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SESQUITERPENES FROM CELASTRUS PANICULATUS SUBSP. PANICULATUS

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Key Word Index—Celastrus paniculatus subsp. paniculatus; Celastraceae; sesquiterpenes; isolation; CO₂ supercritical fluid.

Abstract—Two new β -dihydroagarofuran sesquiterpene polyol esters were isolated from *Celastrus paniculatus* subsp. *paniculatus*. Their structures were deduced on the basis of spectral analysis including 2D NMR. © 1998 Elsevier Science Ltd. All rights reserved

INTRODUCTION

The plants of the Celastraceae are widely distributed in China. Many species have traditionally been used as insecticides [1]. Previous studies on chemical constitutions of Celastraceae plants have disclosed the presence of various β -dihydroagarofuran sesquiterpene polyol esters [2]. Some compounds with a β -dihydroagarofuran sesquiterpene polyol ester skeleton have been isolated from Celastrus paniculatus [3–6]. In the present paper, we report two new β dihydroagarofuran sesquiterpene polyol esters from Celastrus paniculatus subsp. paniculatus. They were identified as 1β , 9α -dibenzoyloxy- 4α -hydroxy- 6α -acetoxy-β-dihydroagarofuran and $1\beta.6\alpha.9\alpha$ -tribenzoyloxy- 4α -hydroxy- β -dihydroagarofuran. Several compounds with a similar structure have been found from some species of Celastraceae [7, 8].

RESULTS AND DISCUSSION

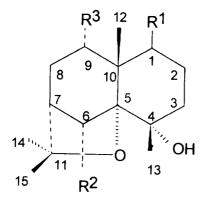
Compound 1 analyzed for $C_{31}H_{36}O_8$ by the spectral data. In its IR spectrum the absorption at 1719 cm⁻¹ showed the existence of ester groups. In the ¹³C NMR spectrum peaks at δ 170.5 (—COO—) and 21.6 and the ¹H NMR spectrum at δ 2.15 (Me)—and the presence of cross-peaks between δ 170.5 (—COO—) and δ 2.15 (Me) in COLOC suggested the presence of one acetate ester. ¹H NMR spectral peaks at δ 7.23–7.79 (10H). ¹³C NMR peaks at δ 127.9–132.9 (C_6H_5), 165.3,

165.4 (—COO—) and the presence of cross-peaks between δ 165.3, 165.4 (—COO—) and δ 7.23–7.79 (10H) in COLOC suggested the presence of two benzoate esters. The 3450 cm⁻¹ absorption in the IR spectrum and a ¹H NMR peak at δ 2.94 (1H) suggested the presence of one hydroxyl group. In addition, the ¹³C NMR and DEPT spectra indicated that compound 1 consisted of four methyl carbons (δ 20.0, 24.1, 25.8, 29.6); three methylene carbons (δ 38.6, 23.5, 31.9); four methine carbons (δ 73.4, 73.4, 49.1, 79.8); and four quaternary carbons (δ 70.5, 84.4, 91.6, 51.8). This parent was identified as β -dihydroagarofuran with four substituents at three methine carbons (δ 73.4, 73.4, 73.4, 79.8) and one quaternary carbon (δ 70.5) [2].

In the COLOC of compound 1, C-5 (δ 91.6) was assigned by its correlation to H-12, H-7; and C-11 (δ 84.4) by its correlation to Me-14, Me-15 and H-6; C-4 (δ 70.5) by its correlation to Me-13, H-3 and H-2; the remaining quaternary carbon C-10 was at δ 51.8. In this kind of compound, the hydroxyl group is usually located at the C-4 position [2].

The presence of cross-peaks between H-1 and the signal at δ 165.3 and the signals at δ 165.3 and δ 7.23–7.79 (protons of the phenyl) allowed the assignment of C_6H_3COO — which connected to C-1. The cross-peaks between H-6 and the signal at δ 170.5 together with the signals at δ 170.5 and 2.15 (protons of OAc) allowed the assignments of CH₃COO— which connected to C-6. The cross-peaks between H-9 and the signal at δ 165.4; and the signals at δ 165.4 and 7.23–7.79 (protons of the phenyl) allowed the assignment of C_6H_3COO — which connected to C-9. Furthermore,

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1
$$R_1 = OBz$$
, $R_2 = OAc$, $R_3 = OBz$;

2
$$R_1 = R_2 = R_3 = OB_Z$$
:

all the ring-protons were assigned unambiguously by ¹H NMR. COLOC and H¹-H¹ COSY. The stereochemistry of the ring-protons was deduced by comparing the coupling constants of protons with those reported in previous reports [9]. Therefore, compound 1 was identified as 1β ,9 α -dibenzoyloxy-4 α -hydroxy-6 α -acetoxy- β -dihydroagarofuran. The ¹H NMR and ¹³C NMR spectral data for compound 1 are presented in Tables 1 and 2.

Compound **2**, $C_{36}H_{38}O_8$ was very similar to compound **1**. The IR spectrum of **2** showed absorption at 1714 cm⁻¹ revealing the existence of ester groups. ¹H NMR peaks at δ 7.25–8.16 (15H) and ¹³C NMR peaks at δ 128.0–133.3 (3× C_6H_5)—and δ 165.3, 165.4, 165.5 (3×—COO—) suggested the presence of three benzoate esters. The presence of cross-peaks between H-1 and the signal at δ 165.3, and the signals at δ 165.3

and δ 7.25–8.16 (protons of the phenyl) allowed the assignments of C_6H_5COO — which connected to C-1. The cross-peaks between H-6 and the signal at δ 165.4 and the signals at δ 165.4 and δ 7.25–8.16 (protons of the phenyl) allowed the assignment of the second C_6H_5COO — which connected to C-6. The cross-peaks between H-9 and the signal at δ 165.5, and the signals at δ 165.5 and δ 7.25–8.16 (protons of the phenyl) allowed the assignment of the third C_6H_5COO — which connected to C-9. The other spectral data of compound **2** were almost the same as those of compound **1**. Thus compound **2** was identified as 1β ,6 α .9 α -tribenzoyloxy-4 α -hydroxyl- β -dihydroagarofuran. The 'H NMR and '3C NMR spectral data for compound **2** are presented in Table 2.

EXPERIMENTAL

General experimental procedures

Mps were determined on a Kolfer apparatus. Optical rotation was determined on Perkin-Elmer 241 apparatus. ¹H NMR, ¹³C NMR, DEPT, COSY and COLOC spectra were recorded on a Bruker AM-400 NMR spectrometer with TMS as internal standard. UV spectra in EtOH were on a UV-210A spectrophotometer. FAB-MS were obtained on VG ZHB-HS mass spectrometer. IR spectra were determined on a NICOLET FT-IR spectrophotometer. SFE was carried out on a modified ISCO-100D supercritical fluid extraction system.

Plant material

The seeds of *Celastrus paniculatus* subsp. *paniculatus* were collected from Xishuang Banna, Yunan province, China, in September and identified by Mr Cheng Biqiang, a senior researcher of the Tropical Botanical Garden of Xishuang Banna, Chinese Acad-

Table 1. H NMR spectral data for compounds 1 and 2 (400 MHz, CDCl₃)

Proton	1	2
1	5.59 (dd, 3.96, 11.93)	5.65 (dd, 3.68, 11.85)
2	1.74, 2.00 (each 1 H, m)	1.80, 2.06 (each 1 H, m)
3	1.61, 2.07 (each 1 H, m)	1.65, 2.11 (each 1 H, m)
6	5.55(s)	5.71 (s)
7	2.50 (m)	2.52(m)
8	2.18, 2.23 (each 1 H, m)	2.30, 2.38 (each 1 H, m)
9	5.10 (d, 6.82)	5.16 (d, 6.98)
Me-12	1.55 (s)	1.59 (s)
Me-13	1.38 (s)	1.26(s)
Me-14	1.50(s)	1.54(s)
Me-15	1.54(s)	1.56(s)
OH	2.94	3.15
OA¢	2.15 (s)	
OBz	7.23 7.80 (m, 10 H)	7.25 8.16 (m, 15 H)

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Carbon 1 2 1 73.4 d73.4 d2 39.0 t38.67 3 23.5 / 23.5 / 4 70.5 s70.8 s 5 91.6 s91.6 s6 73.4 d73.4 d7 49.1 d49.1 d 8 31.97 31.9 t 9 79.8 d80.6 d10 $51.8 \ s$ 51.8 s11 84.4 s 84.5 s $20.0 \ q$ 12 20.1 g13 24.1 q24.0 q25.8 q14 25.9 q29.6 q15 29.7 qOAc 170.5, 21.6 OBz $165.4.\ 165.3\ (2 \times --CO_2--)$ $165.5, 165.4, 165.3 (3 \times --CO_2--)$ 132.9 d, 132.6 d, 133.3 d, 132.9 d, 132.6 d $130.0 \text{ s}, 129.8 \ (2 \times d)$ $130.2 (2 \times d)$, 130.0 s129.3 s. 129.0 ($2 \times d$) 129.8 s, 129.8 $(2 \times d)$ $128.0 \ (2 \times d), \ 127.9 \ (2 \times d)$ 129.3 s, 129.1 $(2 \times d)$

Table 2. ¹³C NMR spectral data for compounds 1 and 2 (400 MHz, CDCl₃)

emy of Sciences. Voucher specimens are deposited at the Tropical Botanical Garden of Xishuang Banna.

Extraction and isolation

Air-dried and pulverized seeds (1.5 kg) of *Celastrus paniculatus* subsp. *paniculatus* were extracted with CO₂ supercritical fluid at 500 bar and 45°C for 2 h, A brown oil was obtained (180 g). Silica gel was used to absorb the oil. After air drying, the silica gel carrying the oil was eluted with CO₂ supercritical fluid at 100, 150, 200, 250, 300, 350, 400 and 450 bar at 45°C. Eight fractions were collected; the last three fractions were combined to give a yellowish extract (1.43 g). The extract was chromatographed on a silica gel (80 g) column with a gradient mixture of petrol and Me₂CO (9:1, 8:2, 7:3...) as eluent. compound 1 (108 mg) and compound 2 (144 mg) can be obtained from 8:2 and 7:3 fractions, respectively.

Compound 1. Colourless crystals (petrol/Me₂CO); mp 197-198°C; $[\alpha]_D^{2.5} = +80.4^{\circ}$ (CH₃Cl, c 0.26). UV $\lambda_{\text{max}}^{\text{McOII}}$ nm: 273.6, 230.8, 201.0. IR $\nu_{\text{max}}^{\text{KBr}}$ cm $^{-1}$: 3450, 2952, 2934. 1719, 1602, 1584, 1490, 1387, 1367, 1283, 1244, 1097. $^{-1}$ H NMR (400 MHz, CDCl₃) and $^{-13}$ C NMR (400 MHz, CDCl₃) see Tables 1 and 2; FAB-MS m/z (rel. int.) 537 ([M+1, 23] $^{+}$ (2), 520 (82), 478 (15), 415 (100), 397 (11), 355 (14), 337 (15), 293 (64), 233 (73).

Compound 2. Colourless crystals (petrol/Me₂CO) mp 281–282 C; $[\alpha]_D^{25} = +78^{\circ}$ (CH₃Cl, *c* 0.27). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 274.2, 231.4, 201.6. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3532,

2985, 2961, 1714, 1600, 1584, 1488, 1389, 1337, 1280, 1262, 1114, 1097. FAB-MS m/z (rel. int.): 599 ([M+1]⁻²⁶), 582 (59), 563 (14), 477 (79), 460 (28), 354 (50), 233 (100). ¹H NMR (400 MHz, CDCl₃) and ¹³C NMR (400 MHz, CDCl₃) see Tables 1 and 2.

 $128.6 \ (2 \times d), \ 128.0 \ (4 \times d)$

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REFERENCES

- Wakabaysshi, N., Wu, W. J., Water, R. M., Redfern, R. E., Mills Jr., G. D., DeMilo, A. B., Lusby, W. R. and Andrzejewski, D., J. Nat. Prod., 1988, 51, 537.
- Bruning, R. and Wagner, H., *Phytochemistry*, 1978, 17, 1821.
- 3. den Hertog Jr., H. H., Kruk, C., Nanavati, D. D. and Sukh Dev., *Tetrahedron Lett.*, 1974, 2219.
- Wagner, H., Heckel, E. and Sonnenbichler, J., Tetrahedron, 1975, 31, 1949.
- Tu, Y. Q., Tong, X., Wu, X.. Chen, Y. Z. and Zi Zh. Li, J. Nat. Prod., 1991, 54, 1383.
- Hong Sang, Hanqing Wang, Tu, Y. Q. and Chen, Y. Z., Phytochemistry, 1991, 30, 1547.
- Liu, J. and Jia, Z., Chin. Sci. Bull., 1989, 34(9), 1639.
- 8. Liu, J. and Wu, D., Planta Med., 1991, 57, 475.
- Takaishi, Y., Ujita, K., Nakano, K. and Tomimatsu, T., Phytochemistry, 1987, 26, 2325.