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FOUR MONOTERPENE ALKALOID DERIVATIVES FROM INCARVILLEA SINENSIS

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Key Word Index—*Incarvillea sinensis*; Bignoniaceae; incarvine D; methoxyincarvillateine; incarvillateine *N*-oxide; incarvine A *N*-oxide; anti-rheumatism.

Abstract—Two new derivatives of monoterpene alkaloids, named incarvine D and methoxycarvillateine, and two novel N-oxides, incarvillateine N-oxide and incarvine A N-oxide were isolated from the aerial parts of *Incarvillea sinensis* LAM. On the basis of chemical and spectroscopic evidence, incarvine D was characterized as the ester of a monoterpene alkaloid, incarvilline, and a monoterpene, tetrahydro-alcohol compound corresponding to Hildebrandt's acid; methoxycarvillateine was established as the methoxy derivative of incarvillateine, and the two oxides were determined to be N-oxide derivatives of incarvillateine and incarvine A, respectively. © 1997 Elsevier Science Ltd

INTRODUCTION

Incarvillea sinensis LAM., one of the well-known Chinese crude drugs, has been used to treat rheumatism and to relieve pain. In studies on its pharmacological active substance(s), we have isolated and characterized five new alkaloids, incarvilline (5), incarvillateine (6) and incarvine A-C [1-4]. The present paper describes the isolation and structure elucidation of two new alkaloids, incarvine D (1) and methoxyincarvillateine (2), and two novel N-oxides, incarvillateine N-oxide (3) and incarvine A N-oxide (4).

RESULTS AND DISCUSSION

Incarvine D (1) showed peaks at m/z (rel. int.) 351 [M]⁺ (66), 183 [incarvilline]⁺ (70), 182 [incarvilline-H]⁺ (100) and 166 [incarvilline-OH]⁺ (100) in its mass spectrum. Its ¹³C NMR signals at δ 57.5, 57.7, 30.6, 37.7, 29.9, 76.0, 41.0, 46.1, 46.2, 17.3 and 14.9 could be assigned, respectively, to C-1, C-3–9, *N*-Me, Me-4 and Me-8 of the monoterpene alkaloid moiety, incarvilline (5). The structure of (5) was previously determined by X-ray analysis [1], and whose absolute con-

figuration was determined recently [5]. The proton signals were also attributed as follows: δ 0.86 (3H, d, J = 6.7 Hz, 4-Me), 0.96 (3H, d, J = 7.3 Hz, 8-Me), 1.55–1.64 (2H, m, 1-Ha, 6-Ha), 1.68 (1H, t, J = 11.6Hz, 3-Ha), 1.91-2.22 (4H, m, 6-Hb, 8, 9, 4-H), 2.23 (3H, s, N-Me), 2.39 (1H, m, 5-H), 2.50 (1H, m, 3-Hb),2.66 (1H, m, 1-Hb), 5.28 (1H, m, 7-H). The remaining carbon signals at δ 12.3 (q), 19.4 (q), 26.2 (t), 29.4 (d), 35.8 (t), 39.7 (t), 60.8 (t), 127.8 (s), 142.2 (d) and 167.8 (s) established the presence of a monoterpene moiety whose ¹H-¹³C COSY led to the assignment of one methyl at δ 0.94 (3H, d, J = 6.7 Hz); one olefinic methyl at 1.83 (3H, s); one olefinic proton at 6.73 (1H, t, J = 7.3 Hz); six methylene protons at 1.30 (1H, m), 1.40 (1H, m), 1.44 (1H, m), 1.55–1.64 (2H, m), 1.91– 2.22 (2H, m) and two oxygenated methylene protons at δ 3.66 (2H, m). The monoterpene moiety was characterized as a tetrahydro-alcohol compound corresponding to Hildebrandt's acid [6], by comparing the chemical shifts with those of incarvine A (7) [3]. The bond between incarvilline (5) and the monoterpene moiety was elucidated by the HMBC spectrum, in which a cross peak between C-1' of the monoterpene moiety and H-7 of the incarvilline (5) moiety was observed. The chemical shift of the signal at $\delta_{\rm H}$ 5.28 (1H, m), assignable to H-7 of 5, also established that the 7-hydroxyl group was acylated. Consequently, the structure of incarvine D (1) has been established as shown in the formula.

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Methoxyincarvillateine (2) showed peaks at m/z(rel. int.) 748 [M]+ (13), 182 (73) and 166 (54) in its EI mass spectrum. The ¹³C NMR signals (Table 1) suggested the presence of two molecules of monoterpene alkaloid moiety, incarvilline (5) [1]. The proton signals also supported the occurrence of the two moles of the incarvilline moiety (Table 2). The remaining carbon signals (Table 1) disclosed the presence of a 3-methoxy-4-hydroxy-benzyl moiety and a 3,5dimethoxy-4-hydroxybenzyl moiety. The signals at δ 41.2, 42.2, 47.9 and 48.0 indicated the presence of a four-membered ring which was confirmed by ¹H-¹H and ¹H-¹³C COSY. Consequently, the structure of 2 has been established as shown in the formula, named methoxyincarvillateine. Its structure was also verified by the aid of ¹H-¹H, ¹H-¹³C COSY and HMBC, and by comparison with incarvillateine (6), whose structure was previously determined by X-ray analysis [2].

Incarvillateine N-oxide (3) seemed as a pure substance according to TLC, UV and FABMS; however, the multiplicity of the 'H and ¹³C NMR spectral data (Table 3) showed it to be a mixture of two N-oxides. Attempts to separate them by HPLC under various conditions were unsuccessful. The positive FAB-mass spectrum of 3 showed an $[M+1]^+$ peak at m/z 735, and intense fragments due to half molecular moieties at m/z 360 and 376, as well as the fragment ion $[M-16]^+$ at m/z 718. These were typical for N-oxides [7]. All other prominent signals were similar to those found in the spectrum of incarvillateine (6) [2].

On comparison of the ¹H and ¹³C NMR spectra of 3 with those of 6 (Table 3), noticeable shifts were observed not only for two of the *N*-Me groups, but also for their neighbouring methylene groups, whereas the other signals were similar to those of 6. The *N*-Me $\delta_{\rm H}$ signals were shifted downfield to δ 3.09 (+0.92 ppm), while their neighbouring methylene groups were also shifted downfield; however, their chemical

2 R=OMe

6 R≕H

shifts could not be assigned accurately due to overlapping of the signals. The 13 C NMR signals of the two N-Me groups were shifted to δ 61.44 (+15.44 ppm) (av.), C-1, -1' to $\delta_{\rm C}$ 66.52 (+9.47 ppm) (av.), and C-3, -3' to $\delta_{\rm C}$ 67.66 (+10.31 ppm) (av.), respectively. These shifts were indicative of the influence of the N-O dipole.

Oxidation of 3 and 6, respectively, with m-chloroperbenzoic acid, yield incarvillateine N,N'-dioxide (8), whose positive FAB mass spectrum showed $[M+1]^+$ at m/z 751, and $[M-O+1]^+$ at m/z 735, suggesting that both N and N' were oxidized. The ${}^{1}H$ and ¹³C NMR spectra of 8 were similar to those of 6 except that all of the N-Me signals were shifted downfield to δ 3.13 (+0.96 ppm), 1-, 1'-Ha to δ 2.78 (+1.36 ppm), 1-, 1'-Hb to δ 3.10 (+0.55 ppm), 3-, 3'-Ha to δ 2.94 (+1.36 ppm) and 3-, 3'-Hb to δ 2.94 (+0.49 ppm) and in the ¹H NMR spectrum, and all of the N-Me signals were shifted downfield to δ 6.15 (+15.5 ppm), C-1, -1' to δ 66.4 (+9.3 ppm) and C-3, -3' to δ 67.6 (+10.2 ppm) in the ¹³C NMR spectrum (Table 3). These facts suggested that 3 was a mixture of 3a and 3b.

The ¹H NMR spectrum of incarvine A *N*-oxide (4) was similar to that of 7 [4] except for the downfield shifts of *N*-Me groups (+1.05 ppm), and their neighbouring methylene groups. The ¹³C NMR spectrum (Table 4) also supported the above facts: two *N*-Me signals were shifted downfield to δ 61.25 (+15.35 ppm), C-1, -1' to δ 67.23 (+10.13 ppm), and C-3, -3' to δ 66.16 (+8.86 ppm). The occurrence of [M+1]⁺ at m/z 545 and [M]⁺ at m/z 544 in the positive FAB mass spectrum of 4, as well as a fragment ion [M-O+1]⁺ at m/z 529, was typical of *N*-oxides [7]. The multiplicity observed in the ¹H NMR spectrum and the duplicate ¹³C-signals indicated that 4 was a mixture of 4a and 4b.

Oxidation of 4 and 7 with *m*-chloroperbenzoic acid, individually, afforded incarvine A N,N'-dioxide (9). This confirmed that 4 was composed of 4a and 4b. The positive FAB mass spectrum of 9 showed $[M+1]^+$ at m/z 561 suggesting both N and N' were oxidized. The 1 H and 13 C NMR spectra of 9 were similar to that of 7 except that all N-Me signals were shifted downfield to $\delta_{\rm H}$ 3.19 (+0.90 ppm), 1-, 1'-Ha to δ 3.00 (+1.34 ppm), 1-, 1'-Hb to δ 3.17 (+0.45 ppm), 3-, 3'-Ha to δ 3.00 (+1.24 ppm) and 3-, 3'-Hb to $\delta_{\rm H}$ 3.10 (+0.52 ppm) and to $\delta_{\rm C}$ 60.96 (+15.06 ppm), $\delta_{\rm C}$ 67.41 (+10.31 ppm), and $\delta_{\rm C}$ 66.31 (+9.01 ppm), respectively (Table 4).

EXPERIMENTAL

General. ¹H and ¹³C NMR: JEOL JNM-GX 500 NMR, int. standard (TMS); EIMS: JEOL DX-303 HF spectrometer; TLC: precoated Kieselgel 60 F₂₅₄ plate (0.2 mm Merck), detection by spraying Dragendorff and 10% aq. H₂SO₄; CC: Kieselgel 60 (70–230 and 230–400 mesh, Merck), Aluminium oxide 90

Table 1. 13C NMR data for compounds 2 and 6

Incarvilline moieties			C ₆ -C ₃ units			
С	6	2	С	6	2	
4, 4′, 8, 8′-Me	14.4	14.4	a, a'-C	40.3	41.2	
	14.8	14.8		41.7	42.2	
	16.8	16.8	b, b'-C	47.2	47.9	
	17.0	17.1		47.8	48.0	
C-6,6′	29.1	29.2	COO, COO′	171.7	171.6	
	29.6	29.7		171.9	171.9	
C-4, 4'	30.1†	30.3†	3", 3"'-OMe	55.6	55.8	
	30.1†	30.3†		55.7	56.2†	
C-5, 5'	37.2†	37.3†	C-2", 2"'	110.8	110.9	
	37.2†	37.3†		110.9	104.4	
C-8, 8'	40.1†	40.3	C-5", 5"'	114.7†	114.6	
	40.1†	40.4		114.7†	147.2	
<i>N</i> , <i>N</i> ′-Me	45.9	46.1	C-6", 6""	119.8	119.9	
	46.0	47.4		120.3	104.6	
C-9, 9'	45.6	45.8	C-1", 1"	130.2	129.7	
	45.7	45.9		130.4	130.5	
C-1, 1'	57.1	57.2	C-3", 3"	145.3	145.2	
	57.3	57.3		145.5	147.1	
C-3, 3'	57.4†	57.5	C-4", 4"'	146.8	146.8	
•	57.4†	57.6	•	146.9	134.3	
C-7, 7'	76.2	76.4	5" (or 5"')-OMe		56.2†	
•	76.5	76.6	` '			

[†] Overlapped signals.

Table 2. ¹H NMR data for compounds 2 and 6

6	2				
0.58 (1H, m, 6-Ha)	0.59 (1H, m, 6-Ha)				
0.60 (3H, d, J = 7.3 Hz, 8-Me)	0.59 (3H, d, J = 7.3 Hz, 8-Me)				
0.71 (3H, d, J = 6.7 Hz, 4-Me)	0.72 (3H, d, J = 6.1 Hz, 4-Me)				
0.76 (3H, d, J = 6.7 Hz, 4'-Me)	0.76 (3H, d, J = 6.7 Hz, 4'-Me)				
0.81 (3H, d, J = 7.0 Hz, 8'-Me)	0.82 (3H, d, J = 7.3 Hz, 8'-Me)				
1.06 (1H, m, 6'-Ha)	1.02 (1H, m, 6'-Ha)				
1.42 (2H, m, 1, 1'-Ha)	1.44 (2H, t, J = 12.2 Hz, 1, 1'-Ha)				
1.58 (4H, m, 3, 3'-Ha, 6, 6'-Hb)	1.59 (3H, m, 3, 3'-Ha, 6-Hb)				
1.72 (2H, m, 8', 8-H)	1.72 (2H, <i>m</i> 6'-H <i>b</i> , 8-H)				
1.84 (2H, m, 9, 9'-H)	1.84 (3H, m, 9, 9', 8'-H)				
1.97 (2H, m, 4, 4'-H)	1.96 (3H, m, 4, 4', 5-H)				
2.13 (2H, m, 5, 5'-H)	2.14 (1H, m, 5'-H)				
2.17 (6H, s, N, N'-Me)	2.19 (6H, s, N, N'-Me)				
2.45 (2H, m, 3, 3'-Hb)	2.46 (2H, m, 3, 3'-Hb)				
2.55 (2H, m, 1, 1'-Hb)	2.57 (2H, m, 1, 1'-Hb)				
3.81–3.89 (2H, m, b, b'-H)	3.83 (2H, <i>m</i> , b, b'-H)				
3.88, 3.89 (each, 3H, s, 3", 3"'-OMe)	3.88, 3.89, 3.90 (each, 3H, s , 3", 3", 5" (or 5")-OMe \times 3)				
4.30-4.38 (2H, m, a, a'-H)	4.34 (2H, m, a, a'-H)				
4.89 (2H, m, 7, 7'-H)	4.90 (2H, m, 7, 7'-H)				
6.77 (6H, m, 2", 2"', 5", 5"', 6", 6"'-H)	6.53 (2H, m, 2"', 6"'-H)				
	6.81 (3H, m, 2", 5", 6"-H)				

^{*} Assignments may be interchanged between two incarvilline moieties.

aktiv (70–230 mesh, Merck) and Bondapak C_{18} (Water Associates, Inc.).

Extraction and separation. The aerial parts of Incarvillea sinensis collected in Hebei province, China, were exhaustively extracted with EtOH. The EtOH extract was concd under red. pres. to a syrup, which was

dissolved in 2% HCl and filtered. The filtrate was adjusted to pH 11 by adding NH₄OH, and the alkaloids extracted into CHCl₃. After removal of solvent *in vacuo*, the residue was repeatedly chromatographed over an Al₂O₃ column with CHCl₃–MeOH (10:0.4), Bondapak C₁₈ with 70–100% MeOH, and a silica gel

Table 3. ¹³C NMR data for compounds 3 (CD₃OD), 6 (CDCl₃) and 8 (CD₃OD+D₂O)

Incarvilline moieties			C ₆ -C ₃ units				
С	6	3	8	С	6	3	8
8, 8'-Me	14.4	15.91	15.9	a, a'-C	40.3	43.21	43.7
	14.8	16.06	16.0		41.7	44.06	44.0
4, 4'-Me	16.8	17.01	17.0	b, b'-C	47.2	48.96	48.8
	17.0	18.05			47.8	50.05	50.0
		18.08				50.08	
C-6, 6'	29.1	31.10	31.1	O, O'-Me	55.6	57.30	57.4
	29.6	31.21	31.4		55.7		55.7
		31.41		C-2", 2"'	110.8	113.53	113.4
		31.53			110.9	113.68	113.5
C-4, 4'	30.1	27.47	27.4	C-5", 5"	114.7	116.98	117.0
		31.80				117.06	
		31.85		C-6", 6""	119.8	121.70	121.5
C-5, 5'	37.2	38.17	38.1		120.3	121.73	122.1
	37.3	39.18				122.26	
		39.23		C-1", 1""	130.2	132.55	132.5
C-9, 9'	45.6	43.86	43.0	- /	130.4	132.58	132.6
C 7, 7	45.7	43.92	44.0			132.66	102.0
		47.24		C-3", 3"	145.3	147.65	147.3
		47.29		_ , _	145.5	147.85	147.5
N, N'-Me	45.9	46.39	61.5	C-4", 4"'	146.8	149.67	149.7
,	46.0	46.42		, .	146.9	149.72	• • • • • • • • • • • • • • • • • • • •
		61.44		COO, COO	171.7	174.06	174.5
C-8, 8'	40.1	41.80	41.7	,	171.9	174.15	174.8
,		42.59				174.38	
C-1, 1'	57.0	58.03	66.4			174.46	
z -	57.1	58.13					
		66.50					
		66.53					
C-3, 3′	57.3	58.53	67.6				
C 3, 3	57.4	58.58					
	· · · ·	67.66					
C-7, 7'	77.2	77.76	78.2				
, .	76.5	78.11	78.5				
	, 3.5	78.25	, 5.5				
		78.50					

column with cyclohexane–MeOH– Et_2NH (30:1:1–5:1:1) to afford compounds 1 (40.1 mg), 2 (62.5 mg), 3 (47.0 mg) and 4 (41.0 mg).

Incarvine D (1). White powder, $[\alpha]_D^{26} - 4.2^{\circ}$ (c = 0.60, CHCl₃). EIMS m/z (rel. int.): 352 [M+1]⁺ (100), 351 [M]⁺ (66), 336 (10), 250 (59), 183 (70), 182 (100), 166 (100), 84 (16), 81 (22), 58 (88); ¹H NMR (CDCl₃); Monoterpene moiety: δ 0.94 (3H, d, J = 6.7 Hz, 7'-Me), 1.30 (1H, m, 6'-Ha), 1.40 (1H, m, 6'-Hb), 1.44 (1H, m, 9'-Ha), 1.55–1.64 (2H, m, 8'-H, 9'-Hb), 1.83 (3H, s, 2'-Me), 1.91–2.22 (2H, m, 5'-H₂), 3.66 (2H, m, 10'-H₂), 6.73 (1H, t, t) = 7.3 Hz, 4'-H). ¹³C NMR (CDCl₃); Monoterpene moiety: δ 167.8 (C-1'), 12.3 (C-2'), 127.8 (C-3'), 142.2 (C-4'), 26.2 (C-5'), 35.8 (C-6'), 19.4 (C-7'), 29.4 (C-8'), 39.7 (C-9'), 60.8 (C-10').

Methoxyincarvillateine (2). White powder, $[\alpha]_D^{20}$ -4.0° (c = 0.61, CHCl₃), EIMS m/z (rel. int.): 748 [M]⁺ (13), 390 (44), 389 (19), 361 (25), 360 (100), 359 (47), 182 (73), 166 (54), 58 (27).

Incarvillateine N-oxide (3). White powder, $[x]_{\rm D}^{\rm 18}$ –19.2° (c=0.50, CH₃OH). Positive FAB-MS m/z (rel. int.): 735 [M+1]⁺ (100), 717 (18), 376 (96), 375 (15), 361 (26), 360 (92); ¹H NMR (CD₃OD): δ 0.66 (2H, d, J=7.9 Hz, 6-Ha), 0.71, 0.75, 0.77, 0.79, 0.80, 0.83, 0.88, 0.91 (24H, d, J=7.3, 7.3, 6.7, 7.3, 6.7, 7.3, 6.7, 7.3, 7.3 Hz, 4, 4'-Me × 2, 8, 8'-Me × 2), 1.04 (2H, m, 6'-Ha), 1.58 (4H, m, 6, 6'-Hb), 1.68–1.90 (10H, m, 1, 3-Ha, 8, 8', 9-H), 2.01, 2.15 (6H, m, 4, 5, 5'-H), 2.28 (6H, s, N-Me), 2.32 (2H, m, 9'-H), 2.57 (4H, m, 3-Hb, 4'-H), 2.67–2.99 (10H, m, 1-Hb, 1', 3'-H), 309 (6H, s, N'-Me), 3.82–3.95 (4H, m, b, b'-H), 3.86, 3.88 (12H, s, OMe), 4.29–4.39 (4H, m, a, a'-H), 4.93 (4H, m, 7, 7'-H), 6.77 (8H, m, 5", 5"', 6", 6"'-H), 6.87, 6.92 (4H, m, 2", 2"'-H).

Oxidation of 3 and 6 with m-chlorperbenzoic acid. A soln of m-chloroperbenzoic acid (3.4 mg) in CHCl₃ (1 ml) was added gradually to the stirred soln of 3 (5.2 mg) in CHCl₃ (1 ml), and stirring continued for 2 hr. After evapn of the solvent, the residue was subjected

Table 4. ¹³C NMR data for compounds 4 (CDCl₃+CD₃OD), 7 (CDCl₃) and 9 (CD₃OD)

Incarvilline moiety				Hildebrant's acid moiety			
С	7	4	9	С	7	4	9
8, 8'-Me	14.6	15.83	15.98	C-1"	167.3	169.29	169.68
		15.92	16.04	C-2"	12.2	12.19	13.42
4, 4'-Me	17.1	17.10	19.67			13.58	
		19.99		C-3"	128.9	129.89	130.55
C-6, 6'	29.6	30.92	31.50			130.04	
	29.7	31.04		C-4"	139.9	142.00	142.97
		31.10				142.36	
C-4, 4'	30.2	26.85	27.67	C-5"	26.4	27.85	28.20
		31.22				27.91	
C-5, 5'	37.3	37.60	38.46	C-6"	39.3	40.69	41.05
,	37.4	38.72		C-7"	18.6	18.23	17.09
		38.78		C-8"	157.9	160.09	161.22
C-9, 9'	45.7	43.21	43.95			160.58	
		43.61		C-9"	116.2	117.23	117.92
		46.43				117.53	
<i>N</i> , <i>N</i> ′- M e	45.9	46.49	60.96	C-10"	166.2	168.02	168.43
		46.64				168.06	
		61.25					
C-8, 8′	40.6	41.33	42.26				
	40.7	41.42					
		42.12					
		42.27					
C-1, 1'	57.1	57.75	67.41				
,		67.23					
C-3, 3'	57.3	58.12	66.31				
		66.16					
C-7, 7'	75.1	76.21	76.87				
	75.9	76.64	77.81				
		77.09					
		77.52					

to CC on silica gel eluted with CHCl₃-MeOH-H₂O (8:2:0.2-7:3:0.5) to provide compound 8 (3.3 mg). By the same method, 8 (28.5 mg) was prepd from incarvillateine (6) (51.7 mg). $[\alpha]_D^{28} - 20.0^{\circ}$ (c = 0.37, $CH_3OH + H_2O$ (1:1)). Positive FAB-MS m/z (rel. int.): $751 [M+1]^+$ (45), 376 (50), 360 (10), 182 (100); ¹H NMR (CD₃OD + D₂O): δ 0.67 (1H, m, 6-Ha), 0.79 (3H, d, J = 7.3 Hz, 8-Me), 0.80 (3H, d, J = 7.3 Hz,8'-Me), 0.83 (3H, d, J = 6.7 Hz, 4-Me), 0.91 (3H, d, 6-Hb), 1.72 (1H, m, 6'-Hb), 1.89 (1H, m, 8-H), 1.96 (1H, m, 8'-H), 2.06 (1H, m, 5-H), 2.15 (1H, m, 5'-H), 2.32 (2H, m, 9, 9'-H), 2.55 (2H, m, 4,4'-H), 2.78 (2H, m, 1, 1'-Ha), 2.94 (4H, m, 3, 3'-H), 3.10 (2H, m, 1, 1'-Hb), 3.13 (6H, s, N, N'-Me), 3.85-3.99 (2H, m, b, b'-H), 3.88, 3.90 (6H, s, O, O'-Me), 4.30–4.41 (2H, m, a, a'-H), 4.94 (2H, m, 7, 7'-H), 6.77 (4H, m, 5", 5"', 6", 6"'-H), 6.87, 6.92 (2H, m, 2", 2"'-H).

Incarvine A N-oxide (4). White powder, $[\alpha]_D^{27} - 14.8^{\circ}$ (c = 0.45, CHCl₃). Positive FABMS m/z (rel. int.): 545 $[M+1]^+$ (200), 544 $[M]^+$ (21), 543 (31), 529 $[M-O+1]^+$ (10), 527 (20), 380 (23), 364 (8), 281 (12), 265 (6); EIMS m/z (rel. int.): 545 $[M+1]^+$, 543, 529 $[M-O+1]^+$ (61), 528 $[M-O]^+$ (51), 364 (18), 347

(18), 182 (76), 166 (65), 164 (40), 163 (100), 110 (42); ¹H NMR (CDCl₃+CD₃OD): δ 0.87, 0.91, 0.93, 0.94, 0.96, 0.98, 1.27, 1.28 (24H, d, J = 7.0, 7.0, 6.5, 8.4, 6.2, 7.3, 7.3, 7.3 Hz, 4, 4′-Me × 2, 8, 8′-Me × 2), 1.57–1.84 (8H, m, 1, 3-Ha, 6, 6′-Ha), 1.84 (6H, s, 2″-Me), 1.84–2.17 (12H, m, 6, 6′-Hb, 4, 9, 8′-H), 2.17 (6H, s, 7″-Me), 2.29–2.46 (10H, m, 5, 5″, 6″-H), 2.36 (6H, s, N-Me), 2.50–2.63 (6H, m, 5′-H, 1, 3-Hb, 9′-H), 2.78–2.94 (4H, m, 4′-H, 3′-Ha), 3.00–3.18 (4H, 1′-Ha, 1′-Hb), 3.34 (8H, s, N'-Me, 3′-Hb), 5.29 (4H, m, 7, 7′-H), 5.68 (2H, s, 9″-H), 6.70 (2H, m, 4″-H).

Oxidation of 4 and 7 with m-chloroperbenzoic acid. A soln of *m*-chloroperbenzoic acid (10.0 mg) in CHCl₃ (1 ml) was added gradually to the stirred soln of 4 (7.0 mg) in CHCl₃ (1 ml), and stirring was continued for 2 hr. After evapn of the solvent, the residue was subjected to CC on silica gel eluted with CHCl₃-MeOH-H₂O (9:1:0.1-7:3:0.5) to provide compound 9 (2.6 mg). By the same method, 9 (2.0 mg) was prepd from incarvine A (7) (5.0 mg). $[\alpha]_D^{27}$ -6.0° (c = 0.44, CH₃OH). Positive FAB-MS m/z (rel. int.): 561 (44) $[M+1]^+$, 380 (7), 281 (7), 182 (100); ¹H NMR (CD₃OD): δ 0.95 (each 3H, d, d = 7.3 Hz, 4, 4'-Me), 0.97, 1.00 (each 3H, d, d = 7.3 Hz, 8, 8'-Me), 1.66

4 a

4 b

9

(2H, *m*, 6, 6'-H*a*), 1.84 (3H, *s*, 2"-Me), 1.94–2.12 (4H, *m*, 6, 6'-H*b*, 8, 8'-H), 2.17 (3H, *d*, *J* = 1.2 Hz, 7"-Me), 2.34 (2H, *m*, 6"-H), 2.45 (6H, *m*, 5, 5', 9, 9',5"-H), 2.71 (2H, *m*, 4, 4'-H*a*), 3.00 (4H, *m*, 1, 1', 3, 3'-H*a*), 3.10 (2H, *m*, 3, 3'-H*b*), 3.17 (2H, *m*, 1, 1'-H*b*), 3.19 (6H, *s*, *N*, *N*'-Me), 5.29 (2H, *m*, 7, 7'-H), 5.71 (1H, *m*, 9"-H), 6.73 (1H, *m*, 4"-H).

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