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MYRICETIN 5,7,3',4',5'-PENTAMETHYL ETHER AND OTHER METHYLATED FLAVONOIDS FROM MURRAYA PANICULATA

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Key Word Index—*Murraya paniculata*; Rutaceae; myricetin 5,7,3',4',5'-pentamethyl ether; flavonol; methoxyflavones; chemotaxonomy.

Abstract—Seven flavonoids, including one new flavonol, were isolated from the leaves of *Murraya paniculata* (Rutaceae). Four were identified as previously known constituents of *M. paniculata* or allied species, namely, 5.7.3',4'.5'-pentamethoxyflavone. 5.7.8.3',4'.5'-hexamethoxyflavone, 3.5.7.3',4'.5'-hexamethoxyflavone and 3.5.7.8,3',4'.5'-heptamethoxyflavone. The structure of the new flavonol was elucidated as 3-hydroxy-5.7.3',4'.5'-pentamethoxyflavone on the basis of spectroscopic studies. Two other flavonoids, a methylated flavanone and its corresponding chalcone isomer were identified as 2-(S)-5.7.3',4'.5'-pentamethoxyflavanone and 2'-hydroxy-3.4.5.4',6'-pentamethoxychalcone, respectively. © 1997 Elsevier Science Ltd. All rights reserved

INTRODUCTION

The dried leaves of Murraya paniculata (L.) Jack (Rutaceae) are frequently used in traditional Indonesian mixed herbal medicines (jamu). In a previous chemical investigation of the root of M. paniculata the first author reported an alkaloid, prenylated coumarins and polymethoxyflavones [1]. More recently, we reported the isolation of highly oxygenated flavonoids [2] and prenyl coumarins [3] from leaves of the same plant. Here, we describe the isolation and characterization of further minor components including a new flavonol.

RESULTS AND DISCUSSION

The chloroform extract of *M. paniculata* leaves was separated by a combination of silica gel and Sephadex LH-20 column chromatography to afford seven flavonoids. Two were readily identified by comparison with authentic markers of 5,7,3',4',5'-pentamethoxy-flavone (1) and 5,7,8,3',4',5'-hexamethoxyflavone (2), which were previously obtained from the root of this plant [1]. Two other polymethoxyflavones were also isolated and characterized as 3,5,7,3',4',5'-hexamethoxyflavone (3) and 3,5,7,8,3',4',5'-heptam-

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ethoxyflavone (4). Compound 3, a permethylated derivative of myricetin, has been reported from the leaves of *M. exotica* [4] and the flowers of *M. paniculata* [5], and compound 4 from the leaves of *M. paniculata* of Sri Lankan origin [6].

Compound 5 was obtained as pale yellow needles of mp 232-234. Its molecular formula was determined to be $C_{20}H_{20}O_8$ based on the high resolution mass spectral analysis. Methylation of 5 with diazomethane afforded a hexamethoxyflavone identical to myricetin permethyl ether (3). The ¹H NMR spectrum of 3 exhibited the presence of five methoxyls, AB aromatic protons [δ 6.37 (d, J = 2.1 Hz) and 6.55 (d, J = 2.1 Hz)], one singlet at δ 7.49 with an integration of two protons, and a broad peak at δ 7.48 that disappeared on deuteration, but a diagnostic signal in a lower field (10–15 ppm) due to a chelated 5-hydroxyl group was not observed. The UV spectrum exhibited a large bathochromic shift (61 nm) on addition with AlCl₃. The above evidence assigned the structure of 5 as 3-hydroxy-5,7,3',4',5'-pentamethoxyflavone, and it structure was further substantiated by comparison of ¹³C resonances of 5 with those of related compounds 1 and 3 as shown in Table 1. Minor constituents that are considered to be biogenetically related to these polymethoxyflavones were also obtained, and characterized as 2-(S)-5,7,3',4',5'-pentamethoxyflavanone (6) and 2'-hydroxy-3,4,5,4',6'-pentamethoxychalcone (7). Compounds 6 and 7 have been reported previously in the Rutaceae, from Neoraputia alba [7] and Mer-

rillia caloxylon [8], respectively. This is the second report of these constituents. The co-occurrence of 6 and 7 along with 1 and 3, which have the same substitution pattern, is noteworthy in view of the biogenesis of polymethoxyflavonoids.

EXPERIMENTAL

General procedure. Mps: uncorr; IR: KBr; UV: MeOH; ¹H (400 MHz) and ¹³C (100 MHz) NMR: CDCl₃ with TMS as int. standard; Si gel CC: Wakogel C-200. C₆H₆-Me₂CO (B-A); Sephadex LH-20 CC: Pharmacia, CHCl₃-MeOH: TLC: Si gel (60 GF₂₅₄, Merck), spots were detected by inspection under UV light (254 and 365 nm) or by the colours developed by spraying with 10% H₂SO₄ followed by heating.

Plant materials. The leaves of Murraya paniculata were collected in 1984–1985 near Sukabumi, Java. Indonesia. A voucher specimen is on deposit at the Herbarium of the University of Tokyo. A further sample was collected near Bandung, Java, Indonesia by the second author (K. F.). Since TLC analysis of both samples showed the same chemical spectrum, chemical analysis was carried out with the plant material collected in Sukabumi.

Extraction and isolation. The CHCl₃ extract of M. paniculata leaves was separated into 12 frs (Fr. I–Fr. XII) by Si gel CC as previously described [2, 3]. Fr. XII (20.69 g) was chromatographed over Si gel (column size 6.5×18.5 cm) with C_6H_6 – Me_2CO (B-A) as eluants increasing amounts of Me_2CO stepwise collecting 400 ml frs. Frs eluted with 6% B-A, on crystallization from MeOH, afforded 20 mg of 2′-hydroxy-3,4,5,4′.6′-pentamethoxychalcone (7), mp 186–187 (lit. [8] 176–178°). Frs eluted with 12% B-A were chro-

matographed over Sephadex LH-20 to yield 2-(S)-5.7.3'.4'.5'-pentamethoxyflavanone (6) (0.18 g), mp 174–175 (recryst. from MeOH), $[\alpha]_D - 36.3^\circ$ (MeOH; c 0.187). Frs eluted with 16% B-A were further separated by Sephadex LH-20 CC to afford 12 mg of 3hydroxy-5,7,3',4',5'-pentamethoxyflavone (5) and 0.14 g of 5,7,8,3',4',5'-hexamethoxyflavone (2), mp196-197 (lit. [1] 195-196). Frs eluted with 20% B-A were separated by Sephadex LH-20 CC to yield 3.5,7,3',4',5'-hexamethoxyflavone (3) (0.96 g), mp 155–157 (lit [5] 148–150°). Frs eluted with 24% B-A were combined and recryst, from MeOH to afford 3.5.7.8.3',4',5'-heptamethoxyflavone (4) (0.36 g), mp 193–194 (lit. [6] 192–193°). Frs eluted with 32% **B-A** afforded ppts, which were recryst. from MeOH to give 5,7,3',4',5'-pentamethoxyflavone (1) (0.63 g), mp 195– 196 (lit. [1] 197).

3-Hydroxy-5,7,3',4'.5'-pentamethoxyflavone Yellow needles from Me₂CO, mp 232–234°; TLC: R_t on Si gel 60GF₂₅₄ (Merck): 0.40 (CHCl₃-Me₂CO 4:1); 0.25 (benzene-EtOAc 4:1); spot colour (365 nm): dark yellow; + NH₃ (365 nm): pale yellow; UV λ_{max}^{MeOH} (log ε): 248 (4.05), 259sh (4.01), 305 (3.81), 355 (4.09); $\lambda_{max}^{MeOH + NaOAc}$: 248, 259sh, 305, 355; $\lambda_{max}^{MeOH + NaOAc + H_3BO_3}$: 248, 259sh, 305, 355; $\lambda_{max}^{MeOH+MeONa}$: 259, 398; $\lambda_{max}^{MeOH - AICl_3}$: 263sh, 272, 335, 416; $\lambda_{max}^{MeOH + AICl_3 + HCl_3}$: 263sh, 272, 335, 416 nm. ¹H NMR (400 MHz, CDCl₃) δ: 3.93 (3H, s, 7-OMe), 3.94 (3H, s, 4'-OMe), 3.97 (6H, s, 3'- and 5'-OMe), 3.99 (3H, s, 5-OMe), 6.37 (1H, d, J = 2.1 Hz, 6-H), 6.55 (1H, d, J = 2.1 Hz, 8-Hz)H), 7.48 (1H, br d, 3-OH, disappeared on deuteration), 7.49 (2H, s, 2'- and 6'-H). ¹³C NMR (100 MHz, CDCl₃) δ : see Table 1. EIMS m/z (rel. int., %): 388 (M⁺, 100), 373 (37), 342 (27), 179 (10), HRMS: [M]⁺ 388.1157 ($C_{20}H_{20}O_8$ requires 388.1159).

Table 1. 13 C NMR (100 MHz, δ ppm, TMS as internal standard) spectral data of flavonoids isolated from M. paniculata leaves

C*	1	3	5	6	C†	7
C-2	160.5	152.3	139.7	79.4	С-β	142.4
C-3	108.9	140.1	138.1	45.8	C-α	127.0
C-4	177.5	174.0	172.0	189.0	C = 0	192.4
C-5	161.0	161.1	160.7	162.3	C-6′	162.5
C-6	96.2	95.8	95.8	93.6	C-5'	91.4
C-7	164.1	164.0	164.6	166.0	C-4'	166.2
C-8	92.9	92.5	92.5	93.3	C-3'	93.9
C-9	159.9	158.8	158.9	164.9	C-2'	168.4
C-10	109.3	109.5	106.2	106.0	C-1′	106.4
C-1'	126.8	126.0	126.4	134.3	C-1	131.2
C-2'	103.5	105.9	105.1	103.3	C-2	105.7
C-3′	153.6	153.1	153.3	153.5	C-3	153.5
C-4′	140.9	141.6	141.7	138.3	C-4	140.2
C-5′	153.6	153.1	153.3	153.5	C-5	153.5
C-6′	103.5	105.9	105.1	103.3	C-6	105.7
3-OMe	V	60.1				
5-OMe	56.4	56.4	56.4	56.2	6′-OMe	55.8
7-OMe	55.8	55.8	55.9	55.6	4'-OMe	55.6
3'-OMe	56.4	56.4	56.4	56.2	3-OMe	56.2
4′-OMe	61.0	61.0	61.0	60.8	4-OMe	61.0
5'-OMe	56.4	56.4	56.4	56.2	5-OMe	56.2

Assignments were aided by ¹³C-¹H COSY and long range ¹³C-¹H COSY spectra.

Methylation of 5. To a MeOH soln of 5 (3 mg) was added excess CH_2N_2 – Et_2O , and the mixt. kept at 0 for 24 hr. Excess CH_2N_2 was decomposed by adding HCOOH, and then the solvent removed. The residue was recryst. from MeOH to give a methyl ether (2.5 mg) which was identical to myricetin permethyl ether (3) in mmp, IR and ¹H NMR.

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^{*} Numbering based on the flavone, flavonol or flavanone skeleton.

[†] Numbering based on the chalcone skeleton.