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# A SECOIRIDOID AND OTHER CONSTITUENTS OF GONOCARYUM CALLERYANUM

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**Key Word Index**—Gonocaryum calleryanum; Icacinaceae; secoiridoid glycoside; gonocaryoside E; kingiside derivative.

Abstract—Investigation of the leaves, branch, stem and root bark of Gonocaryum calleryanum resulted in the isolation of a new secoiridoid glycoside, gonocaryoside E, together with 14 known compounds. Their structures were elucidated by spectral analysis and chemical transformation. Gonocaryoside E was shown to be a derivative of kingiside in which the lactone ring was open, and with the 8-hydroxy group esterified with tiglic acid. Alkaline hydrolysis of these secoiridoids is discussed. © 1998 Published by Elsevier Science Ltd. All rights reserved

#### INTRODUCTION

Gonocaryum callervanum (Baill) Becc is a small evergreen tree which is widely distributed from Indonesia to the Philippines to Taiwan. This plant is the only species of Gonocaryum native to the southern part of Taiwan [1]. The leaves of this plant are used for the treatment of stomach disease in Philippine folk medicine [2]. However, it is apparently not used in Taiwan folk medicine. Kaneko et al. reported the isolation of secoiridoid glycosides and flavonoid glycosides from the leaves of this plant [3]. We were interested to investigate the constituents of G. callervanum due to the lack of phytochemical work on the plants of this genus. We describe herein the isolation and structure determination of a new secoiridoid glycoside. gonocarvoside-E (1a), together with 14 known compounds from leaves, branch, stem and root bark of G. calleryanum.

## RESULTS AND DISCUSSION

Gonocaryoside-E (1a) was isolated as an optically active amorphous powder. It gave rise to a quasimolecular ion peak at m/z 505 [M+H]<sup>+</sup> on FAB-mass spectrometry. High mass measurement established the molecular formula as  $C_{22}H_{32}O_{13}$ . The UV spectrum of compound 1a showed an intense absorption band at 221.8 nm characteristic of a conjugated

carbonyl function. Furthermore, its <sup>1</sup>H and <sup>13</sup>C NMR spectra displayed a carbomethoxyl signal at  $\delta_{\rm H}$  3.57 ( $\delta_C$  166.3 and 50.3) and a trisubstituted double bond at  $\delta_{\rm H}$  7.73 ( $\delta_{\rm C}$  152.3, 109.5) and a ketal signal at  $\delta_{\rm H}$ 6.28 ( $\delta_{\rm C}$  95.3). These data together with a positive H<sub>2</sub>SO<sub>4</sub>-vanillin reaction suggested 1a to be an iridoid. The <sup>1</sup>H NMR signals at  $\delta$  7.06 (q, I = 7.0 Hz), 1.88 (3H, s), and 1.62 (3H, d, J = 7.0 Hz) together with those in the  $^{13}$ C NMR spectrum at  $\delta$  166.5 (s), 137.0 (d), 128.5 (s), 13.5 (q) and 11.5 (q) suggested the presence of a tiglic acid moiety in the molecule. On alkaline hydrolysis, 1a afforded two (1c and 18a) hydrolysis products of which one (1c) was one of the alkaline hydrolysis products (1b, 1c, 1d and 18b) of gonocaryoside-A (2) which was isolated from the same plant as a major product. 18a was identified as tiglic acid by comparison of its spectral data. The connection between 1c and 18a as well as the full 13C NMR and <sup>1</sup>H NMR chemical shift assignments were made by means of COSY, HMQC, NOESY (Fig. 1)

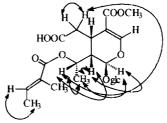


Fig. 1. NOESY spectrum of Gonocaryoside E.

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Table 1. Alkaline hy	drolysis of secoiridoids

Compound	Reagent	Temp.	Time	Product
2	10% NaOH	60–70°C	l hr	1b
2	5% KOH	RT	20 hr	1b
1 b	5% KOH	RΤ	3 day	no reaction
1b	5% KOH	60–70°C	8 hr	no reaction
1b	10% KOH	60–70°C	7 hr	no reaction
1 b	10% NaOH	60−70°C	7 hr	no reaction
1b	5% KOH, HMPA	RT	overnight	1c, 1d, 18b
1a	5% KOH, HMPA	RT	overnight	1c. 18a

Table 2. <sup>2</sup>J, <sup>3</sup>J-correlation of HMBC of Gonocaryoside E

C	$\delta$ , ppm	Н
l	95.3	3 (7.73), 5 (3.80), 8 (5.68), G-1 (5.41)
3	152.3	1 (6.28), 5 (3.80)
4	109.5	3 (7.73), 5 (3.80), 6 (3.10), 6 (2.86)
5	28.9	1 (6.28), 3 (7.73), 6 (3.10), 6 (2.86)
6	34.9	5 (3.80)
7	174.2	5 (3.80), 6 (3.10), 6 (2.86)
8	68.8	1 (6.28), 5 (3.80), 10 (1.47)
9	<b>4</b> 2.7	5 (3.80), 6 (3.10), 10 (1.47)
11	166.3	3 (7.73), OCH <sub>3</sub> (3.57)
1′	166.5	3' (7.06), 4' (1.88), 5' (1.62)
2′	128.5	3' (7.06), 4' (1.88), 5' (1.62)
3′	137.0	4' (1.88), 5' (1.62)
4′	13.5	3' (7.06)
5′	11.5	3' (7.06), 4' (1.88)
G-l	99.8	1 (6.28), G-2 (4.03)
G-2	73.9	G-3 (4.24)
G-3	77.6	G-2 (4.03)
G-4	70.6	G-2 (4.03), G-3 (4.24), G-6 (4.50), G-6 (4.34)
<b>G-</b> 5	77.9	G-1 (5.41), G-4 (4.20)
G-6	61.9	G-4 (4.20)

and HMBC (Table 2). The  $^{1}H^{-13}C$  long range correlation between the H-8 ( $\delta$  5.68) and a C-1′ carbonyl carbon signal at  $\delta$  166.5 indicated that the tiglic acid (**18a**) must be connected with the hydroxyl group at C-8 of **1c**. Based on the above results, the structure of gonocaryoside E was assigned as **1a**.

In our hands, alkaline hydrolysis of 2 was more difficult than reported by Kaneko's group [3]. Using their conditions for 20 hours, we obtained only partial hydrolysis to 1b (Table 1. Entry 2). This result could not be improved by increasing the alkaline concentration or the temperature. However, when HMPA was added to the solution of 1b, the products, 1c. 1d and 18b were successfully obtained.

The known compounds, gonocaryoside A (2) [3], -B (3) [3],  $\beta$ -amyrin acetate (4) [4], oleanolic aldehyde-3-acetate (5) [5],  $\beta$ -amyrin (6) [6], erythrodiol-3-acetate (7) [7], oleanolic acid (8) [8], 3-acetyl oleanolic acid (9) [5, 7].  $\beta$ -sitosteryl- $\beta$ -D-glucoside (11a) [6], stigmasteryl- $\beta$ -D-glucoside (11b) [6], ursolic acid (10) [9], 5-hydroxymethyl-2-furaldehyde (12) [10] and betu-

nilic acid (15) [8] were isolated and characterized from the root and stem bark of G. calleryanum. Gonocaryoside A (2) [3],  $\beta$ -amyrin (6) [6], oleanolic acid (8) [8],  $\alpha$ -amyrin (13) [11] and lupeol (14) [6, 12] were isolated and identified from the branch of the same plant. On the other hand, gonocaryoside A (2) [3],  $\beta$ -amyrin (6) [6],  $\alpha$ -amyrin (13) [11], lupeol (14) [6, 12], ursolic acid (10) [9], methylparaben (16) [12], p-hydroxybenzaldehyde (17) [12], 2S, 3S-angliceric acid (18b) [3], cis-methylcaffeate (19) [13] and trans-methyl caffeate (20) [13] were also isolated from the leaves of the same plant. This is the first report of the separation of  $\alpha$ -amyrin (13),  $\beta$ -amyrin (6) and lupeol (14) by HPLC.

#### EXPERIMENTAL

## General

UV: MeOH; IR: CHCl<sub>3</sub>, except where noted; <sup>1</sup>H (400 or 200 MHz) and <sup>13</sup>C (100 or 50 MHz) NMR: TMS as int. standard; CC: silica gel (70–230 mesh, 230–400 mesh, Merck), highly porous polymer resin (Diaion HP-20, Mitsubishi Chem. Ind., Japan) and Sephadex LH-20 (Pharmacia Fine Chem.); HPLC: KR-100-5-C18 (4.6 mm × 25 cm, Kromasil Sweden). The mobile phase was MeOH at a flow-rate of 1.0 ml/min. The UV detector wavelength was set at 210 nm. All solvent systems for chromatography were homogeneous.

## Plant material

The bark of stem and root, leaves and branch of G. calleryanum (Baill) Becc. were collected from Hengchun Peninsula, Pingtung Hsien, Taiwan in December, 1990. The plant was identified by Prof. Chang-Shen Kuoh of this University. A voucher specimen was deposited in the Herbarium of Cheng Kung University, Tainan, Taiwan, R.O.C.

## Extraction and separation

A. The dried stem and root bark (900 g) were extracted with hot MeOH ( $\times$ 3). The combined

MeOH extract was concd in vacuo to give a brown syrup which was partitioned between H<sub>2</sub>O and CHCl<sub>3</sub>. The CHCl<sub>3</sub> layer (23 g) was fractionated on a silica gel column eluted with a gradient of benzene and Me2CO to afford 20 frs. Each fr. was rechromatographed on silica gel and prep. TLC or subjected to recrystallization. Pure compounds were obtained as follows. Frs. 3 and 4 gave 4 (158 mg). Frs. 5, 6 and 7 gave 5 (95 mg). Fr. 9–13 gave 6 (17.6 g) and 7 (81 mg). Frs. 15 and 16 gave 7 (5 mg) and 8 (2 mg), respectively. Fr. 20 gave 11a and 11b mixture (18 mg), 9 (42 mg), 8 and 10 mixture (52 mg), successively. The H<sub>2</sub>O layer was concd in vacuo to give a residue (227 g) which was chromatographed on silica gel column eluted with a gradient of CHCl3 and MeOH to afford frs. Frs. 1-4 were combined and rechromatographed on a silica gel column eluted with nhexane-EtOAc (5:1) to give 6 (18 mg). Similarly, the combined frs. 5-11 eluting with CHCl<sub>3</sub>-MeOH (9:1) gave 15 (1 mg). 11a and 11b mixture (36 mg), 8 and 10 mixture (16 mg). Frs. 12-15 were also combined and rechromatographed over a silica gel column eluted with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (100:20:1) to give 2 (25.1 g) and 1a (32 mg). Frs. 16 and 17 were chromatographed on Diaion HP-20 eluting with a gradient of  $H_2O$  and MeOH to give 12 (8 mg), 3 (0.3 g) and 2 (5.6 g), successively.

B. The dried and powdered branch (190 g) was extracted with hot MeOH (×3). The combined MeOH extracts were concd. under red. press. to give a brown syrup which was directly subjected to silica gel column CC and eluted with a gradient of CHCl<sub>3</sub> and MeOH to afford 6 frs. Fr. 2 was separated by HPLC using MeOH as eluent to give 14 (82 mg), 6 (11 mg) and 13 (28 mg), successively. Frs. 3 and 4 gave 8 (284 mg). Frs. 5 and 6 afforded 2 (5.2 g).

C. The powdered leaves (360 g) was extracted with hot MeOH ( $\times$ 3). After concentration of the combined extracts in vacuo, the residue was chromatographed on a silica gel column developed with C<sub>6</sub>H<sub>6</sub>, CHCl<sub>3</sub>, CHCl<sub>3</sub>-MeOH (9:1), CHCl<sub>3</sub>-MeOH (5:1) and MeOH to give Frs. 1-12, 13-19, 20-24, 25-30 and 31, successively. Frs. 14-17 gave a mixture of triterpenoids which was separated by HPLC on RP-18 (MeOH) to afford 14 (68 mg, 19.08 min), 6 (8 mg, 24.21 min) and 13 (31 mg, 27.83 min), respectively. Frs. 20-22 gave 16 (5 mg). Frs. 24 was rechromatographed on silica gel using n-hexane-Me<sub>2</sub>CO (5:1) to give unknown c (3 mg). Frs. 25-27 afforded 10 (26 mg). Frs. 28 was purified by silica gel, sephadex LH-20 and prep. TLC to give 17 (4 mg), 2 (0.3 g), 18b (13 mg), a mixture of **19** and **20** (12 mg), **3** (81 mg), successively. Frs. 31 gave 2 (2.3 g). Gonocaryoside E (1a), colorless amorphous powder,  $[\alpha]_D = 49.68^{\circ}$  (c, 1.263, MeOH). FAB-HRMS: m/z 505.1923 ([M+H]  $(C_{12}H_{33}O_{13})$ , requires: m/z 505.1921); UV  $\lambda_{max}$  nm: 221.8, 280.6, 320.3; IR  $v_{\text{max}}$  cm<sup>-1</sup>: 3404, 1713, 1640, 1505; FAB-MS (rel. int.) m/z: 505 ([M+H]<sup>+</sup>, 2.8), 460 (4), 391 (14), 307 (25), 289 (12), 154 (100), 137 (72), and 136 (71); <sup>1</sup>H NMR ( $C_5D_5N$ ):  $\delta$  7.73 (1H, s,

H-3), 7.06 (1H, q, J = 7.0 Hz, H-3'), 6.28 (1H, d, J = 6.6 Hz, H-1), 5.68 (1H, m, H-8), 5.41 (1H, d. J = 7.8 Hz, G-1), 4.50 (1H, dd, J = 11.8, 2.2 Hz, G-6), 4.34 (1H, dd, J = 11.8, 5.2 Hz, G-6), 4.24 (1H, t, J = 8.9 Hz, G-3), 4.20 (1H, t, J = 8.9 Hz, G-4), 4.03 (1H, t, J = 8.9 Hz, G-2), 3.96 (1H, m, G-5), 3.80 (1H, m, G-5),m, H-5), 3.57 (3H, s, 11-OMe), 3.10 (1H, dd, J = 16.2, 4.7 Hz, H-6), 2.86 (1H, dd, J = 16.2, 7.8 Hz, H-6), 2.56 (1H, m, H-9), 1.88 (3H, s, 5'-Me), 1.62 (3H, d, J = 7.0 Hz, 4'-Me, 1.47 (3H, d, J = 6.4 Hz, 10-Me); <sup>13</sup>C NMR ( $C_5D_5N$ ):  $\delta$  174.2 (C-7), 166.5 (C-1'), 166.3 (C-11). 152.3 (C-3), 137.0 (C-3'), 128.5 (C-2'), 109.5 (C-4), 99.8 (G-1), 95.3 (C-1), 77.9 (G-5), 77.6 (G-3), 73.9 (G-2), 70.6 (G-4), 68.8 (C-8), 61.9 (G-6), 50.3 (OMe), 42.7 (C-9), 34.9 (C-6), 28.9 (C-5), 18.6 (C-10), 13.5 (C-4'), 11.5 (C-5').

## Alkaline hydrolysis of compound 2

A. A soln of 2 (500 mg) in 5% aq. KOH (12.5 ml) was allowed to stand for 20 hr at room temp. The reaction mixture was neutralized with Ion exchange I (Merck) ion-exchange resin. After removal of the solvent, the residue was purified by  $C_{18}$  CC using  $H_2O$ as eluent to give 1b (426 mg). Compound 1b: colorless amorphous powder,  $[\alpha]_D - 123.9^\circ$  (c, 0.4, MeOH). UV  $\lambda_{\text{max}}$  nm: 231.4; IR  $\nu_{\text{max}}$  cm<sup>-1</sup>: 3404, 1720, 1645, 1595, 1515; FAB-MS (m/z): 525 ([M+H]<sup>+</sup> 1.3), 413 (27), 195 (20), 173 (20), 157 (33), 131 (50), 115 (100): <sup>1</sup>H NMR  $(C_5D_5N)$ :  $\delta$  7.90 (1H, s, H-3), 6.00 (1H, d, J = 5.6 Hz, H-1), 5.35 (1H, d, J = 7.8 Hz, G-1), 4.72 (1H, m, H-8), 4.50 (1H, br d, J = 11.6 Hz, G-6), 4.354.20 (4H, m, H-3', G-6, G-3, G-4), 4.07 (1H, t, J = 7.8)Hz, G-2), 3.98 (1H, m, G-5), 3.60 (1H, m, H-5), 3.28 (1H, dd, J = 16.4, 7.2 Hz, H-6), 3.12 (1H, dd, J = 16.4)4.0 Hz, H-6), 2.51 (1H, m, H-9), 1.83 (3H, s, 5'-Me), 1.58 (3H, d, J = 7.0 Hz, 4'-Me), 1.50 (3H, d, J = 6.0Hz, 10-Me);  ${}^{13}$ C NMR ( $C_5D_5N$ ):  $\delta$  17.6 (C-10), 17.9 (C-4'). 22.3 (C-5'), 26.9 (C-5), 33.7 (C-6), 38.8 (C-9), 62.1 (C-6"), 70.9 (C-4"), 72.8 (C-3'), 74.2 (C-2"), 74.6 (C-8), 77.8 (C-3"), 78.1 (C-5"), 80.2 (C-2'), 93.4 (C-1), 100.3 (C-1"), 111.4 (C-4), 152.0 (C-3), 168.5 (C-11), 171.3 (C-7), 180.4 (C-1').

B. A soln. of **1b** (544 mg) in 5% aq. KOH (25 ml) and 15 drops HMPA was stirred at room temp overnight. The reaction mixture was treated as just described to give **1c** (356 mg), **1d** (16 mg) and **18b** (126 mg) which were identified by comparison of their spectral data with literature values [3].

## Alkaline hydrolysis of compound 1a

A. A soln. of **1a** (20 mg) in 5% aq. KOH (2 ml) and 1 drop HMPA was stirred at room temp overnight. The reaction mixture was treated as above to give **1c** (92 mg) and **18a** (26 mg) which were identified by comparison of their spectral data with literature values [3, 14]. Compound **1c**: colorless amorphous powder,  $[\alpha]_D$  118.17° (c, 0.5, MeOH). UV  $\lambda_{max}$  nm: 231.8: IR  $v_{max}$  cm<sup>-1</sup>: 3395, 1718, 1560, 1546, 1400,

1259, 1076; FAB-MS  $(m_f z)$ : 391  $([M+H]^+, 14)$ , 229 (30), 155 (28), 154 (100), 139 (22), 138 (42), 137 (88), 136 (85), 125 (28), 123 (20), 107 (38);  ${}^{1}H$  NMR (D<sub>2</sub>O):  $\delta$  7.43 (1H, s, H-3), 5.57 (1H, d, J = 4.8 Hz, H-1), 4.70 (1H, m. H-8), 4.60 (1H, d, J = 8.8 Hz, G-1), 3.78 (1H, m. H-8)dd, J = 12.2, 1.4 Hz, G-6), 3.58 (1H, dd, J = 12.2, 5.6 Hz, G-6), 3.35 (1H, m, G-5), 3.32 (1H, t, J = 8.8 Hz, G-3), 3.24 (1H, t, J = 8.8 Hz, G-4), 3.21 (1H, m, H-5). 3.16 (1H, t, J = 8.8 Hz, G-2), 2.93 (1H, dd, J = 16.6, 6.6 Hz, H-6, 2.64 (1H, dd, J = 16.6, 4.8 Hz, H-6), 2.49 (1H, m, H-9), 1.36 (3H, d, J = 6.8 Hz, 10-Me);  ${}^{13}$ C NMR (D<sub>2</sub>O):  $\delta$  17.5 (C-10), 26.6 (C-5), 33.8 (C-6), 39.0 (C-9), 61.5 (C-6'), 70.4 (C-4'), 73.5 (C-2'), 76.5 (C-3'), 76.6 (C-8), 77.2 (C-5'), 94.4 (C-1), 99.2 (C-1'), 111.2 (C-4), 154.8 (C-3), 170.7 (C-11), 177.2 (C-7). Compound 1d: colorless amorphous powder,  $[\alpha]_D$  – 14.69° (c, 0.8, MeOH). UV  $\lambda_{max}$  nm : 225.6; 1R  $v_{\text{max}} \text{ cm}^{-1}$ : 3419, 1731, 1651, 1564, 1384, 1076; FAB-MS m/z: 391 ([M+H]<sup>+</sup> 12), 176 (56), 167 (21), 154 (52), 149 (100), 138 (28), 137 (50), 136 (59), 123 (20), 113 (42), 109 (21), 107 (38), 105 (29);  ${}^{1}H$  NMR (D<sub>2</sub>O):  $\delta$  6.92 (1H, s, H-3), 5.64 (1H, d, J = 5.4 Hz, H-1), 4.79 (1H, d, J = 8.0 Hz, G-1), 4.08 (1H, m, H-8), 3.87 (1H, m, H-8)dd, J = 12.5, 2.0 Hz, G-6). 3.68 (1H, dd, J = 12.5. 5.7 Hz, G-6), 3.46 (2H, m, G-5, G-3), 3.37 (1H, t, J = 9.3Hz, G-4), 3.30 (1H, t, J = 9.3 Hz, G-2), 3.14 (1H, m. H-5), 2 58 (1H, dd, J = 15.8, 4.2 Hz, H-6), 2.23 (1H, dd, J = 15.8, 10.4 Hz, H-6), 2.01 (1H, q, J = 5.5 Hz, H-9), 1.25 (3H, d, J = 6.6 Hz, 10-Me); <sup>13</sup>C NMR  $(D_2O)$ :  $\delta$  20.9 (C-10), 30.3 (C-5), 38.1 (C-6), 43.8 (C-9), 60.6 (C-6'), 65.9 (C-8), 69.6 (C-4'), 72.7 (C-2'), 75.5 (C-3'), 76.3 (C-5'), 95.5 (C-1), 98.5 (C-1'), 117.4 (C-

4), 145.9 (C-3), 175.5 (C-11), 181.3 (C-7). Compound **18b**: colorless amorphous powder,  $[\alpha]_D - 2.25^\circ$  (c, 3.67, MeOH). IR  $v_{\text{max}}$  cm<sup>-</sup>: 3360, 1732, 1558. 1541; FAB-MS m/z: 135 ([M+H]] $^\circ$ , 2.7), 115 (100);  $^\circ$ H NMR (D<sub>2</sub>O): 3.85 (IH, q, J = 6.4 Hz, H-3), 1.29 (3H, s, 2-CH<sub>2</sub>), 1.05 (3H, d, J = 6.4 Hz, 3-CH<sub>3</sub>). Compound **18a**: colorless amorphous powder. IR  $v_{\text{max}}$  cm<sup>-1</sup>: 1672, 1560, 1508, 1429, 1294; EI-MS m/z: 100 ([M] $^\circ$ , 58), 85 (22), 83 (81), 82 (26), 71 (20), 69 (20), 57 (34), 55 (100), 54 (22);  $^\circ$ H NMR (D<sub>2</sub>O): 6.82 (1H, q, J = 4.3 Hz, H-3), 1.75 (3H, d, J = 4.3 Hz, 3-CH<sub>3</sub>), 1.74 (3H, s, 2-CH<sub>3</sub>).

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