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EUDESMANOLIDES AND OTHER CONSTITUENTS OF INULA THAPSOIDES*

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Key Word Index—*Inula thapsoides*; Compositae; sesquiterpene lactones; eudesmanolides; diterpenes; steroids.

Abstract—Three new eudesmanolides, 1β -hydroxy- 11α , 13-dihydroalantolactone, 1α -hydroxy- 11α , 13-dihydroalantolactone and 1β , 4α -dihydroxy- 11α , 13-dihydroalantolactone, along with the known compounds sclareol, 7,11,15-trimethyl-3-methylenehexadecan-1,2-diol, β -sitositerol, stigmasterol and stigmasterol 3β -D-glucoside have been isolated from the whole plant extract of *Inula thapsoides*. © 1997 Elsevier Science Ltd

INTRODUCTION

In our studies on Turkish Inula species, we have isolated guaianolides and eudesmanolides from Inula anatolica [1] and I. graveolens [2, 3] as well as some sesquiterpene acids, steroids and flavonoids. The whole plant extract of Inula thapsoides (Bieb. ex Willd. Sprengel) subsp. thapsoides (Syn: Conyza thapsoides Bieb. ex Willd.) gave seven new guaianolides [4]. Further investigation of I. thapsoides has led to the isolation of three new eudesmanolides, 1\beta-hydroxy- 11α , 13-dihydroalantolactone (1), 1α -hydroxy- 11α , 13dihydroalantolactone (2) and 1β , 4α -dihydroxy- 11α , 13-dihydroalantolactone (3), in addition to the known compounds, sclareol [5], 7,11,15-trimethyl-3methylene-hexadecan-1,2-diol [6], β-sitosterol, stigmasterol and stigmasterol 3β -D-glucoside. We now report on the structure determination of the three new eudesmanolides (1-3) by spectroscopic methods including X-ray analysis for compound 1.

RESULTS AND DISCUSSION

Compound 1 was the most abundant eudesmanolide of the plant extract. The IR spectrum contained absorptions bands for a hydroxyl at 3440 cm⁻¹, a lactone at 1780 cm⁻¹ and unsaturation at 1675 cm⁻¹. The ¹H NMR spectrum showed a methyl singlet at δ 1.21 and two methyl doublets at δ 1.12 (J = 7.5 Hz)

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- R=β-OH
- 2 R=α-OH
- 1a R=β-OAc
- 2a R=α-OAc

and 1.36 (J = 7.5 Hz). However, the characteristic doublets of exocyclic methylene protons were not observed. The absorption of the lactone carbonyl in the IR spectrum at 1780 cm⁻¹ and the signal at δ 179.8 in the ¹³C NMR spectrum of 1 (Table 1) established the presence of a five membered saturated lactone. In the ¹H NMR spectrum (Table 1), an olefinic proton at δ 5.26 (d, J = 3.5 Hz), a lactone proton at δ 4.90 (ddd, J = 2, 3.5 and 6.5 Hz) and an oxymethine proton at δ 3.29 (ddd, J = 1, 4 and 11.2 Hz, H-1) were observed. Irradiation of H-7 at δ 2.64 led to the sequence of H-6 to H-9 as shown by the collapse of the protons at δ 5.26 (dd, H-6), 4.90 (ddd, H-8) and 2.49 (d, H-9) into a singlet, a doublet of doublet and a doublet, respectively, while on irradiation of H-1 at δ 3.29, the C-2 proton signals were simplified. Compound 1 gave a monoacetyl derivative (1a). In the ¹H NMR spectrum of 1a (see Experimental) the signal at δ 3.29 was shifted to δ 4.53 and an acetyl methyl singlet appeared at δ 2.08 verifying the presence of a secondary hydroxyl group. Its β orientation was deduced from the splitting pattern of H-1 as well as

^{*}Dedicated to Professor Ayhan Ulubelen (University of Istanbul, Turkey) on the occasion of her sixtyfifth Birthday. §Author to whom correspondence should be addressed.

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Table 1. ¹ H and ¹³ C NMR data of eudesmanolides 1–3 (CDCl	Tal	ble 1. ¹ H and	¹³ C NMR	data of e	eudesmanolides	1–3	(CDCI	í,
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	1 (¹³ C)	1 (¹ H)	2 (¹H)	3 (¹H)	3 (13C)
1	80.3	3.29 dd (4, 11.2)	3.46 dd (5.5, 10)	3.29 ddd (1.5, 4.5, 12)	77.9
2	29.9	1.86 m, 1.56 m	1.95 dd, 1.52 m	1.88 m, 1.53 m	30.1
3	39.3	1.61 m, 1.72 m	1.70 m	obscured	41.7
4	37.8	2.45 m	2.42 m	_	72.3
5	147.7	_	_	·	147.9
6	121.4	5.26 d (3.5)	5.32 d (3.5)	5.61 d (3.5 Hz)	122.1
7	42.3	2.64 ddd (1.5, 3.5, 6)	2.67 ddd (1.5, 3.5, 6)	2.69 m	43.6
8	76.0	4.90 ddd (2, 3.5, 6.5)	4.88 ddd (2.5, 3.5, 6.5)	4.91 ddd (2.5, 3.5, 7.5)	78.1
9	39.3	2.49 dd (3.5, 15 Hz)	2.24 dd (2.5, 15.5)	2.53 dd (4.5, 15), 1.63 m	40.3
		1.58 dd (4, 15 Hz)	1.60 dd (3.5, 15.5)		
10	38.6	_	·	_	39.7
11	43.5	2.42 dq (7, 7.5)	2.46 dq (7, 7.5)	2.46 dq (7, 7)	43.6
12	179.8	_		_	177.3
13	16.1	1.36 d (7.5)	1.37 d(7.5)	1.39 d(7)	14.6
14	22.0	1.21 s	1.30 s	1.26 s	22.4
15	22.4	1.12 d(7.5)	1.13 d(7.5)	1.37 s	16.5

by comparing the ¹H NMR spectral data with those of 1β -hydroxyalantolactone [7] and similar eudesmanolides [8]. The placement of a double bond between C-5 and C-6 was also compatible with the dihydroalantolactone derivatives [9]. The ¹³C NMR (APT) spectrum displayed 15 carbons (3 methyl, 6 methine, 3 methylene and 3 quaternary). The HREI mass spectrum proved the proposed structure i.e. $[M]^+$ at m/z 250.1585 in agreement with the molecular formula $C_{15}H_{22}O_3$. HETCOR experiments allowed the unambiguous assignment of all proton-bearing carbons. The orientation of the C-11 methyl group was deduced from the J value of H-11 with H-7 and by comparison with 11,13-dihydroeudesmanolides [9, 10]. Since the acetate of 1 (1a) gave suitable crystals, an X-ray analysis was performed. Thus, the stereochemistry of all centres was clearly deduced. Based on all the spectral data, the structure of 1 was determined as 1β -hydroxy- 11α , 13-dihydroalantolactone.

Compound 2 showed ¹H NMR spectral properties very similar to those of 1. A proton geminal to a secondary hydroxyl group was observed at δ 3.46 with a splitting pattern of J = 5.5 and 10.5 Hz and attributed to an a hydroxyl group at C-1. The data given in the literature [11] for 1α -hydroxylated eudesmanolides were in a good agreement with the chemical shift and multiplicity of H-1. The EI mass spectrum gave the same molecular ion peak ([M]⁺ = m/z 250) as compound 1, while the acetate of 2 (2a) showed a [M]⁺ peak at m/z 292 confirming the presence of one secondary hydroxyl group in the molecule. Thus, 2 was identified as 1α -hydroxy- 11α , 13-dihydroalantolactone.

The third eudesmane (3) also gave a ¹H NMR spectrum similar to that of compound 1. However, there was a significant difference in the chemical shift of the C-15 methyl (δ 1.37, s) compared with those of the

other methyls [δ 1.39 (J = 7 Hz, H₃-13) and 1.26 (s, H_3 -14)]. A secondary hydroxyl group at δ 3.29 with the same J values (ddd, J = 1, 4, 11.5 Hz) and an olefinic proton at δ 5.61 were observed. The location of the double bond was deduced to be between C-5 and C-6 as in the first two compounds. Its downfield chemical shift compared with that in compound 1 could be explained by the presence of a hydroxyl group at C-4. The quaternary oxygenated carbon at δ 72.3 in the ¹³C NMR (APT) spectrum confirmed the presence of a tertiary hydroxyl group at C-4. The HREI mass spectrum showed a molecular ion peak at m/z 266.1529 corresponding to molecular formula C₁₅H₂₂O₄, and the most prominent fragment ion was observed at m/z 248 [M-H₂O]⁺. Compound 3 was deduced to be 1β , 4α -dihydroxy- 11α , 13-dihydroalantolactone.

EXPERIMENTAL

General. IR: CHCl₃; ¹H and ¹³C NMR: 200 and 50.32 MHz; MS: VG ZabSpec GC-MS instruments.

Plant material. Inula thapsoides subsp. thapsoides were collected in July 1992 from central Turkey (Yıldızeli, Sivas) and identified by Prof. Semra Kurucu. A specimen is deposited in the Herbarium of Faculty of Pharmacy, University of Ankara (AEF: 17512).

Extraction and isolation. Whole plant (1.7 kg) of I. thapsoides was extracted with petrol-Et₂O-EtOH (1:1:1) at room temp. and the solvent evapd. in vacuo to give 32 g extract. The extract was chromatographed on silica gel (350 g, 100-200 mesh) eluting with petrol, and a gradient of petrol-EtOAc up to 100% EtOAc, followed by MeOH. From the first frs collected with petrol-EtOAc (9:1), compounds 1 (500 mg), 2 (11 mg) and 3 (18 mg) were isolated and purified by repeated prep. TLC.

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 1β -Hydroxy-11α,13-dihydroalantolactone (1). IR $\lambda_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3010, 2930, 1780, 1675, 1470, 1390, 1360, 1340, 1280, 1240, 1225, 1200, 1080, 1050, 1030, 990, 940, 920, 900, 880, 855 and 760; HREIMS m/z: 250.1585 (calc. 250.1568 for C₁₅H₂₂O₃); CIMS m/z (rel. int.): 251 [M+1]+ (82), 233 [M+1-H₂O]+ (58), 205 (19), 187 (12), 159 (23), 57 (100); ¹H and ¹³C NMR: Table 1.

Acetyl derivative of 1 (1a). IR $\lambda_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1775, 1730, 1635, 1475, 1380, 1345, 1260, 1240, 1080, 1050, 995, 950, 890 and 760; ¹H NMR (CDCl₃) δ : 1.35 (3H, d, J = 7.5 Hz, H-13), 1.28 (3H, s, H-14), 1.13 (3H, d, J = 7.5 Hz, Me-15), 2.08 (3H, s, OAc), 2.43 (1H, dq, J = 7 and 7 Hz, H-11), 2.23 (1H, dd, J = 4.5 and 15 Hz, H-9), 1.53 (1H, dd, J = 3 and 15 Hz, H-9'), 2.63 (1H, ddd, J = 2 and 3.5 and 6 Hz, H-7), 4.53 (1H, dd, J = 3.8 and 11.5 Hz, H-1a), 4.87 (1H, ddd, J = 3, 3.5 and 6 Hz, H-8), 5.30 (1H, d, d, d = 3.5 Hz, H-6).

 1α -Hydroxy-11α,13-dihydroalantolactone (2). IR $\lambda_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3440, 3020, 2935, 1780, 1670, 1480, 1385, 1360, 1340, 1280, 1245, 1230, 1200, 1080, 1030, 990, 945, 900, 880, 855 and 760; ¹H NMR: Table 1; EIMS (70 eV) m/z (rel. int.): 250 [M]⁺ (3), 232 [M-H₂O]⁻ (55), 207 (22), 186 (24), 172 (11), 159 (29), 57 (100).

Acetyl derivative of **2** (2a). ¹H NMR (CDCl₃) δ 1.15 (3H, d, J = 7 Hz, H-15), 1.35 (3H, s, Me-14) and 1.39 (3H, d, J = 7 Hz, H-13), 2.06 (3H, s, OAc), 4.88 (1H, dd, J = 5 and 10 Hz, H-1), 4.87 (m, H-8), 5.33 (1H, d, J = 3 Hz, H-6); EIMS (70 eV) m/z (rel. int.): 292 [M]⁺ (1), 291 [M-1]⁺ (6), 290 [M-2]⁺ (30), 256 (21), 251 [M-OAc]⁺ (20), 248 [M-OAc-1]⁺ (100), 230 (37), 215 (18), 185 (17), 159 (47), 131 (22), 119 (23), 97 (26), 83 (31), 57 (44).

 1β , 4α-Dihydroxy-11α,13-dihydroalantolactone (3). IR $\lambda_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3430, 3010, 2890, 1775, 1660, 1470, 1380, 1350, 1265, 1230, 11020, 1080, 1030, 990, 885, 765; ^{1}H and ^{13}C NMR: Table 1; HREIMS m/z: 266.1529 (calc. 266.1518 for C₁₅H₂₂O₄), 248 [M-H₂O]⁺ (100), 233 (51), 215 (233-H₂O)⁺ (25), 187 (18), 160 (45), 59 (80), 57 (75); EIMS (70 eV) m/z (rel. int.): 266 [M]⁺ (10), 248 (70), 233 (18), 230 (29), 207 (23), 205 (28), 187 (21), 175 (71), 157 (100), 147 (87), 131 (86), 121 (73), 105 (75), 91 (87), 79 (53), 47 (73).

X-ray analysis. All data were collected on a Rigaku AFC6S diffractometer with graphite monochromated Cu-K α radiation and a constant speed ω -20 scan technique with weak reflections rescanned a maximum of 4 times. Crystal and refinement data were as follows: formula: C₁₇H₂₄O₄; F.W.: 292.37; crystal colour, habit: colourless, prismatic; crystal system: monoclinic; space group: P2₁ (#4); crystal dimensions: $0.20 \times 0.15 \times 0.30$ mm; lattice: primitive; lattice parameters: (a) 6.219 (3) Å, (b) 9.112 (1) Å, (c) 14.261 (2) Å; b(°): 96.63 (2); Z value: Z; V_1 : 802.7(4) Å³; D(calc): 1.210 g cm⁻³; radiation: (1 = 1.54178 Å; abs.coefm): $6.89 \,\mathrm{cm}^{-1}$; F(000): 316.00; temp: 23.0°; diffractometer: Rigaku AFC6S; scan mode: w-20; 2q range (max): 157.4°; total data: 3527; unique data: 1783 (R = 0.035); observed data used: 177 8[1 > 0.006 int (I)]; no. of parameters refined: 287; max. shift/error in

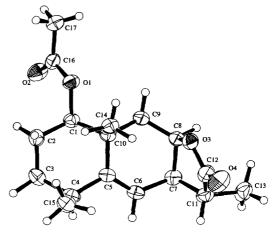


Fig. 1. Molecular structure of 1a

final cycle: 0.11; max. resid. density: $0.20 \, \mathrm{e}^{-/}$ Å³; $R \, Rw$: 0.051; 0.062; Goodness of Fit Indicator: 1.35. Unit cell parameters were obtained from a least-squares refinement of 1 carefully centred reflections in the range $53.13 < 2q < 53.13^{\circ}$ corresponded to a primitive monoclinic cell. Lorentz-polarization, a Y-scan empirical absorption correction and anisotropic extinction corrections were applied. The structure was solved by direct methods [12] and refined and analysed by teXsan [13] and PLATON [14]. The coordinates, distances and angles have been deposited with the Cambridge Structural Data Base.

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