



# DEPENDENCE OF THE $^1\text{H}$ NMR CHEMICAL SHIFTS OF RING F RESONANCES ON THE ORIENTATION OF THE 27-METHYL GROUP OF SPIROSTANE-TYPE STEROIDAL SAPOGENINS\*

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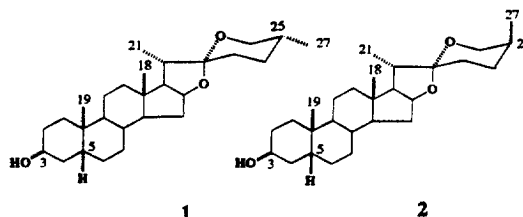
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**Key Word Index**—Steroidal sapogenin; 2D NMR, 25*R*/25*S* stereochemical assignments.

**Abstract**—A relationship between the  $^1\text{H}$  NMR chemical shifts of the ring F resonances and orientation of the  $\text{H}_3$ -27 group has been derived for the establishment of 25*R*- and 25*S*-stereochemistry in spirostane type of steroidal sapogenins. © 1997 Elsevier Science Ltd

## INTRODUCTION

The steroidal sapogenins and their glycosides (steroidal saponins), widely distributed in various plant families [2], are attracting attention, as some have shown biological activities as inhibition to platelet aggregation [3], reduction of blood glucose level [4], local antitussive effect for dry cough [5], molluscicidal activities [6] and inhibition of the proliferation of various kinds of human malignant tumor cells *in vitro* [7]. The variation in spirostane structure generally arises from the configuration at C-5, i.e. A/B ring junction, and the configuration of the methyl group at C-25. Depending upon the equatorial or axial orientation of the 27-methyl group, the spirostanoids have been grouped in to 25*R* or 25*S* types, respectively. The question of the establishment of 25*R*/25*S* stereochemistry has been earlier addressed by the consideration of IR absorption bands [8, 9], 27-methyl group  $^1\text{H}$  NMR chemical shift [10] and  $^{13}\text{C}$  NMR chemical shifts [11, 12]. Although, the  $^{13}\text{C}$  NMR approach has been widely employed for the stereochemical establishment of these compounds [13] the relative insensitivity of  $^{13}\text{C}$  NMR as compared with  $^1\text{H}$  NMR encourages the search for a relationship between the 25*R*/25*S*-stereochemistry and  $^1\text{H}$  NMR data.



## RESULTS AND DISCUSSION

To derive a primary set of  $^1\text{H}$  NMR data, a pair of epimeric 5β-spirostanes: smilagenin (25*R*-5β-spirostan-3β-ol (1)) and sarsasapogenin (25*S*-5β-spirostan-3β-ol (2)) were investigated by two-dimensional techniques [14].  $^1\text{H}$ - $^1\text{H}$  COSY and TOCSY led to the partial assignment of the  $^1\text{H}$  NMR resonances, which were correlated with the corresponding one-bond coupled  $^{13}\text{C}$  signals using HMQC. A combined application of HMQC-TOCSY, HSQC-RELAY and HMBC experiments further led to prove  $^1\text{H}$  NMR identification of the remaining  $^1\text{H}$  assignments as well as of the quaternary carbons. The  $^{13}\text{C}$  assignments of 2 were verified by INADEQUATE experiments. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR assignments for 1 and 2 are listed in Table 1.

Since spectral data for both compounds were measured under the same experimental conditions, any alteration in the  $^1\text{H}$  NMR shielding pattern must be due to the difference in  $\text{H}_3$ -27 stereochemistry. A comparison of the  $^1\text{H}$  NMR data for 1 and 2 (Table 1) reveals that the  $^1\text{H}$  NMR resonances exhibit very similar  $^1\text{H}$  NMR chemical shifts for all of the rings except

\* Part 46 in the series, 'NMR spectral Investigations.' For part 45 see ref. [1].

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Table 1.  $^{13}\text{C}$  and  $^1\text{H}$  NMR chemical shift assignments of smilagenin 1 and sarsasapogenin 2

| Carbon | $^{13}\text{C}$ | 1<br>$^1\text{H}$ | 2<br>$^{13}\text{C}$ | $^1\text{H}$ |
|--------|-----------------|-------------------|----------------------|--------------|
| 1      | 29.95           | 1.40, 1.52        | 29.95                | 1.39, 1.50   |
| 2      | 27.82           | 1.40, 1.58        | 27.79                | 1.40, 1.50   |
| 3      | 67.12           | 4.11              | 67.12                | 4.11         |
| 4      | 33.54           | 1.33, 1.98        | 33.51                | 1.32, 1.97   |
| 5      | 36.58           | 1.72              | 36.51                | 1.72         |
| 6      | 26.56           | 1.16, 1.91        | 26.56                | 1.15, 1.90   |
| 7      | 26.54           | 1.07, 1.58        | 26.54                | 1.04, 1.39   |
| 8      | 35.28           | 1.58              | 35.27                | 1.58         |
| 9      | 39.85           | 1.33              | 39.85                | 1.32         |
| 10     | 35.28           | —                 | 35.28                | —            |
| 11     | 20.90           | 1.25, 1.40        | 20.90                | 1.25, 1.39   |
| 12     | 40.30           | 1.16, 1.72        | 40.31                | 1.15, 1.72   |
| 13     | 40.70           | —                 | 40.70                | —            |
| 14     | 56.48           | 1.16              | 56.47                | 1.15         |
| 15     | 31.80           | 1.25, 1.98        | 31.74                | 1.25, 1.97   |
| 16     | 80.93           | 4.40              | 81.02                | 4.40         |
| 17     | 62.27           | 1.77              | 62.09                | 1.75         |
| 18     | 16.49           | 0.76              | 16.50                | 0.76         |
| 19     | 23.92           | 0.98              | 23.92                | 0.97         |
| 20     | 41.61           | 1.86              | 42.12                | 1.81         |
| 21     | 14.51           | 0.97              | 14.34                | 0.99         |
| 22     | 109.26          | —                 | 109.74               | —            |
| 23     | 341.39          | 1.58, 1.67        | 25.94                | 1.39, 1.87   |
| 24     | 28.80           | 1.46, 1.64        | 25.77                | 2.02, 1.39   |
| 25     | 30.31           | 1.64              | 27.08                | 1.69         |
| 26     | 66.86           | 3.48, 3.38        | 66.13                | 3.95, 3.28   |
| 27     | 17.14           | 0.79              | 16.05                | 1.08         |

ring F. The most noteworthy feature is that the equatorial protons of H<sub>2</sub>-23, H<sub>2</sub>-24 and H<sub>2</sub>-26 resonate 0.2 to 0.5 ppm to lower field in **1** than in **2**, the axial protons resonate 0.07 to 0.19 ppm to higher field (Fig.

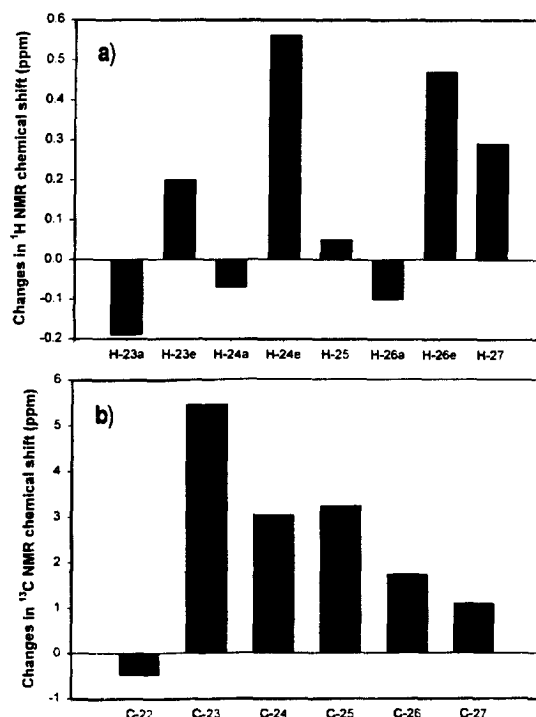


Fig. 1. Changes in the (a)  $^1\text{H}$  and (b)  $^{13}\text{C}$  NMR chemical shifts for ring-F resonances. The bars represents the difference in chemical shifts ( $\Delta\delta = \delta_1 - \delta_2$ ).

1). This implies that an axial orientation of the 27-Me group is reflected not only in its own appearance at 0.29 ppm to lower field, but also in the dispersion of the  $^1\text{H}$  NMR chemical shifts of the methylene protons occupying  $\beta$  and  $\gamma$ -positions [Fig. 1(a)]. In general, the difference between the chemical shifts for geminal protons at positions C-24 and C-26 was three to four times higher in 25*S*-compounds than in 25*R*-epimers.

Table 2.  $^1\text{H}$  NMR chemical shift data for ring F resonances for some steroidal sapogenins and steroidal saponins<sup>+</sup>

|                    | A <sup>+</sup><br>[18] | B <sup>+</sup><br>[19] | C <sup>+</sup><br>[20] | D <sup>+</sup><br>[21] | E <sup>+</sup><br>[22] | F <sup>+</sup><br>[23] | G <sup>+</sup><br>[15] | H <sup>+</sup><br>[16] | I <sup>+</sup><br>[17] | J <sup>+</sup><br>[17] | K <sup>+</sup><br>[24] | L <sup>+</sup><br>[25] |
|--------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|
| H <sub>2</sub> -23 |                        |                        |                        |                        |                        | 1.60                   | 1.58                   | 1.58                   |                        |                        |                        | 1.96                   |
|                    |                        |                        |                        |                        |                        | 1.72                   | 1.68                   | 1.68                   |                        |                        |                        | 1.39                   |
| H <sub>2</sub> -24 |                        |                        |                        |                        |                        | 1.55                   | 1.56                   | 1.64                   |                        |                        |                        | 2.07                   |
|                    |                        |                        |                        |                        |                        | 1.55                   | 1.42                   | 1.43                   |                        |                        |                        | 1.46                   |
| H-25               |                        |                        |                        |                        |                        | 1.56                   | 1.62                   | 1.61                   |                        |                        |                        | 1.72                   |
| H <sub>2</sub> -26 | 3.58                   | 3.59                   | 3.56                   | 3.59                   | 3.52                   | 3.57                   | 3.35                   | 3.38                   | 4.07                   | 4.05                   | 4.04                   | 3.97                   |
|                    | 3.50                   | 3.50                   | 3.47                   | 3.52                   | 3.52                   | 3.46                   | 3.42                   | 3.36                   | 3.38                   | 3.37                   | 3.35                   | 3.31                   |
| H <sub>2</sub> -27 | 0.70                   | 0.70                   | 0.70                   | 0.70                   | 0.69                   | 0.71                   | 0.80                   | 0.76                   | 1.08                   | 1.08                   | 1.06                   | 1.13                   |

<sup>+</sup> A (25*R*)-5 $\alpha$ -spirostane-3 $\beta$ -ol (Tigogenin) 3-*O*-{*O*- $\alpha$ -L-rhamnopyranosyl-(1  $\rightarrow$  2)-*O*- $\beta$ -D-xylopyranosyl-(1  $\rightarrow$  2)-*O*-[ $\beta$ -D-xylopyranosyl-(1  $\rightarrow$  3)]-*O*- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  4)- $\beta$ -D-galactopyranoside}; B (25*R*)-5 $\alpha$ -spirostane-2 $\alpha$ ,3 $\beta$ -diol (gitogenin); C (25*R*)-5 $\alpha$ -spirostane-2 $\alpha$ ,3 $\beta$ -diol (gitogenin)-3-*O*- $\beta$ -D-galactopyranoside; D (25*R*)-5 $\alpha$ -spirostane-1 $\beta$ ,3 $\beta$ -diol (Brisbagenin); E (25*R*)-5 $\alpha$ -spirostane-3 $\beta$ , 17-diol (Penogenin) 3-*O*-{*O*- $\alpha$ -L-rhamnopyranosyl-(1  $\rightarrow$  2)-*O*-[ $\beta$ -D-galactopyranosyl-(1  $\rightarrow$  3)- $\beta$ -D-glucopyranoside]; F (25*R*)-2 $\alpha$ ,3 $\beta$ -dihydroxy-5 $\alpha$ -spirost-9(11)-en-12-one-3-*O*-{*O*- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  2)-*O*-[*O*- $\alpha$ -L-rhamnopyranosyl-(1  $\rightarrow$  4)- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  3)]-*O*- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  4)-*O*- $\beta$ -D-galactopyranoside}; G (25*R*)-3 $\beta$ -acetoxy-5 $\alpha$ -spirost-12-one (Hecogenin acetate); H (25*R*) spirost-5-ene-3 $\beta$ -ol (Diosgenin); I (25*S*)-5 $\alpha$ -spirostane-3 $\beta$ -ol (Neotigogenin); J (25*S*)-spirost-5-ene-2 $\alpha$ ,3 $\beta$ -diol (Neogitogenin); K (25*S*) spirost-5-ene-1 $\beta$ ,3 $\beta$ -diol(25*S*)-ruscogenin-1-*O*-{*O*- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  3)-*O*- $\alpha$ -L-rhamnopyranosyl-(1  $\rightarrow$  2)-*O*- $\beta$ -D-glucopyranoside}; L (25*S*)spirost-5-ene-1 $\beta$ ,3 $\beta$ -diol (25*S*)-ruscogenin-3-*O*-{*O*- $\beta$ -D-fucopyranosyl-(1  $\rightarrow$  3)-*O*- $\alpha$ -L-rhamnopyranoside}.

A literature search [13, 15–25] showed that the chemical shifts for  $\text{H}_{2-24}$  have not been frequently reported but the reported values for  $\Delta\delta_{\text{H}_{26e}-\text{H}_{26a}}$  were consistent with the above observations (Table 2). A similar comparison of the  $^{13}\text{C}$  NMR chemical shifts for **1** and **2** [Fig. 1(b)] showed that all the carbon resonances except C-22 occur at lower field in **1** than in **2**.

It is worthwhile to mention that substitution in rings A–D does not usually affect the chemical shifts of ring F. However, substitution in ring F will modify the  $^1\text{H}$  NMR shielding pattern, so the  $^1\text{H}$  NMR shielding behaviour described will be of general applicability for ring-F unsubstituted  $22\alpha$ -spirostanoids.

#### EXPERIMENTAL

The 1D and 2D NMR experiments were carried out in  $\text{CDCl}_3$  in 5 mm tube at ambient temp. ( $20^\circ$ ) with an indirect detection probe on a Varian Unity 500 MHz NMR spectrometer. The chemical shifts are expressed on the  $\delta$  scale.

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