

PII: S0031-9422(97)00175-1

TRITERPENOID SAPONINS FROM THE LEAVES OF ILEX LATIFOLIA

MING-AN OUYANG, HAN-QING WANG, YU-QING LIU† and CHONG-REN YANG†*

Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences, Lanzhou 730000, People's Republic of China; † Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650204, People's Republic of China

(Received in revised form 20 January 1997)

Key Word Index—Ilex latifolia; Aquifoliaceae; triterpene; saponins; latifoloside A-E; pomolic acid; siaresinolic acid; ilexgenin B.

Abstract—Five new triterpenoid saponins latifolosides A-E were isolated from the leaves of Ilex latifolia, along with a known compound. Their chemical structures have been elucidated on the basis of the chemical and spectral methods. © 1997 Elsevier Science Ltd. All rights reserved

INTRODUCTION

Ilex latifolia Thunb, one of the species in the Ilex genus used in the tea Ku-Ding Cha [1], has been used in China as a diuretic and remedy for sore throat, weight loss and hypertension [2, 3]. In a previous paper [4, 5], we reported the identity of the triterpenoid glycosides from I. kudincha. As a part of our continuing phytochemical research on plants in the genus of *Ilex*, this paper deals with the isolation and structural elucidation of five new triterpenoid saponins, latifoloside A(1), B(2), C(3), D(4) and E(5), along with a known compound (6) from the leaves of I. latifolia.

RESULTS AND DISCUSSION

The butanol soluble fraction of the methanol extract of the leaves of I. latifolia was repeatedly chromatographed on silica gel to yield six saponins 1-6. By comparing of the ¹H and ¹³C NMR signals with reported data, compound (6) was identified as kudinoside G, isolated from I. kudincha [4]. Three compounds 1, 2, 4 and the other three compounds 3, 5, 6 were isomers, respectively. Their IR spectra showed ester group absorption (1730 cm⁻¹) together with strong hydroxyl absorption (3650–3100 cm⁻¹) and a C = C double bond (1640 cm⁻¹).

Compound 1 was a colourless powder and its molecular formula was determined as C₄₇H₇₆O₁₇ by the negative FAB-mass spectrum together with its NMR spectrum (DEPT). Cellulase treatment of 1

cated the presence of three anomeric signals; one aarabinopyranosyl unit [H-1: δ 4.88 (d, J = 5.2 Hz), C-1: δ 104.8], one α -rhamnopyranosyl unit [H-1: δ 6.11, br s, C-1: δ 101.9], and one β -glucopyranosyl unit [H-1: δ 6.28 (d, J = 8.0 Hz), C-1: δ 96.0]. Compound 1 afforded L-arabinose, D-glucose, and L-rhamnose (1:1:1) (by HPLC) on acid hydrolysis. For the aglycone moiety, a glycosylation shift was observed for C-3 signal (+10.7 ppm, from δ 78.4 to δ 89.1). In the ¹³C NMR spectrum, the anomeric carbon of the glucose group at δ 96.0 suggested that 1 had a 28-O-glycosidic linkage, which was proved by the ¹³C NMR signal at δ 177.2 (C-28), alkaline hydrolysis and the HMBC spectrum (see Fig. 1). Alkaline hydrolysis of 1 gave compound (8) and D-glucose. The ¹H NMR and ¹³C NMR spectra of 8 indicated the presence of one αarabinopyranosyl unit [H-1: δ 4.87 (d, J = 5.3 Hz), C-1: δ 104.9] and one α -rhamnopyranosyl unit [H-1: δ 6.20, br s, C-1: δ 102.0]. Comparison of the ¹³C NMR spectrum of 8 with that of ziyuglucoside 9 [7] showed a glycosylating shift for the C-2 signal of the arabinopyranosyl moiety, demonstrating that an α-rhamnopyranosyl group is located at the C-2 hydroxyl of arabinose. In the HMBC spectrum of 1, there were three characteristic cross-peaks between C-3 of the aglycone and the anomeric proton of arabinose, between the C-2 of arabinose and the anomeric proton of rhamnose, and between the quaternary carbon C-28 and the anomeric proton of glucose. Thus, compound 1 was identified as 3-O-[α-L-rhamnopyranosyl (1-2)]- α -L-arabinopyranosyl pomolic acid 28-O- β -Dglucopyranoside, and named as latifoloside A.

gave an aglycone (7), which was identified as pomolic

acid by comparison of its spectral properties with ref-

erence data [6]. The ¹H, ¹³C NMR spectra of 1 indi-

^{*} Author to whom correspondence should be addressed.

Fig. 1.

Compound 2 gave a quasi-molecular ion peak [M-H]⁻ at m/z 911, which corresponded to $C_{47}H_{76}O_{17}$ in the negative FAB-mass spectrum and ¹³C NMR (DEPT). Cellulase treatment of 2 provided an aglycone (10), which was identical with siaresinolic acid,

and a mixture of L-arabinose, D-glucose and L-rhamnose (1:1:1) as determined by HPLC analysis. The IR spectrum of 2 showed a similar absorption pattern to that of latifoloside A. Comparison of the ¹³C NMR data for 2 with that of the sugar moiety of 1 showed

that 2 and 1 were the same sugar chain. Thus, B(2) was concluded to the 3-O-[α -L-rhamnopyranosyl(1-2)]- α -L-arabinopyranosyl siaresinolic acid 28-O- β -D-glucopyranoside (13 C NMR Table 1).

The element composition of compound 3 was proved to be C₅₃H₈₆O₂₂ by the negative FAB-mass spectrum and ¹³C NMR (DEPT). Treatment of 3 with cellulase afforded siaresinolic acid by comparing the ¹³C NMR data, and a mixture of L-arabinose, D-glucose and L-rhamnose (1:2:1). Comparison of the NMR data for 3 with that for 6 showed that 3 contained the same oligosaccharic sequence as 6 (Table 1). Consequently, latifoloside C (3) was identified as 3-O-[α-L-rhamnopyranosyl(1-2)]-[β-D-glucopyranosyl(1-3)]-α-L-arabinopyranosyl siaresinolic acid 28-O-β-D-glucopyranoside.

The negative fast atom bombardment mass spectrum of compound 4 gave a quasi-molecular ion peak m/z 911 [M-1]⁺, corresponding to $C_{47}H_{76}O_{17}$. Cellulase treatment of 4 furnished an aglycone (11) and a mixture of L-arabinose, D-glucose and L-rhamnose (1:1:1) by HPLC. In the ¹H NMR spectrum of 4 and 11, one methyl δ 0.98 (3H, d, J = 6.6 Hz) was observed, which showed that the methyl connected with the carbon of methine. The NOESY experiment on the aglycone (11) exhibited two characteristic cross peaks between the signal assignable to the methine proton at C-18 (δ 3.16) and the signal assignable to the methyl proton at C-29 (δ 1.40), and the signal of the methyl proton at C-30 (δ 0.96). Comparison of the NMR data for 11 with that for 7 showed that resonances of the D- and E-ring carbons were significantly different (Table 1). The chemical shifts (7 vs 11) of signals due to C-18 (-7.2 ppm), C-22 (-4.2ppm) and C-29 (+2.7 ppm) revealed the C-30 methyl group to be β (axial) in place of the α -(equatorial) methyl in 7. Hence, the aglycone of 4 was formulated as $30(s)-3\beta$, 19α -dihydroxyurs-12-en-28-oic acid, which has the same aglycone as the triterpene ilexgenin B [6, 8]. As latifolosides A, B, D were isomers, latifoloside D (4) was formulated as 3-O- $[\alpha-L$ -rhamnopyranosyl (1-2)]- α -L-arabinopyranosyl ilexgenin B 28-O- β -glucopyranoside.

By the same deduction, compounds 3, 5, 6 were isomers and 5 and 4 have the same aglycone (ilexgenin B). Thus, 5 was determined to be $3-O-[\alpha-L-rhamno-pyranosyl(1-2)]-[\beta-D-glucopyranosyl(1-3)-]-\alpha-L-arabinopyranosyl ilexgenin B <math>28-O-\beta-D-glucopyranoside$ (^{13}C NMR Table 1).

EXPERIMENTAL

All mps were determined on a Beijing Micromelting apparatus and are uncorr. IR were run with a Perkin–Elmer 683 spectrometer. 1 H, 13 C NMR and 2D NMR were measured by Bruker AM 400 Hz spectromers with pyridine- d_5 as a solvent and TMS as an int. standard. FAB-MS were taken on a ZAM-HB system spectrometer. CC and TLC were performed on silica

Table 1. ¹³C NMR spectral data for latifoloside A-E (400 MHz, pyridine-d₅)

		WITIZ,	pyriaine	-u ₅) 		
Carbon	Α	(6)	В	С	D	E
1	39.1	39.3	39.0	39.0	39.2	39.3
2	26.8	26.8	26.7	26.7	27.0	27.0
3	89.1	88.4	89.2	88.3	89.1	88.4
4	39.6	39.7	39.7	39.7	39.7	39.8
5	56.1	56.2	56.2	56.2	56.2	56.2
6	18.9	18.8	18.9	18.8	18.9	18.9
7	33.6	33.6	33.1	33.1	32.1	32.0
8	40.7	40.7	40.2	40.3	40.6	40.6
9	47.9	47.9	48.4	48.4	48.0	48.0
10	37.1	37.1	37.3	37.3	37.2	37.2
11	24.2	24.2	24.2	24.2	24.2	24.2
12	128.5	128.6	123.7	123.6	127.8	127.8
13	139.4	139.4	144.4	144.4	139.0	138.9
14	42.2	42.2	42.2	42.2	42.3	42.3
15	29.4	29.4	29.2	29.2	29.4	29.4
16	26.2	26.3	28.1	28.1	24.9	24.8
17	48.8	48.8	46.6	46.6	48.6	48.5
18	54.6	54.6	44.7	44.7	47.4	47.4
19	72.8	72.8	81.1	81.1	73.6	73.6
20	42.2	42.2	35.7	35.7	42.9	42.9
21	26.7	26.8	29.0	29.0	26.8	26.8
22	37.9	37.9	33.3	33.3	33.7	33.6
23	28.3	28.3	28.2	28.2	28.3	28.3
24	16.8	16.9	17.0	17.1	17.2	17.2
25	15.8	15.9	15.7	15.7	15.9	15.9
26	17.5	17.5	17.6	17.7	17.7	17.6
27	24.7	24.7	24.8	24.2	242.5	24.5
28	177.2	177.3	177.5	177.5	177.5	177.3
29	27.2	27.2	28.7	28.8	29.9	29.8
30	17.1	17.2	25.0	25.0	16.3	16.2
Sugar						
3- <i>O</i> -Ara		4040	1040	1010	1010	1040
1	104.8	104.8	104.9	104.9	104.9	104.8
2	76.1	74.7	76.1	74.7	76.2	74.7 82.1
3	74.2	82.1	74.2	82.3	74.2	68.2
4	68.6	68.2	68.7	68.3 64.6	68.8	64.8
5 D1	64.5	64.9	64.7	04.0	64.7	04.0
Rha	101.0	102.0	101.9	102.0	101.9	102.0
1	101.9	102.0		102.0		72.5
2	72.7 72.8	72.5 72.6	72.4 72.6	72.6 72.5	72.5 72.5	72.6
3 4	74.1	74.2	74.1	74.2	74.1	74.2
5	70.0	70.2	70.1	70.1	70.1	70.2
6	18.7	18.7	18.7	18.7	18.8	18.7
Glc	10.7	10.7	10.7	10.7	10.0	10.7
l l		104.7		104.7		104.7
2		75.1		75.1		75.1
3		78.3		78.3		78.3
4		71.3		71.5		71.6
5		78.8		78.6		78.7
6		62.6		62.6		62.6
28- <i>O</i> -Glc		02.0		54.0		J
1	96.0	96.0	96.0	96.0	96.0	96.0
2	73.7	74.0	73.7	74.0	73.8	74.0
3	79.0	79.0	79.0	79.0	79.0	79.0
3 4	79.0	71.5	71.3	71.2	71.2	71.2
5	79.3	79.4	79.3	79.4	79.4	79.4
6	62.5	62.4	62.2	62.2	62.3	62.3
·	32.3	32.4				

gel, RP-8 and RP-18 using the following solvent systems: a) CHCl₃-MeOH-H₂O (7:3:0.5), CHCl₃-MeOH-H₂O (65:35:9) and MeOH-H₂O (6:4-7:3) for saponins; b) CHCl₃-MeOH-H₂O (7:3:1) lower-layer 9 ml+1 ml HOAc for sugars. Detection: for saponins, spraying with 5% H₂SO₄ following by heating for 5 min at 105°, for sugar, aniline-phthalate reagent.

Plant material. Ilex latifolia plants were collected in the Hunan Province of China in the Summer of 1993 and identified by Prof. Chong-Ren Yang. A voucher specimen is deposited in the Herbarium of Kunming Institute of Botany, Chinese Academy of Science.

Extraction and isolation of saponins. The dry leaves (800 g) were extracted x 3 with MeOH at 50° for 8 hr, and the solvent was removed under red. pres. The combined extract (100 g) was suspended in H₂O and the aq. suspension was extracted with CHCl₃ and *n*-BuOH, respectively. The *n*-BuOH layer was evapd to dryness to give a residue (50 g) which was chromatographed on silica gel (1.5 kg, 200–300 mesh) with CHCl₃-MeOH-H₂O (7:3:0.5) to give 20 fr. Fr. 10 and fr. 14 were sepd on HPLC [ODS, eluting with MeOH-H₂O (8:2-6:4). Flow rate: 5 ml min⁻¹; Injection: 0.4 ml (10 mg ml⁻¹)] to afford latifoloside A (1, 80 mg), B (2, 50 mg), C (3, 75 mg), D (4, 80 mg), E (5, 140 mg) and F (6, 125 mg).

Latifoloside A (1). Colourless powder, mp (207–210°, $C_{47}H_{76}O_{17}$ IR v_{max}^{KBr} cm⁻¹: 3433 (OH), 2933 (C-H), 1734 (C=O), 1647 (C=C), 1458, 1386, 1074, 1028. FAB-MS m/z: 911 [M-H]⁻, 749 [M-H-162]⁻, 603 [M-H-162-146]⁻, 471 [M-H-162-146-132]⁻, 453 [M-H-162-146-132-H₂O]⁻. ¹H NMR: δ 0.89 (s, 3-H), 1.05 (d, J = 6.4 Hz, 3H), 1.07 (s, 3H), 1.39 (s, 3H), 1.75 (s, 3H), 2.92 (1H, br s, 18β-H), 3.23 (1H, dd, J = 11.3, 4.2 Hz, 3α-H), 5.55 (1H, br s, 12-H), 4.88 (1H, d, J = 5.2 Hz, C-1-H of Ara), 6.11 (1H, br s, C-1-H of Rha), 1.61 (3H, d, J = 6.2 Hz, C-6-H of Rha), 6.28 (1H, d, J = 8.0 Hz, C-1-H of Glc). ¹³C NMR data see Table 1.

Latifoloside B (2). Colourless powder, mp 225–228°, $C_{47}H_{76}O_{17}$; IR v_{max}^{KBr} cm⁻¹: 3404 (OH), 2935 (C-H), 1730 (C=O), 1643 (C=C), 1450, 1380, 1072, 1030. FAB-MS m/z: 911 [M-H]⁻, 749 [M-H-162]⁻, 603 [M-H-164-146]⁻, 453 [M-H-162-146-132]⁻¹. ¹H NMR: δ 0.87 (s, 3H), 0.97 (s, 3H), 1.10 (s, 3H), 1.12 (s, 3H), 1.14 (s, 3H), 1.19 (s, 3H), 1.64 (s, 3H), 3.29 (dd, J = 11.2, 4.5 Hz, 3 α -H), 5.50 (br s, 12-H), 3.54 (t-like), 3.50 (br s, 18 α -H), 4.86 (1H, d, J = 5.5 Hz, C-1-H of Ara), 6.11 (1H, br s, C-1-H of Rha), 1.61 (3H, d, d) = 6.2 Hz, C-6-H of Rha), 6.30 (1H, d, d) = 7.9 Hz, C-1-H of Glc); ¹³C NMR data see Table 1.

Latifoloside C (3). Colourless powder, mp 231–234°, $C_{53}H_{86}O_{22}$, IR v_{max}^{KBr} cm⁻¹: 3420 (OH), 2930 (C-H), 1731 (C=O), 1637 (C=C), 1454, 1386, 1070, 1026. FAB-MS m/z: 1073 [M-1]⁻, 911 [M-1-162]⁻, 749 [M-1-2-162]⁻, 765 [M-1-162-146]⁻, 603 [M-1-146-2×162]⁻, 453 [M-1-146-2×162-132-H₂O]⁻. ¹H NMR: δ 0.87 (s, 3H), 0.97 (s, 3H),

1.10 (s, 3H), 1.12 (s, 3H), 1.14 (s, 3H), 1.19 (s, 3H), 1.64 (s, 3H), 3.30 (1H, dd, J = 11.3, 4.3 Hz, 3 α -H), 3.51 (br s, 18 α -H), 5.50 (br s, 12-H), 4.86 (1H, d, J = 5.7 Hz, C-1-H of Ara), 5.09 (1H, d, J = 8.1 Hz, C-1-H of Glc), 6.16 (1H, br s, C-1-H of Rha), 1.62 (d, J = 6.8 Hz, C-6-H of Rha), 6.36 (1H, d, J = 7.5 Hz, C-1-H of Glc); ¹³C NMR data see Table 1.

Latifoloside D (4). Colourless powder, mp 212–215°, $C_{47}H_{76}O_{17}$, IR ν_{max}^{KBr} cm⁻¹: 3420 (OH), 2930 (C-H), 1733 (C=O), 1641 (C=C), 1450, 1387, 1074, 1026. FAB-MS m/z: 911 [M-1]⁻, 749 [M-1-162]⁻, 603 [M-1-162-146]⁻, 471 [M-1-162-146-132]⁻, 453 [M-1-162-146-132-H₂O]⁻. ¹H NMR: δ 0.88 (s, 3H), 0.96 (dd, J=6.6 Hz, 3H), 1.06 (s, 3H), 1.13 (s, 3H), 1.16 (s, 3H), 1.40 (s, 3H), 1.71 (s, 3H), 3.18 (dd, J=11.4, 4.4 Hz, 3α-H), 3.16 (s, 18α-H), 5.49 (br s, 12-H), 5.27 (s, 19β-OH), 4.88 (1H, J=5.1 Hz, C-1-H of Ara), 6.30 (1H, d, J=7.9 Hz, C-1-H of Glc), 6.11 (1H, br s, C-1-H of Rha), 1.60 (3H, d, J=5.7 Hz, C-6-H of Rha). ¹³C NMR data: see Table 1.

Latifoloside E (5). Colourless powder, mp 228–230°, C₅₃H₈₆O₂₂, IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3430 (OH), 2932 (C-H), 1730 (C=O), 1454, 1388, 1072. FAB-MS m/z: 1073 [M-1]⁻, 911 [M-1-162]⁻, 765 [M-1-162-146]⁻, 749 [M-1-2×162]⁻, 603 [M-1-146-2×162]⁻, 471 [M-1-146-2×162-132]⁻, 453 [M-1-146-2×162-132-H₂O]⁻. ¹H NMR: δ 0.87 (s, 3H), 0.96 (dd, J = 7.0 Hz, 3H), 1.12 (s, 3H), 1.16 (s, 3H), 1.17 (s, 3H), 1.39 (s, 3H), 1.71 (s, 3H), 3.27 (dd, J = 11.2 Hz, 4.3 Hz, 3a-H), 3.16 (s, 18α-H), 5.22 (s, 19β-OH), 5.50 (br s, 12-H), 4.85 (1H, d, J = 5.3 Hz, C-1-H of Ara), 5.08 (1H, d, J = 7.6 Hz, C-1-H of Glc), 6.13 (1H, dr s, C-1-H of Rha), 1.61 (3H, dr, dr = 6.1 Hz, C-6-H of Rha), 6.31 (1H, dr, dr = 8.0 Hz, C-1-H of Glc). ¹³C NMR data: see Table 1.

Compound (6). Colourless powder, mp 228–230°, $C_{53}H_{86}O_{22}$, $IR v_{ma}^{KBr} cm^{-1}$: 3428 (OH), 2932 (C-H), 1734 (C=O), 1638 (C=C), 1454, 1389, 1073, 1026. FAB-MS m/z: 1073 $[M-1]^-$, 911 $[M-1-162]^-$, 765 $[M-1-162-146]^-$, 749 $[M-1-2\times162]^-$, 603 $[M-1-146-2\times162]^-$, 453 $[M-1-146-2\times162]^-$, 603 [M-1] NMR: δ 0.87 (s, 3H), 1.06 (d, J=6.4] Hz, 3H), 1.12 (s, 3H), 1.16 (s, 3H), 1.18 (s, 3H), 1.39 (s, 3H), 1.70 (s, 3H), 1.61 (3H, d, J=6.0] Hz, C-6-H of Rha), 4.85 (1H, d, J=5.4] Hz, C-1-H of Ara), 3.27 (1H, dd, J=11.4, 4.5] Hz, 3 α -H), 5.54 (1H, br, s, 12-H), 5.08 (1H, d, J=7.7] Hz, C-1-H of Glc), 6.14 (1H, br, s, C-1-H) of Rha), 6.28 (1H, d, J=8.0] Hz, C-1-H of Glc).

Alkaline hydrolysis of latifoloside A (1). LiOH (6 mg) was added to a soln of latifoloside A (1, 24 mg) in H₂O (3.0 ml). The reaction mixt. was heated with stirring at 40° for 10 hr, then cooled to ambient temp. and the solvent was removed on a rotary evaporator to give 8 (18 mg). Compound 8 was purified by CC (silica gel, 3 g, CH₂Cl₂–MeOH, 3:1) to afford a hydrolysate (15 mg) and D-glucose.

Acid hydrolysis of latifoloside A-E. A soln of each compound (10 mg) in 5% H₂SO₄ in 50% EtOH was

heated at 100° for 10 hr. The reaction mixt. was diluted with H₂O, neutralized with 2% NaOH and evapd *in vacuo* to dryness. The mole ratio and D/L of each sugar was determined using RI detection (Waters 410) and chiral detection (Shodex OR-1), respectively, in HPLC (Shodex RS pak DC-613, MeCN-H₂O, 3:1, 1 ml min⁻¹, 70°) by comparison with authentic sugars (10 mM each of L-Ara, D-Glc and L-Rha). Each sugar gave a peak as follows: L-Ara, 6.0 min; D-Glc, 7.4 min and L-Rha, 4.8 min.

Enzymatic hydrolysis of latifoloside A (1). Latifoloside A (1) (35 mg) was taken in EtOH–H₂O (1:9) and 0.01 M NaH₂PO₄ buffer (pH 4.0), 5 ml of each, incubated with crude cellulase (50 mg, Sigma) for two weeks at 37° and worked-up as usual. The crude genin was chromatographed on a silica gel column with CHCl₃–MeOH–H₂O (250:40:1) giving pomolic acid (7, 12 mg) which was identical with an authentic sample of pomolic acid based on comparison of ¹H and ¹³C NMR data.

Enzymatic hydrolysis of latifoloside B (2) and C (3). Enzymatic hydrolysis of latifoloside B (2) (25 mg) or C (3) (30 mg) was carried out in the same way as for 1 to give siaresindic acid (9, 8 mg), which had the same NMR data as an authentic sample.

Enzymatic hydrolysis of latifoloside D (4) and E (5). Latifoloside D (4) (20 mg), or E (5) (30 mg), was hydrolysed in the same way as for 1 to give ilexgenin

B (10, 8 mg), which had the same NMR data as an authentic sample.

Acknowledgements—The authors are grateful to the staff of the Laboratory of Kunming Institute of Botany, Chinese Academy of Sciences for financial support and their valuable advice.

REFERENCES

- He, Z. D., Ueda, S., Akaji, M. and Yang, C. R., Phytochemistry, 1994, 36, 709.
- Ahai, R. K., in *Iconographia Cormophytorum Sini*coru. Beijing Science, Beijing, 1985, p. 641.
- Jang Su Medical College, in Chinese Medicine Dictionary. Shanghai People Publishing House, Shanghai, 1975, p. 1288.
- Ouyang, M. A., Yang, C. R., Chen, Z. L. and Wang, H. Q., *Phytochemistry*, 1996, 41, 871.
- 5. Ouyang, M. A., Wang, H. Q., Chen, Z. L. and Yang, C. R., *Phytochemistry*, 1996, **43**, 443.
- Kakuno, T., Yoshikawa, K., Arihara, S., *Phyto-chemistry*, 1992, 31, 3553.
- Yoshioks, I., Sugawara, T., Ohsuka, A. and Kita-gawa, I., Chemical and Pharmaceutical Bulletin, 1971, 19, 1700.
- 8. Hidka, K., Iyo, M. and Matsuda, Y., Phytochemistry, 1987, 26, 2033.