



(+)-3'-O-ACETYLISOPTELEFLORINE, A QUINOLINE ALKALOID FROM *ORIXA JAPONICA*

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Key Word Index—*Orixia japonica*; Rutaceae; stem; quinoline alkaloid; (+)-3'-O-acetylisoptyeleflorine.

Abstract—A new quinoline alkaloid, (+)-3'-O-acetylisoptyeleflorine, was isolated from a methanol extract of the stems of *Orixia japonica*. Its structure was elucidated by a combination of NMR analyses and chemical reactions.

INTRODUCTION

Orixia japonica Thunb. is a shrub widely distributed in Japan, Korea and China. The stems and leaves of this plant were formerly used in Japan as an insecticide for livestock. Previously, several new quinoline alkaloids have been obtained from extracts of this species [1-3]. In the present paper we report the isolation and structural elucidation of a new quinoline alkaloid, (+)-3'-O-acetylisoptyeleflorine (1), isolated from a methanol extract of the stems of *O. japonica*.

RESULTS AND DISCUSSION

The UV spectrum of compound 1, $\lambda_{\max}$ (MeOH) 254 and 313 nm, suggested a quinoline or quinolone moiety; it was concluded to be a quinoline because the $\epsilon_{\max}$ of the latter absorption maximum was small (3700) [3]. In the ^{13}C NMR spectrum of 1, 18 signals, including four methyl, two methylene, three methine and nine quaternary carbons were observed (Table 1) and in the EI-mass spectrum, the $[\text{M}]^+$ appeared at m/z 345. Thus, the molecular formula of this compound was estimated to be $\text{C}_{18}\text{H}_{19}\text{NO}_6$, which was subsequently established by HREI-mass spectrometry (observed 345.1208; calculated 345.1213). In the ^1H NMR spectrum of 1 (Table 1), a methyl signal attributed to an acetyl moiety (δ 1.97) was observed, along with a geminal methyl (δ 1.55 and δ 1.62), methoxyl (δ 4.22), methylenedioxy (δ 6.15 (2H)), three sp^3 signals coupled to each other (δ 3.49, δ 3.62 and δ 4.97) and two coupled aromatic signals (δ 6.95 and δ 7.58). The existence of an acetyl moiety was also proved by the existence of an ion at m/z 303 $[\text{M} - 42]^+$ in the EI-mass spectrum and a band at 1737 cm^{-1} in the IR spectrum. Together, these results indicated that this quinoline alkaloid possessed a dihydrofuran or dihydropyran moiety and the structure of this compound was deduced as one

Table 1. ^1H and ^{13}C NMR assignments of (+)-3'-O-acetylisoptyeleflorine (1)

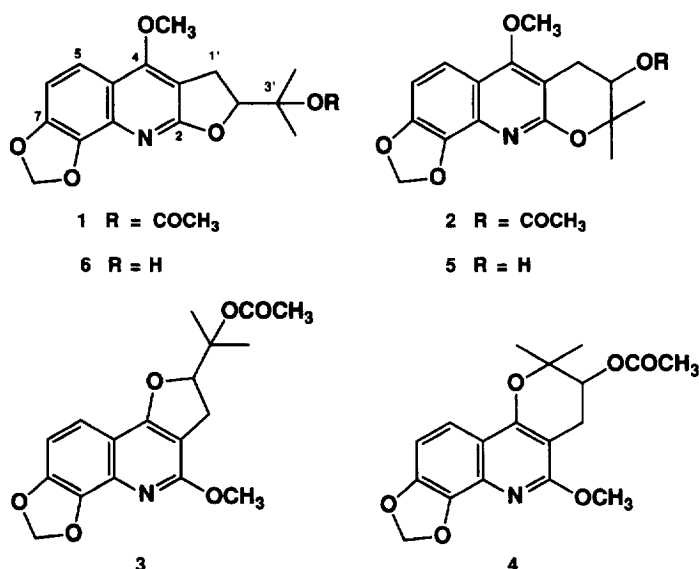
C	δ_{C}	δ_{H}
2	169.5 s	—
3	99.7 s	—
4	159.3 s	—
4a	116.7 s	—
5	116.1 d	7.58 d (9)*
6	106.7 d	6.95 d (9)
7	147.9 s	—
8	140.2 s	—
8a	133.9 s	—
1'	29.0 t	3.49 dd (7, 16) 3.62 dd (9, 16)
2'	84.1 d	4.97 dd (7, 9)
3'	82.2 s	—
4' (3'-CH ₃)	20.9 q	1.55 3H, s
5' (3'-CH ₃)	22.4 q	1.62 3H, s
C ₄ -OCH ₃	58.4 q	4.22 3H, s
-OCH ₂ O-	102.2 t	6.15 2H, d (each 2)
COCH ₃	22.4 q	1.97 3H, s
COCH ₃	170.3 s	—

*Figures in parentheses are coupling constants (Hz).

of the isomers 1-4. Among these, compound 2 is attributed to the monoacetyl derivative of pteleflorine (5) [4].

In the long-range ^{13}C - ^1H COSY spectrum, δ_{C} 159.3, assigned to C₄, was coupled to δ_{H} 3.49 (H-1'), 4.22 (OCH₃) and 7.58 (H-5) (Fig. 1). Consequently, the position of a methylenedioxy moiety (between C-7 and C-8) was confirmed and the angular isomers 3 and 4 were excluded as the structure of this compound. Conversely, when δ_{H} 4.22 (C-4, OCH₃) was irradiated, a 1% NOE was observed for δ_{H} 7.58 (H-5) and 2% and 3% NOE effects were observed for δ_{H} 3.62 and 3.49 (H₂-1'), respectively.

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These observations also supported the linear system of this compound.

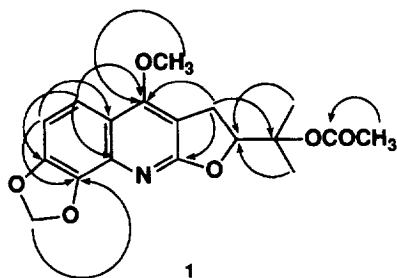


Fig. 1. ¹³C-¹H correlations in the long-range C-H COSY spectrum of 3'-O-acetylisopteleflorine (1).

Compound **1** was hydrolyzed with dilute alkali to give an alcohol **6** (C₁₆H₁₇NO₅, *M_r* 303). In the ¹H NMR spectrum of **6** almost all of the signals appeared similar to the parent compound, except for the absence of the methyl signal of the acetyl moiety of **1**. The oxygen-bearing methine signal (δ_H 4.64) in **6** appeared at lower field compared with that of C-2'-H (δ_H 3.74) in pteleflorine (**5**) [3] and was not changed dramatically compared with that of **1** (δ_H 4.97). Consequently, the acetyl moiety was not attached to the oxygen next to the methine moiety. In addition, when the alcohol **6** was treated with Ac₂O/pyridine, no acetylation occurred and, thus, the alcohol moiety was established as tertiary. From these observations the structure was concluded to be a linear furanoquinoline. Assignments of the ¹H NMR and attached ¹³C NMR signals were achieved by a ¹³C-¹H COSY spectrum and by comparison with existing data [3]. Unambiguous assignments of the quaternary carbons of **1** were accomplished using the long-range ¹³C-¹H COSY spectral data, as shown in Fig. 1 and Table 1.

Compound **6** has been synthesized previously as a by-product in the synthesis of pteleflorine (**5**) [4], but **1** has

not been reported previously. Compounds **6** and **1** were designated as (–)-isopteleflorine and (+)-3'-O-acetylisopteleflorine, respectively. Until now, *O*-methylbal-fouronium salt (as a perchlorate) is the only other alkaloid isolated from *O. japonica* [5] possessing a dihydrofuran moiety.

EXPERIMENTAL

General. ¹H NMR and ¹³C NMR were recorded on a JEOL JNM GX-500 or HITACHI R-3000 instrument; TMS was used as int. standard and chemical shifts are recorded in δ units. Wacogel C-200 (Wako Pure Chemical Ind., Ltd.) and Cosmosil 75C₁₈-OPN (Nacalai Tesque, Inc.) were used for CC and DC-Fertigplatten Kieselgel 60 F₂₅₄ (Merck) for prep. TLC.

Plant material. *O. japonica* plants were collected near Sendai, Japan in May, 1991. The plant was identified by S.F. and a voucher specimen is deposited in the herbarium of this department.

Extraction and isolation. Air-dried stems (5.9 kg) were extracted with *n*-hexane at room temp. (3 ×) to yield a residue (36 g), which was further extracted with MeOH to yield 556 g. The MeOH extract was partitioned between CHCl₃-H₂O to yield CHCl₃-solubles (128.1 g). Part of this fr. (62.4 g) was chromatographed over silica gel (1.25 kg) using CHCl₃ to give 80 frs. Frs. 38–43 (8.05–10.20 l, 5.85 g) were further purified by silica gel CC (250 g) using *n*-hexane-EtOAc (7:4) and Cosmosil 75C₁₈-OPN CC (10 g) using MeOH-H₂O (2:1) followed by prep. TLC using *n*-hexane-EtOAc (1:1), to give (+)-3'-O-acetylisopteleflorine (**1**) (*R_f* = 0.47, 56.1 mg, 0.002% yield).

(+)-3'-O-Acetylisopteleflorine (**1**). Oil. UV λ_{max} (MeOH): 254 (ε_{max} 39300), 313 (3700) nm. IR ν_{max} (KBr): 2990, 1737, 1638, 1539, 1480, 1420, 1274, 1111, 1041, 980 cm⁻¹. EI-MS *m/z* (rel. int.): 345 ([M]⁺, 50), 303 (4), 285 (49), 270 (100), 255 (12), 244 (33), 232 (10), 202 (7), 186 (6), 135 (5). HREI-MS: obsd *m/z* 345.1208, calcd

for $C_{18}H_{19}NO_6$ 345.1213. $[\alpha]_D^{25} + 6.4^\circ$ ($CHCl_3$, c 0.7). 1H and ^{13}C NMR ($CDCl_3$); see Table 1.

Hydrolysis of (+)-3'-O-acetylisopecteflorine (1). To (+)-3'-O-acetylisopecteflorine (1) (15 mg) in MeOH (3 ml) was added aq. 2 N KOH (1 ml) and the mixt. stirred at room temp. After 26 hr, the reaction mixt. was dil. with H_2O (20 ml), and extracted with $CHCl_3$ (3×20 ml). The combined $CHCl_3$ layers were washed with H_2O (3×20 ml), dried (Na_2SO_4) and evapd *in vacuo* to provide a gummy residue (12.9 mg) which was purified by prep. TLC using *n*-hexane-EtOAc (1:1) to give (–)-isopecteflorine (6) (10.9 mg, 83% yield).

(–)-Isopecteflorine (6). Oil. UV λ_{max} (MeOH): 255 (ϵ_{max} 40 400), 314 (3900) nm. IR ν_{max} (KBr): 3350, 2940, 1636, 1593, 1539, 1481, 1420, 1367, 1350, 1271, 1110, 1042, 981 cm^{-1} . EI-MS m/z (rel. int.): 303 ($[M]^+$, 82), 285 (11), 270 (25), 244 (100), 231 (17), 217 (24), 202 (24), 186 (15), 174 (10), 156 (6), 149 (16), 116 (11), 103 (8). $[\alpha]_D^{21} - 15.2^\circ$ ($CHCl_3$, c 0.2). 1H NMR ($CDCl_3$): δ 1.26 and δ 1.42 (each 3H, s, C-3'-CH₃ $\times$ 2), 3.54 (1H, dd, $J = 17, 8$ Hz, H-1'), 3.59 (1H, dd, $J = 17, 9$ Hz, H-1'), 4.23 (3H, s, C-4-OCH₃), 4.63 (1H, dd, $J = 9, 8$ Hz, H-2'), 6.15 (2H, each 1H, d, $J = 2$ Hz, -OCH₂O-), 6.95 (1H, d, $J = 9$ Hz, H-6), 7.60 (1H, d, $J = 9$ Hz, H-5).

Acetylation of (–)-isopecteflorine (6). Compound 6 (2.2 mg) was dissolved in pyridine (0.5 ml), Ac_2O

(0.25 ml) added and the soln stirred under a N_2 atmosphere for 24 hr. The reaction mixt. was dil. with H_2O (20 ml) and extracted with $CHCl_3$ (3×20 ml). The combined $CHCl_3$ layers were washed with H_2O (3×20 ml), dried (Na_2SO_4) and coned *in vacuo* to give unreacted 6 (1.8 mg).

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