CH <sub>2</sub> )	0.97, 1.78	Glucose (internal) (Glc)		
2	25.88 (CH <sub>2</sub> )	1.95, 2.02	1	103.97 (CH)	5.24 (1H, d,
3	82.63 (CH)	4.15 (1H, m)		,	J = 7.5  Hz
4	43.88 (C)		2	76.30 (CH)	3.91
5	47.84 (CH)	1.52	3	77.64 (CH)	4.16
6	17.71 (CH <sub>2</sub> )	1.50, 1.76	4	78.48 (CH)	4.37
7	31.55 (CH <sub>2</sub> )	1.31, 1.40	5	76.48 (CH)	3.70
8	42.05 (C)	_	6	63.19 (CH <sub>2</sub> )	4.05 (2H, m)
9	53.75 (CH)	2.03		se (Fuc)	` ' '
10	36.33 (C)	<del></del>	1	104.14 (CH)	4.89 (1H, d,
11	132.07 (CH)	5.53 (1H, dd,	_	,	J = 7.5  Hz
		J = 10.4  Hz  and  1.5  Hz)	2	77.24 (CH)	4.62
12	131.75 (CH)	5.93 (1H, d,	3	84.82 (CH)	4.03
	101110 (011)	J = 10.4  Hz)	4	77.26 (CH)	4.15
13	84.75 (C)	_	5	70.53 (CH)	3.59
14	43.88 (C)	_	6	17.25 (CH <sub>3</sub> )	1.38 (3H, d,
15	31.09 (CH <sub>2</sub> )	0.89, 1.84		(3/	J = 6.2  Hz
16	25.68 (CH <sub>2</sub> )	1.01, 1.94	Glucose (terminal) (Glc)		
17	41.73 (C)		1	105.02 (CH)	5.56 (1H, d,
18	51.49 (CH)	1.69		()	J = 7.2  Hz
19	37.37 (CH <sub>2</sub> )	1.26, 1.68	2	75.98 (CH)	4.08
20	31.77 (C)	<del></del>	3	78.87 (CH)	4.18
21	31.55 (CH <sub>2</sub> )	$1.27 \times 2$	4	72.06 (CH)	4.31
22	26.08 (CH <sub>2</sub> )	1.20, 1.49	5	77.24 (CH)	3.62
23	64.19 (CH <sub>2</sub> )	3.72, 4.38	6	61.37 (CH <sub>2</sub> )	4.25, 4.30
	(12)	(AB, q, J = 11.6  Hz)		27	(each 1H, m)
24	12.75 (CH <sub>3</sub> )	1.06 (3H, s)	Rhan	nose (Rha)	
25	18.70 (CH <sub>3</sub> )	0.93 (3H, s)	1	102.82 (CH)	5.82 (1H, broad
26	19.63 (CH <sub>3</sub> )	0.94 (3H, s)		` '	$s, W_{1/2} = 1.8 \text{ Hz}$
27	19.90 (CH <sub>3</sub> )	1.31 (3H, s)	2	72.82 (CH)	4.85
28	77.13 (CH <sub>2</sub> )	3.32, 3.73	3	72.62 (CH)	4.51
	. ( - 12)	(AB, q, J = 11.2  Hz)	4	73.98 (CH)	4.34
29	33.66 (CH <sub>3</sub> )	0.90 (3H, s)	5	70.44 (CH)	4.94
30	23.62 (CH <sub>3</sub> )	0.80 (3H, s)	6	18.58 (CH <sub>3</sub> )	1.73 (3H, d,
-	(3)		-	(	J = 6.2  Hz

<sup>\*</sup>C<sub>5</sub>D<sub>5</sub>N, TMS as internal standard, ppm (multiplicity).

due to four hexose residues. The  $^{13}$ C NMR and DEPT spectra of 1 showed that 30 carbons of the aglycone consisted of six methyls ( $\delta$ 12.75, 18.70, 19.63, 19.90, 23.62, 33.66), nine methylenes, 7 aliphatic methines, including

an oxymethine ( $\delta$ 82.63), and seven quaternary carbons. There were also two olefinic methines ( $\delta$ 131.75 and 132.07) for an endocyclic olefinic bond, an oxy-substituted quaternary carbon atom ( $\delta$ 84.93, C-13) and an

<sup>†</sup>From the one-bond HETCOR spectrum and COSY spectrum.

oxy-methylene ( $\delta$ 77.13, C-28) indicative of a 13 $\beta$ ,28-epoxyoleanane skeleton [5]. The <sup>13</sup>C NMR resonances due to the aglycone moiety were in good agreement with those of  $13\beta$ ,28-epoxyolea-11-en-3 $\beta$ -23-diol (16-dehydrosaikogenin G) [2, 3]. The appearance of four anomeric signals at  $\delta$ 105.02, 104.14, 103.97 and 102.82, evident in the <sup>13</sup>C NMR spectra and correlating with the <sup>1</sup>H NMR resonances at  $\delta$ 5.56, 4.89, 5.24 and 5.82, respectively, in the HETCOR spectrum, further confirmed the existence of a tetrasaccharide moiety in 1 [4-6]. In addition there were 16 methine resonances between  $\delta 69-82$ , two hydroxymethylene resonances ( $\delta 63.19$ ,  $\delta 1.37$ ) and two methyl resonances ( $\delta$ 17.25 and 18.58) thus supporting the existence of two hexopyranose and two 6deoxyhexopyranose residues. Both the hexopyranose residues were identified as glucopyranose whereas the 6-deoxyhexopyranose moieties were identified as fucopyranose and rhamnopyranose by analysis of the COSY spectrum [4, 7]. Moreover, on acid hydrolysis compound 1 yielded rhamnose, fucose and glucose in the ratio of 1:1:2. The anomeric configuration of both of the glucopyranose and fucopyranose units was determined as  $\beta$  due to: (1) the appearance of anomeric resonance as a doublet ( ${}^3J_{1,2} = 7.2 - 7.5 \text{ Hz}$ ) in the  ${}^1H$  NMR spectrum; (2) the <sup>13</sup>C NMR chemical shift of the anomeric carbon resonances ( $\delta$ 105.02, 104.14 and 103.97); and (3) the NOE cross-peaks between H-1, H-3, H-5 in the twodimensional NOESY spectrum [4, 8].

The interglycosidic linkage and sequence in the sugar chain was established by the negative ion, FAB mass spectrum together with two-dimensional NOE spectroscopy and also in view of the glycosylation effects observed in the <sup>13</sup>CNMR spectrum [4, 9]. The fragment ions at m/z 925 and m/z 909 were in accord with the terminal position of the rhamnopyranose and glucopyranose in the glycosidic part of 1. The fragment ion at m/z 763 further confirmed the loss of the rhamnose and glucopyranose. Therefore, these were terminal sugar residues directly attached to the fucopyranosyl residue and in agreement with the fragment ion at m/z 619. In the two-dimensional NOESY spectrum, the cross-peaks were observed between H-1 of a terminal glucose ( $\delta$  5.56) and H-2 ( $\delta$ 4.62) of the fucose moiety, and H-1 of rhamnose ( $\delta$ 5.82) and H-3 ( $\delta$ 4.03) of the fucose residue. The H-1 of fucose ( $\delta$ 4.89) exhibited a NOESY cross peak with the H-4 ( $\delta$ 4.37) of the internal glucose moiety revealing a  $(1 \rightarrow 4)$  interglycosidic linkage. Observation of NOESY cross-peaks between H-3 of the aglycone ( $\delta 4.15$ ) and H-1 of glucose ( $\delta$ 5.24) led to the identification of glucose as the internal sugar involved in linking the aglycone and the 2,3-disubstituted fucopyranosyl residue. Thus, the sequence of the monosaccharide residues and their interglycosidic linkage could be deduced as  $\{\beta$ -D-Glc(1  $\rightarrow$  2),  $\alpha$ -L-Rha(1  $\rightarrow$  3)}- $\beta$ -D-Fuc-(1  $\rightarrow$  4)-Glc, which was in full agreement with the <sup>13</sup>C NMR chemical shifts. The C-2 and C-3 of the fucose were observed at  $\delta$ 77.24 and  $\delta$ 84.82, respectively, revealing significant deshielding (4.54 and 10.92 ppm) in comparison with the reported values for methyl- $\beta$ -D-fucopyranoside [4] due to the  $\alpha$ -effect of glycosylation [6, 9]. Moreover, analogous chemical shift values have been reported for the C-2 and C-3 of fucopyranose in 2,3-diglycosylated fucopyranosylcontaining naturally occurring glycosides [2, 3]. The C-4 of the internal glucose ( $\delta$ 78.48) was also in complete accordance with the glycosidation at this position [2, 3, 6, 10].

The structure of scrokoelziside A was thus deduced to be 3-O-{[ $\alpha$ -L-rhamnopyranosyl-(1  $\rightarrow$  3),  $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  2)]- $\beta$ -D-fucopyranosyl-(1  $\rightarrow$  4)- $\beta$ -D-glucopyranosyl}-13 $\beta$ ,28-epoxyolean-11-en-23-ol. It is notable that scrokoelziside A is similar to thapsuine A from *Verbascum thapsus* and *V. lychnitis* [11–13], ilwensisaponin 1 from *Scrophularia ilwensis* [14], mimengoside A from *Buddleia officinalis* [2] and songarosapin C from *V. songaricum* [3] and *V. nigrum* [15], as it has same aglycone and the same sugar composition but a different arrangement of sugar sequence. This is in accord with a taxonomically interesting relationship between the plants of the genera *Verbascum* and *Scrophularia* [15].

## EXPERIMENTAL

Mps: uncorr. TLC: silica gel G (SISCO). Spots were visualized by spraying with 1% Ce(SO<sub>4</sub>)<sub>2</sub> in 1M H<sub>2</sub>SO<sub>4</sub>. CC: silica gel (60–120 mesh) (SISCO). Flash CC: EF-10 (EYELA) A.S.C. silica gel (230–400 mesh). PC: Whatman paper No. 1. GC (steel column 2 m × 4 mm packed with 3% OV-225 on GC Q. FID. temp.  $80-120^{\circ}$ C; N<sub>2</sub> at 50 ml min  $^{-1}$ ).  $^{1}$ H,  $^{13}$ C and other NMR experiments were carried out in CDCl<sub>3</sub> and/or C<sub>5</sub>D<sub>5</sub>N, using TMS as int. standard. Chemical shifts were expressed in  $\delta$  values. Mass spectra were recorded at 70 eV. The matrix for the FAB-MS was glycerin. The isolation and experimental conditions were same as reported earlier [1].

Extraction and Isolation. The CHCl<sub>3</sub> fr. (57.5 g) of the aerial parts of the plant [1] was chromatographed over silica gel G with a stepwise increase of MeOH content in CHCl<sub>3</sub>. The residue (18.1 g) the 25–100% MeOH–CHCl<sub>3</sub> frs was subjected to CC over silica gel with MeOH–CHCl<sub>3</sub> (10–30%) gradient. The fractions eluted with MeOH–CHCl<sub>3</sub> (18–25%) yielded 1, which, on flash CC over silica gel (MeOH–CHCl<sub>3</sub>, 18–19%), provided compound 1 (1.32 g) as an amorphous powder  $[\alpha]_{\rm D}^{27} + 27^{\circ}$  (c 0.8; pyridine).

Scrokoelziside A (1). IR  $v_{max}^{KBr}$  cm<sup>-1</sup>: 3400 (OH), 2920, 1070, FAB-MS negative-ion 3 kV, m/z (rel. int.): 1071 [M - 1]<sup>-</sup> (100), 1069 (15), 927 (19), 925 (26), 909 (18), 763 (40), 619 (12); <sup>1</sup>H and <sup>13</sup>C NMR (400 MHz: C<sub>5</sub>D<sub>5</sub>N): see Table 1.

Hydrolysis of compound 1. Compound 1 (50 mg) was treated with 50% methanolic HCl (10 ml for 5 hr) under reflux and worked up as usual. The aq. hydrolysate was neutralized with BaCO<sub>3</sub> and filtered. The filtrate was evapd under red. press. to dryness and acetylated with pyridine-Ac<sub>2</sub>O. GC analysis showed, in comparison with reference compounds, the presence of peracetylrhamnose, peracetylfucose and peracetylglucose (1:1:2).

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