

DITERPENE ACIDS FROM *CONYZA INCANA*

AHMED M. GALAL, ESSAM ABDEL-SATTAR,* FAROUK S. EL-FERALY, JABER S. MOSSA,
 MESELHY R. MESELHY,† SHIGETOSHI KADOTA† and TSUNEO NAMBA†

Department of Pharmacognosy, College of Pharmacy, King Saud University, Riyadh 11451, P.O. Box 2457, Saudi Arabia; †Research Institute for Wakan-Yaku, Toyama Medical and Pharmaceutical University, 2630-Sugitani, Toyama 930-01, Japan

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Key Word Index—*Conyza incana*; Compositae; diterpene acids; *ent*-labdane; NMR; MS.

Abstract—The aerial parts of *Conyza incana* yielded three new diterpene acids: the two alicyclic furano diterpenes, namely, *E,E* and *E,Z*-incanic acids and an *ent*-labdane succinate ester. The structures of these compounds were elucidated by spectroscopic methods. © 1998 Elsevier Science Ltd. All rights reserved

INTRODUCTION

More than 1200 diterpenes have been identified from approximately 550 Compositae taxa. The genus *Conyza* (tribe Asterae) comprises about 50 species, which are mainly distributed in tropical and subtropical areas. The diterpenes of the genus *Conyza* are mainly of labdane, clerodane, *seco*-clerodane and alicyclic types [1]. *C. incana* Willd was studied for its essential oil by GC/MS [2]. In addition, the flavonoids chrysophenol B and penduletin [3] were isolated from its aerial parts. As a continuation of the photochemical investigation of diterpene-containing plants of Saudi flora, the aerial parts of *C. incana* have been investigated, resulting in the isolation of three diterpenes, which are the subject of this paper.

RESULTS AND DISCUSSION

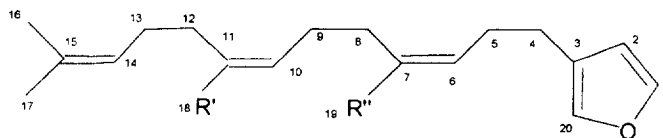
The CH₂Cl₂ extract of the aerial parts of *C. incana* afforded three diterpenes: the two furano alicyclic acids **1** and **5** and the *ent*-labdane succinate ester **6**. The structure of compound **1** was deduced from its molecular formula C₂₀H₂₆O₅ (high resolution MS), fragmentation pattern (see Experimental) and from the ¹H NMR and ¹³C NMR data (Tables 1 and 2). The presence of a β-substituted furan ring followed from the presence of a fragment ion at *m/z* 81 (C₅H₅O, pyrylium ion) in its EIMS and from the characteristic pattern of the downfield signals in its ¹H NMR spectrum (δ 6.27, 7.24, 7.35) (Table 1). The fast methy-

lation of **1** to **1a** using diazomethane indicated the presence of two carboxylic groups (two methoxy groups at δ 3.74 and 3.72 in the ¹H NMR). The ¹H NMR and ¹³C NMR spectra of **1** were consistent with an alicyclic diterpene acid with a terminal furan ring. The structure of **1** and its methyl ester **1a** showed close resemblance to those of centipedic acid (**2**), a diterpene isolated from *Centipeda orbicularis* [4] and microglossic acid (**3**), a diterpene acid isolated from *Microglossa zeylanica* [5]. Also, the unsaturated side chain of **1** appeared to be identical in composition and stereochemistry to the alkyl group on C-3 of oleaxillaric acid (**4**), a diterpene acid isolated from *Olearia axillaris* [6]. This fact was ascertained by comparing their relevant proton chemical shift values, which were almost indistinguishable (Table 1). The chemical shift values of H-6 and H-10 (δ 6.99 and 6.89, respectively) clearly indicated an *E* configuration for both double bonds as in **4** and related compounds with *E* configurations [5–7]. The name *E,E*-incanic acid is now used to describe compound **1**.

Compound **5** (C₂₀H₂₆O₅) showed very similar spectral data (Table 1, 2) to those of **1**. The main difference was the upfield shift of H-10 in **1** by 0.99 ppm, which suggested a reversed stereochemistry around the C-10/11 double bond from *E* in **1** to *Z* in **5** [6]. This was further confirmed by comparison with compounds having a similar configuration [4, 7–9] and also by noting the deshielded chemical shift value of C-12 (δ 34.53) compared to δ 28.21 in the isomeric compound **1**. The relative deshielding of C-12 can be attributed to the lack of steric compression at this position due to the *Z* configuration around the C-10/11 double bond [10]. Therefore, compound **5** was identified as the *E,Z* isomer of **1** and is named *E,Z*-incanic acid.

Compound **6**, C₂₄H₃₆O₆ (HRMS), was found to

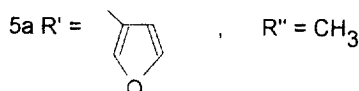
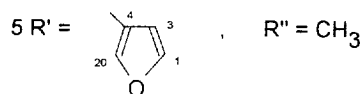
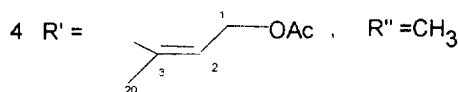
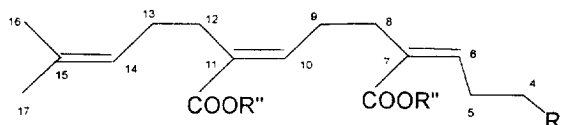
* Author to whom correspondence should be addressed.
 Present Address: Department of Pharmacognosy, College of Pharmacy, El-Kasr El-Aini Street, Cairo 11562, Egypt.



1 $R' = R'' = \text{COOH}$, 1a $R' = R'' = \text{COOCH}_3$

2 $R' = \text{COOH}$, $R'' = \text{H}$

3 $R' = R'' = \text{COOH}$, 6,7 dihydro



have a β -substituted butanolide ring as indicated by a positive reaction with Raymond reagent [11], which was further confirmed by the presence of an absorption band at 1775 cm^{-1} in the IR spectrum. The IR spectrum also showed absorption bands at 3090 , 1640 and 890 cm^{-1} , suggesting the presence of an exocyclic methylene group [11], whose presence was further substantiated from the ^1H NMR data (Table 1). The ^1H NMR spectrum of **6** showed signals at δ 5.85 (*d*, $J = 1.3$, H-14) and two *dd* at δ 4.69 and 4.75 ($J = 17, 1.3$, H-16), which are consistent with the values of a β -substituted butanolide ring [11], similar to that of the steroidal cardenolides [12]. The ^1H NMR spectrum of **6** also showed a close resemblance to those of the minor lactone alcohol **7** isolated from *Xanthocephalum linearifolium* [13] and daniellol succinate (**8**) isolated from *C. welwitschii* [14]. The EIMS showed the mass ion at m/z 318 $[\text{M}-100]^+$ indicating the loss of a succinic acid moiety. The presence of the latter was further confirmed by the multiplet signal at δ 2.63 (^1H NMR) assigned for the two CH_2 groups

and the two carboxyl signals at δ 177.63 and 172.2 (^{13}C NMR). The presence of the succinyl group at C-19 caused a downfield shift of the respective protons (δ 4.27 and 3.87) relative to those of the alcohol **7** [13] and similar values to those of **8** [14]. The stereochemistry of **6** was ascertained as an *ent*-labdane based on its negative optical rotation [14]. Therefore, compound **6** was identified as the 19-succinate ester of **7**.

EXPERIMENTAL

General. Mps: uncorr.; IR: KBr and neat; ^1H and ^{13}C NMR were: CDCl_3 using a JEOL GNM-GX400 spectrophotometer (400 and 100 MHz, respectively); HRMS and EIMS: JEOL JMS D-300. $[\alpha]_D$: at amb. temp.; MPLC: 40 g of RP C18 silica gel (25–40 μ , 2.5 $\times$ 22.5 cm), flow rate 5 ml/min; TLC: silica gel 60 F_{254} , CH_2Cl_2 -MeOH (10:1) as solvent system; visualization using *p*-anisaldehyde/ H_2SO_4 as spray reagent.

Plant material. The aerial parts of *C. incana* Willd were collected in Abha, Saudi Arabia in August 1995.

Table 1. ^1H NMR spectral data of the diterpenes isolated from *C. incana* and oleaoxillaric acid (**4**)

	1	1a	4	5	5a	6	6a
1	7.35 <i>t</i>	7.35 <i>t</i>	4.58 <i>d</i>	7.36 <i>t</i>	7.38 <i>t</i>	β 1.08 <i>ddd</i> α 1.80 <i>m</i>	1.07 <i>ddd</i> 1.80 <i>m</i>
2	6.27 <i>dd</i>	6.26 <i>dd</i>	5.36 <i>br.d</i>	6.27 <i>dd</i>	6.31 <i>dd</i>	1.50–1.52 <i>m</i>	1.50–1.52 <i>m</i>
3	—	—	—	—	—	1.02 <i>m</i> 1.78 <i>m</i>	1.05 <i>m</i> 1.76 <i>m</i>
4	2.60 <i>t</i>	2.57 <i>t</i>	2.15 <i>br.t</i>	2.55 <i>t</i>	2.40 <i>t</i>	—	—
5	2.49 <i>br.q</i>	2.45 <i>br.q</i>	2.29 <i>m</i>	2.41 <i>br.q</i>	2.46 <i>br.q</i>	1.25 <i>dd</i>	1.25 <i>dd</i>
6	6.99 <i>t</i>	6.82 <i>t</i>	6.76 <i>t</i>	6.93 <i>t</i>	6.81 <i>t</i>	α 1.36 <i>dq</i> β 1.88 <i>m</i>	1.35 <i>dq</i> 1.85 <i>m</i>
7	—	—	—	—	—	β 1.94 <i>ddd</i> α 2.42 <i>br.d</i>	1.94 <i>m</i> 2.43 <i>ddd</i>
8	2.43 <i>br.t</i>	2.42 <i>t</i>	3.41 <i>t</i>	2.44 <i>t</i>	2.57 <i>t</i>	—	—
9	2.29 <i>br.q</i>	2.26 <i>br.q</i>	2.29 <i>m</i>	2.64 <i>br.q</i>	2.48 <i>br.q</i>	1.65 <i>br.s</i>	1.62 <i>br.s</i>
10	6.89 <i>t</i>	6.67 <i>t</i>	6.71 <i>br.t</i>	5.90 <i>t</i>	5.80 <i>t</i>	—	—
11	—	—	—	—	—	1.65–1.8 <i>m</i>	1.7–1.8 <i>m</i>
12	2.33 <i>br.t</i>	2.28 <i>t</i>	2.29 <i>m</i>	2.24 <i>t</i>	2.24 <i>t</i>	A 2.26 <i>m</i> B 2.56 <i>m</i>	2.25 <i>m</i> 2.55 <i>m</i>
13	2.08 <i>q</i>	2.05 <i>q</i>	2.05 <i>dt</i>	2.10 <i>q</i>	2.08 <i>q</i>	—	—
14	5.11 <i>t</i>	5.12 <i>t</i>	5.10 <i>br.t</i>	5.08 <i>t</i>	5.10 <i>t</i>	5.85 <i>br.s</i>	5.85 <i>t</i>
16	1.66 <i>br.s</i>	1.66 <i>br.s</i>	1.66 <i>br.s</i>	1.67 <i>br.s</i>	1.70 <i>br.s</i>	A 4.69 <i>dd</i> B 4.75 <i>dd</i>	4.68 <i>dd</i> 4.75 <i>dd</i>
17	1.58 <i>b.s</i>	1.57 <i>br.s</i>	1.57 <i>br.s</i>	1.57 <i>br.s</i>	1.50 <i>br.s</i>	A 4.47 <i>br.s</i> B 4.89 <i>br.s</i>	4.47 <i>br.s</i> 4.88 <i>br.s</i>
18	—	—	—	—	—	0.97 <i>s</i>	0.96 <i>s</i>
19	—	—	—	—	—	A 3.87 <i>d</i> B 4.27 <i>d</i>	3.87 <i>d</i> 4.26 <i>d</i>
20	7.24 <i>br.s</i>	7.23 <i>t</i>	1.71 <i>br.s</i>	7.23 <i>t</i>	7.27 <i>t</i>	0.70 <i>s</i>	0.70 <i>s</i>
Succinic acid moiety:							
(CH ₂) ₂	—	—	—	—	—	2.63 <i>m</i>	2.62 <i>m</i>
OCH ₃	—	3.74 <i>s</i>	—	3.74 <i>s</i>	3.73 <i>s</i>	—	3.7 <i>s</i>
	—	3.72 <i>s</i>	—	3.71 <i>s</i>	3.71 <i>s</i>	—	—
OAc	—	—	2.04 <i>s</i>	—	—	—	—

J [Hz]; Compound **1**, **1a**, **5** and **5a**: 4,5 = 5,6 = 8,9 = 9,10 = 12,13 = 13,14 = $\sim$ 7; 1,2 = 1,20 = 1.8; Compound **6** and **6a**: 1 α , 1 β = 1 β , 2 α = 12.0; 1 α , 2 α = 1 β , 2 β = 5; 5, 6 α = 12.8; 5, 6 β = 1.8; 6 α , 7 α $\sim$ 4; 6 α , 6 β = 6 α , 7 β = 12.5; 6 β , 7 β = 5; 14,16A = 14,16 β $\sim$ 1.3; 16A, 16B = 17.4; 19A, 19B = 11.

A voucher specimen (# 12959) was deposited in the herbarium of the College of Pharmacy, King Saud University, Riyadh, Saudi Arabia.

Isolation procedure. The dried ground aerial parts (1 kg) of *C. incana* were percolated with CH₂Cl₂ (6 l) at room temp then evaporated to give 63 g of a dark green extract. A portion of the CH₂Cl₂ extract (50 g) was partitioned between MeCN and *n*-hexane to give 35 g and 9 g, respectively. The MeCN fraction (4 g) was dissolved in EtOAc (75 ml) and extracted with 2% NaHCO₃ soln (3 $\times$ 100 ml). The combined alkaline extracts were adjusted to pH 4 using HOAc and the acidic soln was shaken with EtOAc (3 $\times$ 100 ml). The combined EtOAc extracts were washed with H₂O and filtered on anhydrous Na₂SO₄ then evaporated under vacuum (1.3 g). In a typical run, the EtOAc extract (500 mg) was subjected to MPLC over RP C18 using MeOH–MeCN–H₂O (5:2:3) as eluent, to afford compound **1** (140 mg, *R*_f 0.36), followed by compound **5** (73 mg, *R*_f 0.47).

The EtOAc phase left after shaking with NaHCO₃

soln, was dried over anhydrous Na₂SO₄ to leave 1.7 g. The residue was subjected to silica gel CC using CH₂Cl₂–MeOH (9.5:0.5) as the solvent system, followed by MLPC rechromatography using RP C18 and MeOH–MeCN–H₂O (2:1:1) as eluent (flow rate 3 ml/min), to give 113 mg of compound **6**.

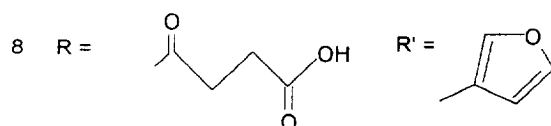
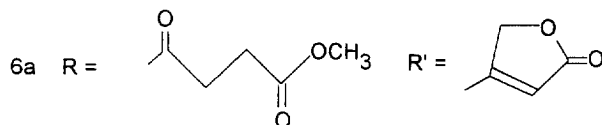
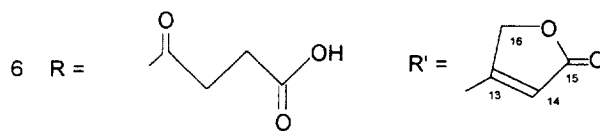
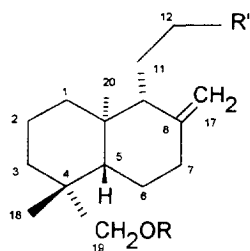
E,E-Incanic acid (1). White amorphous powder, mp 81–83°, [α]_D + 6.7° (*c* 0.102; MeOH). IR ν_{max} (KBr) cm^{−1}: 3420 *br* (COOH), 1670 (CO), 1630, 1425; ^1H and ^{13}C NMR: Tables 1 and 2; EIMS *m/z* (rel. int.): 346 [M]⁺ (6), 328 [M–H₂O]⁺ (35), 277 [M–C₅H₉]⁺ (15), 179 [M–C₁₀H₁₅O₂]⁺ (10), 95 [M–C₁₄H₁₉O₄]⁺ (18), 81 [C₅H₉O]⁺ (85) and 69 [C₅H₉]⁺ (100).

E,Z-Incanic acid methyl ester (1a). Colourless oil, [α]_D + 10.3° (*c* 0.07; MeOH). ^1H and ^{13}C NMR: Tables 1 and 2; EIMS *m/z* (rel. int.): 374 [M]⁺ (15), 342 [M–MeOH]⁺ (57), 310 [M–2XMeOH]⁺ (32), 229 (41), 179 [M–C₁₀H₁₅O₂]⁺ (22), 81 [C₅H₉O]⁺ (100) and 69 [C₅H₉]⁺ (96).

E,Z-Incanic acid (5). Colourless oil, [α]_D + 8.9° (*c* 0.092; MeOH). IR ν_{max} (neat) cm^{−1}: 3500 *br*

Table 2. ^{13}C NMR spectral data of diterpenes isolated from *C. incana*

	1	1a	5	5a	6	6a
1	143.03 <i>d</i>	141.99	142.97	42.93	38.87 <i>t</i>	38.93
2	110.76 <i>d</i>	110.75	110.76	110.81	18.80 <i>t</i>	18.86
3	123.48 <i>s</i>	123.84	123.32	123.45	36.01 <i>t</i>	36.04
4	23.90 <i>t</i>	24.02	24.06	26.90	37.38 <i>s</i>	37.41
5	29.49 <i>t</i>	29.24	29.46	29.03	56.11 <i>d</i>	56.08
6	145.28 <i>d</i>	142.63	144.06	142.39	24.35 <i>t</i>	24.42
7	132.40 <i>s</i>	132.49	131.10	131.88	38.32 <i>t</i>	38.35
8	25.63 <i>t</i>	26.08	25.94	24.08	147.00 <i>s</i>	147.06
9	26.60 <i>t</i>	26.90	28.54	29.09	56.01 <i>d</i>	56.17
10	144.06 <i>d</i>	141.46	141.23	140.51	39.56 <i>s</i>	39.63
11	131.04 <i>s</i>	132.28	133.13	132.28	21.32 <i>t</i>	21.35
12	28.21 <i>t</i>	28.00	34.53	34.70	27.37 <i>t</i>	27.56
13	27.70 <i>t</i>	27.70	27.76	27.70	170.98 <i>s</i>	170.83
14	123.69 <i>d</i>	123.59	123.27	123.38	115.16 <i>d</i>	115.28
15	132.07 <i>s</i>	131.99	132.46	132.28	174.26 <i>s</i>	174.08
16	25.63 <i>q</i>	25.60	25.66	25.70	73.14 <i>t</i>	73.08
17	17.62 <i>q</i>	17.60	17.71	17.61	106.96 <i>t</i>	106.99
18	173.42 <i>s</i>	168.49	174.48	168.34	27.48 <i>q</i>	27.39
19	173.05 <i>s</i>	167.86	173.54	168.04	67.01 <i>t</i>	66.92
20	139.08 <i>d</i>	139.05	139.02	139.05	15.10 <i>q</i>	15.16
Succinic acid moiety:						
(CH ₂) ₂	—	—	—	—	28.94 <i>t</i>	29.21
—	—	—	—	—	28.94 <i>t</i>	28.94
COOH	—	—	—	—	177.63 <i>s</i>	172.72
—	—	—	—	—	172.20 <i>s</i>	172.32
CH ₃ O	—	51.77 <i>q</i>	—	51.74 <i>q</i>	—	51.83
—	—	51.70 <i>q</i>	—	51.13 <i>q</i>	—	—



(COOH), 1680 (CO), 1630, 1420; ^1H and ^{13}C NMR: Tables 1 and 2; EIMS m/z (rel. int.): 346 $[\text{M}]^+$ (5), 328 $[\text{M}-\text{H}_2\text{O}]^+$ (35), 277 $[\text{M}-\text{C}_5\text{H}_9]^+$ (12), 179 $[\text{M}-\text{C}_{10}\text{H}_{15}\text{O}_2]^+$ (7), 95 $[\text{M}-\text{C}_{14}\text{H}_{19}\text{O}_4]^+$ (18), 81 $[\text{C}_5\text{H}_5\text{O}]^+$ (85) and 69 $[\text{C}_5\text{H}_9]^+$ (100).

E,E-Incanic acid methyl ester (5a). Colourless oil, $[\alpha]_{\text{D}} + 8.8^\circ$ (c 0.082; MeOH). ^1H and ^{13}C NMR: Tables 1 and 2; EIMS m/z (rel. int.): 374 $[\text{M}]^+$ (5), 342 $[\text{M}-\text{MeOH}]^+$ (28), 310 $[\text{M}-2\text{XMeOH}]^+$ (25), 229 (40), 179 $[\text{M}-\text{C}_{10}\text{H}_{15}\text{O}_2]^+$ (16), 107 (30), 81 $[\text{C}_5\text{H}_5\text{O}]^+$ (100) and 69 $[\text{C}_5\text{H}_9]^+$ (96).

Compound 6. Colourless oil, $[\alpha]_{\text{D}} - 25.33^\circ$ (c 0.107; MeOH). IR ν_{max} (neat) cm^{-1} : 3450 br (COOH), 3090, 1640, 890 (exocyclic methylene group), 1775 (γ -butanolide ring); ^1H and ^{13}C NMR: Tables 1 and 2; EIMS m/z (rel. int.): 418 $[\text{M}]^+$ (4), 400 $[\text{M}-\text{H}_2\text{O}]^+$ (6), 318 $[\text{M}-\text{succinate}]^+$ (13), 300 $[\text{M}-\text{H}_2\text{O}-\text{succinate}]^+$ (45), 287 $[\text{M}-\text{CH}_2\text{OCO}(\text{CH}_2)_2\text{COOH}]^+$ (38), 203 (75), 135 (86), 119 (57), 98 $[\text{C}_5\text{H}_6\text{O}_2]^+$ (100) and 69 (85).

Compound 6a. Colourless oil, $[\alpha]_{\text{D}} - 26.9^\circ$ (c 0.086; MeOH). ^1H and ^{13}C NMR: Tables 1 and 2; EIMS m/z (rel. int.): 432 $[\text{M}]^+$ (5), 318 $[\text{M}-\text{methyl succinate}]^+$ (4), 300 $[\text{M}-\text{H}_2\text{O}-\text{methyl succinate}]^+$ (66), 287 $[\text{M}-\text{CH}_2\text{OCO}(\text{CH}_2)_2\text{COOCH}_3]^+$ (54), 203 (88), 135 (99), 115 (100), 98 $[\text{C}_5\text{H}_6\text{O}_2]^+$ (92) and 55 (82).

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