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# A HOMOISOFLAVANONE FROM PTEROCARPUS MARSUPIUM\*

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Abstract—A novel 6,7,3',4'-tetraoxygenated homoisoflavonoid, characterized as 6-hydroxy-7-O-methyl-3-(3-hydroxy-4-O-methyl benzyl)chroman-4-one, has been isolated from the ether-soluble portion of the heartwood of *Pterocarpus marsupium*. This is the first report of the isolation of a homoisoflavonoid from a *Pterocarpus* species. Copyright © 1997 Published by Elsevier Science Ltd

### INTRODUCTION

Pterocarpus marsupium Roxb is well known for its commercial importance and is used as a substitute for teak. It is commonly known as Vengisa of bijasal and possesses medicinal properties [1-4] which made it useful in the Ayurvedic and Unani systems of medicine. A phytochemical investigation of the heartwood of P. marsupium has led to the isolation of a new homoisoflavonoid, pteromarsupone (1), besides several known compounds such as trans-pterostilbene [5, 6], liquiritigenin [7], isoliquiritigenin [7], 1,3bis[4-hydroxyphenyl]-propan-2-ol [8], 1-(2,4-dihydroxyphenyl)propan-2-ol [9] and 8-hydroxy-4'methoxy isoflavanone-7-O-glucopyranoside [10]. Compound 1 belongs to a small class of natural products (homoisoflavonoids) some of which are known for their antimutagenic [11] and antiinflammatory [12] properties and are of chemotaxonomic interest [13].

### RESULTS AND DISCUSSION

Compound 1 gave a molecular ion peak [M]<sup>+</sup> at m/z 330. Its 1R spectrum showed characteristic absorptions of carbonyl and hydroxyl groups at 1 640 and 3400 cm<sup>-1</sup>, respectively. The <sup>1</sup>H NMR spectrum showed the presence of two methoxyl groups and contained characteristic peaks at  $\delta$  4.14 (dd, 1H) and 4.39 (dd, 1H) for C-2 protons, at  $\delta$  2.83 (m, 1H) for a C-3 proton, and at  $\delta$  2.60 (dd, 1H) and 3.15 (dd, 1H) for C-9 benzylic protons, indicating that 1 possessed a

homoisoflavonoid skeleton. This was supported by a benzylic carbon peak at  $\delta$  31.88 in its <sup>13</sup>C NMR spectrum. On acetylation, 1 gave a diacetate, showing the presence of two hydroxyl groups. Thus 1 was a tetraoxygenated 3-benzyl-4-chromanone. The substitution pattern in ring A and B was deduced by coupling the mass fragmentation pattern with the UV and <sup>1</sup>H NMR data. The base peak at m/z 137 was due to the hydroxymethoxy tropylium ion from ring B. The <sup>1</sup>H NMR spectrum showed signals at  $\delta$  6.75 (d,  $J_{\text{meta}} = 1.8 \text{ Hz}$ ), 6.85 (d,  $J_{\text{ortho}} = 7.93 \text{ Hz}$ ) and 6.71 (dd,  $J_{\text{meta}} = 1.8 \text{ and } J_{\text{ortho}} = 7.93 \text{ Hz}$ ) for C-2', C-5' and C-6' protons, respectively, indicating a 3',4'-dioxygenation pattern. The hydroxyl group was placed at C-3' and the methoxyl group at C-4' on the basis of the UV data and the NOE effects. Irradiation of the methoxyl signals at  $\delta$  3.81 caused a NOE enhancement of H-5' and H-8, thereby suggesting them to be at C-4' and C-7 positions. Again a peak at m/z 193 supported the presence of another set of hydroxyl and methoxyl groups in ring A. Since compound 1 did not

give a positive FeCl<sub>3</sub> test and its UV spectrum in methanol showed no bathochromic shift on treatment with either aluminium chloride or sodium acetate, it was inferred that it contained a free-OH at C-6 rather than at C-5. This was supported further by peaks at  $\delta$  6.50 (s, 1H) and 7.25 (s, 1H) for H-8 and H-5, respectively. Based on the above spectral characteristics, the new homoisoflavonoid, pteromarsupone, has been assigned the structure 6-hydroxy-7-O-methyl-3-(3-hydroxy-4-O-methyl benzyl) chroman-4-one. This

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compound has not been reported from any other natural source. This is the first report of the isolation of a homoisoflavonoid from *Pterocarpus* species and the oxygenation pattern that 1 contains is also rare occurring among earlier known homoisoflavonoids.

### **EXPERIMENTAL**

Mps: uncorr; IR: KBr; UV: MeOH; <sup>1</sup>H NMR: 250 MHz; <sup>13</sup>C NMR: 62.9 MHz; EIMS technique was used for MS analysis. CC: silica gel (60–80 mesh); TLC; silica gel G.

Plant material. The heartwood of Pterocarpus marsupium was collected locally and identified by Professor C. R. Babu, Department of Botany, University of Delhi, Delhi (India).

Extraction and isolation. The dried heartwood of P. marsupium (1.0 kg) was crushed and extracted with different solvents of increasing polarity for several hours in a soxhlet. The Et<sub>2</sub>O-soluble fraction of the solvent free EtOAc extract was concentrated in vacuo and the residue subjected to CC on silica gel. The column was initially eluted with  $C_6H_6$  and then with  $C_6H_6$ -EtOAc with an increasing concentration of EtOAc. Elution with  $C_6H_6$ -EtOAc (7:3) gave fractions that mainly contained compound 1 along with some minor components. The separation and purification of 1 was achieved by prep. TLC using  $C_6H_6$ -EtOAc (7:3) as the developing solvent.

6-Hydroxy-7-O-methyl-3-(3-hydroxy-4-O-methyl benzyl)-chroman-4-one (1). Recrystallized from CHCl<sub>3</sub>-MeOH as needles (8 mg), mp. 187°,  $R_f = 0.66$  and 0.74 in  $C_6H_6$ -EtOAc (7:3) and  $CH_2Cl_2$ -CH<sub>3</sub>OH (9:1), respectively, [α] – 71° (MeOH, c = 0.2). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 276. IR  $\nu_{\text{max}}^{\text{KBr}}$  cm <sup>-1</sup>: 3 400, 1 640; <sup>1</sup>H NMR (CD<sub>3</sub>)<sub>2</sub>CO: δ 2.60 (1H, dd, H-9), 2.83 (1H, m, H-3), 3.15 (1H, dd, H-9), 3.81 (6H, s, 2 × -OCH<sub>3</sub>), 4.14 (1H, dd, H-2), 4.39 (1H, dd, H-2), 6.50 (1H, s, H-8), 6.71 (1H, dd, J = 1.8 and 7.93 Hz, H-6'), 6.75 (1H, d, J<sub>m</sub> = 1.8 Hz, H-2'), 6.85 (1H, d, J<sub>o</sub> = 7.93 Hz, H-5'), 7.25 (1H, s, H-5); <sup>13</sup>C NMR (CD<sub>3</sub>)<sub>2</sub>CO: δ 31.88 (C-9), 47.45 (C-3), 55.76 and 55.93 (2 × -OCH<sub>3</sub>), 70.30 (C-2), 100.06 (C-4a), 110.60 (C-8), 112.03 (C-5'), 116.01 (C-2'), 116.10 (C-5), 120.25 (C-6'), 132.06 (C-1'), 146.40

(C-3', C-4' and C-6), 152.26 (C-8a), 157,26 (C-7), 191.82 (C-4); EIMS (probe) 70 eV, m/z (rel. int.): 330 [M]<sup>+</sup> (68), 313 (5), 194 (10), 193 (19), 167 (24), 149 (9), 138 (12), 137 (100), 122 (6), 94 (4), 72 (4), 55 (4), 28 (3), 18 (30). Acetate (pyridine–Ac<sub>2</sub>O, 24 hr at room temp) mp: 174°.

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#### REFERENCES

- Mitra, J. and Joshi, T., Phytochemistry, 1982, 21, 2429.
- The Wealth of India, A Dictionary of Indian Raw Materials and Industrial Products, VIII, 305.
  Publication and Information Directorate, CSIR, New Delhi, 1972.
- 3. Ojha, K. N., Pabrai, R. P. and Venkatachalam, K., Indian Journal of Pharmacology, 1949, 11, 188.
- 4. Gupta, S. S., Indian Journal of Medical Research, 1963, 51, 716.
- Mathew, T., Subba Rao, A. V. and Subba Rao, N. V., Current Science, 1977, 46, 337.
- Adinarayana, D. and Syamasundar, K. V., Phytochemistry, 1982, 21, 1083.
- Adinarayana, D. and Syamasundar, K. V., Seligmann, O. and Wagner, H., Z. Naturforsch, 1982, 37C, 145.
- Subba Rao, A. V., Mathew, J. and Sankaram, A. V. B., Phytochemistry, 1984, 23, 897.
- Mathew, J. and Subba Rao, A. V., Phytochemistry, 1984, 23, 1814.
- Mitra, J. and Joshi, T., Phytochemistry, 1983, 22, 2326.
- Wall, M. E., Wami, M. C., Manikumar, G., Taylor, H. and McGivney, R., Journal of Natural Products, 1989, 52, 774.
- Della, L. R., Del, N. P., Tubaro, A., Barone, G. and Parrilli, M., *Planta Medica*, 1989, 55, 587.
- Heller, W. and Tamm, Ch., Fortschr. Chem. Org. Naturst., 1981, 40, 105.