

3-HYDROXYTALATISAMINE FROM *ACONITUM NASUTUM*\*

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**Key Word Index**—*Aconitum nasutum*; Ranunculaceae; norditerpenoid alkaloids; 3-hydroxy-talatisamine.

**Abstract**—Together with five known alkaloids, a new norditerpenoid alkaloid was isolated from the above-ground parts of *Aconitum nasutum*. The new compound was identified as 3-hydroxytalatisamine on the basis of 1D and 2D NMR techniques.

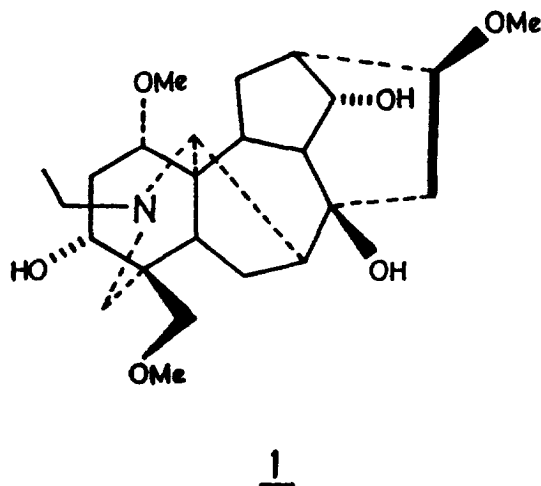
## INTRODUCTION

*Aconitum* species are used in traditional medicine, especially in China [1–3], while in Turkey, due to their high toxicity, they are not included in folk medicine, but used only as pain relievers under a physician's control [4]. Of the four *Aconitum* species found in Turkey, we have recently studied *A. orientale* [5]. In the present paper, we have investigated the alkaloids of *A. nasutum*, which was collected from eastern Turkey, near the Russian border. There is only one previous investigation of this species by a Russian group [6], but aconosine was not obtained from the Turkish collection. We have isolated five known diterpenoid alkaloids,

talatisamine [7], lycoctonine [8], columbidine [9], 14-acetylalatisamine [10] and anthranoyllycoctonine [11], and one new alkaloid 3-hydroxytalatisamine (**1**). In this report, we describe the isolation and structural identification of **1**.

## RESULTS AND DISCUSSION

The molecular formula of **1**,  $C_{24}H_{39}NO_6$  ( $m/z$  437.2770, calc. 437.2777), was derived from its HREI mass spectrum and  $^{13}C$  NMR (DEPT) spectrum, which showed four methyl quartets, seven methylene triplets, nine methine doublets (for 10 methine groups) and three quaternary carbon singlets. There were six oxygen-bearing C atoms represented by five signals ( $\delta$  72.0 *d*, 72.8 *s*, 75.6 *d*, 78.5 *t* and 82.3 *d*), three of them carrying three methoxyl groups; the other three, therefore, should have hydroxyl groups. The  $^1H$  NMR spectrum together with the  $^{13}C$  NMR spectrum provided the most information for **1**. The signals [ $\delta_H$  1.10 (3H, *t*,  $J = 7$  Hz,  $NCH_2CH_3$ );  $\delta_C$  13.5] showed the presence of an *N*-ethyl group, together with three methoxyl groups at [ $\delta_H$  3.28 (3H, *s*, C-18');  $\delta_C$  59.5 and 3.35 (6H, *s*, C-1' and C-16');  $\delta_C$  56.3 and 56.5] indicating that **1** is a norditerpenoid alkaloid. Other important signals were at [ $\delta_H$  2.9 (1H, *br d*,  $J = 10$  Hz, H-18a) and 3.1 (1H, *br d*,  $J = 10$  Hz, H-18b);  $\delta_C$  78.5 C-18], indicating that one of the methoxyl groups was at C-18 [ $\delta_H$  3.4 (2H, *m*, H-1 and H-16);  $\delta_C$  82.3 (C-1) and (C-16), the two other methoxyl groups being at C-1 and C-16 ( $\delta$  82.3). The signal at [ $\delta_H$  4.1 (1H, *dd*,  $J = 2$  and 4.5 Hz, H-14 $\beta$ ;  $\delta_C$  75.6] on  $D_2O$ -exchange collapsed the double of doublets at  $\delta$  4.1 to a triplet ( $J = 4.5$  Hz), thus indicating that one of the hydroxyl groups must be placed at C-14. The second hydroxyl was placed at C-8 following from the singlet signal



\*This paper is dedicated to Prof. Dr Rudolf Hansel (Münich, Germany) on the occasion of his 75th birthday.

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observed at  $\delta$  72.8. There was another doublet signal at  $\delta$  72.0, which should correspond to the carbon atom carrying the third hydroxyl group. This group could be placed at one of the following places, C-2, C-3, C-6, C-12 or C-15; the signal for C-15 with a  $\beta$ -OH appears at  $\delta$  69.0 and at 78–82 in the case of  $\alpha$ -OH [13]. Although an hydroxyl group at C-12 is rather rare in norditerpenoid alkaloids, nevertheless it appears at  $\delta$  76–77 [12, 13]; if the hydroxyl group is at C-6 and  $\beta$  it is observed at  $\delta$  77–80 with  $\alpha$  being  $\delta$  70–71. In norditerpenoid alkaloids, there is no example known with a C-2 carrying a hydroxyl group. When the hydroxyl group is at C-3 the chemical shift would be *ca*  $\delta$  71–72 and it would induce an upfield shift on the chemical shift of C-1 and a downfield shift on the chemical shift of C-2 and C-4. In the present case, a signal at [ $\delta_{\text{H}}$  3.72 (1H, *t*,  $J = 4$  Hz);  $\delta_{\text{C}}$  72.0] was attributed to a C-3 $\alpha$  hydroxyl group; however, the chemical shift of C-1 ( $\delta$  82.3) was lower than expected and the chemical shifts of C-2 ( $\delta$  37.7) and C-4 ( $\delta$  45.8) were higher than those with no hydroxyl group at C-3. Similar compounds with a 3 $\alpha$ -hydroxyl group induced the same chemical shifts [14–16]. A HETCOR experiment revealed the correlations between carbons and protons (Table 1). A selective INEPT experiment showed long-range correlations through three bonds and, in some cases, two and four bonds. Irradiation of the proton at  $\delta$  4.1 (H-14) enhanced C-8 and C-10,

irradiation of H-12 ( $\delta$  1.3) enhanced C-9, C-11 and C-14, and that of H-3 ( $\delta$  3.72) enhanced C-1, C-4 and C-5, while irradiation of H-9 (H-10)  $\delta$  2.28 (2.30) enhanced C-1 and C-13 and irradiation of H-19 ( $\delta$  2.47) enhanced C-3, C-5 and C-18. A NOESY experiment showed that the stereochemistry of the compounds was similar to that of talatisamine (2), and H-3 $\beta$  showed NOE with H-6 $\beta$  ( $\delta$  1.8) and H-7 $\beta$  ( $\delta$  2.2). All  $^{13}\text{C}$  NMR signals of 1 were similar to those of (2) except for C-1 to C-5, indicating the presence of a hydroxyl group at C-3. Spectral data are in agreement with the proposed structure for 1.

#### EXPERIMENTAL

*General.* IR spectra were recorded in  $\text{CHCl}_3$ .  $^1\text{H}$  and  $^{13}\text{C}$  NMR were measured using a Bruker AM 400, 2D experiments on a 500-MHz Bruker ORX.

*Plant material.* Above-ground parts of *A. nasutum* Fisch. et Reicht. were collected from Ardahan (Ardanuc Pass) near the Russian border in August 1984 and identified by Dr R. Ilarslan. A voucher is deposited in the Herbarium of the Faculty of Sciences, University of Ankara, No. Ilarslan 1675.

*Extraction and separation.* Powdered plant material (0.5 kg) was extracted with EtOH by percolation and the extract obtained evapd to dryness at 35° *in vacuo*. The residue was treated with 0.5 N  $\text{H}_2\text{SO}_4$  and ex-

Table 1.  $^1\text{H}$  and  $^{13}\text{C}$  NMR data for compound 1 and  $^{13}\text{C}$  NMR data for compound 2

C	1		2
	$^1\text{H}$ NMR	$^{13}\text{C}$ NMR	( $^{13}\text{C}$ NMR)
1	3.40 <i>m</i>	82.3 <i>d</i>	86.1
2	2.37 <i>dd</i> 2.42 <i>dd</i>	37.7 <i>t</i>	25.7
3	3.72 <i>t</i>	72.0 <i>d</i>	32.6
4	—	45.8 <i>s</i>	38.6
5	2.30 <i>br d</i>	48.5 <i>d</i>	37.7
6	1.80 <i>dd</i>	24.6 <i>t</i>	24.8
7	2.20 <i>d</i>	46.0 <i>d</i>	45.7
8	—	72.8 <i>s</i>	72.7
9	2.28 <i>m</i>	47.0 <i>d</i>	46.9
10	2.30 <i>m</i>	45.2 <i>d</i>	45.7
11	—	48.8 <i>s</i>	48.6
12	1.30 <i>dd</i>	28.7 <i>t</i>	28.6
13	2.00 <i>m</i>	45.3 <i>d</i>	45.7
14	4.10 <i>dd</i>	75.6 <i>d</i>	75.7
15	1.80 <i>dd</i>	39.1 <i>t</i>	39.2
16	3.40 <i>m</i>	82.3 <i>d</i>	82.2
17	3.30 <i>s</i>	62.8 <i>d</i>	62.8
18	3.10 <i>br d</i> 2.90 <i>br d</i>	78.5 <i>t</i>	79.4
19	2.47 <i>d</i> 2.37 <i>d</i>	53.2 <i>t</i>	53.1
NCH <sub>2</sub>	2.80 <i>q</i>	49.5 <i>t</i>	49.4
CH <sub>3</sub>	1.10 <i>t</i>	13.5 <i>q</i>	13.6
C-1 OMe	3.35 <i>s</i>	56.3 <i>q</i>	56.1
C-16 OMe	33.5 <i>s</i>	56.5 <i>q</i>	56.3
C-18 OMe	3.28 <i>s</i>	59.6 <i>q</i>	59.3

tracted with  $\text{CHCl}_3$ . NaOH (5%) was then added to the acidic soln (pH 10) and the soln extracted with  $\text{CHCl}_3$ . The  $\text{CHCl}_3$  extract was evapd to dryness, yielding 1.5 g of a crude alkaloidal mixt. which was chromatographed over basic alumina (column  $3 \times 50$  cm). The alkaloidal mixts obtained were further sepd and/or purified by prep. silica gel and alumina TLC (Merck). The following compounds were obtained: **2** (90 mg), 14-acetylaltatisamine (8 mg), columbidine (12 mg), lycotonine (14 mg), anthranoyllycotonine (7 mg) and **1** (10 mg).

**3-Hydroxyaltatisamine (1).** Amorphous.  $[\alpha]_D -14.3^\circ$  (c 0.3,  $\text{CHCl}_3$ ).  $\text{IR}_{\text{max}}^{\text{CHCl}_3} \text{ cm}^{-1}$ : 3404, 2928, 2823, 2361, 1634, 1455, 1381, 1295, 1226, 1160, 1090, 990, 944, 905, 753.  $^1\text{H}$  and  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ): Table 1. HREIMS  $m/z$  (rel. int. %): 437.2770  $[\text{M}]^+$  ( $\text{C}_{24}\text{H}_{39}\text{NO}_6$ ) (6), 406  $[\text{M} - \text{OMe}]^+$  (96), 376  $[\text{M} - 2 \times \text{OMe} + \text{H}]^+$  (100), 69 (3), 58 (4), 55 (2).

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