



REVISED STRUCTURE FOR FIVE ACETOPHENONES FROM *CYNANCHUM TAIWANIANUM*

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Key Word Index—*Cynanchum taiwanianum*; Asclepiadaceae; acetophenones; biacetophenone dimer; cynandiones A–D; cynanchone A.

Abstract—The structures of the acetophenones cynandione A–D and cynanchone A isolated from *Cynanchum taiwanianum* have been revised following re-examination of their NMR spectral properties. Copyright © 1997 Elsevier Science Ltd

INTRODUCTION

In previous papers [1, 2] we have reported the isolation and structural elucidation of five novel acetophenones, cynandione A–D (1'–4') and cynanchone A (5'). In a continuing study of the cytotoxic principles of this species, a 2,5-dihydroxyacetophenone (6) has been isolated. Examination of the NMR spectral properties of 1'–5' and 6 shows that the chemical shifts of the A ring of 1'–5' were in agreement with those of the corresponding data for 6 (Table 1); therefore, the characterizations of 1'–5' are incorrect. In the present paper, we report revised structures for 1'–5'.

RESULTS AND DISCUSSION

The chemical shifts for the A and B rings of 1 were in agreement with the corresponding data for 6, 2,4-dihydroxyacetophenone (7) [3] and published data [4]. By comparing the aromatic proton chemical shifts of 1 and its peracetate (1a) (Table 1), it was found that the signal at δ 6.49 experienced a significant downfield shift of +0.82 ppm, whereas the signal at δ 7.78 only

showed a slight downfield shift of +0.17 ppm. The other aromatic proton signals of 1 at δ 6.79 and 6.94 showed almost the same significant downfield shift, +0.46 and +0.45 ppm (Table 1), respectively, which is in accordance with the acetylation of hydroxyl groups in the *para* or *ortho* positions to the respective proton [5]. Based on the above evidence, 1' was revised to be 2,3'-diacetyl-3,6,2',6'-tetrahydroxybiphenyl (1). The two-dimensional spectrum of 1a (Fig. 1) also supported this revision.

In the ^1H NMR spectra of 2–5, the *ortho* coupled aromatic proton signals of the A ring, were all similar to those of the corresponding signals in 1. The two-dimensional NMR spectra of 2, 4 and 5 (Fig. 1) also indicated that structures 2', 4' and 5' should be revised to be 2, 4 and 5, and established the revision of ^{13}C NMR spectral assignments (Table 1). The ^{13}C NMR spectrum of 3 was assigned by comparison with those of corresponding data for 2 and 4, and also indicated that the structure of 3' should be revised to be 3. The mass spectrum of revised structures (Scheme 1) also supported the revision of structures 1'–5' to 1–5.

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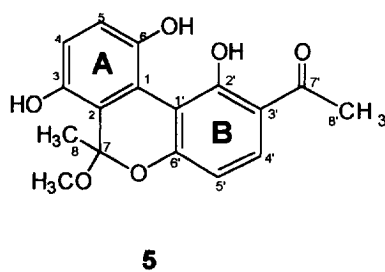
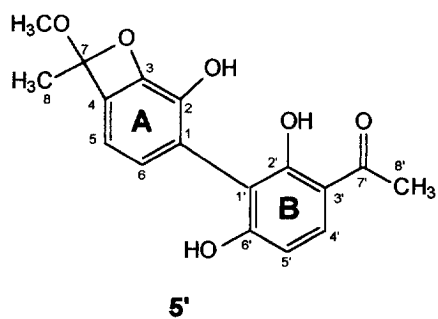
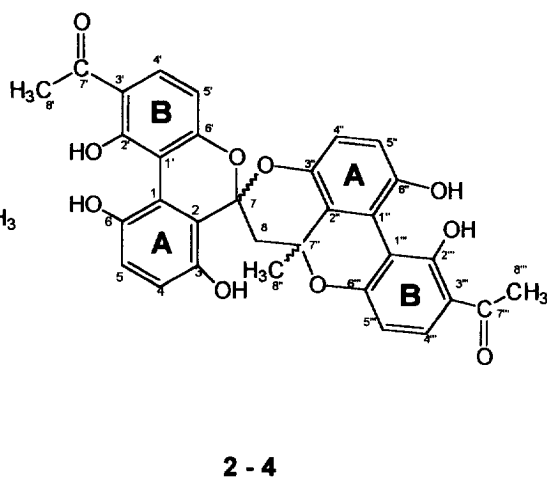
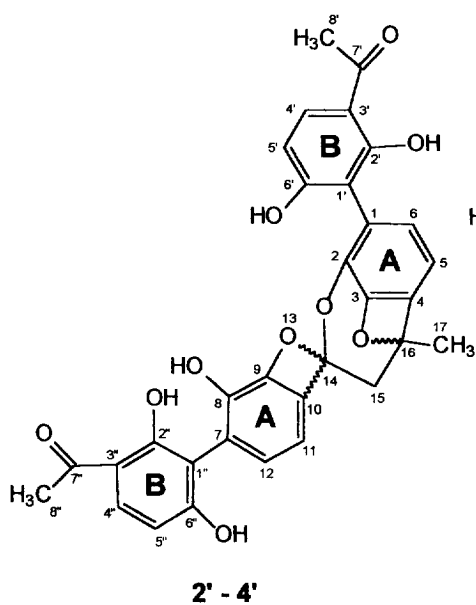
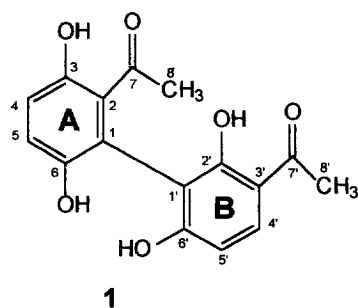
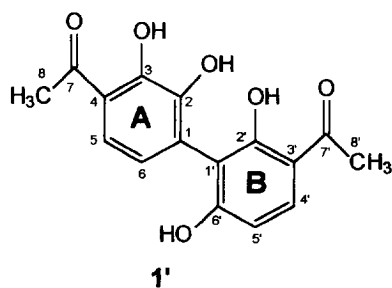


Table 1. NMR data of compounds 1–7

1 (CD ₃ OD)*		1a (CDCl ₃)+		2 (pyridine- <i>d</i> ₅)+		3 (pyridine- <i>d</i> ₅)*		4 (CDCl ₃)†		5 (CDCl ₃)†		6 (acetone- <i>d</i> ₆)‡		7 (DMSO- <i>d</i> ₆)§		
C	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H	δ _C	δ _H
1	128.1 ^{''}		122.4 ^{''}		122.0 ^{''}		122.0 ^{''}		122.8 ^{''}		118.9		120.7		112.8	
2	120.6		135.8		117.5		117.8		115.0		115.3		156.5		164.9	
3	152.6		144.1		149.1		148.9		147.6		147.1		119.5		102.3	
4	118.5	6.79 <i>d</i>	124.3	7.25 <i>d</i>	122.1	7.39 <i>d</i>	122.2	7.41 <i>d</i>	122.8	8.27 (OH)	122.0	7.03 <i>d</i>	125.9	6.78 <i>d</i>	164.2	6.25 <i>d</i>
5	122.0	6.94 <i>d</i>	125.0	7.37 <i>d</i>	119.6	7.29 <i>d</i>	119.5	7.31 <i>d</i>	120.4	7.10 <i>d</i>	118.9	6.92 <i>d</i>	130.6	7.00 <i>d</i>	108.1	6.37 <i>dd</i>
6	149.3		145.9		147.8		147.9		150.7	8.95 (OH)	146.0		116.4	7.22 <i>d</i>	133.6	7.74 <i>dd</i>
7	207.7		199.4		101.3		100.9		102.6		104.7		205.9		205.6	
8	31.2	2.18 <i>s</i>	30.7	2.07 <i>s</i>	41.1	3.30 <i>d</i>	46.1	3.43 <i>d</i>	44.4	2.77 <i>d</i>	24.0	1.60 <i>s</i>	26.9	2.58 <i>s</i>	26.2	2.52 <i>s</i>
9						4.85 <i>d</i>		5.08 <i>d</i>		2.93 <i>d</i>						
10					112.6		112.4		111.5		110.9					
11					158.7		158.9		158.2	15.6 (OH)	158.5	15.6 (OH)				
12					116.8		115.5		114.7		115.0					
13					133.2		133.0	7.74 <i>d</i>	132.1	7.70 <i>d</i>	131.8	7.77 <i>d</i>				
14					110.8	7.78 <i>d</i>	120.1	6.79 <i>d</i>	111.3	6.58 <i>d</i>	110.9	6.74 <i>d</i>				
15					160.7	6.84 <i>d</i>	161.3		161.0		159.0					
16					205.1		205.0 ^{''}		204.2 ^{''}		204.6					
17					26.4 ^{''}	2.53 <i>s</i>	26.3	2.55 <i>s</i>	26.3	2.68 <i>s</i>	26.2	2.68 <i>s</i>				
1'	114.8 ^{''}		122.5 ^{''}		113.1		113.3		114.7		114.7					
2'	164.0		147.8		120.9		120.9		122.8 ^{''}		122.8 ^{''}					
3'	113.4		128.7		149.1		142.3 ^{''}		147.6		147.6					
4'	134.3	7.78 <i>d</i>	131.0	7.95 <i>d</i>	118.9	6.97 <i>d</i>	111.2	6.85 <i>d</i>	120.3	7.02 <i>d</i>	120.3					
5'	109.0	6.49 <i>d</i>	120.1	7.31 <i>d</i>	121.0	7.17 <i>d</i>	120.6	7.20 <i>d</i>	120.5	7.08 <i>d</i>	120.5					
6'	164.0		152.0		143.1		143.2		146.1	7.40 (OH)	146.1					
7'	204.9		196.4		75.6		75.2		72.8		72.8					
8'	26.6	2.56 <i>s</i>	29.0	2.56 <i>s</i>	24.8	2.00 <i>s</i>	26.2	1.79 <i>s</i>	25.6	1.40 <i>s</i>	26.2					
1''					114.1		114.0		115.0		115.0					
2''					157.9		157.8		158.2	15.6 (OH)	158.2					
3''					115.7		116.6		114.7		114.7					
4''					132.6		132.4	7.81 <i>d</i>	131.8	7.74 <i>d</i>	131.8					
5''					111.6	7.83 <i>d</i>	111.6	6.47 <i>d</i>	111.5	6.45 <i>d</i>	111.5					
6''					159.2	6.80 <i>d</i>	159.0		158.2		158.2					
7''					205.1		205.1 ^{''}		205.2 ^{''}		205.2 ^{''}					
8''					26.7	2.57 <i>s</i>	26.7	2.58 <i>s</i>	26.2	2.68 <i>s</i>	26.2					
OCH ₃																
											51.6	3.52 <i>s</i>				
(1a)	OCOMe 20.6, 20.7(2), 20.8, 2.06, 2.08, 2.12, 2.30															
	OCOMe 168.0, 168.4, 168.8, 169.1															

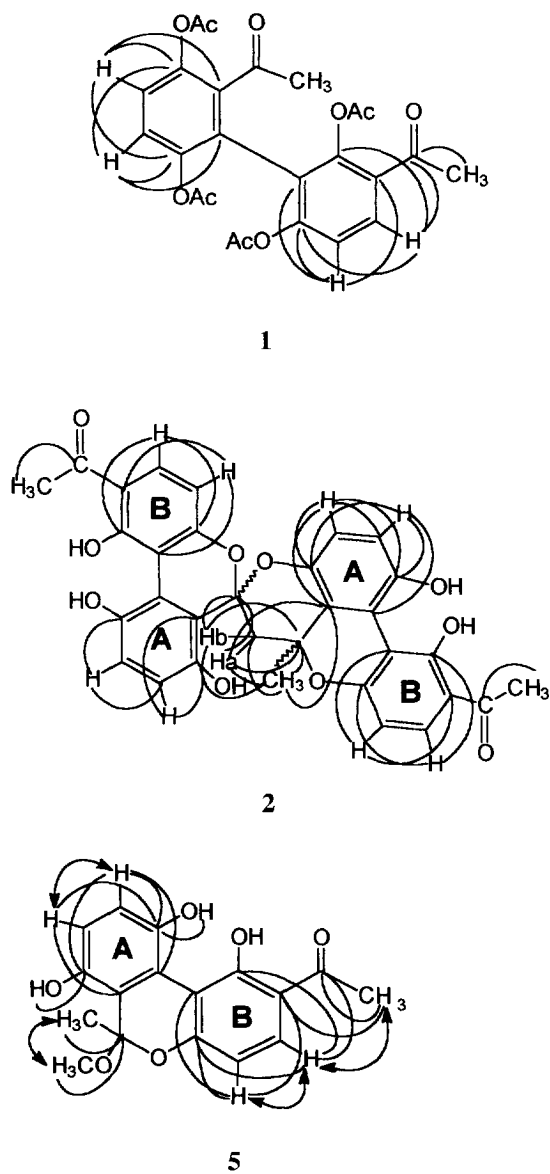
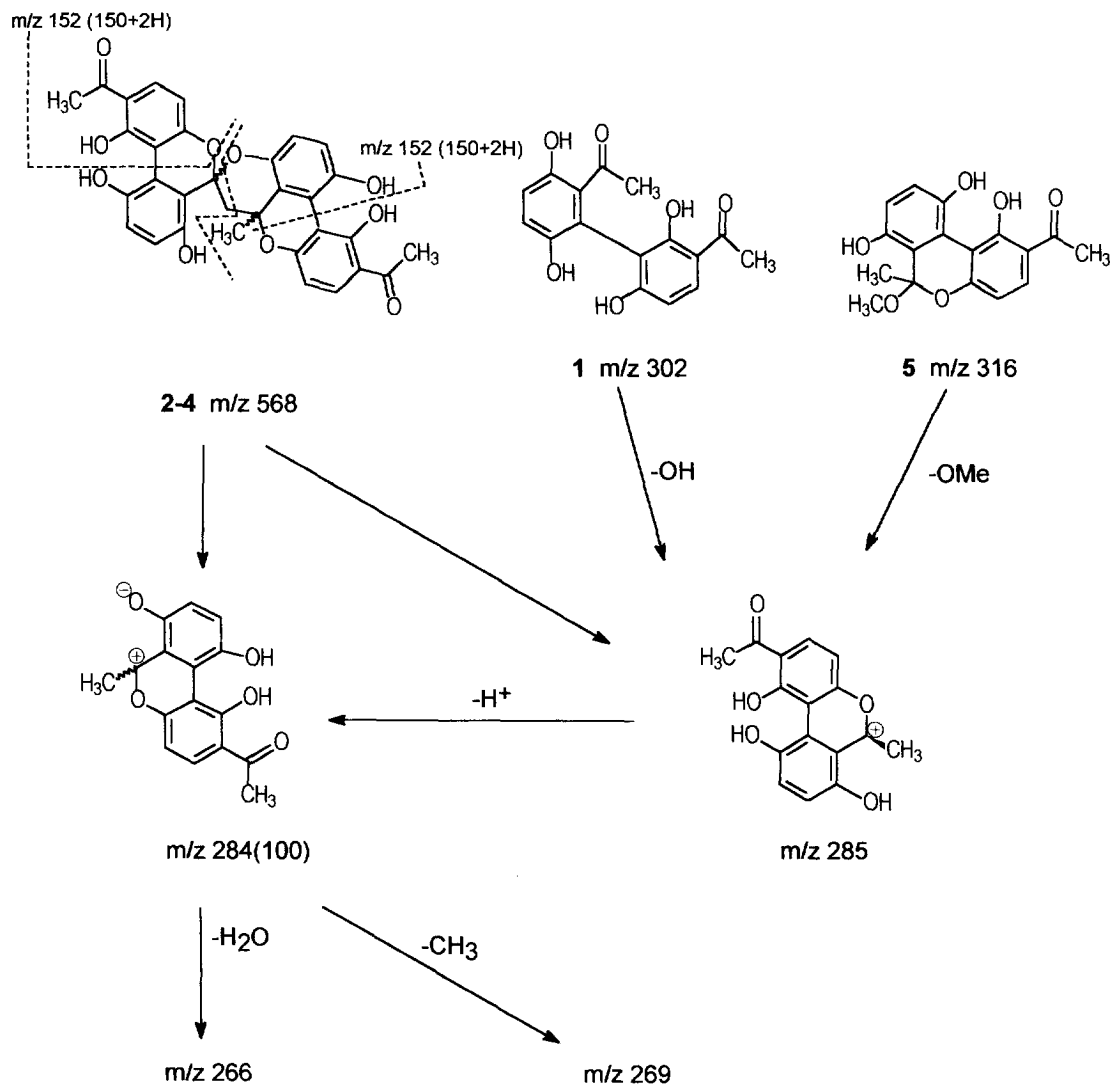


Fig. 1. (—) HMBC of compounds **1**, **2** and **5** and (---) NOESY of **5**.



Scheme 1. Mass spectral fragmentation of compounds 1-5.

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REFERENCES

- Huang, P. L., Lu, C. M., Yen, M. H., Wu, R. R. and Lin, C. N., *Phytochemistry*, 1995, **40**, 537.
- Huang, P. L., Lu, C. M., Yen, M. H., Wu, R. R. and Lin, C. N., *Phytochemistry*, 1996, **41**, 293.
- Li, J., Kadota, S., Kawata, Y., Hattori, M., Xu, G. J. and Namba, T., *Chemical Pharmaceutical Bulletin*, 1992, **40**, 3133.
- Bieman, K., in *Spectral Data for Structure Determination of Organic Compounds*. Springer, Berlin, 1989, p. c120.
- Steglich, W. and Losel, W., *Tetrahedron*, 1969, **25**, 4301.