PII: S0031-9422(97)00537-2

CLITORIACETAL 11-*O-β*-D-GLUCOPYRANOSIDE FROM *CLITORIA FAIRCHILDIANA*

BERNADETE P. DA SILVA, ROBSON R. BERNARDO and JOSÉ P. PARENTE*

Núcleo de Pesquisas de Produtos Naturais, Universidade Federal do Rio de Janeiro, 21941-590 Rio de Janeiro, Brazil

(Received 19 May 1997)

Key Word Index—*Clitoria fairchildiana*; Leguminosae; roots; rotenoid glycoside; clitoriacetal 11-O- β -D-glucopyranoside.

Abstract—A new rotenoid glycoside, clitoriacetal 11-O- β -D-glucopyranoside, was isolated from the roots of *Clitoria fairchildiana*. Its structure was established by spectroscopic and chemical methods. © 1997 Elsevier Science Ltd

INTRODUCTION

The genus Clitoria has 60 reported species [1], some of which are used for their healing properties and, in particular, C. macrophylla for the treatment of skin diseases in Thailand. From its roots stemonacetal and clitoriacetal have been isolated [2]. The latter was reported as the major component and possessed remarkable anti-inflammatory and antipyretic activities [3]. Moreover, from the roots of C. macrophylla 6-deoxyclitoriacetal was isolated, which in vitro tests showed strong cytotoxic activity against cultured P-388 lymphocytic leukaemia cells [4]. Clitoria fairchildiana, known in Brazil as sombreiro, is widely grown as a shade tree but no medicinal use is reported [5]. In the present work, we report on the isolation of a new rotenoid glycoside, clitoriacetal 11-O-β-Dglucopyranoside (1), from the roots of C. fairchildiana.

RESULTS AND DISCUSSION

Fractionation of a methanol extract by adsorption chromatography on silica gel afforded compound 1 that was obtained as white amorphous powder, which gave a positive Rogers–Calamari test for rotenoids [6] but did not give any colour with FeCl₃. Its molecular formula was calculated as $C_{25}H_{28}O_{14}$ by a combination of LSI mass (negative ion mode) m/z 551 [M-H]⁺ and ¹³C NMR spectral data (Table 1). The UV spectrum of 1 showed single maxima at 286 nm (4.33) and 320 nm (3.80). The chromatographic behaviour, UV spectrum, IR 3436 (OH), 1670 cm⁻¹ (>C = O), 1613, 1573 and 1511 cm⁻¹ (aromatic), ¹H NMR δ 5.50 (1H,

bs, H-6, after D₂O-exchange) and 4.57 (1H, bs, H-6a) [2, 7], 13 C NMR δ 70.02 (C, C-12a) and 77.48 (CH, C-6a) [4, 8] and 90.85 (CH, C-6) spectral data, established that 1 is a rotenoid glycoside. Two doublets at δ 5.50 (J = 6.3 Hz) and 7.60 (J = 6.3 Hz) integrating for single protons and the D₂O-exchange of the latter indicated the presence of a 6-hydroxy group. The ¹H NMR spectrum displayed, in addition to signals for three methoxyl groups, H-6a, H-6 and 6-OH of a rotenoid nucleus, two doublets at δ 6.10 and 6.34 for H-8 and H-10, respectively. Two singlets at δ 6.78 and 6.41 integrating for single protons were assigned to H-1 and H-4 of the aglycone moiety, respectively. Based on the H-1 chemical shift value (δ 6.78), the B/C ring junction in 1 was determined to be cis [9-11]. The appearance of a broad singlet at δ 5.50, attributed to H-6, and a broad singlet at δ 4.57, assigned to H-6a suggested the presence of two hydroxyl groups at C-6 and C-12a and that H-6 exists in 1 as H-6ax [11, 12]. The assignments of the chemical shifts of the methoxyl groups (Table 1) were established by 2D-NMR correlations (Table 2) and by comparison of the data with those of 6-deoxyclitoriacetal [4]. A doublet at δ 4.70 (J = 7.2 Hz) integrating for a single proton was assigned to H-1 of a glucose, indicating β -linkage.

The ¹³C NMR spectrum of 1 was consistent with

^{*} Author to whom correspondence should be addressed.

Table 1. 1 H and 13 C NMR spectral data for compound 1 in DMSO- d_{δ}

Attribution	δ ¹³ C	$\delta^{-1}H(J=Hz)$	DEPT
1	110.9	6.78 s	СН
1a	109.8		C
2	143.4		\boldsymbol{C}
2-OMe	55.9	3.83 s	CH_3
3	150.4		\boldsymbol{C}
3-OMe	55.8	3.73 s	CH_3
4	102.2	6.41 s	CH
4a	147.6		C
6	90.9	5.50 d(6.3)	CH
6-OH		7.60 d(6.3)	
6*	90.9	5.50 bs	CH
6a	77.5	4.57 bs	CH
7a	160.0		C
8	95.1	6.10 d(2.3)	CH
9	165.6		C
9-OMe	55.7	3.65 s	CH_3
10	97.1	6.34 d(2.3)	CH
11	161.7		C
lla	103.8		C
12	188.5		C
12a	70.0		C
1'	100.8	4.70 d(7.2)	CH
2′	73.4		CH
3′	77.5		CH
4′	69.9		CH
5′	76.4		CH
6′	60.9		CH ₂

^{*} After D₂O-exchange.

the UV, IR and ¹H NMR data. Three quartets resonated at δ 55.7, 55.8 and 55.9. They were assigned to the carbons of the three methoxy-substituents at C-9, C-3 and C-2, respectively. The signal at δ 188.5 was attributed to the carbonyl carbon. The resonance of the aromatic moiety was assigned by DEPT (Table 1),

HETCOR (¹³C) (Table 2) and by comparison with data from the literature [4, 8, 13, 14]. The proposed structure 1 was fully supported by its ¹³C NMR spectrum, which exhibited peaks for 25 carbon atoms (Table 1).

On acid hydrolysis, compound 1 yielded clitoriacetal (2) [2] as the aglycone and glucose. UV and IR spectra of 2 were in accordance with those reported in the literature [2]. The ¹H NMR spectral data of 2 (Table 3) were identified from direct comparison with

Table 3. ¹H and ¹³C NMR data of compound 2 in CDCl₃

Attribution	2		
	δ ¹³ C	δ ¹ H (J = Hz)	DEPT
1	109.0	6.59 s	CH
la	107.6		C
2	144.8		C
2-OMe	56.3	3.82 s	CH_3
3	151.8		C
3-OMe	55.8	3.73 s	CH ₃
4	101.8	6.42 s	CH
4 a	147.9		C
6	91.5	5.52 d (6.4)	CH
6-OH		5.40 d (6.4)	
6*	91.5	5.52 bs	CH
6a	77.0	4.64 <i>bs</i>	CH
7a	160.8		C
8	94.6	5.87 d(2.3)	CH
9	169.0		C
9-OMe	55.8	3.78 s	CH_3
10	95.8	5.95 d(2.3)	CH
11	164.2		C
11-OH		11.72 s	C
11a	100.0		CH_3
12	194.1		C
12a	69.5		C

^{*} After D₂O-exchange.

Table 2. Summary of 2D-NMR correlations of compounds 1 and 2

Н	1 and 2 COSY (¹ H)	HETCOR (13C)	COSY (¹H) LR
1	4	I	
2-OMe		2-OMe	1
3-OMe		3-OMe	4
4	1	4	
6	6a	6	
6a	6	6a	
8	10	8	
9-OMe		9-OMe	8, 10
10	8	10	
	1		
1′	2′	1'	
2′			
3′			
4′		4′	
5′			
6′		6′	

Scheme 1. Mass spectral fragmentation of clitoriacetal (2).

those of clitoriacetal [2], COSY (1H) and COSY (1H) LR experiments (Table 2). The ¹³C NMR spectrum of 2 was consistent with the UV, IR and ¹H NMR data. Two quartets resonated at δ 55.8 (OMe-3 and OMe-9) and 56.3 (OMe-2). The signals at δ 69.5, 7.0 and 91.5 were assigned to C-12a, C-6a and C-6, respectively. The signal at δ 194.13 was attributed to the carbonyl carbon. The resonance of the aromatic moiety of clitoriacetal (2) was assigned by DEPT (Table 3), HETCOR (13C) (Table 2) and by comparison with data from the literature [4, 8, 13, 14]. The mass spectrum of 2 exhibited a $[M]^+$ at m/z 390, strongly suggesting that 2 is clitoriacetal. Confirmation of this observation was supported by the HR mass spectrum, which indicated a molecular formula C₁₉H₁₈O₉. Two prominent fragment ion peaks at m/z 167 and 224, due to cleavage between the B/C rings (Scheme 1), revealed that ring D possessed methoxyl and hydroxyl groups (m/z 167), and rings A and B had two methoxyl and two hydroxyl groups (m/z 224). Clitoriacetal (2) obtained from the hydrolysate indicated that 1 has a sugar moiety attached to C-11. The sugar was identified as glucose and its absolute configuration determined by GC of its TMSi (-)-2-butylglycosides [15]; D-glucose was detected. Consequently, on the basis of UV, IR, ¹H and ¹³C NMR spectroscopy, LSI mass spectrometry and chemical reactions, the structure of compound 1 was established as (+)-(6S, 6aR, 12aR)-6,12a-dihydroxy-2,3,9-trimethoxy-rotenoid 11- $O-\beta$ -D-glucopyranoside (clitoriacetal 11-O-β-D-glucopyranoside).

EXPERIMENTAL

General. Mps are uncorr. Optical rotations were measured at 20°. IR spectra were recorded in KBr discs. ¹H NMR were run at 200 MHz, CDCl₃ or DMSO-d₆, TMS as int. standard. ¹³C NMR-edited

DEPT spectra were obtained at 50 MHz from CDCl₃ or DMSO-d₆ solns. GC was carried out with FID, using a glass capillary column (0.31 mm×25 m) SE 30. EIMS was recorded at 70 eV. Negative LSIMS was carried out using HMPA-glycerol as matrix, 35 kV anodic voltage, 8 kV accelerating voltage using Cs ions. Silica gel columns (230–400 mesh ASTM, Merck) was used for CC. TLC was performed on silica gel (Merck) using the following solvent systems: (A) CHCl₃-MeOH (4:1) for rotenoid glycoside and (B) CHCl₃-MeOH (19:1) for rotenoid aglycone and © n-BuOH-pyridine-H₂O (6:4:3) for sugars. Compounds 1 and 2 were detected under UV of 254 and 366 nm and by spraying with orcinol-H₂SO₄. Sugar was detected by spraying with orcinol-H₂SO₄.

Plant material. Roots of C. fairchildiana Howard were collected at Ilha do Fundão, Rio de Janeiro, on February 1995, and identified by Luci Mendonça de Senna. A voucher specimen (no. R190871) is deposited at the Museu Nacional, Rio de Janeiro, Brazil

Isolation of constituents. Dried and powdered roots (200 g) were extracted with cold MeOH. Evapn of solvent gave a residue (56 g) a part of which (30 g) was submitted to CC (90×3 cm) on silica gel which was eluted with CHCl₃-MeOH mixts of increasing polarity (up to 10% MeOH). Frs eluted with CHCl₃-MeOH (9:1) yielded 1, TLC homogeneous compound (1.5 g), $R_f = 0.50$, dark blue colour with orcinol-H₂SO₄, negative FeCl₃ test and positive Rogers-Calamari tests.

Clitoriacetal 11-O-β-D-glucopyranoside (1). Amorphous powder (MeOH), mp 165–167° [α]_D+100° (DMSO; c 0.001). UV $\lambda_{\rm max}^{\rm EtOH}$ nm (log ε): 220 (4.40), 286 (4.33), 320 (sh, 3.80). IR $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 3436 (OH), 2939, 1670 (>C = O), 1613, 1573, 1511, 1445, 1412, 1343, 1268, 1220, 1170, 1148, 1101, 1076, 973, 955, 905, 864, 830, 781, 741. $^{\rm 1}$ H and $^{\rm 13}$ C NMR: Tables

1 and 2. Negative LSIMS, m/z (rel. int.): 551 [M-H]⁻ (53), 389 [M-163] (100).

Acid hydrolysis of 1. The compound (1, 100 mg) was hydrolysed with 1 M HCl in dioxane-H₂O (1:1, 10 ml) at room temp. for 10 min under N₂. After diluting with H₂O (90 ml), the aglycone was extracted with CHCl3-MeOH (4:1) and evapd to dryness in vacuo. The residue was dissolved in MeOH and the soln on concn yielded a white compound which on further recrystallization gave clitoriacetal (2, 53 mg), mp 130°. $[\alpha]_D + 258^\circ$ (CHCl₃, c 0.275), (lit [2] $[\alpha]_D + 259.5^\circ$ (CHCl₃, c 0.275). Clitoriacetal was analysed by silica gel-TLC in the above described system. After spraying with orcinol- H_2SO_4 it gave a yellow spot at $R_f = 0.45$. Compound 2 gave a brown colour with FeCl₃. UV $\hat{\lambda}_{\text{max}}^{\text{EiOH}}$ nm (log ε): 230 (4.43), 292 (4.30), 320 (sh, 3.88); (AlCl₃): 220 (4.95), 295 (4.00), 310 (sh, 4.08). IR $v_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3425 (OH), 3007, 2937, 2850, 1641 (>C = O), 1575, 1510, 1451, 1410, 1364, 1268, 1195, 1161, 1104, 1039, 975, 957, 907, 867, 830, 753. ¹H and ¹³C NMR: Table 3. EIMS (probe) 70 eV m/z (rel. int.): 390 [M] (18), 372 (3), 343 (9), 224 (100), 208 (39), 167 (92); HRMS found: $[M]^+$ 390.3490, $C_{19}H_{18}O_9$ requires 390.3493. The aq. layer was adjusted to pH 6 by addition of NaHCO₃. After lyophilization, sugar was extracted with pyridine and analysed by silica gel-TLC in the above described system. After spraying, glucose gave a blue spot at R_f 0.70. For co-chromatography glucose was used. The configuration of the sugar was established by capillary GC of its TMSi (−)-2-butyl glycosides [15].

Acknowledgements—The authors are grateful to Messrs Eduardo M.B. da Silva, Alexandre F. Neves and Mrs Maria C.P. Lima for recording spectra. Financial support from CNPq and CAPES is also acknowledged.

REFERENCES

- 1. Fantz, P. R., Economic Botany, 1991, 45, 511.
- Taguchi, H., Kanchanapee, P. and Amatayakul, T., Chemical and Pharmaceutical Bulletin, 1977, 25, 1026.
- Thai Medicinal Plant Research Project, Division of Medical Research in the Department of Medical Sciences, 1972, 14, 1.
- Lin, L. J., Ruangrungsi, N., Cordell, G. A., Shieh, H. L., You, M. and Pezzuto, J. M., Phytochemistry, 1992, 31, 4329.
- Rizzini, C. T., Arquivos do Jardim Botânico do Rio de Janeiro, 1973, XVII, 171.
- Rogers, H. D. and Calamari, J. A., Industrial Engineering and Chemistry, Analytical Edition, 1936, 8, 135.
- Krupadanam, G. L. D., Sarma, P. N., Srimannarayana, G. and Rao, N. V. S., Tetrahedron Letters, 1977, 2125.
- Shawl, A. S., Mengi, N., Misra, L. N. and Vishwapaul, *Phytochemistry*, 1988, 27, 3331.
- 9. Crombie, L. and Lown, J. W., Journal of the Chemical Society, 1962, 775.
- Büchi, G., Crombie, L., Godin, P. J., Kaltenbronn, J. S., Siddalingaiah, K. S. and Whiting, D. A., Journal of the Chemical Society, 1961, 2843.
- Carlson, D. G., Weisleder, D. and Tallent, W. H., Tetrahedron, 1973, 29, 2731.
- 12. Inouye, H., Tobita, S., Akiyama, Y., Ito, K. and Shingu, T., *Chemical and Pharmaceutical Bulletin*, 1973, **21**, 846.
- Messana, I., Ferrari, F. and Sant'Ana, A. E. G., *Phytochemistry*, 1986, 25, 2688.
- Crombie, L., Kilbee, G. W. and Whiting, D. A., Journal of the Chemical Society, Perkin Trans-actions 1, 1975, 1497.
- 15. Gerwig, G. J., Kamerling, J. P. and Vliegenthart, J. F. G., *Carbohydrate Research*, 1978, **62**, 349.