

Phytochemistry, Vol. 42, No. 2, pp. 461-464, 1996 Copyright © 1996 Elsevier Science Ltd Printed in Great Britain. All rights reserved 0031-9422/96 \$15.00 + 0.00



WILFRIED A. KÖNIG,\* ANGELA RIECK, YÜCEL SARITAS, INGO H. HARDT and KARL-HEINZ KUBECZKA†

Institut für Organische Chemie, Universität Hamburg, D-20146 Hamburg, Germany; †Abteilung für Pharmazeutische Biologie, Universität Hamburg, D-20146 Hamburg, Germany

(Received in revised form 11 December 1995)

Key Word Index—Meum athamanticum; Bazzania trilobata; liverwort; enantioselective twodimensional capillary GC; (-)- $\alpha$ -barbatene; (+)- $\beta$ -barbatene;  $\beta$ -bazzanene; isobarbatene; isobazzanene.

**Abstract**—The sesquiterpene hydrocarbons  $\beta$ -bazzanene and  $\alpha$ - and  $\beta$ -barbatene, typical constituents of liverworts (Hepaticae), were identified for the first time as constituents of a higher plant in the roots of Meum athamanticum (L.) Jacq. In addition, isobazzanene and isobarbatene, together with a variety of common sesquiterpene hydrocarbons, were identified. For  $\alpha$ - and  $\beta$ -barbatene and isobarbatene the opposite configurations from those in liverworts were determined by enantioselective GC using modified cyclodextrins as chiral stationary phases. The configuration of  $\beta$ -bazzanene and isobazzanene could not be assigned. Isobarbatene was found for the first time as a natural compound.

## INTRODUCTION

Pergamon

The sesquiterpene skeletons of  $\beta$ -bazzanene (1) and of the barbatenes (2, 3) are believed to be unique to the liverworts [1].  $\beta$ -Barbatene (3) has been detected in virtually every leafy liverwort of the Jungermanniales, but has never been found in higher plants. In the course of our investigation of the essential oil from Meum athamanticum, a strongly aromatic umbellifer of the mountains of Western and Central Europe, in addition to known constituents [2, 3] some uncommon compounds were found. The hydrocarbon fraction of the hydrodistillate of M. athamanticum roots, which contains a large variety of sesquiterpenes was investigated by GC-mass spectrometry. In addition to major peaks of (-)- $\beta$ -elemene (6), (+)-bicyclogermacrene (7), (-)germacrene D (8), (-)- $\alpha$ -chamigrene (9), (+)- $\beta$ bisabolene (10) and germacrene B (11), the mass spectra indicated the presence of some unusual constituents.

## RESULTS AND DISCUSSION

The sesquiterpene hydrocarbons listed in Fig. 1 were identified by their mass spectra and retention indices on an unpolar stationary phase (CpSil 5). Most remarkably, among the many common sesquiterpene hydrocarbons, the members of a series of compounds so far

enantiomeric excess of the (-)-enantiomer, as determined by enantioselective capillary gas chromatograpy) could also be identified in the essential oil of M. athamanticum (Fig. 1) by comparing their GC retention times and their characteristic mass spectra with those of

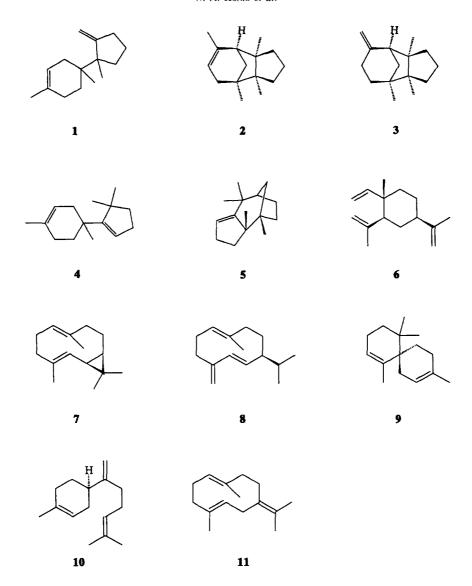
authentic reference compounds. While  $\beta$ -bazzanene (1)

Much smaller amounts of  $\beta$ -bazzanene (1) [11],

isobazzanene (4) and isobarbatene (5) (with a 70%

only known as constituents of liverworts [4, 5] were unambiguously identified by their characteristic EImass spectra. To confirm their structures  $\alpha$ - and  $\beta$ barbatene were isolated by preparative GC using cyclodextrin derivatives as stationary phases [6]. The 'H NMR spectra of the isolated fractions clearly confirmed the structures derived from the mass spectra. They were identical to those of  $\alpha$ - and  $\beta$ -barbatene samples which were isolated by preparative GC from Bazzania trilobata and with published data [7]. Capillary GC with modified cyclodextrins as chiral stationary phases [8] indicated the presence of enantiomers with a configuration opposite to those isolated from B. trilobata. This was also confirmed by measuring the sense of optical rotation of the isolated fractions. Two-dimensional capillary GC [9] further confirmed these findings. The  $\alpha$ - and  $\beta$ -barbatene fractions, respectively, were transferred from a capillary with a nonpolar stationary phase (CpSil 5) to a second capillary with heptakis (2,6-di-Omethyl-3-O-pentyl)- $\beta$ -cyclodextrin [10] and compared by co-injection with the corresponding isomers isolated from B. trilobata under identical experimental conditions (Fig. 1).

<sup>\*</sup>Author to whom correspondence should be addressed.



has been found as a common constituent of liverworts [1], isobazzanene (4) was identified as a constituent of B. fauriana and B. angustifolia [12] and could be prepared from  $\beta$ -bazzanene (1) by rearrangement under acidic conditions [13]. Isobarbatene (5) has been obtained as a stable trifluoroacetic acid rearrangement product from  $\alpha$ - (2) or  $\beta$ -barbatene (3) but never observed as a natural product [14].

## EXPERIMENTAL

Plant material. Bazzania trilobata was collected in Southern Germany (Fichtelgebirge, Bischofsgrün-Karches) in November 1993. Meum athamanticum was harvested in July 1994 in the Botanical Garden of the University of Hamburg.

Hydrodistillation. The essential oils of M. athamanticum (L.) Jacq. and of B. trilobata (L.) S. Gray was prepared by steam distillation (6 and 2 hr, respectively) of aq. homogenates of fresh and green plants using n-hexane as collection solvent. Because of the greatly differing weights, the fresh material was not weighed.

Enantioselective capillary GC. Capillary columns with cyclodextrin derivatives were prepared as described in ref. [15].

Two-dimensional GC [16]. The essential oil samples were injected on a 25 m (0.25 mm i.d.) capillary column containing nonpolar CpSil 5 (Chrompack) in a Siemens Sichromat 2 gas chromatograph at 50° and programmed at a rate of 3° min<sup>-1</sup> to 200°. Sample transfer was performed after 33.83 min (the  $R_r$  of  $\alpha$ -barbatene) and after 35.21 min (the  $R_r$  of  $\beta$ -barbatene) to a 25-m capillary column containing heptakis (2,6-di- $\alpha$ -methyl-3- $\alpha$ -pentyl)- $\alpha$ -cyclodextrin (50% in polysiloxane OV1701, w/w) which was kept isothermally at 100°. The chromatograms from both columns were recorded with a two-channel Merck-Hitachi model 2500 integrator. H<sub>2</sub> at an entrance pressure of 80 kPa

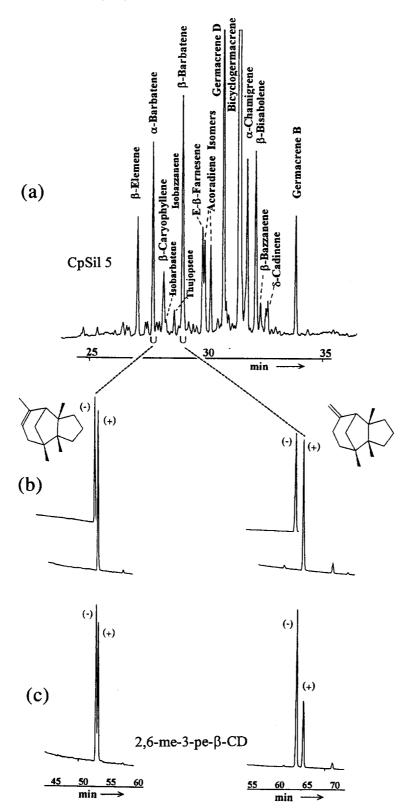


Fig. 1. Gas chromatographic investigation of the essential oil of the roots of *Meum athamanticum* (L.). (A) Partial gas chromatogram of the sesquiterpene portion on a 25 m fused silica capillary with methyl-polysiloxane CpSil 5; (B) gas chromatograms of *M. athamanticum* fractions of  $\alpha$ - and  $\beta$ -barbatene transferred from the CpSil 5 capillary to a 25 m fused silica capillary with heptakis (2,6-di- $\theta$ -methyl-3- $\theta$ -pentyl)- $\theta$ -cyclodextrin and comparison with  $\alpha$ - and  $\theta$ -barbatene from the liverwort *Bazzania trilobata* transferred under identical conditions; (C) co-injection of *M. athamanticum* sample with  $\alpha$ - and  $\theta$ -barbatene from *B. trilobata* and transfer of these fractions to the chiral capillary column.

for the CpSil 5 capillary and 65 kPa for the cyclodextrin capillary was used as a carrier gas.

Preparative GC. Isolation of 2 and 3 was performed by prep. GG on a Varian 1400 instrument, equipped with a stainless steel column  $(1.8 \text{ m} \times 4.3 \text{ mm})$  with 5% heptakis  $(2,6-\text{di}-O-\text{methyl}-3-O-\text{pentyl})-\beta$ -cyclodextrin/OV-1701 (1:1, w/w) on Chromosorb W-HP (phase A) [6]. The rearrangement products isobazzanene (4) and isobarbatene (5) were isolated using a stainless steel column  $(2.0 \text{ m} \times 5.3 \text{ mm})$  with 2.5% heptakis  $(6-O-\text{dimethylthexylsilyl}-2,3-\text{di}-O-\text{methyl})-\beta$ -cyclodextrin-SE-52 (20:80) on Chromosorb G-HP (phase B). He was used as carrier gas at a flow rate of 240 ml min<sup>-1</sup>.

NMR spectra (400 MHz) were measured in CDCl<sub>3</sub> using TMS as int. standard.

GC-EI-MS (70 eV) measurements were carried out on a HP 5890 gas chromatograph coupled to a VG Analytical VG 70-250S mass spectrometer.

β-Bazzanene (1). <sup>1</sup>H NMR: δ 5.29 (1H, m), 4.94 (1H, m), 4.79 (2H, d, J = 3 Hz), 1.64 (3H, bs), 1.02 (3H, s), 0.84 (3H, s); MS (EI, 70 eV), m/z (rel. int.): 204 [M]<sup>+</sup> (1), 109 (100), 108 (42), 93 (24), 67 (37), 41 (21).

α-Barbatene (2). α-Barbatene isolated from M. athamanticum was not completely separated from β-elemene. Therefore, the optical rotation was not measured. [α]<sub>D</sub><sup>22</sup> +55.3° (c 0.001, CHCl<sub>3</sub>, α-barbatene from B. trilobata); <sup>1</sup>H NMR: δ 5.19 (=CH, m), 1.65 (=C-CH<sub>3</sub>, s), 1.00 (3H, s), 0.89 (3H, s), 0.84 (3H, s); EI-MS (70 eV), m/z (rel. int.): 204 [M]<sup>+</sup> (14), 109 (28), 108 (79), 96 (37), 95 (60), 94 (35), 93 (100), 81 (28).

β-Barbatene (3).  $[\alpha]_D^{22}$  -22.5° (c 0.009, CHCl<sub>3</sub>, β-barbatene from B. trilobata),  $[\alpha]_D^{22}$  +19.4° (c 0.001, CHCl<sub>3</sub>, β-barbatene from M. athamanticum); <sup>1</sup>H NMR: δ 4.60 (1H, m), 4.57 (1H, m), 1.03 (3H, s), 0.90 (3H, s), 0.84 (3H, s); EI-MS (70 eV), m/z (rel. int.): 204 [M]<sup>+</sup> (3), 111 (36), 108 (79), 96 (100), 95 (73), 94 (50), 93 (91), 81 (66), 79 (38).

Isobazzanene (4). EI-MS (70 eV), m/z (rel. int.): 204 [M]<sup>+</sup> (9), 189 (18), 136 (50), 121 (100), 107 (21), 93 (43), 91 (22), 79 (20), 41 (31).

Isobarbatene (5). (-)- $\beta$ -Barbatene, 3 mg, (from B. trilobata) was dissolved in 0.5 ml n-hexane and a small amount of strong acidic ion exchange resin Amberlyst 15 was added. The reaction mixt, was stirred for 36 hr at 70°. Isobarbatene was isolated by prep. GC using

phase A. Optical rotation measurement yielded a positive value ( $\alpha = +29^{\circ}$  of approx. 1 mg sample in 1.5 ml n-hexane). <sup>1</sup>H NMR:  $\delta$  5.25 (=CH, dd, J = 31 Hz, J = 2.0 Hz), 1.15 (3H, s), 1.09 (3H, s), 1.01 (3H, s), 0.94 (3H, s); EI-MS (70 eV), m/z (rel. int.): 204 [M]<sup>+</sup> (7), 124 (42), 123 (100), 122 (21), 109 (21), 107 (23), 105 (23), 93 (23), 91 (29), 81 (68), 79 (23), 55 (23), 41 (37).

## REFERENCES

- Ohta, Y., Andersen, N. H. and Liu C.-B. (1977) Tetrahedron 33, 617.
- 2. Stahl, E. and Bohrmann H. (1967) Naturwissenschaften 54, 118.
- Kubeczka, K. H., Formacek, V. and Grünsfelder M. (1980) Planta Med. 39, 271.
- Asakawa, Y. (1982) in Progress in the Chemistry of Organic Natural Products (Herz, W., Grisebach, H. and Kirby, G. W., eds) Vol. 42, p. 1. Springer, Wien.
- Huneck, S. (1983) in New Manual of Bryology (Schuster, R. M, ed.), p. 1. The Hattori Bot. Lab. Nichinan, Japan.
- Hardt, I. H. and König, W. A. (1994) J. Chromatogr. A 666, 611.
- Andersen, N. H., Costin, C. R., Kramer Jr, C. M., Ohta, Y. and Huneck, S. (1973) *Phytochemistry* 12, 2709.
- König, W. A., Rieck, A., Hardt, I. H., Gehrcke, B., Muhle, H. and Kubeczka, K.-H. (1994) J. High Res. Chromatogr. 17, 315.
- Hardt, I. H., Rieck, A. Fricke, C. and König, W. A. (1995) Flavour Fragr. J. 10, 165.
- König, W. A., Gehrcke, B., Icheln, D., Evers, P., Dönnecke, J. and Wang, W. (1992) J. High Res. Chromatogr. 15, 367.
- 11. Matsuo, A. (1971) Tetrahedron 27, 2757.
- Wu, C.-L., Tsai, R. S. and Wu, C. M. (1982) Tamkang J. 19, 487.
- Wu, C.-L. and Liu, S. (1981) Tetrahedron 39, 2657.
- 14. Andersen, N. H., Tseng, C.-L. W., Moore, A. and Ohta, Y. (1978) *Tetrahedron* 34, 47.
- König, W. A., Krüger, A., Icheln, D. and Runge, T. (1992) J. High Res. Chromatogr. 15, 184.
- Fricke, C., Rieck, A., Hardt, I. H., König, W. A. and Muhle, H. (1995) *Phytochemistry* 39, 1119.