



A SECOIRIDOID GLUCOSIDE FROM *FRAXINUS ANGUSTIFOLIA*

IHSAN CALIS, MOHAMMED HOSNY*† and MOHAMMED F. LAHLOUB‡

Department of Pharmacognosy, Faculty of Pharmacy, Hacettepe University, TR-06100 Ankara, Turkey; †Department of Pharmacognosy, Faculty of Pharmacy, Al-Azhar University, Cairo, Egypt; ‡Department of Pharmacognosy, Faculty of Pharmacy, Mansoura University, Mansoura, Egypt

(Received in revised form 25 September 1995)

Key Word Index—*Fraxinus angustifolia*; Oleaceae; secoiridoid; nuezhenide; angustifolioside C.

Abstract—The known glucoside nuezhenide, and a new secoiridoid glucoside, angustifolioside C [6'-O- β -D-glucopyranosyl]-neoleuropein], were isolated from a methanolic extract of the leaves of *Fraxinus angustifolia*. Their structures were elucidated on the basis of chemical and spectral data.

INTRODUCTION

The Oleaceae family is a rich source of secoiridoid, phenylpropanoid and lignan glucosides. A number of secoiridoid glycosides including oleoside, 10-hydroxyoleoside, secologanoside, morroniside as well as secoiridoid lactone types [1-3] have been identified in Oleaceous plants. Members of the genus *Fraxinus* have been used in folk medicine [4]. Among various *Fraxinus* secondary metabolites examined, some secoiridoids exhibited a wide spectrum of biological activities [5]. Of particular interest are oleuropein and its derivatives, which showed hypotensive and related cardiovascular activities [6, 7]. Several plants of the genus *Fraxinus* have been reported to contain secoiridoid compounds, namely, oleuropein, from the leaves of *F. japonica* [8], ligstroside, from the bark of *F. griffithii* [9], and 10-hydroxyligstroside, from the bark of *F. excelsior* [10]. More complex dimeric secoiridoids consisting of two oleoside moieties linked to glucose and/or tyrosol, designated GI-3 and GI-5, accompanied by nuezhenide, have been identified as secondary metabolites in embryos of *F. americana* [11, 12]. Excelsioside was isolated from the leaves of *F. excelsior* [13]. Two secoiridoid glucosides, frachinoside [14] and fraxiformoside [15], have also been reported from *F. chinensis* and *F. formosana*, respectively.

Previously, we have studied the secoiridoid constituents in a methanolic extract of the leaves of *F. angustifolia*. This led to the isolation of several new secoiridoids including: ligstral (3), angustifolioside A (4), angustifolioside B (5), ligstrobutyl (6), oleobutyl (7), penta-O-methyl-ligstroside (8), and tetra-O-methyl-oleoside-

dimethylester (9), along with the two known secoiridoid glucosides, ligstroside (10) and oleuropein (11) [16, 17]. We now wish to report on the isolation of a new secoiridoid glucoside, angustifolioside C (1), together with a known glucoside nuezhenide (2).

RESULTS AND DISCUSSION

The water soluble portion of the methanolic extract of the leaves of *F. angustifolia* was fractionated by solvent partitioning. A combination of polyamide column chromatography (CC) and silica gel CC followed by further purification through reversed phase chromatography and Sephadex LH-20 CC led to the isolation of nuezhenide (2) and a new glucoside, angustifolioside C (1).

Compound 1 was obtained as a white amorphous powder. Its FAB-mass spectrum showed a quasimolecular ion peak, $[M + Na]^+$, at m/z 847, indicating its M_r to be 824. It also showed fragments at m/z 685 and 501 attributed to $[M + Na - \text{glucose}]^+$ and $[M + H - (\text{glucose} \times 2)]^+$, respectively. These results were compatible with the molecular formula $C_{38}H_{48}O_{20}$. The UV and IR spectra suggested the presence of an enol-ether system conjugated with a carbonyl group (230 nm; 1715, 1700 and 1630 cm^{-1}), which are characteristics of many iridoid and secoiridoid skeletons [1]. In addition, absorptions due to the catechol chromophore (281 nm; 1520 and 1450 cm^{-1}) were observed. The proton signals due to the secoiridoid moiety of 1 present in its 1D and 2D ^1H - ^1H -homonuclear NMR spectra were found to be closely related to those of 4 [17], multiroside and multifloroside [18].

Compound 1 gave rise to signals for two 3,4-dihydroxyphenethyl groups, unlike 4, which contains a signal for a carbomethoxyl group [17], and these were in good agreement with the reported data for multifloroside [18]

*Author to whom correspondence should be addressed.

Table 1. ^{13}C NMR spectral data for angustifolioside C (1) (CD_3OD , 90.50 MHz), multirosid and multifloroside*

C	Secoiridoid moiety			Phenolic moiety			Glucosyl moiety		
	1	Multiroside	Multifloroside	1	Multiroside	Multifloroside	1	Multiroside	Multifloroside
1	94.70 <i>d</i>	94.65 <i>d</i>	94.57 <i>d</i>	1'	66.68 <i>t</i>	66.57 <i>t</i>	1'''	100.80 <i>d</i>	100.86 <i>d</i>
3	155.03 <i>d</i>	155.06 <i>d</i>	155.02 <i>d</i>	2'	35.96 <i>t</i>	35.32 <i>t</i>	2'''	74.73 <i>d</i>	74.81 <i>d</i>
4	109.83 <i>s</i>	109.06 <i>s</i>	109.26 <i>s</i>	3'	131.85 <i>s</i>	135.15 <i>s</i>	3'''	78.36 <i>d</i>	78.26 <i>d</i>
5	31.78 <i>d</i>	32.23 <i>d</i>	32.17 <i>d</i>	4'	117.88 <i>d</i>	117.12 <i>d</i>	4'''	71.35 <i>d</i>	71.30 <i>d</i>
6	41.14 <i>t</i>	41.14 <i>t</i>	41.07 <i>t</i>	5'	145.20 <i>s</i>	145.38 <i>s</i>	5'''	78.08 <i>d</i>	78.11 <i>d</i>
7	173.15 <i>s</i>	172.99 <i>s</i>	173.07 <i>s</i>	6''	144.84 <i>s</i>	144.78 <i>s</i>	6'''	62.67 <i>t</i>	62.70 <i>t</i>
8	126.65 <i>d</i>	129.40 <i>d</i>	129.33 <i>d</i>	7''	119.79 <i>d</i>	119.28 <i>d</i>	1''''	104.50 <i>d</i>	104.52 <i>d</i>
9	130.54 <i>s</i>	130.92 <i>s</i>	130.64 <i>s</i>	8'	122.92 <i>d</i>	121.45 <i>d</i>	2'''	74.88 <i>d</i>	74.64 <i>d</i>
10	13.55 <i>q</i>	59.13 <i>t</i>	59.19 <i>t</i>	1''	67.34 <i>t</i>	66.39 <i>t</i>	3'''	78.22 <i>d</i>	77.78 <i>d</i>
11	168.20 <i>s</i>	168.41 <i>s</i>	167.96 <i>s</i>	2''	36.42 <i>t</i>	35.39 <i>t</i>	4'''	71.30 <i>d</i>	71.27 <i>d</i>
COCH ₃	—	52.02 <i>q</i>	—	3''	133.08 <i>s</i>	130.92 <i>s</i>	5'''	77.92 <i>d</i>	77.49 <i>d</i>
				4''	116.62 <i>d</i>	116.35 <i>d</i>	6'''	62.46 <i>t</i>	62.33 <i>t</i>
				5''	146.54 <i>s</i>	146.14 <i>s</i>			
				6''	148.15 <i>s</i>	144.78 <i>s</i>			
				7''	117.32 <i>d</i>	116.97 <i>d</i>			
				8''	121.04 <i>d</i>	121.32 <i>d</i>			

*Date taken from ref. [18].

and neoleuropein [19] with similar 3,4-dihydroxyphenethyl moieties in the secoiridoid skeleton. These were evidenced by the presence of two aromatic ABX spin systems centred between $\delta 6.56$ – 7.04 and two aliphatic ABX₂ spin systems at $\delta 4.26$ *dt*, 4.15 *dt* and 2.78 *t* and at $\delta 4.41$ *t*, 4.29 *t* and 2.87 *t*. In the ¹³C NMR spectrum of **1**, two singlets at $\delta 168.20$ and 173.15 indicated the presence of two carbonyl groups at C-11 and C-7, respectively. The ¹H–¹³C HMBC spectrum of **1** revealed a significant correlation of the carbonyl resonance at $\delta 168.20$ (s) with the proton of the vinylic singlet ($\delta 7.48$, *s*, H-3). The other carbonyl signal at $\delta 173.15$ (s), assigned to C-7, was secured by its correlation with one of the methylene protons (H-6a $\delta 2.44$, *dd*, *J* = 14.2 and 9.5 Hz). One of the glucosyl carbon signals at $\delta 100.80$ (C-1'') showed a weak correlation with the proton singlet of H-1 ($\delta 5.89$ *brs*), suggesting the location of one sugar residue at C-1. Moreover, H-C(7) exhibited correlation to the 'd' at $\delta 4.74$ (H-C-1''), suggesting that the second glucosyl unit was attached to the aromatic hydroxyl group at C-6'. Cross comparison of the ¹³C-NMR spectral data for **1** (Table 1) with those for **4** [17] and multiroside [18] revealed close correspondence in every aspect to those signals arising from the secoiridoid diglucoside moiety. Moreover, an additional set of carbon signals corresponding to a 3,4-dihydroxyphenethoxyl group was observed, as in multifloroside [18] (Table 1). This finding was also supported by the positive FAB-mass spectrum, which showed a quasimolecular ion peak $[M + Na]^+$ at *m/z* 847, indicating an increase of 122 mass units in comparison with that of **4**. The base fragment ion at *m/z* 154 in the EI mass spectrum of **1** also attested to the presence of this side chain.

These features suggested that **1** was a diglucoside of an oleoside-type secoiridoid in which the carbomethoxyl group of **4** was replaced by a 3,4-dihydroxyphenethyl group. It is well known that glycosidation produces an upfield shift of the signal of oxygenated phenolic C-atoms and downfield shifts of the signals for *ortho*- and *para*-related C-atoms [20]. The resonances for H-4', H-7' and H-8' ($\delta 7.09$, H-4'; 7.13 , H-7'; and 6.77 , H-8'), for the glycosylated dihydroxyphenylethanol moiety, were shifted upfield by $\delta 0.23$, 0.35 and 0.21 , respectively. A similar spectral difference was observed for periclymenoside and periclymenosidic acid, which have the same substituents [21, 22]. Similar glycosidation effects were observed for the carbon signals of C-3' and C-7' (downfield shifts of $\Delta = 1.05$ and $\Delta = 2.69$, respectively) (Table 1). This result also suggested that the second glucosyl unit was attached to the aromatic hydroxyl group at C-6'. This is in full agreement with the data reported for **4** [17] and multiroside [18], which have the same substitution pattern for the phenethyl moiety.

Acetylation of **1** gave the fully acetylated undecaacetate **1a**. As expected, the FAB-mass spectrum contained a $[M + H]^+$ ion at *m/z* 1287 corresponding to the molecular formula C₆₀H₇₀O₃₁. The main fragment peaks recorded were at *m/z* 978 $[(M + Na) - 331]^+$ for the loss of a [2,3,4,6-*O*-acetyl glucose oxonium ion]⁺. The ¹H NMR spectrum of **1a** revealed, besides those

signals for the secoiridoid diglucoside residue, the presence of 11 acetyl signals belonging to three aromatic ($\delta 2.31$, 2.29 and 2.28) and eight aliphatic ($\delta 2.13$, 2.10 , 2.08 , 2.06 , 2.02 , 1.99 , 1.90 and 1.75) acetyl groups, thus confirming the presence of two glucosyl and two 3,4-dihydroxyphenethyl moieties, the downfield shifts for the signals of two glucose units indicating that both are terminal. This spectral evidence pointed to structure **1** for angustifolioside C, a new example of a secoiridoid diglucoside which has not been isolated formerly.

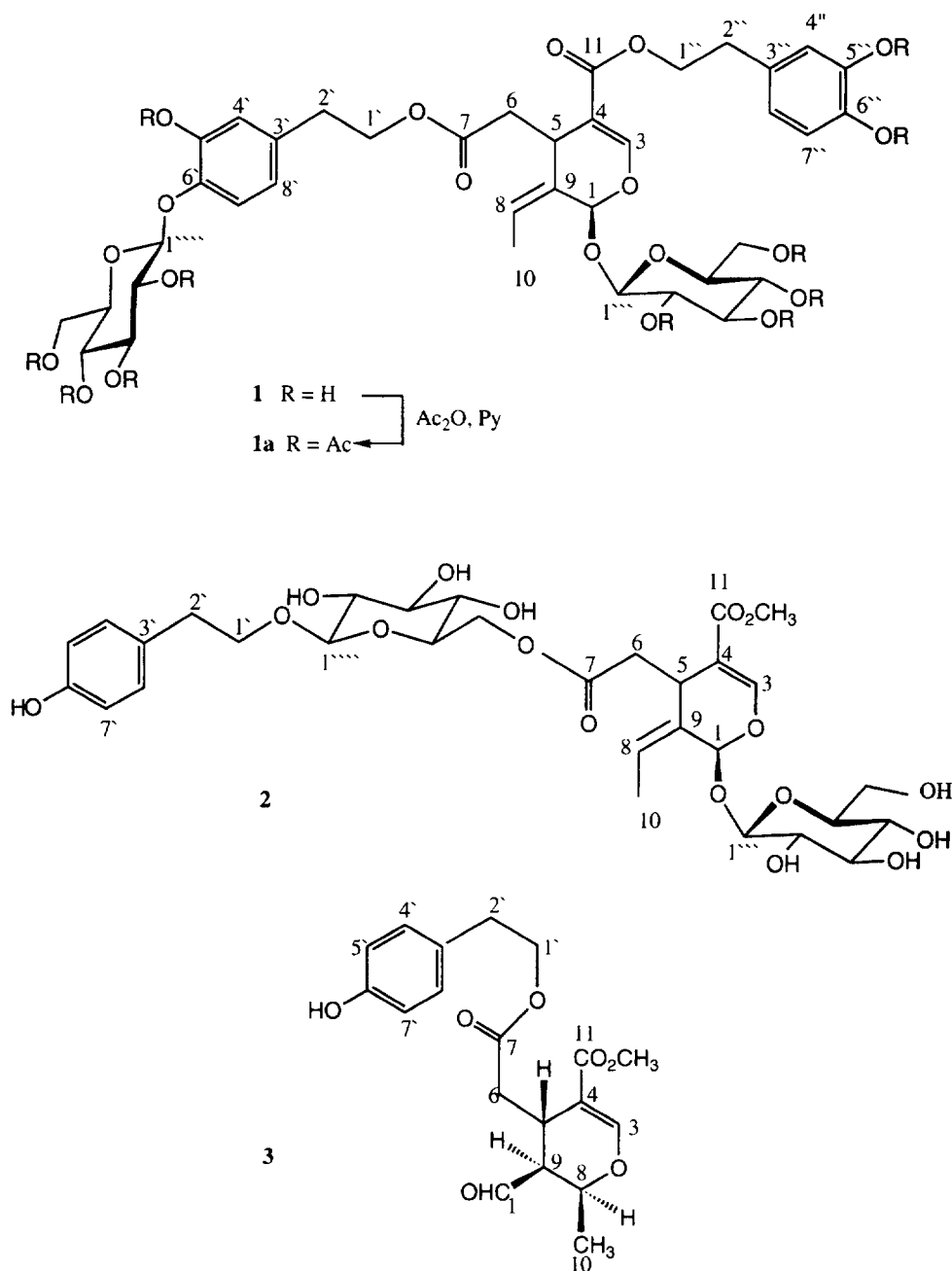
Compound **2** was obtained as an amorphous white powder. Its molecular formula C₃₄H₄₂O₁₇ was estimated from a weak $[M + 1]^+$ peak at *m/z* 687 and an intensive $[M + Na]^+$ peak at *m/z* 709 in the FAB-mass spectrum. The presence of an aromatic moiety and a conjugated ester group was indicated by the absorptions in the UV (238 and 273 nm) and IR (1730, 1705, 1630 and 1520 cm⁻¹) spectra of **2**. Its ¹H and ¹³C NMR spectra contained signals characteristic of those for oleoside and one tyrosyl glucoside. A comparison of the spectral data for **2** with those reported in the literature confirmed that **2** was nuezhenide [23, 24].

EXPERIMENTAL

General experimental procedures. MPLC: Sephalyte C18, 40 μ m (Analytichem); TLC: Silica gel 60 F₂₅₄ (Merck) plates; CC: Silica gel 60 (0.063–0.2 mm, Merck), Polyamide (ICN Biomedicals) or Sephadex LH-20 (Pharmacia). Secoiridoids were detected by spraying with 1% vanillin/H₂SO₄, followed by heating at 100° for 5 min. ¹H and ¹³C NMR: Bruker NMR 360 MHz and Varian NMR 600 MHz high field spectrometers equipped with an IBM Aspect-2000 processor and with software VNMR version 4.1b, respectively, using TMS as int. standard in appropriate solvents.

Plant material. The leaves of *F. angustifolia* Vohl were collected in September 1989 from the Royal Plant Garden, Al-Ribat, Morocco. The plant material was kindly identified by Prof. Dr Lotfi Bolous and Dr Mohammed El-Gibaly (Plant Taxonomy Unit, National Research Centre, Cairo, Egypt). A voucher specimen has been deposited in the Herbarium of the Pharmacognosy Department, Faculty of Pharmacy, Al-Azhar University, Cairo, Egypt.

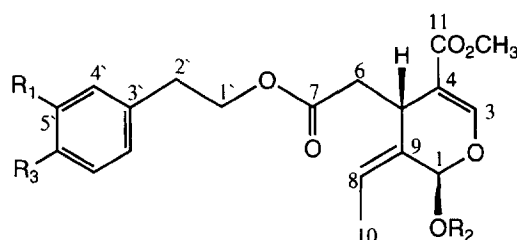
Extraction and Isolation. Air-dried powdered leaves of *F. angustifolia* (600 g) were subjected to exhaustive extraction with MeOH (4 $\times$ 2.5 l). The combined MeOH extracts were concd *in vacuo* at 40° to dryness (103 g). The conc. MeOH extract was dissolved in H₂O (500 ml) and filtered through Celite. The filtrate and washings were combined and defatted with petrol. The defatted crude extract (89 g) was extracted successively with *n*-BuOH satd with H₂O. Then *n*-BuOH-soluble frs were pooled and concd *in vacuo* at 40° to afford a residue which was chromatographed on a polyamide column (130 $\times$ 2.5 cm) with H₂O, followed by increasing concns of MeOH to yield five main frs (A: H₂O; B: 25% MeOH; C: 50% MeOH; D: 75% MeOH; and E: MeOH).



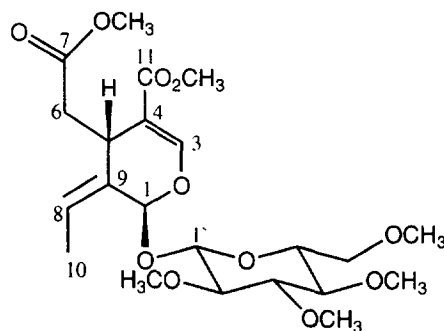
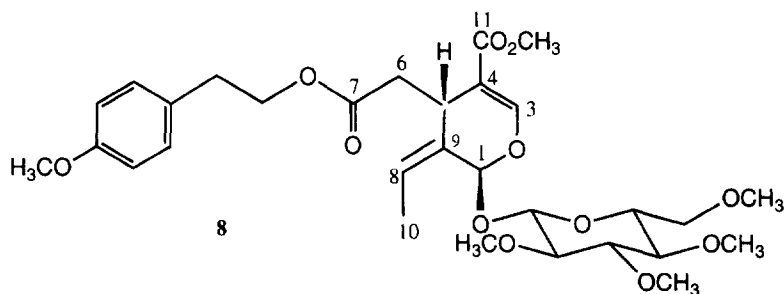
Fr. A (10 g) was rechromatographed on silica gel with CHCl₃-MeOH-H₂O (80:20:1-15:10:1) to give 7 frs (A1-A7). Fr. A3 was subjected to CC on silica gel eluting with a CH₂Cl₂-MeOH gradient (4:1-7:3), and three frs (A3_a-A3_c) were collected. The second fr. was purified by MPLC (325 × 18 mm) using a gradient of MeOH in H₂O (5-35%) to yield a mixt. of **1** and **2**, which were then sepd on a silica gel column eluted with CHCl₃-MeOH-H₂O (40:10:1-70:30:3) to afford **1** and **2**. Final purification of **1** and **2** on Sephadex LH-20 columns eluted with MeOH yielded **1** (29 mg) and **2** (22 mg), respectively.

Anqustifolioside C (**1**). Amorphous powder; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 230, 281; IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1715

(ester), 1700 (C=O), 1630 (C=C), 1520 (arom. ring); FAB-MS m/z : 847 [M + Na]⁺, 685 [M + Na - Glc]⁺, 501 [M + Na - 2Glc]⁺; ¹H NMR (360, 600 MHz, CD₃OD): secoiridoid moiety: δ 5.89 (*br s*, H-1), 7.48 (*s*, H-3), 3.98 (*dd*, $J = 9.5, 4.5$ Hz, H-5), 2.44 (*dd*, $J = 14.2, 9.5$ Hz, H-6a), 2.72 (*dd*, $J = 14.5, 4.5$ Hz, H-6b), 6.19 (*br t*, $J = 7.0$ Hz, H-8), 1.72 (*dd*, $J = 7.0, 1.5$ Hz, H-1), phenolic moiety: δ 4.26 (*dt*, $J = 11.0, 6.7$ Hz, H-1'a), 4.15 (*dt*, $J = 11.0, 6.7$ Hz, H-1'b), 2.78 (*t*, $J = 6.7$ Hz, H-2'), 7.09 (*d*, $J = 1.7$ Hz, H-4'), 7.13 (*d*, $J = 8.0$ Hz, H-7'), 6.77 (*dd*, $J = 8.0, 1.7$ Hz, H-8'), 4.41 (*t*, $J = 7.0$ Hz, H-1''a), 4.29 (*t*, $J = 7.0$ Hz, H-1''b), 2.87 (*t*, $J = 7.0$ Hz, H-2''), 6.86 (*d*, $J = 1.7$ Hz, H-4''), 6.78 (*d*, $J = 8.0$ Hz, H-7''), 6.56



	R ₁	R ₂	R ₃
4	OH	Glc	OGlc
5	H	Glc	OGlc
6	OH	CH ₂ CH ₂ CH ₂ CH ₃	H
7	OH	CH ₂ CH ₂ CH ₂ CH ₃	OH
10	H	Glc	H
11	OH	Glc	H



9

(*dd*, $J = 8.0, 1.7$ Hz, H-8''), glucosyl moiety: δ 4.81 (*d*, $J = 7.9$ Hz, H-1'''), 5.18 (*dd*, $J = 7.8, 9.2$ Hz, H-2'''), 3.25–3.52 (2H, *m*, overlapping, H-3''' and H-3'''), 3.68 (2H, *dd*, $J = 12.0, 1.5$ Hz, H-6'''a, H-6'''a), 3.86 (*dd*, $J = 12.0, 5.0$ Hz, H-6'''b), 4.74 (*d*, $J = 7.6$ Hz, H-1'''), 3.89 (*dd*, $J = 12.0, 5.5$ Hz, H-6'''b); ^{13}C NMR: Table 1.

Undecaacetyl angustifolioside C (1a). Treatment of **1** (12 mg) with Ac_2O (0.5 ml) and pyridine (0.5 ml) at room temp. overnight, followed by CC over silica gel using $\text{C}_6\text{H}_6\text{--Me}_2\text{CO}$ (3:1) gave the undecaacetate **1a** as an amorphous powder; IR^{KBr} $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 1760 (ester),

1710 (C=O), 1630 (C=C), FAB-MS m/z : 1287 $[\text{M} + \text{H}]^+$, 1309 $[\text{M} + \text{Na}]^+$, 978 $[(\text{M} + \text{Na}) - 331]^+$; ^1H NMR (360, 600 MHz, CDCl_3): secoiridoid moiety: δ 5.67 (*br s*, H-1), 7.41 (*s*, H-3), 3.89 (*dd*, $J = 9.0, 4.5$ Hz, H-5), 2.38 (*dd*, $J = 14.5, 9.0$ Hz, H-6a), 2.65 (*dd*, $J = 14.5, 4.5$ Hz, H-6b), 5.98 (*q*, $J = 7.0$ Hz, H-8), 1.81 (*dd*, $J = 7.0, 1.5$ Hz, H-10), phenolic moiety: δ 4.18 (*dt*, $J = 11.0, 7.0$ Hz, H-1'a), 4.24 (*dt*, $J = 11.0, 7.0$ Hz, H-1'b), 2.90 (*t*, $J = 6.7$ Hz, H-2'), 7.04 (*d*, $J = 2.0$ Hz, H-4'), 6.86 (*d*, $J = 8.0$ Hz, H-7'), 7.04 (*dd*, $J = 8.0, 2.0$ Hz, H-8'), 4.30 (*m*, H-1''a), 4.36 (*m*, H-1''b), 2.94 (*t*, $J = 7.0$ Hz, H-2''), 7.17 (*d*, $J = 2.0$ Hz, H-4''), 7.08 (*d*, $J = 8.0$ Hz,

H-7''), 7.08 (*dd*, $J = 8.0, 2.0$ Hz, H-8''), glucosyl moiety: δ 5.04 (*d*, $J = 7.9$ Hz, H-1'''), 5.34 (*dd*, $J = 9.2, 9.4$ Hz, H-3'''), 5.15 (*dd*, $J = 9.4, 10.0$ Hz, H-4'''), 3.79 (*m*, H-5'''), 4.14 (*dd*, $J = 12.2, 2.8$ Hz, H-6'''a), 4.19 (*dd*, $J = 12.0, 5.0$ Hz, H-6'''b), 5.25 (*d*, $J = 7.6$ Hz, H-1'''), 5.12 (*dd*, $J = 7.5, 9.2$ Hz, H-2'''), 5.36 (*dd*, $J = 9.2, 9.4$ Hz, H-3'''), 5.13 (*dd*, $J = 9.4, 10.0$ Hz, H-4'''), 3.86 (*ddd*, $J = 10.0, 5.5, 2.5$ Hz, H-5'''), 4.08 (*dd*, $J = 12.4, 2.5$ Hz, H-6'''a), 4.30 (*dd*, $J = 12.4, 5.5$ Hz, H-6'''b), 2.31, 2.29, 2.28 [each 3H, *s*, (3 $\times$ arom. acetoxy)], 2.13, 2.10, 2.08, 2.06, 2.02, 1.99, 1.90, 1.75 (each 3H, *s*, (8 $\times$ aliph. acetoxy)].

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