

ALKALOIDS OF *SARCOCOCCA SALIGNA*

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**Key Word Index**—*Sarcococca saligna*; Buxaceae; saracocinaene (3 $\alpha$ -dimethylamino-20 $\alpha$ -*N*-methyl-*N*-acylamino-pregna-5,16-diene); saracodine (3 $\alpha$ -dimethylamino-20 $\alpha$ -*N*-methyl-*N*-acylamino-pregnane); pachyaximine-A (3 $\beta$ -methoxy-20 $\alpha$ -dimethylamino-pregn-5-ene); spectroscopic studies.

**Abstract**—A new alkaloid, saracocinaene (3 $\alpha$ -dimethylamino-20 $\alpha$ -*N*-methyl-*N*-acylamino-pregna-5,16-diene), and two known alkaloids, saracodine and pachyaximine-A, were isolated from *Sarcococca saligna*. The structures of these steroidal bases were established on the basis of detailed spectroscopic techniques. Copyright © 1996 Published by Elsevier Science Ltd

## INTRODUCTION

*Sarcococca saligna* Muel (syn. *Sarcococca pruniformis* Lindl.) is an evergreen shrub found widely distributed in the northwest region of Pakistan [1]. The leaves of the herb enjoy considerable reputation as a remedy for different diseases and for the treatment of fever and rheumatism in the indigenous system of medicines. Our studies of the crude ethanolic extract of the aerial parts of *S. saligna* also showed good antibacterial activity against *Pseudomonas pseudomalliae*, *Shigella boydii* and *Carnebacterium diphtheria* bacterial strains.

A number of steroidal alkaloids which induce a non-recoverable fall in the blood pressure in dogs and are toxic to *Paramoecia* as well as exhibiting other activities have been isolated from the leaves of the plant [2–8]. The leaves of the plant also contain betulin [9, 10].

The present investigation has resulted in the isolation of a new alkaloid, saracocinaene (**1**), along with the known bases saracodine (**2**) and pachyaximine-A (**3**). Compound **3** was isolated for the first time from this plant. Compound **2** is the major constituent of this plant, which was first isolated by Kohli *et al.* [7].

## RESULTS AND DISCUSSION

An ethanolic extract of the aerial parts of *S. saligna* was evaporated and partitioned between chloroform and aqueous acid solution at various pH values. The acidic fraction was subjected to repeated column chromatography to afford compounds **1–3**.

Compound **1** was isolated as a white solid. The

HREI mass spectrum of **1** exhibited the  $[M]^+$  peak at  $m/z$  393.3260 analysing for  $C_{26}H_{42}N_2O$  (calc. 398.3296). Hence, **1** ( $C_{26}H_{42}N_2O$ ) possessed seven degrees of unsaturation. Four of these were accounted for by the tetracyclic structure of a pregnane type steroid, two were due to endocyclic double bonds and one due to a carbonyl function. The UV spectrum showed only terminal absorption. The IR spectrum ( $CHCl_3$ ) showed an intense absorption at  $1620\text{ cm}^{-1}$  characteristic of an amide function.

The HREI mass spectrum of **1** showed the molecular ion peak at  $m/z$  398.3260. A peak at  $m/z$  383.3020 was due to the loss of a methyl group from the  $[M]^+$  ion. The compound showed the base peak at  $m/z$  84.0830 ( $C_5H_{10}N$ ), representing the cleavage of ring A along with the  $N(CH_3)_2$  substituent in the ring A [4].

The  $^1H$  NMR spectrum of **1** showed doubling of signals for various protons due to the hindered rotation of the C-20 amide function. The methyl signals at  $\delta$  0.71/0.77 and 1.04/1.06 were due to quaternary methyl groups [11]. Another three-proton doublet at  $\delta$  1.14/1.26 was due to the secondary C-21 methyl group. A three-proton singlet at  $\delta$  2.66/2.69 was due to *N*-methyl and another one-proton singlet at  $\delta$  5.27 was characteristic of a  $\Delta^5$ -double bond. A one-proton triplet at  $\delta$  5.66 was assigned to the C-16 olefinic proton. The small coupling constant indicated a *gauche* conformation of the olefinic proton with respect to the C-15 methylenic protons. Similar doubling of the *N*-acetyl ( $\delta$  2.05/2.17) signal and various other neighbouring proton signals were also observed due to the restricted rotation. A one-proton quartet at  $\delta$  4.40/5.40 showed COSY-45° interaction with the C-21 methyl group at  $\delta$  1.14/1.26, confirming it to be the C-20 methine proton.

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Extensive 2D-NMR experiments (COSY 45°, NOESY, HOHAHA, HMQC and HMBC) [12] further confirmed the above mentioned assignments. Thus, compound **1**, called saracosinaene, was identified as 3 $\alpha$ -dimethylamino-20 $\alpha$ -*N*-methyl-*N*-acylamino-pregna-5,16-diene.

Compound **2** was found to be identical to saracodine, which was synthesized by Goutarel *et al.* [13]. The HREI mass spectrum of **2** exhibited the  $[M]^+$  peak at  $m/z$  402.3579 analysing for  $C_{26}H_{46}N_2O$  (calc. 402.3609) with five degrees of unsaturation. Four of these were accounted for by the tetracyclic structure of a pregnane-type steroid, while one was due to a carbonyl function. The UV spectrum was inconclusive. The IR spectrum ( $CHCl_3$ ) showed intense absorption at  $1620\text{ cm}^{-1}$  characteristic of an amide function. The  $^1H$  NMR spectrum of the compound also showed doubling of peaks due to the restricted rotation of the amide

function. Two quarternary methyl signals at  $\delta$  0.69/0.72 and 0.79/0.80 were due to the C-18 and C-19 methyl groups, respectively. A three-proton doublet at  $\delta$  1.05/1.15 ( $J = 7.2\text{ Hz}$ ) was due to the C-21 methyl group. Two three-proton singlets at  $\delta$  2.01/2.68 and 2.71/2.76 were due to  $COCH_3$  and  $NCH_3$  groups, respectively. A one-proton double quartet at  $\delta$  3.60/4.70 was due to the C-20 methine proton, which showed coupling with the C-21 methyl group. A six-proton broad singlet at  $\delta$  2.20 was due to the *N*-dimethyl group. These chemical shifts were further confirmed by HMQC, HMBC and HOHAHA NMR spectroscopic techniques. Thus, compound **2** was identified as saracodine (3 $\alpha$ -dimethylamino-20 $\alpha$ -*N*-methyl-*N*-acylamino-pregnane).

Compound **3** was obtained as a white solid. The HREI mass spectrum of **3** exhibited the  $[M]^+$  peak at  $m/z$  359.32065, corresponding to the molecular composition  $C_{24}H_{41}NO$  (calc. 359.3188). Hence, compound **3** possessed five degrees of unsaturation. Four of these were accounted for by the tetracyclic pregnane-type structure and one by a double bond. The IR spectrum showed peaks at 2902, 2810, 1665, 1590, 1455 and  $1090\text{ cm}^{-1}$ . The  $^1H$  and  $^{13}C$  NMR assignments were in agreement with the known compound 3 $\beta$ -methoxy-20 $\alpha$ -dimethylamino-pregn-5-ene earlier isolated from *Pachysandra axillaris* [8] and given the trivial name pachyaximine-A.

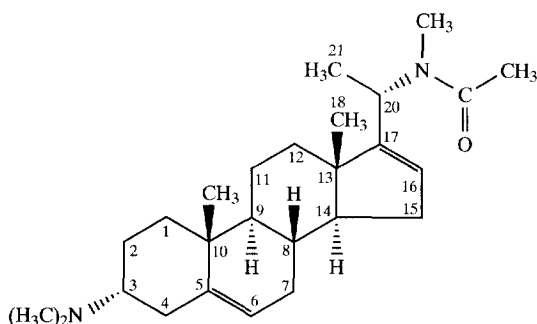
## EXPERIMENTAL

**General experimental procedure.** IR spectra: JASCO 302-A spectrophotometer. UV spectra: Hitachi U 3200 spectrophotometer. EI, FD and HREI MS: JMS HX100 (with data system) and JMS-DA 500 mass spectrometers.  $^1H$  and  $^{13}C$  NMR spectra: Bruker NMR spectrometer at 500 and 125 MHz, respectively, at room temp. Chemical shift values ( $\delta$ ) in ppm and coupling constants ( $J$ ) in Hz. Standard pulse sequences were used for COSY, HOHAHA, DEPT, HMQC and HMBC experiments.

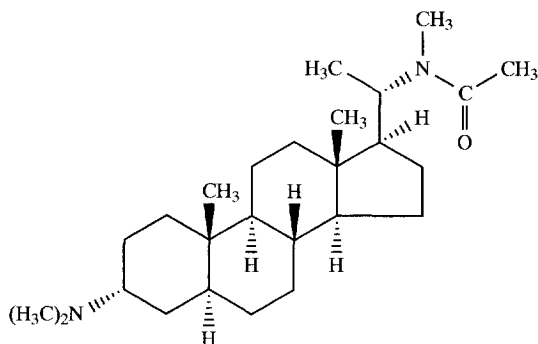
**Chromatographic conditions.** TLC (precoated silica G-25), UV254 plates; CC, silica gel, 230–400 mesh. Visualization of TLC plates was at 250 and 336 nm and Dragendorff's spray reagent was used for detection.

**Plant material.** Aerial parts of *S. saligna* (40 kg) were collected from Kuldana, Murree Hills, Pakistan, in October 1992.

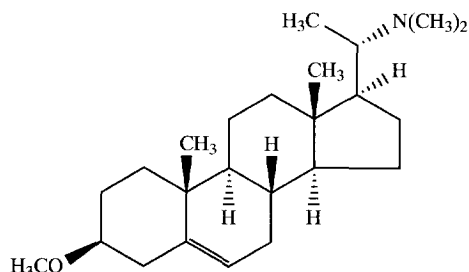
**Extraction and isolation.** The EtOH extract of plant aerial parts (40 kg) was evapd to a gum (2 kg). Total alkaloids (700 g) were obtained by extraction into 10% AcOH. Partial sepn of the alkaloids was achieved by extraction with  $CHCl_3$  at different pH values (3.5, 8.5). The fr. obtained at pH 3.5 (560 g) was subjected to CC on silica gel. Elution with  $CHCl_3$  and then with  $CHCl_3$ -MeOH yielded several frs. A fr. obtained by CC on elution with  $CHCl_3$ -MeOH (41:9) (1.5 g) was purified by prep. TLC with *n*-hexane-EtOAc-Et<sub>3</sub>NH (85:13:2) to afford **1** (5.5 mg). A fr. obtained by CC on elution with  $CHCl_3$ -MeOH (47:3) (3.0 g) was



1



2



3

Table 1.  $^{13}\text{C}$  NMR chemical shifts (in ppm) of compounds 1–3

No.	1		2		3	
	$\delta$	Multiplicity*	$\delta$	Multiplicity*	$\delta$	Multiplicity
1	35.7	$\text{CH}_2$	33.0	$\text{CH}_2$	37.2	$\text{CH}_2$
2	25.3	$\text{CH}_2$	28.7	$\text{CH}_2$	28.0	$\text{CH}_2$
3	62.7	CH	61.8	CH	80.4	CH
4	29.7	$\text{CH}_2$	31.9	$\text{CH}_2$	38.7	$\text{CH}_2$
5	141.0	C	39.5	CH	140.9	C
6	121.0	CH	24.9	$\text{CH}_2$	121.6	CH
7	30.1	$\text{CH}_2$	31.8	$\text{CH}_2$	31.9	$\text{CH}_2$
8	33.5	CH	35.4	CH	31.9	CH
9	49.8	CH	54.0	CH	50.2	CH
10	37.3	C	36.1	C	36.9	C
11	20.2	$\text{CH}_2$	20.7	$\text{CH}_2$	21.0	$\text{CH}_2$
12	34.5	$\text{CH}_2$	39.7	$\text{CH}_2$	39.6	$\text{CH}_2$
13	46.1	C	41.7	C	41.4	C
14	58.0	CH	56.6	CH	56.9	CH
15	31.2	$\text{CH}_2$	23.7/23.8†	$\text{CH}_2$	24.1	$\text{CH}_2$
16	127.4	CH	26.01†	$\text{CH}_2$	27.7	$\text{CH}_2$
17	154.3	C	53.7/54.2	CH	54.5	$\text{CH}_2$
18	15.9	Me	12.6/12.8	Me	12.1	Me
19	19.8	Me	12.1	Me	19.4	Me
20	44.6/51.1	CH	49.4/55.4	CH	61.1	CH
21	16.1/17.5	Me	18.2/19.0	Me	9.9	Me
NMe	27.8/29.8	Me	26.5/29.4	Me	—	—
NCOMe	22.2/22.6	Me	22.0/22.2	Me	—	—
N(Me <sub>2</sub> )	43.7	Me	43.8	Me	39.9	Me
(OMe)	—	—	—	—	55.6	Me
NCOMe	170.0/170.1	C	169/170.0	C	—	—

\*Multiplicities were determined by DEPT techniques.

†Assignments may be interchangeable.

again purified by CC using  $\text{CHCl}_3$ –MeOH (9:1) to yield a crystalline solid which was recrystallized using  $\text{CH}_2\text{Cl}_2$  and *i*-octane mixt. to give a crystalline compound **2** (32 mg) ( $1.49 \times 10^{-3}\%$ ). A fr. obtained on elution with  $\text{CHCl}_3$ –MeOH (43:7) was again purified, on a small column, using *n*-hexane–EtOAc–Et<sub>2</sub>NH (85:13:2) as eluent to give a pure crystalline solid compound **3** (206 mg) ( $0.51 \times 10^{-3}\%$ ).

**Saracocinaene (1)**. Solid, mp 129–130°;  $[\alpha]_D^{25}$  170.0 (c 0.05,  $\text{CHCl}_3$ ); IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ , 1620 (amide); MS  $m/z$  (rel. int.): 398  $[\text{M}]^+$  (20), 383  $[\text{M} - \text{Me}]^+$  (4), 84.1 (100).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 500 MHz):  $\delta$  0.71/0.77 (3H, s, Me-18), 1.04/1.06 (3H, s, Me-19), 2.05/2.17 (3H, s, Ac), 2.20 (6H, s, NMe<sub>2</sub>), 2.66/2.69 (3H, s, NMe<sub>3</sub>), 1.14/1.26 (3H, d,  $J = 6.8$  Hz, Me-21), 5.27 (1H, m, H-6), 5.66 (1H, t, H-16), 4.40/5.40 (1H, q, H-20).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz);  $\delta$ : see Table 1.

**Saracodine (2)**. Crystals mp 246–248°,  $[\alpha]_D^{25}$  –11.9 ( $\text{CHCl}_3$ ); IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ , 1620 (amide); MS  $m/z$  (rel. int.): 402  $[\text{M}]^+$  (26), 302 (4), 110 (35), 100 (7), 84 (100), 58 (18).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 500 MHz at room temp.):  $\delta$  0.69/0.70 (3H, s, Me-18), 0.79/0.80 (3H, s, Me-19);  $\delta$  1.05/1.15 (3H, d,  $J = 6.6/6.7$  Hz, Me-21), 2.01/2.68 (3H, s, Ac), 2.71/2.76 (3H, s, NMe), 2.20 (6H, br s, NMe<sub>2</sub>), 3.60/4.70 (1H, dq, H-20).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz);  $\delta$ : see Table 1.

**Pachyaximine-A (3)**. Crystalline solid mp 141–142°,  $[\alpha]_D^{25}$  11.90 ( $\text{CHCl}_3$ ); IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ , 2902, 2810, 1665,

1590, 1455, 1090; MS  $m/z$  (rel. int.): 359  $[\text{M}]^+$  (2.4), 344 (4), 84 (7), 72 (100), 55 (3).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 500 MHz):  $\delta$  1.85/1.03 (2H, m, H-1), 1.41/1.90 (2H, m, H-2) 3.05 (1H, ddd,  $J = 15.0, 15.0, 9.0, 4.5$  Hz, H-3), 2.38/2.50 (2H, m, H-4) 5.35 (1H, t, H-6) 1.53/1.97 (2H, m, H-7), 1.47 (1H, m, H-8), 0.92 (1H, m, H-9), 1.42/1.50 (2H, m, H-11), 1.14/1.91 (2H, m, H-12), 1.04 (1H, m, H-14), 1.06/1.59 (2H, m, H-15), 1.47/1.83 (2H, m, H-16), 1.35 (1H, m, H-17), 0.66 (3H, s, Me-18), 0.90 (3H, s, Me-19), 2.41 (1H, m, H-20), 0.86 (3H, d,  $J = 4.5$  Hz, Me-21), 2.15 (6H, br s, NMe<sub>2</sub>), 3.34 (3H, s, OMe).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 125 MHz);  $\delta$ : see Table 1.

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