



C-p-HYDROXYBENZOYLGLYCOFLAVONES FROM *CITRULLUS COLOCYNTHIS*

GALAL T. MAATOOQ, SALEH H. EL-SHARKAWY*, M. S. AFIFI and JACK P. N. ROSAZZA†

Dept. of Pharmacognosy, Faculty of Pharmacy, University of Mansoura, Mansoura 35516, Egypt; †Center for Biocatalysis and Bioprocessing, College of Pharmacy, University of Iowa, Iowa City, IA 52242, U.S.A.

(Received in revised form 30 April 1996)

Key Word Index—*Citrullus colocynthis*; Cucurbitaceae; C-linked flavone C-glucoside; *C-p*-hydroxybenzoylisoorientin; vitexin; isovitexin 4'-*O*-glucoside.

Abstract—In a flavonoid investigation of the fruits and aerial parts of *Citrullus colocynthis* six flavone C-glycosides were identified. The fruits contained isovitexin, iso-orientin and iso-orientin 3'-methyl ether, while the aerial parts contained three new *C-p*-hydroxybenzyl derivatives, viz. 8-*C-p*-hydroxybenzoylisovitexin, 6-*C-p*-hydroxybenzoylvitexin, and 8-*C-p*-hydroxybenzoylisovitexin 4'-*O*-glucoside. Their chemical identity was established by NMR spectroscopic methods including 2D-NMR, as well as UV and MS analyses. Copyright © 1996 Elsevier Science Ltd

INTRODUCTION

The fruits of *Citrullus colocynthis* Schrad (Cucurbitaceae) are well known for their medicinal properties [1]. Several cucurbitacins have been reported [2, 3] but there are no records of the flavonoid constituents of *C. colocynthis*. However, flavone C-glycosides have been isolated from other members of the Cucurbitaceae [4–6].

The present paper describes the isolation, purification and structure elucidation of the glycoflavones of the fruits and aerial parts of *C. colocynthis*.

RESULTS AND DISCUSSION

The methanolic extract of powdered *C. colocynthis* fruits was purified by column chromatography (CC) on a Sephadex LH20 to give two fractions rich in flavonoids. Further purification afforded isoorientin 3'-methyl ether (**1**), isoorientin (**2**), and isovitexin (**3**) [7]. The aerial parts were treated similarly on silica gel CC and further purification on preparative TLC resulted in three new flavone C-glycosides, viz. 8-*C-p*-hydroxybenzoylisovitexin (**4**), 6-*C-p*-hydroxybenzoylvitexin (**5**), and 8-*C-p*-hydroxybenzoylisovitexin 4'-*O*- β -*D*-glucoside (**6**). The identity of the isolated flavonoids was achieved using extensive UV, MS and NMR spectroscopic analyses. The complete ^{13}C NMR data are given in Table 1, and additionally confirmed by 2D-COSY, HETCOR, Selective INEPT experiments for the first time.

Isoorientin 3'-methyl ether (**1**) from the fruits

showed similar spectral data to isoorientin (**2**), [8] and its $[\text{M}]^+$ showed m/z 462 indicating the addition of a methyl group to **2**. Addition of AlCl_3 and AlCl_3/HCl reagents to **1** gave bathochromic shifts of 53 nm (from 337 to 390 nm) [9]. The absence of the catechol at ring B was confirmed by the $\text{NaOAc}/\text{H}_3\text{BO}_3$ spectrum which gave no bathochromic shift at 337 nm band while a 55 nm bathochromic shift in band-I of NaOAc spectrum showed the absence of a free 4'-hydroxyl and the 5.5 nm bathochromic shift of band-II indicated the presence of a free 7-hydroxyl. This was supported by the presence of 336 nm shoulder in the NaOAc spectrum [10, 11]. The ^1H NMR of **1** showed a doublet at δ_{H} 4.62 (1H, $J = 9.8$ Hz, $\text{H}_{1''}$) suggesting an anomeric proton of a β -linked sugar [4] and the presence of the methyl group was supported by δ_{H} 3.9 singlet and δ_{C} 55.84 signals. The signal at δ_{C} 108.72 ppm was assigned to a C_6 -linked to the sugar (normally between 96–98 ppm) and its connectivity was confirmed by a selective INEPT experiment where irradiation of the anomeric proton at δ_{H} 4.62 enhanced the carbon signals at δ_{C} 108.72 (C_6) and δ_{C} 163.24 (C_7).

Compounds **2** and **3** were identified by comparison of their physical properties and spectral analyses with reported data [7, 8].

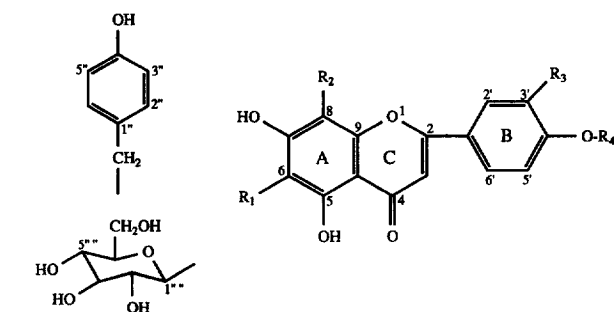
The EI-MS of **4**, from the aerial parts, gave a molecular ion at m/z 538 for $\text{C}_{28}\text{H}_{26}\text{O}_{11}$. The ^1H NMR spectral data (Table 1) showed four doublets (each of 2 protons) in two aromatic AB systems, viz. δ_{H} 7.8 and 6.92 ($J = 7.5$ Hz) corresponding to $\text{H}_{2'',6''}$ and $\text{H}_{3',5'}$ of ring B, and δ_{H} 7.07 and 6.63 ($J = 7.8$ Hz) corresponding to $\text{H}_{2'',6''}$ and $\text{H}_{3',5'}$ of ring D. The ^{13}C NMR data (Table 1) showed δ_{C} 127.99, 128.70, 115.81, and 114.77 shifts assigned to $\text{C}_{2'',6''}$ and $\text{C}_{2'',6''}$ and $\text{C}_{3',5'}$,

*Author to whom correspondence should be addressed.

Table 1. ^{13}C NMR data of isolated flavonoids (90.56 MHz, $\text{DMSO}-d_6$, TMS as int. standard)*

C	1	2	3	4	5	6
2	163.32 <i>s</i>	163.48 <i>s</i>	163.71 <i>s</i>	162.91 <i>s</i>	163.00 <i>s</i>	162.80 <i>s</i>
3	103.03 <i>d</i>	102.88 <i>d</i>	102.85 <i>d</i>	103.20 <i>d</i>	102.13 <i>d</i>	103.45 <i>d</i>
4	181.85 <i>s</i>	181.85 <i>s</i>	182.07 <i>s</i>	181.46 <i>s</i>	181.72 <i>s</i>	182.12 <i>s</i>
5	160.48 <i>s</i>	160.61 <i>s</i>	160.82 <i>s</i>	157.25 <i>s</i>	157.13 <i>s</i>	157.25 <i>s</i>
6	108.72 <i>s</i>	108.87 <i>s</i>	108.87 <i>s</i>	107.90 <i>s</i>	106.95 <i>s</i>	107.73 <i>s</i>
7	163.24 <i>s</i>	163.03 <i>s</i>	163.97 <i>s</i>	160.97 <i>s</i>	160.80 <i>s</i>	160.76 <i>s</i>
8	93.64 <i>d</i>	93.45 <i>d</i>	94.00 <i>d</i>	106.87 <i>s</i>	108.10 <i>s</i>	106.76 <i>s</i>
9	156.21 <i>s</i>	156.06 <i>s</i>	156.46 <i>s</i>	153.86 <i>s</i>	152.07 <i>s</i>	153.69 <i>s</i>
10	103.29 <i>s</i>	103.36 <i>s</i>	103.43 <i>s</i>	103.20 <i>s</i>	107.10 <i>s</i>	103.56 <i>s</i>
1'	121.32 <i>s</i>	121.40 <i>s</i>	121.15 <i>s</i>	121.19 <i>s</i>	121.39 <i>s</i>	123.86 <i>s</i>
2'	109.99 <i>d</i>	113.23 <i>d</i>	128.54 <i>d</i>	127.99 <i>d</i>	127.87 <i>d</i>	127.90 <i>d</i>
3'	147.90 <i>s</i>	145.53 <i>s</i>	116.27 <i>s</i>	115.81 <i>d</i>	115.80 <i>d</i>	116.47 <i>d</i>
4'	150.59 <i>s</i>	149.54 <i>s</i>	161.52 <i>s</i>	160.94 <i>s</i>	160.80 <i>s</i>	160.08 <i>s</i>
5'	115.68 <i>d</i>	115.96 <i>d</i>	116.27 <i>d</i>	115.81 <i>d</i>	115.80 <i>d</i>	116.47 <i>d</i>
6'	120.23 <i>d</i>	18.93 <i>d</i>	128.54 <i>d</i>	127.99 <i>d</i>	127.87 <i>d</i>	127.90 <i>d</i>
1''	—	—	—	130.83 <i>s</i>	130.92 <i>s</i>	130.21 <i>s</i>
2'',6''	—	—	—	128.70 <i>d</i>	128.77 <i>d</i>	128.87 <i>d</i>
3'',5''	—	—	—	114.77 <i>d</i>	114.66 <i>d</i>	114.87 <i>d</i>
4''	—	—	—	155.07 <i>s</i>	152.07 <i>s</i>	155.16 <i>s</i>
CH_2 -bridge	—	—	—	26.90 <i>t</i>	27.02 <i>t</i>	26.77 <i>t</i>
1'''	72.98 <i>d</i>	72.96 <i>d</i>	73.33 <i>d</i>	74.26 <i>d</i>	74.38 <i>d</i>	74.03 <i>d</i>
2'''	70.48 <i>d</i>	70.52 <i>d</i>	70.76 <i>d</i>	71.42 <i>d</i>	71.43 <i>d</i>	71.74 <i>d</i>
3'''	78.83 <i>d</i>	78.88 <i>d</i>	79.17 <i>d</i>	77.98 <i>d</i>	78.00 <i>d</i>	77.61 <i>d</i>
4'''	70.16 <i>d</i>	70.14 <i>d</i>	70.44 <i>d</i>	68.93 <i>d</i>	68.94 <i>d</i>	68.96 <i>d</i>
5'''	81.43 <i>d</i>	81.47 <i>d</i>	81.86 <i>d</i>	80.78 <i>d</i>	80.69 <i>d</i>	80.92 <i>d</i>
6'''	61.36 <i>t</i>	61.32 <i>t</i>	61.60 <i>t</i>	59.65 <i>t</i>	59.66 <i>t</i>	59.72 <i>t</i>
1'''	—	—	—	—	—	99.64 <i>d</i>
2'''	—	—	—	—	—	72.86 <i>d</i>
3'''	—	—	—	—	—	76.88 <i>d</i>
4'''	—	—	—	—	—	69.41 <i>d</i>
5'''	—	—	—	—	—	76.29 <i>d</i>
6'''	—	—	—	—	—	60.42 <i>t</i>
$\text{O}-\text{CH}_3$	55.84 <i>q</i>	—	—	—	—	—

*Multiplicities were determined by DEPT pulse sequence.



Compound	R_1	R_2	R_3	R_4
1	glucose	H	OMe	H
2	glucose	H	OH	H
3	glucose	H	H	H
4	glucose	<i>p</i> -hydroxybenzyl	H	H
5	<i>p</i> -hydroxybenzyl	glucose	H	H
6	glucose	<i>p</i> -hydroxybenzyl	H	glucose

C_{3''},_{5''}, respectively. The two singlets at δ_H 6.71 and 4.02 were assigned to H₃ and CH₂-bridge protons and the δ_H 4.79 ($J = 7.5$ Hz) doublet was assigned to the anomeric proton of the β -linked sugar [4]. DEPT experiment confirmed the resonance at δ_C 26.90 and 59.65 for the CH₂-benzyl and C₆-sugar, respectively. The COSY-45° spectrum showed that the doublet at δ_H 7.80 (H_{2',6'}) was coupled to that at δ_H 6.92 (H_{3',5'}) forming an AB system in ring-B. The doublet at δ_H 7.07 (H_{2'',6''}) was also coupled to δ_H 6.63 (H_{3'',5''}) forming the second AB-system in the *p*-hydroxybenzyl moiety. The CH₂-bridge protons at δ_H 4.02 was weakly coupled to the doublet at δ_H 7.07 (H_{2'',6''}) which confirmed the presence of a *p*-hydroxybenzyl moiety. The selective INEPT experiment showed that, irradiation (7 Hz) of the CH₂-bridge proton signal at δ_H 4.02 enhanced δ_C 160.98 (C₇), 153.90 (C₉), 106.87 (C₉) and 103.20 (C₁₀) indicating the flavonoid-C₈ connection to the *p*-hydroxybenzyl moiety.

From the previous discussion, it can be concluded that the β -D-glucose is C-linked to the flavonoid skeleton at C₆ and **4** is 8-*C-p*-hydroxybenzoyliso-vitexin, which is isolated from this plant for the first time.

The ¹H, ¹³C NMR (Table 1), DEPT, HETCOR and COSY-45°, spectra of **5** are similar to those of **4**. Compound **5** was identified as 6-*C-p*-hydroxybenzoyl-vitexin.

The mass spectrum of **6** showed a [M]⁺ of *M/Z* 700 for C₃₄H₃₆O₁₆. Both ¹H and ¹³C NMR data (Table 1) displayed similar spectral patterns to **4**; an extra doublet at δ_H 5.02 ($J = 7.2$ Hz) was assigned to the anomeric proton of a β -linked sugar (H_{1''}) [10]. The carbon spectrum showed 11 aliphatic hydroxy carbons between 59 and 81 ppm, six for the C-linked glucose (as in **4**) and the remaining five signals coincided with chemical shifts for *O*- β -D-glucose [10] and the sixth signal resonated at δ_C 99.64 (C_{1''}). The COSY-45° experiment indicated that the doublets at δ_H 7.95 (H_{2',6'}) and δ_H 7.07 (H_{2'',6''}) were coupled to δ_H 7.19 (H_{3',5'}), and δ_H 6.64 (H_{3'',5''}) doublets, respectively. The singlet at δ_H 4.06 (CH₂ bridge) was weakly coupled to the δ_H 7.07 doublet. A selective INEPT (9 Hz) irradiation of the δ_H 4.06 singlet enhanced δ_C 106.76 (C₈), 128.27 (C_{2'',6''}), 130.21 (C_{1''}) and 153.69 (C₉) indicating that the CH₂-bridge of the *p*-hydroxybenzyl moiety is linked to the flavonoid skeleton at C₈. Irradiation (9 Hz) of δ_H 4.81 (H_{1''}) of the anomeric doublet of the C-linked sugar enhanced δ_C 107.82 (C₆). Attempts to irradiate δ_H 5.03 (H_{1''}) of the *O*-linked sugar was unsuccessful; however, a ROESY experiment demonstrated that this anomeric signal was coupled to a δ_H 7.19 (H_{3',5'}) doublet and suggested its location in close vicinity to H_{3',5'}. Thus, the *O*-linkage must be at C₄ and the singlet at δ_H 6.92 (H₃) was coupled to the doublets at δ_H 7.95 (H_{2',6'}), and δ_H 7.19 (H_{3',5'}). Acid and enzymic hydrolysis of **6** gave D-glucose (comparative TLC) and **4** (co-TLC, UV, MS). These results confirm the structure of **6** as 8-*C-p*-hydroxybenzoyliso-vitexin 4'-*O*- β -D-glucoside.

The new compounds **4–6** represent a new class of flavone C-glycosides, in which the *p*-hydroxybenzoic acid moiety is attached through a carbon atom to the flavonoid skeleton. Glycoflavones acylated with *p*-hydroxybenzoic acid through an oxygen atom to the C-sugar are already known from families other than Cucurbitaceae, e.g. 2''-*O-p*-hydroxybenzoylvitexin from *Vitex lucens* (Verbenaceae) [12] and 2''-*O-p*-hydroxybenzoyliso-orientin and its 4'-*O*-glucoside from *Gentiana asclepiadea* (Gentianaceae) [13].

EXPERIMENTAL

Plant material. Aerial parts and fruits of *C. colocynthis* Schrad were collected early October, 1989 during the flowering and fruiting stage from Gabel El-Magharba, Sinai, Egypt. The plants were identified and authenticated by Professor I. Mashaly, of the Botany Department, University of Mansoura, Egypt, and a voucher specimen has been deposited at the Department of Pharmacognosy (# 12.10.89), University of Mansoura Herbarium.

Extraction and isolation. Powdered fruits (300 g) and aerial parts (600 g), of *C. colocynthis* were extracted with 70% MeOH (5 l). The concd extract was purified on an Amberlite Column (500 g), using H₂O and MeOH as eluants. The concd MeOH eluants gave 4.35 g and 10.1 g, respectively. The MeOH fruit extract was sepd by CC Sephadex LH 20 (Pharmacia) with 50% MeOH. Similar frs were pooled to afford 38 mg of **1**, and 232 mg of a mixt. of two compounds which upon further purification on Rp-C18 CC and eluting with (MeOH–H₂O–HCOOH, 20:80:1) gave 52 mg of **2**, and 78 mg of **3**. The aerial parts extract was purified on silica gel CC (MeOH–CH₂Cl₂, 1:4). Collected frs were concd and further purified on prep. TLC on silica gel containing 10% ZnSO₄ using MeOH–EtOAc (9:1) to afford 25 mg of **4**, 12 mg of **5**, and 15 mg of **6**, in addition to isovitexin.

Mps are uncorr. MS: Kratos MS-50 triple analyser using xenon as a carrier gas and 3-nitrobenzyl alcohol (3-NBA). HR-EI-MS: VG ZAB HF (70 eV). ¹H NMR 360 and 600 MHz, ¹³C NMR 90.56 and 150 MHz, DMSO-*d*₆. TMS as int. standard, chemical shifts (ppm) and *J* (Hz).

Compound 1. Isoorientin 3'-methyl ether. Yellow amorphous powder, mp 212–213°, UV $\lambda_{\max}^{\text{MeOH}}$ nm: 273 and 337; +MeONa: 260 sh, 278, 336 sh, 408; +AlCl₃: 262 sh, 280, 363, 390; +AlCl₃–HCl: 276, 352, 389; +NaOAc: 278, 319, 392; and +NaOAc–H₃BO₃: 274, 333. MS: *m/z* (rel. int.) 491 [M + 29]⁺ (3), 463 [M + 1]⁺ (36), 445 [M – H₂O]⁺ (26), 427 [M – 2H₂O]⁺ (1.7), 314 [M – 148]⁺ (3), and 313 [M – 149]⁺ (5), [165] (12), [151] (11), and [147] (31). ¹H NMR: δ_H 7.57 (*d*, 1H, $J = 8.6$ Hz, H_{6'}), 7.56 (*s*, 1H, H_{2'}), 6.96 (*d*, 1H, $J = 8.6$ Hz, H_{5'}), 6.55 (*s*, 1H, H_{3'}), 6.91 (*s*, 1H, H₈), 4.62 (*d*, 1H, $J = 9.7$ Hz, H_{1''}), and 3.90 (*s*, 3H, OCH₃). ¹³C NMR (Table 1).

Compound 4. 8-*C-p*-Hydroxybenzoyliso-vitexin. Yellow needles, mp 232–235°. UV $\lambda_{\max}^{\text{MeOH}}$ nm: 277, 327;

+MeONa: 286, 335, 404; +AlCl₃: 283, 310, 354, 399; +AlCl₃/HCl: 284, 309, 350, 388 sh; +NaOAc: 285, 305 sh, 394; and +NaOAc-H₃BO₃: 278, 323 sh, 360 sh. MS: *m/z* (rel. int.) 512 [M + 1-H₂O]⁺ (1.5), and 121 [C₇H₅O₂] (100); FAB-MS (3-NBA) *m/z* 561 [M + Na], and 539 [M + 1]⁺ HR-FAB-MS: *m/z* 539.1544 [M + H]⁺, calculated for C₂₈H₂₇C₁₁, 539.1545. ¹H NMR: δ_H 7.80 (*d*, 2H, *J* = 7.50 Hz, H_{2',6'}), 7.07 (*d*, 2H, *J* = 7.8 Hz, H_{2'',6''}), 6.92 (*d*, 2H, *J* = 7.50 Hz, H_{3',5'}), 6.63 (*d*, 1H, *J* = 7.70 Hz, H_{3'',5''}), 6.71 (*s*, 1H, H₃), 4.79 (*d*, 1H, *J* = 7.50 Hz, H_{1''}), and 4.02 (*s*, 2H, CH₂-bridge), HETCOR demonstrated that δ_H 4.02 (CH₂) is correlated to δ_C 26.90, δ_H 3.60 (H_{6''}) to δ_C 59.65 (C_{6''}), δ_H 3.45 (H_{2''}) to δ_C 71.42 (C_{2''}), δ_H 4.79 (H_{1''}) to δ_C 74.26 (C_{1''}), δ_H 3.30 (H_{3''}) to δ_C 77.90 (C_{3''}), δ_H 3.20 (H_{5''}) to δ_C 80.78 (C_{5''}), δ_H 6.71 (H₃) to δ_C 103.20 (C₃), δ_H 6.63 (H_{3',5'}) to δ_C 114.77 (C_{3',5'}), δ_H 6.92 (H_{2',6'}) to δ_C 127.99 (C_{2',6'}), and δ_H 7.07 (H_{2'',6''}) to δ_C 128.70 (C_{2'',6''}). ¹³C NMR (Table 1).

Compound 5. 6-*C-p*-Hydroxybenzoylvitexin. Amorphous powder, mp 242–243°, UV spectral data were identical to **4**. FAB-MS. *m/z* 539 [M + 1]⁺ and 561 [M + Na]⁺, and HR-FAB-MS: *m/z* [M]⁺ 538.1544 (calc. C₂₈H₂₆O₁₁, 538.1545). ¹H NMR: δ_H 7.77 (*d*, 2H, *J* = 7.30 Hz, H_{2'',6''}), 7.07 (*d*, 2H, *J* = 7.0 Hz, H_{2',6'}), 6.90 (*d*, 2H, *J* = 7.30 Hz, H_{3',5'}), 6.61 (*d*, 1H, *J* = 7.70 Hz, H_{3'',5''}), 6.62 (*s*, 1H, H₃), 4.75 (*d*, 1H, *J* = 9.20 Hz, H_{1''}), and 3.98 (*s*, 2H, CH₂-bridge), DEPT, HETCOR and COSY-45° spectra are similar to those of **4**. ¹³C NMR (Table 1).

Compound 6. 8-*C-p*-Hydroxybenzoylisovitexin 4'-*O*-β-D-glucoside. Yellow needles, mp 224–224°. UV λ_{max}^{MeOH} nm: 279, 322; +MeONa: 289, 333, 393; +AlCl₃: 267, 288, 344, 378 sh; +AlCl₃-HCl: 265, 287, 343, 372, +NaOAc: 284, 301 sh, 280 sh; +NaOAc-H₃BO₃: 278, 326 sh. MS (rel. int.) *m/z* 683 [M - H₂O + 1]⁺ (0.3), and 520 [M - O - glc - H₂O]⁺ (0.6) indicating [M]⁺ of 700. HR-FAB-MS, *m/z* 701.2082, 100% (calcd 701.2070 for C₃₄H₃₆O₁₆ + 1). ¹H NMR: δ_H 7.95 (*d*, 2H, *J* = 6.80 Hz, H_{2',6'}), 7.19

(*d*, 2H, *J* = 6.80, H_{3',5'}), 7.07 (*d*, 2H, *J* = 7.70 Hz, H_{2'',6''}), 6.92 (*s*, 1H, H₃), 6.64 (*d*, 1H, *J* = 7.90 Hz, H_{3'',5''}), 5.03 (*d*, 1H, *J* = 7.20 Hz, H_{1''}) 4.81 (*d*, 1H, *J* = 9.70 Hz, H_{1''}), and 4.06 (*s*, 2H, CH₂-bridge), DEPT, HETCOR and COSY-45° spectra are identical to those of **4**. ¹³C NMR (Table 1).

Acknowledgement—The authors thank Mr J. Snyder of the University of Iowa for technical assistance.

REFERENCES

1. *The National Formulary* (1949) The British Medicinal Association, The Pharmaceutical Society of Great Britain, p. 73.
2. Faust, R. E., Cwalina, G. E. and Ramsted, E. J. (1958) *J. Pharm. Assoc.* **47**, 1.
3. Lavie, D., Willner, D., Belkin, W. and Harely, W. G. (1959) *Acta Unio. Intr. Contra. Cancrum* **15**, 177.
4. Monties, B., Bouilant, M. and Chopin, J. (1976) *Phytochemistry* **15**, 1053.
5. Itokawa, H., Oshida, Y., Ikuta, A., Inatomi, H. and Ikegami, S. (1981) *Phytochemistry* **20**, 2421.
6. Bauer, R., Berganza, L. H., Seligmann, O. and Wagner, H. (1985) *Phytochemistry* **24**, 1587.
7. Wagner, H., Horhammer, L. and Kiraly, C. (1970) *Phytochemistry* **9**, 897.
8. Harborne, J. B. and Mabry, T. J. (1982) in *The Flavonoids: Advances in Research*. University Press, Cambridge, p. 19.
9. Markham, K. R. and Mabry, T. J. (1968) *Phytochemistry* **7**, 1197.
10. Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) in *The Systematic Identification of Flavonoids*, Springer, New York, p. 41.
11. Harborne, J. B., Mabry, T. J. and Mabry, H. (1975) *The Flavonoids*, Vol. I. Academic Press, London.
12. Horowitz, R. M. and Gentili, B. (1966) *Chem. Ind.* 625.
13. Goetz, M. and Jacot-Guillarmod, A. (1978) *Helv. Chim. Acta* **61**, 1343.