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CYCLOARTANE TRITERPENOIDS FROM AGLAIA HARMSIANA

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Abstract—Two new and two known cycloartane-type triterpenoids were isolated from the leaves of *Aglaia harmsiana*. The structures were determined using ¹H, ¹³C and 2D NMR techniques. © 1997 Elsevier Science Ltd

INTRODUCTION

We recently reported on the isolation of two cycloartane-type triterpenoids (1,2) from the leaves of *Aglaia* harmsiana Perkins [1]. We have now obtained additional two cycloartane-type triterpenoids (3, 4)together with cycloartenol and (24R)-cycloartane- 3β ,24,25-triol (5) [2] from the same source. This paper deals with the structural elucidations of these compounds.

RESULTS AND DISCUSSION

After repeated column chromatography and HPLC separations of the EtOAc-soluble part of an EtOH extract, two cycloartane-type triterpenoids (3, 4), were obtained together with compounds 1 and 2 [1], cycloartenol, and (24R)-cycloartane- 3β ,24,25-triol (5) [2]. Compound 3 was assigned to the molecular formula C₃₀H₅₂O₄ (HR EI-MS) and had IR absorptions at 3450 cm⁻¹ due to hydroxyl group. The ¹H NMR spectrum of 3, analysed with the aid of COSY and NOESY, showed the presence of five tertiary methyls, a secondary methyl (δ 1.02, d, J = 6.5 Hz), and doublets at δ 0.37 and 0.62 (1H each, J = 4.0 Hz), characteristic of non-equivalent protons of a cyclopropyl methylene group. In addition, signals due to two methine protons geminal to a hydroxyl (δ 3.73, dd, J = 10.3, 1.4 Hz and 4.31, dd, J = 11.7, 4.4 Hz) and methylene protons geminal to a hydroxyl (δ 3.79 and 4.24, 2H, ABq, J = 10.5 Hz) were observed. The ¹³C NMR spectrum showed 30 carbons. The multiplicities of each carbon were determined by CH-COSY experi-

Compound 4 was assigned to the molecular formula $C_{30}H_{52}O_3$ (HR EI-MS) and had IR absorptions at 3450 cm⁻¹ due to hydroxyl group. The ¹H NMR spectrum analysed with the aid of 2D NMR techniques, showed the presence of six tertiary methyls, a secondary methyl (δ 0.89, d, J = 6.6 Hz), and a cyclopropyl methylene group (δ 0.35 and 0.52, 1H each, d, J = 3.9 Hz). In addition, signals due to two methine protons geminal to a hydroxyl (δ 3.29, dd, J = 9.8, 1.7 Hz and 3.47, t, J = 2.6 Hz) were present. The ¹³C NMR

ments, which revealed the presence of six methyls, 12 methylenes, six methines, and six quaternary carbon atoms. In the EI-mass spectrum, compound 3 exhibited important and prominent fragments at m/z 331, 320, 175, and 95, which are characteristic fragmentation ions of 9,19-cycloartane-type triterpenoids with two hydroxyls in the side-chain moiety and two hydroxyls in ring A [1-4]. On the basis of the precise ¹H and ¹³C NMR analyses, compound 3 was deduced to be a cycloartane-type triterpenoid with a 24,25glycol side-chain and either a 3β ,28-diol or a 3β ,29diol system in A ring. This structure was also supported from the HMBC experiments. Thus, compound 3 showed prominent cross-peaks between H- 3α (δ 4.31) and a hydroxymethylene carbon (δ 67.5) and between H-24 α (δ 3.73) and two methyl carbons (δ 26.0 and 26.1). Confirmation for the 3 β ,28-diol arrangement came from the following evidence. The carbon and proton chemical shifts of rings A, B, C, and D in 3 are close agreements with those of 1 [1]. While the carbon and proton chemical shifts of the side-chain moiety in 3 are essentially the same as compounds 2 [1] or 5 [2]. Further comparisons of the ¹H and ¹³C NMR spectral data of 3 and other analogous cycloartane-type triterpenoids [5-7] confirmed the structure of 3 as (24R)-cycloartane-3 β ,24,25,28-tetrol.

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Table 1. 13C NMR spectral data of compounds 3-5

C	3*	4 †	5 †	C	3*	4†	5†
1	32.4	27.5	32.0	16	26.7	26.3	26.5
2	31.0	28.8	30.4	17	52.9	52.3	52.4
3	73.9	77.1	78.9	18	18.5	18.0	18.1
4	44.8	39.6	40.5	19	30.1	29.8	29.9
5	41.6	41.1	47.2	20	37.0	36.4	36.4
6	21.3	21.1	21.1	21	18.9	18.5	18.5
7	28.5	28.1	28.2	22	34.5	33.6	33.6
8	48.3	48.1	48.0	23	29.4	28.6	28.8
9	19.9	19.9	20.0	24	80.0	79.7	79.7
10	26.1	26.5	26.1	25	72.8	73.2	73.2
11	26.2	26.3	26.0	26	26.1ª	23.3	23.3
12	33.3	33.6	33.0	27	$26.0^{\rm a}$	26.6	26.6
13	45.6	45.3	45.3	28	67.5	25.9	25.5
14	49.1	48.9	48.8	29	11.5	21.3	14.0
15	35.9	35.5	35.6	30	19.6	19.3	19.4

^{*} Measured in C₅D₅N.

spectrum showed 30 carbons consisting of seven methyls, 11 methylenes, six methines, and six quaternary carbon atoms. In the EI-mass spectrum, compound 4 showed important and prominent fragments at m/z 315, 297, 203, 175 and 95, which are characteristic fragmentation ions of 9,19-cycloartane-type triterpenoids with two hydroxyls in the side-chain moiety and one hydroxyl in ring A [1-4]. The above spectral data are very similar to those of 5 [2]. However, in the ¹H NMR of 5, the 3α -axial H appeared at δ 3.28 as a double doublet (J = 10.1, 4.4Hz) while in 4, the proton signal due to H-3 appeared at δ 3.47 as a triplet (J = 2.6 Hz). This result indicates the presence of a 3α -axial hydroxyl in 4 [8, 9]. Furthermore, in the ¹³C NMR spectrum of 4, the expected shielding and deshielding effects of the 3α-axial hydroxyl on ring A carbons were present [8-10]. On the basis of the above evidence, the structure of 4 was concluded to be (24R)-cycloartane- 3α , 24, 25-triol.

EXPERIMENTAL

General. Mps: uncorr., ¹H NMR: 400 MHz; ¹³C NMR: 100 MHz and TMS as int. standard; IR: CHCl₃; HPLC: JAIODS-120T column with a differential refractometer.

Plant materials. The leaves of Aglaia harmsiana Perkins were harvested in 1993 at the Herbarium Bogoriense, Java, Indonesia and voucher specimens have been deposited at the Herbarium of the Faculty of Pharmaceutical Sciences, Setsunan University.

Extraction and isolation. The crushed leaves (700 g) were extracted with 95% EtOH and the solvent evapd. The EtOH extract (30.2 g) was suspended with H₂O and the aq. suspension extracted with EtOAc. The EtOAc extract (25.0 g) was chromatographed on silica gel with hexane–EtOAc containing increasing amounts of EtOAc and a fr. containing 1, 4, and 5 (1.0 g), and a fr. containing 3 (0.3 g) were sepd in that

[†] Measured in CDCl3.

^a Assignments may be interchangeable in each vertical column.

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order. Each fr. was further sepd by repeated HPLC to afford 1 [1] (150 mg), 3 (91 mg), 4 (28 mg), and 5 [2] (570 mg).

(24R)-*Cycloartane*-3 β ,24,25,28-*tetrol* (3). Amorphous powder, [α]₂^{D5} +22.8° (MeOH; c 0.50); IR v_{max} cm⁻¹: 3450, 2960, 1260, 1030; EI- and HR EI-MS m/z (rel. int.): 476.3863 (M⁺, C₃₀H₅₂O₄ requires 476.3865, 2), 458 (9), 443 (10), 440 (8), 425 (10), 331 (8), 320 (17), 175 (51), 95 (93), 59 (100); ¹H NMR (C₅D₅N): δ 0.37 (1H, d, J = 4.0 Hz) and 0.62 (1H, d, J = 4.0 Hz) (H₂-19), 0.89 (3H, s, Me-30), 1.02 (3H, d, J = 6.5 Hz, Me-21), 1.04 (3H, s, Me-18), 1.17 (3H, s, Me-29), 1.53, 1.56 (3H each, s, Me-26 and Me-27), 3.73 (1H, dd, J = 10.3, 1.4 Hz, H-24 α), 3.79 and 4.24 (1H each, ABq, J = 10.5 Hz, H₂-28), 4.31 (1H, dd, J = 11.7, 4.4 Hz, H-3 α); ¹³C NMR: Table 1.

(24R)-Cycloartane- 3α ,24,25-triol (4). Mp 201–202° (MeOH); $[\alpha]_D^{20}+35.9^\circ$ (MeOH; c 0.33); IR v_{max} cm⁻¹: 3450, 2940, 1280; EI- and HR EI-MS m/z (rel. int.): 460.3910 (M⁺, $C_{30}H_{52}O_3$ requires 460.3916, 10), 442 (18), 427 (22), 409 (23), 315 (24), 297 (28), 203 (44), 175 (91), 95 (100); ¹H NMR (CDCl₃): δ 0.35 (1H, d, J = 3.9 Hz) and 0.52 (1H, d, J = 3.9 Hz) (H₂-19), 0.88 (3H, s, Me-29), 0.89 (3H, d, d = 6.6 Hz, Me-21), 0.90, (3H, d = 8, Me-30), 0.95 (3H, d = 8, Me-26 and Me-27), 3.29

(1H, dd, J = 9.8, 1.7 Hz, H-24 α), 3.47 (1H, t, J = 2.5 Hz, H-3 β); ¹³C NMR: Table 1.

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REFERENCES

- Inada, A., Murata, H., Inatomi, Y., Nakanishi, T. and Darnaedi, D., Journal of Natural Products, 1995, 58, 1143.
- 2. Greca, M. D., Fiorentino, A., Monaco, P. and Previtera, L., *Phytochemistry*, 1994, **35**, 1013.
- Balakrishna, K. and Kundu, A. B., Journal of Natural Products, 1990, 53, 523.
- Furlan, M., Roque, N. F. and Filho, W. W., *Phytochemistry*, 1993, 32, 1519.
- Anjaneyulu, V., Prasad, K. H., Ravi, K. and Connolly, J. D., *Phytochemistry*, 1985, 24, 2359.
- Vishnoi, S. P., Shoeb, A. and Kapil, R. S., Planta Medica, 1988, 54, 40.
- Nyemba, A. M., Mpondo, T. N., Connolly, J. D. and Rycroft, D. S., *Phytochemistry*, 1990, 29, 994.
- 8. Januario, A. H., Da Silva, M. F. G. F., Vieira, P. C. and Fernandes, J. B., *Phytochemistry*, 1992, 31, 1251.
- Achenbach, H. and Frey, D., *Phytochemistry*, 1992, 31, 4263.
- Chen, T. K., Ales, D. C., Baenziger, N. C. and Wiemer, D. F., Journal of Organic Chemistry, 1983, 48, 3525.