

COUMARINS FROM *NASSAUVIA CUMINGII*

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Key Word Index—*Nassauvia cumingii*; Asteraceae; 5-methyl coumarins.

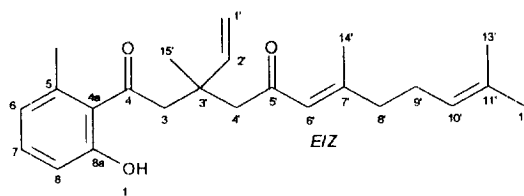
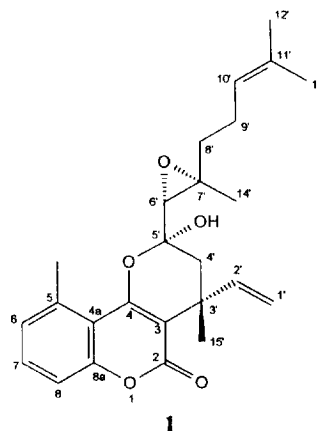
Abstract—The aerial parts of *Nassauvia cumingii* gave a new phenolic compound, presumably derived from a coumarin. The complete relative configuration of a known coumarin was established. © 1997 Elsevier Science Ltd

INTRODUCTION

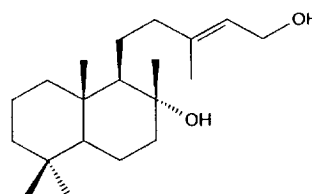
The genus *Nassauvia* is placed in the subtribe Nassauviinae and is represented by 37 species distributed in the Andes from southern Bolivia to Tierra del Fuego and the Malvinas Islands [1]. In Chile, there are 24 species, 10 of which have been investigated chemically [2–8]. As many other genera of the tribe, they are characterized by the occurrence of isocedrenes and 4-hydroxy-5-methyl coumarins. Continuing with our studies of Chilean Asteraceae, we have now examined *N. cumingii*.

RESULTS AND DISCUSSION

The aerial parts of *N. cumingii* H. et A afforded the coumarin **1**, the phenolic compound **2E** and the diterpene **3**. The coumarin has already been described as a constituent of *Triptilion benaventei* [2] and of *N. digitata* [4]. The relative stereochemistry within the cyclic part and in the side-chain was correctly assigned. However, due to free rotation of the side-chain, two diastereomers are possible which would fit the spectral data. Detailed NOE investigation (Table 1) now allowed complete assignment of the relative stereochemistry. In the ¹H NMR spectrum (Table 1), the proton of the hydroxyl group appeared as a doublet due to a ⁴J long-range coupling with the axial H-4' (a consequence of the W-in the plane-orientation of the four bonds). Within the side-chain, the configuration of the epoxide ring was deduced from a NOE effect between H-6' (epoxide proton) and H-8'



2E **2Z**
 $\Delta^{6'E}$ $\Delta^{6'Z}$



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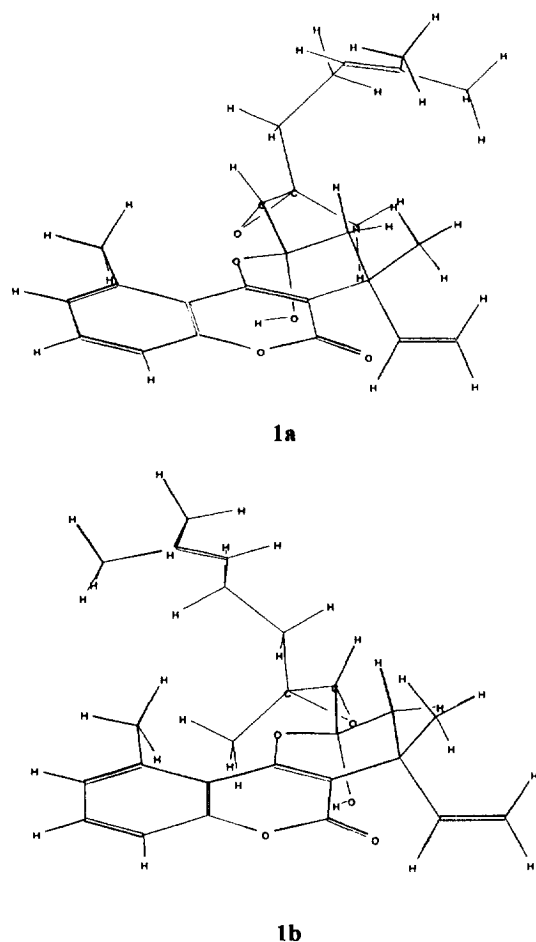


Fig. 1. Calculated conformations of diastereomers **1a** and **1b**; the view from bottom to top of the coumarin part. In both structures the zig-zag (W)-orientation of the hydroxyl proton and the axial proton in the pyrane part, as well as the close spatial proximity of the hydroxyl group and the epoxide oxygen leading to a strong hydrogen bonding, is clearly visible. Only the structure **1a** is in accordance with NOE results.

Compound 1. MS: m/z (rel. int.) 410.209 $[M]^+$ (5) (calcd for $C_{25}H_{30}O_5$ 410.209), 271 $[M\text{-side-chain}]^+$ (25), 228 $[RDA]^+$ (100), 135 $[ArCO]^+$ (55), 95 (45).

Compounds 2E and 2Z. MS: m/z (rel. int.) 368.236 $[M]^+$ (45) (calcd for $C_{24}H_{32}O_3$ 368.235), 350 $[M-H_2O]^+$ (70), 245 $[M-C_9H_{15}]^+$ (45), 233 $[M-Ar^*CO]^+$ (15), 219 $[M-ArCOCH_2]^+$ (70), 203 $[M-C_9H_{15}COCH_2]^+$ (90), 187 (70), 151 $[C_9H_{15}CO]^+$ (75), 149 $[ArCOCH_2]^+$ (55), 135 $[ArCO]^+$ (100), 123 $[C_9H_{15}]^+$ (45), 69 $[C_5H_9]^+$ (60). *Ar = $C_6H_3(OH)(CH_3)$.

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Table 2. 1H NMR data of compounds **2E/Z** and ^{13}C NMR data of compound **2E**

H	2E ($\Delta^\circ E$)	2Z ($\Delta^\circ Z$)	C	2E ($\Delta^\circ E$)
3	3.24 s	3.23 s		
6	6.69 br d	6.69 br d	3	51.8 t
7	7.19 dd	7.19 dd	4	207.9 s
8	6.78 br d	6.78 br d	4a	125.3 s
9	2.47 br s	2.47 br s	5	132.5 s
1'c	4.97 d	4.96 d	6	122.9 d
1't	4.96 d	4.95 d	7	132.8 d
2'	6.01 dd	6.01 dd	8	115.7 d
4' ₁	2.84 d	2.82 d	8a	159.1 s
4' ₂	2.78 d	2.74 d	9	23.0 q
6'	6.02 br s	6.02 br s	1'	111.5 t
8'	2.10 m	2.54 t	2'	145.9 d
9'	2.10 m	2.10 m	3'	39.5 s
10'	5.05 m	5.11 br t	4'	51.9 t
12'	1.68 br s	1.66 br s	5'	200.9 s
13'	1.60 br s	1.59 br s	6'	124.6 d
14'	2.08 d	1.83 d	7'	158.6 s
15'	1.24 s	1.24 s	8'	41.3 t
OH	10.32 br s	10.35 br s	9'	26.1 t
			10'	123.0 d
			11'	137.4 s
			12'	25.6 q
			13'	17.7 q
			14'	19.5 q
			15'	25.5 q

$J(\text{Hz})$: 6,7 = 7,8 = 8; 1'c,2 = 10.5; 1't,2 = 17.5; 4'₁,4'₂ = 15.5; 6',14 = 1; comp. **2Z**: 8',9' = 9',10' = 7.5;

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