



## GALACTOMANNANS AND ARABINANS FROM SEEDS OF CAESALPINIACEAE

CARMEN L. O. PETKOWICZ, MARIA-RITA SIERAKOWSKI, JOANA LÉA M. S. GANTER and FANY REICHER\*

Departamento de Bioquímica e Departamento de Química da Universidade Federal do Paraná,  
 CP 19046, 81531-990 Curitiba, PR, Brazil

(Received in revised form 10 February 1998)

**Key Word Index**—*Schizolobium parahybae*; *Schizolobium amazonicum*; *Cassia fastuosa*; Caesalpiniaceae; chemotyping; seed; polysaccharides; galactomannan; arabinan

**Abstract**—*Schizolobium parahybae* and *S. amazonicum* seeds yielded galactomannans with identical 3.0:1 Man:Gal ratios and with the same D-galactose distribution along the main chain. Although the galactomannan from seeds of *Cassia fastuosa* showed the same Man:Gal ratio, its fine structure differed significantly from that of the two *Schizolobium* species as shown by the analysis of oligosaccharides (DP 2 to 6) obtained by partial acid hydrolysis. Seed coats of *S. parahybae* and *S. amazonicum* furnished similar unusual neutral linear  $\alpha$ -L-arabinofuranan (1 $\rightarrow$ 5) linked, as determined by methylation analysis, optical rotation and  $^{13}\text{C}$  NMR spectroscopy. On the other hand, *C. fastuosa* seed coats furnished two acidic arabinans. These results in terms of using galactomannans and arabinans in chemotyping, support the suggestion of Rizzini that *S. parahybae* and *S. amazonicum* are not different species. © 1998 Published by Elsevier Science Ltd. All rights reserved

### INTRODUCTION

Endosperms of leguminous seeds have thickened cell walls with massive deposits of galactomannans. They are storage polysaccharides with a (1 $\rightarrow$ 4)- $\beta$ -linked-D-mannan backbone, which carries single-unit (1 $\rightarrow$ 6)- $\alpha$ -linked-D-galactosyl branches, and have commercial importance [1]. It has been suggested that the yield and mannose:galactose ratio (Man:Gal) of such galactomannans could be used as markers in the taxonomy of the families of Fabales (Leguminosae) [1–4]. Man:Gal ratios between 1.0:1 (high Gal) and ca 5:1 (low Gal) can occur [1]. In general, low galactose galactomannans are obtained from the seeds of species belonging to the more primitive Caesalpiniaceae (Leguminosae Caesalpinoideae) and medium to high galactose galactomannans from more recent Fabaceae (Leguminosae Faboideae) [3]. However, galactomannans from seeds of different botanical sources can have identical degrees of galactose substitution but with differences in the statistical distribution of galactose residues along the mannan backbone [5].

*Schizolobium parahybae* (Vell.) Blake (guapuruvu) [6], previously called *Schizolobium parahybum*, and *Schizolobium amazonicum* Ducke (pinho cuiabano) are native Caesalpiniaceae species which grow in southern Brazil and the Amazon region, respectively [7].

They have been considered to be identical by Rizzini [8], although no synonym was found in the Index Kewensis [9]. Isolated endosperms from both species yielded ca 50% of galactomannan with identical 3.0:1 mannose to galactose ratios [10, 11]. The polysaccharides have closely related fine structures and D-galactose distribution along the main chain, as shown by the pattern of oligosaccharides obtained on partial acid hydrolysis [11].

On the other hand, in a preliminary investigation, we isolated a linear (1 $\rightarrow$ 5) linked  $\alpha$ -L-arabinofuranan from the seeds of *S. parahybae* [12]. Its structure contrasts with the highly branched arabinans previously found in cell walls of roots [13, 14], seeds [15], pollen tubes [16], bark [17] and fruit [18, 19] of various species, which are generally associated with pectic material [20].

Arabinans have received less attention than galactomannans and could also be helpful in chemotyping. Recently, fat-mimetic properties were described for enzymatically debranched arabinans [21]. Chemical and NMR analysis suggest that it is linear, as is the polymer obtained from material precipitated from apple juice concentrate, and which may have been formed from a more complex polymer by partial enzymolyses [22]. The ability of the linear arabinan to match the rheological properties of high-fat products has resulted in a considerable number of opportunities for its use in fat replacement. It may also be employed in the food industry and has potential

\*Author to whom correspondence should be addressed.

practical applications in cosmetics, pharmaceuticals and toiletries [21]. The fat-mimetic properties might be attributed to its polymorphism and tendency to form microcrystals in the solid state [23].

We now report the characterization of a neutral linear arabinan from *S. amazonicum* and two acidic arabinans from *Cassia fastuosa* Willd [24] which is also a member of the Caesalpiniaceae, in parallel with a comparison of the galactomannan of the latter and those of *Schizolobium* species.

## RESULTS AND DISCUSSION

In previous investigations, we characterized the galactomannans from the seeds of *S. parahybae* and *S. amazonicum* which were obtained by water extraction at 4° [10, 11]. They had the same Man:Gal ratio (3.0:1). When the residues of cold aqueous extractions were treated sequentially with water at higher temperatures (25° and 60°), lower yields of less substituted galactomannans were obtained (4.0:1 and 4.3:1, respectively). The seeds of *C. fastuosa* also furnished, by cold aqueous extraction, a galactomannan with Man:Gal ratio 3.0:1, as determined by GC of derived alditol acetates. When water at 25° and 60° were used for extraction, galactomannans with Man:Gal ratios of 4.0:1 were obtained [25].

The  $^{13}\text{C}$  NMR spectrum of the galactomannan from *C. fastuosa* obtained with cold water extraction confirmed a main chain of units of (1→4)-linked  $\beta$ -D-Manp substituted at O-6 by single-unit side-chains of  $\alpha$ -D-Galp. The Man:Gal ratio of 3.1:1 was also confirmed by the areas of the signals at  $\delta$  101.2 (C-1 of  $\beta$ -D-Manp residues) and that at  $\delta$  100.0 (C-1 of  $\alpha$ -D-Galp units). The  $\text{CH}_2$  atoms ( $\delta$  61.7 and  $\delta$  62.3) from unbranched Man and Gal units respectively, are well documented [26]. The C-4 (Man) pattern of the lines showed clear splitting with different intensities of the O-substituted resonances depending on the substitution of the nearest neighbour (Fig. 1). The peak at the lowest field  $\delta$  78.0 can be assigned to two-substituted contiguous D-Manp units and the peak at high field ( $\delta$  77.5) originates from contiguous unsubstituted D-Manp residues. The intermediate peak ( $\delta$  77.8) represents superimposition of signals originating from diads in which only one of the two D-Manp units is substituted [11, 27].

Galactomannan from *C. fastuosa* (3.0:1) was submitted to partial acid hydrolysis under the same conditions used for galactomannans from *S. amazonicum* and *S. parahybae* (both 3.0:1) [11]. Table 1 shows the oligosaccharides obtained by mild acid hydrolysis from the three galactomannans. The yield of each DP determined by GPC was similar for the three polymers. The major fraction for each of the three galactomannans was the monosaccharide (ca 34%). Identification of the oligosaccharides with DP-2 to 6 was performed by correlation of their retention time on HPLC in the light of their  $^1\text{H}$  and  $^{13}\text{C}$  NMR analyses. As previously reported [11], on analysis of oli-

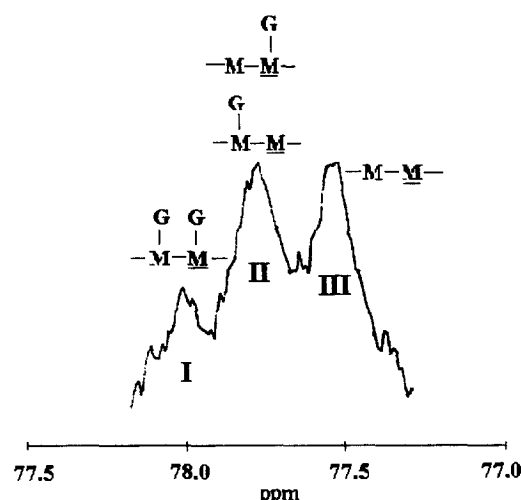


Fig. 1.  $^{13}\text{C}$  NMR spectral region of C-4 of D-Manp units in *C. fastuosa* seed galactomannan. Chemical shifts given are for the underlined M (Man) residues.

Table 1. Oligosaccharides with DPs of 2-6 produced by partial acid hydrolysis of the galactomannan of *C. fastuosa* (*C.f.*) compared with those of *S. parahybae*\* (*S.p.*) and *S. amazonicum*\* (*S.a.*).

DP	Yield <sup>†</sup>	Compound <sup>‡</sup>	Galactomannan source		
			<i>C.f.</i>	<i>S.p.</i>	<i>S.a.</i>
DP 2	15	G-M	37.5	17.0	18.0
		M-M	62.5	83.0	82.0
DP 3	14	G <sup>1</sup> M <sub>2</sub> <sup>§</sup>	39.0	14.5	15.1
		G <sup>2</sup> M <sub>2</sub>	24.2	30.8	32.5
		M <sub>3</sub>	36.8	54.6	52.4
DP 4	14	G <sup>1,2</sup> M <sub>2</sub>	22.8	10.9	10.5
		G <sup>1</sup> M <sub>3</sub>	40.2	32.4	31.5
		G <sup>2</sup> M <sub>3</sub>	16.3	24.8	27.8
		M <sub>4</sub>	20.7	31.8	30.1
DP 5	12	G <sup>1,2</sup> M <sub>3</sub> <sup>  </sup>	20.9	9.4	7.6
		G <sup>1,3</sup> M <sub>3</sub>	19.3	17.9	15.7
		G <sup>1</sup> M <sub>4</sub>	39.3	45.0	42.0
		G <sup>4</sup> M <sub>4</sub>	8.7	10.9	16.4
		M <sub>5</sub>	11.7	16.7	18.2
DP 6	11	1	31.8	17.7	23.2
		2	19.0	21.8	17.3
		3	36.7	43.5	39.9
		4	6.8	8.4	10.2
		5	5.7	8.4	9.3

\*Ganter *et al.* [11].

<sup>†</sup>Yield (average value of each DP of *C.f.*, *S.p.* and *S.a.*) determined by GPC; DP 1 ~ 34%.

<sup>‡</sup>By HPLC.

<sup>§</sup>G<sup>1</sup>M<sub>2</sub> compound correspond to M—M. The numerical subscript corresponds to the number of D-Manp units and the superscript corresponds to the position of the substituent.

<sup>||</sup>This pentasaccharide may be G<sup>1,2</sup>M<sub>3</sub> or G<sup>2,3</sup>M<sub>3</sub>.

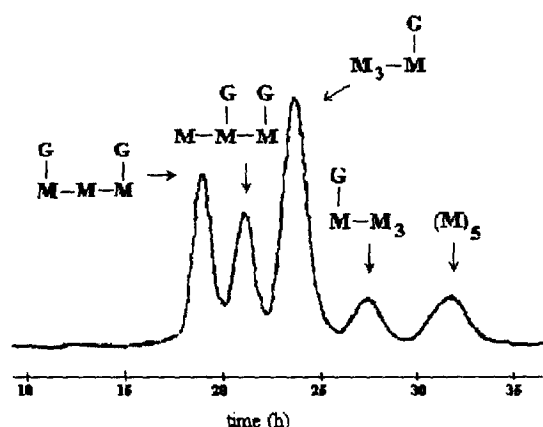


Fig. 2. HPLC chromatogram of the DP-5 series of galactomannan from *C. fastuosa*.

gosaccharide distribution, those from *S. amazonicum* and *S. parahybae*, with each isolated degree of polymerization (2 to 6) were similar. On the other hand, in spite of the same Man:Gal ratio, the distribution of the oligosaccharides in *C. fastuosa* galactomannan is quite different (Table 1); Fig. 2 shows the HPLC profile of the DP-5 series of the galactomannan from *C. fastuosa*. On comparison of the Man homooligosaccharides with DPs of 2–6 (Table 1), it can be seen that their relative proportion is lower than that found for *Schizolobium* species. This is in agreement with the resonances of C-4 of the D-Manp residues in their  $^{13}\text{C}$  NMR spectra. For *S. amazonicum*, the proportion of the peak areas of I, II and III is 3%, 54% and 43% [11] and for *C. fastuosa* 6%, 57% and 37%, respectively (Fig. 1). Thus, peak III, corresponding to two contiguous unsubstituted D-Manp residues is less intense for *C. fastuosa* galactomannan. These results indicate that the unsubstituted regions along the main chain are less abundant in the galactomannan of *C. fastuosa*, when compared to that

of the *Schizolobium* species. Consequently, the fine structure of galactomannan from *C. fastuosa* differs significantly from those of *S. amazonicum* and *S. parahybae*. Dea *et al.* [5] showed that galactomannans from distinct botanical sources with the same Man:Gal ratio, produced a different array of oligosaccharides when treated with  $\beta$ -endo-D-mannanase of *Bacillus subtilis*.

The seed coats of *S. amazonicum* and *C. fastuosa* were defatted and then extracted with water at 25° and at 80°. The residues arising from aqueous extractions were treated with 2M NaOH by the procedure of Whistler and Feather [28], yielding HA and HB. The monosaccharide composition of the seed coat fractions for *S. amazonicum* and *C. fastuosa* are shown in Table 2 and Table 3, respectively. Although other monosaccharides were present, arabinose was the predominant component of all the isolated fractions. As xylose, mannose, galactose and glucose occurred in both the aqueous and alkaline extracts, they are probably components of different polysaccharides from the seed coat cell walls and not contaminants from the endosperm. For *S. amazonicum*, fraction S<sub>3</sub>, extracted with 2M NaOH, the only monosaccharide component present was arabinose (arabinan S<sub>3</sub>).

However, for *C. fastuosa*, the hot water (C<sub>2</sub>, Table 3) and 4M alkaline extractions (C<sub>3</sub>) afforded polymers in which arabinose was the only neutral sugar; colorimetric data showed the presence of 20% and 7% of uronic acid, respectively. The presence of uronic acid was confirmed by characteristic absorption at  $1740\text{ cm}^{-1}$  for C=O group in their IR spectra as well as by PC. In contrast, uronic acid was not detected in polysaccharide S<sub>3</sub>.

The three arabinans (S<sub>3</sub>, C<sub>2</sub> and C<sub>3</sub>) were homogeneous on Sephacryl S-300 column chromatography and were then characterized. The arabinan from *S. amazonicum* (S<sub>3</sub>) had a  $M_r$  of 15100,  $[\alpha]_D^{25} -130^\circ$ ; methylation products (Table 4), were 2,3-Me<sub>2</sub>-Ara (99%) and 2,3,5-M<sub>3</sub> Ara (1%) corresponding to the non-reducing end. Its  $^{13}\text{C}$  NMR spectrum was ident-

Table 2. Yields and neutral sugar composition of polysaccharides obtained from the seed coats of *Shizolobium amazonicum* by water and alkali treatments.

F*	Extractant	Yield* g %	Monosaccharide† (mol %)					
			Rha	Ara	Xyl	Man	Gal	Glc
S <sub>1</sub>	H <sub>2</sub> O, 25°	2.5	3.7	43.6	5.3	21.6	20.8	5.0
S <sub>2</sub>	H <sub>2</sub> O, 80°	0.6	1.4	61.7	5.9	12.4	14.1	4.5
S <sub>3</sub>	2M NaOH (HA)	0.2	-	100.0	-	-	-	-
S <sub>4</sub>	2M NaOH (HB)	0.9	1.9	25.2	17.8	22.4	21.5	11.2
S <sub>5</sub>	4M NaOH (HA)	3.3	tr	57.7	7.7	16.7	5.1	12.8
S <sub>6</sub>	4M NaOH (HB)	1.0	tr	13.8	11.3	36.1	14.2	24.5

\*Fraction.

†Based on lipid free-seed coats.

‡Determined by GC of derived alditol acetates.

tr. trace

Table 3. Yields and neutral sugar composition of polysaccharides obtained from the seed coats of *C. fastuosa* by water and alkali treatments.

F*	Extractant	Yield† g %	Monosaccharide (mol %) <sup>‡</sup>					
			Rha	Ara	Xyl	Man	Gal	Glc
C <sub>1</sub>	H <sub>2</sub> O, 25°	0.8	3.0	38.0	36.4	10.3	11.2	1.1
C <sub>2</sub>	H <sub>2</sub> O, 80°	5.1	tr	99.0	tr	tr	tr	-
C <sub>3</sub>	2M NaOH (HA)	1.5	1.1	56.8	28.8	2.4	7.3	3.5
C <sub>4</sub>	2M NaOH (HB)	4.6	2.5	41.5	41.1	3.1	9.2	2.6
C <sub>5</sub>	4M NaOH (HA)	2.8	tr	99.0	tr	tr	tr	tr
C <sub>6</sub>	4M NaOH (HB)	4.6	7.9	50.3	11.3	6.9	15.6	8.0

\*Fraction.

†Based on lipid-free seed coats.

‡Determined by GC of derived alditol acetates.  
tr. trace.Table 4. Methylation analysis of arabinans from seed coats of *S. amazonicum* (arabinan S<sub>3</sub>) and *C. fastuosa* (arabinan C<sub>2</sub> and C<sub>5</sub>).

Partially O-Methylated alditol acetate	mol %		
	Arabinan S <sub>3</sub>	Arabinan C <sub>2</sub>	Arabinan C <sub>5</sub>
2,3,4-Me <sub>3</sub> -Ara	-	1.0	-
2,3,5-Me <sub>3</sub> -Ara	1.0	7.9	4.0
2,3-Me <sub>2</sub> -Ara	99.0	83.2	96.0
3-Me-Ara	-	2.9	-
2-Me-Ara	-	4.2	-
Ara	-	0.7	-

ical to that of *S. parahybae* arabinan, confirming the same  $\alpha$ -(1→5) linked arabinofuran structure [12].

The occurrence of arabinans C<sub>2</sub> and C<sub>5</sub> (Table 3) and the predominance of arabinose in all the isolated polysaccharide fractions suggest for *C. fastuosa*, the presence of a family of arabinosyl polymers, as already described for other species [13, 17, 29].

The *M*<sub>r</sub> for arabinan C<sub>2</sub> was 22900. This is similar to those found in the literature for other arabinans. In contrast, arabinan C<sub>5</sub> has a *M*<sub>r</sub> of 4600; the possibility of degradation by a  $\beta$ -elimination process during alkali extraction was avoided by the presence of NaBH<sub>4</sub> during this step. Lower *M*<sub>r</sub> values were also obtained for arabinans from *Sinapis alba* seeds [30] and from stem bark of *Rosa glauca* [15, 18, 29].

Arabinans C<sub>2</sub> and C<sub>5</sub> had negative specific rotations of  $[\alpha]_D^{25} - 62^\circ$  and  $[\alpha]_D^{25} - 100^\circ$  (0.5 M NaOH), respectively. Values of specific rotations for all arabinans range from  $-178^\circ$  to  $-83^\circ$ . The lowest value found,  $-62^\circ$ , for arabinan C<sub>2</sub> is probably due to the presence of 20% of  $\alpha$ -linked uronic acid units.

Methylation analysis of C<sub>5</sub> (Table 4) showed that the main chain is linear, as indicated by 2,3-Me<sub>2</sub>-Ara (96%) and the relatively high yield of 2,3,5-Me<sub>3</sub>-Ara (nonreducing ends) confirms the low *M*<sub>r</sub> of the polymer.

Arabinan C<sub>2</sub> is branched at position O-2, O-3 and O-2,3 as indicated by the presence of 3-Me-Ara (2.9%), 2-Me-Ara (4.2%) and Ara (0.7%), respectively, by arabinosyl moieties (2,3,5-Me<sub>3</sub>-Ara, 7.9%) and probably by uronic acid. Determination of branches by uronic acid was not reliable, probably due to the formation of acid-stable aldobiuronic moieties.

The <sup>13</sup>C NMR spectra of the arabinans are in agreement with the results of the methylation analysis. Several authors have reported <sup>13</sup>C assignments for branched arabinans [14, 17, 29]. Backinowsky *et al.* [31] obtained a spectrum for a synthetic linear (1→5)- $\alpha$ -arabinan with a *M*<sub>r</sub> of 2000–3000. Based on these assignments and those of the linear  $\alpha$ -L-arabinofuran from *S. parahybae*, the spectra of arabinans C<sub>2</sub> and C<sub>5</sub> were interpreted. The <sup>13</sup>C NMR spectrum of arabinan C<sub>5</sub> is similar to that of S<sub>3</sub>, with the exception of a small signal in the C-1 region ( $\delta$  100.8), which can be attributed to  $\alpha$ -linked uronic acid. For arabinan C<sub>2</sub> a strong peak at  $\delta$  107.9 can be assigned to C-1 of unbranched (1→5) linked  $\alpha$ -L-arabinofuranosyl units, that at 100.8 to the C-1 of 4-O-methyl  $\alpha$ -D-glucopyranosyl (uronic acid), signals at  $\delta$  171.2 and  $\delta$  53.3 to C-6 and OCH<sub>3</sub> of 4-O-methyl substituted glucuronosyl acid residues, respectively [32]; signals at  $\delta$  81.3,  $\delta$  81.0,  $\delta$  77.2 and  $\delta$  67.5 are

Table 5. Characterization of arabinans from three species of the Caesalpinaceae.

Source	Extractant	$[\alpha]_D^{25}$	$M_r$	Uronic acid* %
<i>S. parahybae</i> <sup>†</sup>	2M NaOH or DMSO	-133°	18.000	-
<i>S. amazonicum</i> (S <sub>3</sub> )	2M NaOH	-130°	15.100	-
<i>C. fastuosa</i> (C <sub>2</sub> )	H <sub>2</sub> O 80°C	-62°	22.900	20
<i>C. fastuosa</i> (C <sub>3</sub> )	4M NaOH	-100°	4.600	7

\*Determined according to ref. [35].

<sup>†</sup>Zawadzki-Baggio *et al.* [12].

assigned to C-4, C-2, C-3 and C-5, respectively, of (1 → 5) linked  $\alpha$ -L-arabinofuranosyl units. Small signals at  $\delta$  106.2 and  $\delta$  105.2 are due to C-1 of substituted  $\alpha$ -L-arabinofuranosyl units, in agreement with methylation data. Table 5 compares the data for the arabinans.

The presence of an unusual neutral linear arabinan in *S. amazonicum* can be taxonomically important, since it has been found to date only in *S. parahybae* and not in *C. fastuosa*. These results, in terms of using the galactomannans and arabinans in chemotyping, support the suggestion of Rizzini [8] that *S. amazonicum* and *S. parahybae* are not different species. DNA homology tests should be used as final confirmation.

## EXPERIMENTAL

### General

Polysaccharides were hydrolysed with 1 M TFA (5 hr, 100°). After removal of acid, hydrolysates were reduced with NaBH<sub>4</sub> and then acetylated in Ac<sub>2</sub>O-pyridine (1:1, 16 hr, at 25°). The resulting alditol acetates were analyzed by GC at 220° (FID and inj. temp. 250°), with a DB-210 capillary column (0.25 mm i.d. × 30 m), film thickness 0.25  $\mu$ m and N<sub>2</sub> carrier gas. Methylated polysaccharides were hydrolysed with 72% w/w aq. H<sub>2</sub>SO<sub>4</sub> (1 hr, 0–4°), following addition of H<sub>2</sub>O to a final acid conc of 8% (5 hr, 100°). The soln was neutralized with BaCO<sub>3</sub> and the alditol acetates of partially methylated sugars analyzed by GC and by GC-MS on an OV-225 capillary column (0.25 mm i.d. × 30 m) linked to an ion trap, MS operated at 70 eV. The column was programmed from 50° to 220° at 40° min<sup>-1</sup>. Whatman N° 1 paper and C<sub>6</sub>H<sub>6</sub>-*n*BuOH-pyridine-H<sub>2</sub>O (1:5:3:3) were used for PC; sugars were detected using the alkaline AgNO<sub>3</sub> reagent [33]. <sup>13</sup>C NMR were recorded at 100 MHz, at 40°, with DDS (sodium 4,4-dimethyl-4-silapentane-1-sulphonate) as int. standard ( $\delta$  = 0). Samples were dissolved in 0.5 M NaOD in D<sub>2</sub>O or D<sub>2</sub>O, depending on their solubility in H<sub>2</sub>O. Specific rotations were measured at 25° in 0.5 M NaOH. Total carbohydrate was estimated using the PhOH-H<sub>2</sub>SO<sub>4</sub> method [34] and uronic acid according to ref. [35].

### Plant material

Seeds of *S. parahybae* were collected on the campus of the Universidade do Rio dos Sinos, São Leopoldo, Rio Grande do Sul, and seeds of *C. fastuosa* were collected on the Campus of Universidade Federal do Paraná, Curitiba, Paraná. These species were identified by Prof. Olavo Araujo Guimarães (Departamento de Botânica da Universidade Federal do Paraná). Voucher specimens are deposited in the Herbarium (UPCB) of the Departamento de Botânica da Universidade Federal do Paraná, under the numbers 8749 for *S. parahybae* and 5904 for *C. fastuosa*. Seeds of *S. amazonicum* were collected at the National Forest of Tapajós, Pará, and supplied by EMBRAPA (Empresa Brasileira de Agropecuária)—CPATU—Belém, Pará, where a voucher specimen is deposited in the Herbarium (IAN), under the number 158478.

### Polysaccharide isolation

Whole seeds were boiled in H<sub>2</sub>O for 30 min and then kept at 4° for 48 hr until swelling took place. Thereafter, the endosperm, seed coat, and embryo were separated manually. The dry endosperm was milled and submitted to successive extractions at 4°, 25° and 60°, as previously described [10]. Seed coats were milled and treated with C<sub>6</sub>H<sub>6</sub>-EtOH (2:1) in a Soxhlet. Defatted and air-dried materials were submitted to aq. extraction at 25° and 80°, yielding S<sub>1</sub> and S<sub>2</sub> for *S. amazonicum* or C<sub>1</sub> and C<sub>2</sub> for *C. fastuosa*, respectively. The residues were extracted with 2 M NaOH and NaBH<sub>4</sub> (10 mg) at 25° and hemicellulose A (HA) then pptd by acidification to pH 5 with 50% (v/v) HOAc [28] (fr. S<sub>3</sub> for *S. amazonicum* and C<sub>3</sub> for *C. fastuosa*) and hemicellulose B (HB) isolated from the supernatant by pptn. with EtOH (fr. S<sub>4</sub> for *S. amazonicum* and C<sub>4</sub> for *C. fastuosa*). Residues were treated with 4 M NaOH and NaBH<sub>4</sub> (10 mg) at 25° and HA and HB then obtained as described above. These procedures furnished frs S<sub>5</sub> and S<sub>6</sub> for *S. amazonicum* and C<sub>5</sub> and C<sub>6</sub> for *C. fastuosa*.

### Partial acid hydrolysis

Galactomanannan from *C. fastuosa* was solubilized in H<sub>2</sub>O and H<sub>2</sub>SO<sub>4</sub> added to a final conc of 25 mM.

The sol was refluxed for 5 hr, neutralized with BaCO<sub>3</sub> and filtered. The vol. of filtrate was reduced and EtOH (3 vols) added. The supernatant was conc and the resulting oligosaccharides analyzed by gel permeation chromatography (GPC) and HPLC. GPC of oligosaccharides was effected in a thermostated column (65°; 210 cm × 1.5 cm i.d.) filled with polyacrylamide gel (Biogel P2, 200–400 mesh; BioRad) using dist. H<sub>2</sub>O as eluant (flow rate, 35 ml h<sup>-1</sup>). Oligosaccharides were purified by ion-exchange HPLC on a "Phenomenex-Rezex-RDC-BR-Oligosaccharide" at 50° and/or a Nucleosil 5 µm C-18 column (Societe Française Chromato Colonne) with H<sub>2</sub>O as eluant. Monosaccharide analyses were carried out on a CHO-682 column (Interchim) eluted with pure H<sub>2</sub>O at 85°. Samples were desalted before injection using a mixed resin (Amberlite M 3). RI was used for detection and peak areas were determined by electronic integration.

#### *Gel filtration of arabinan frs S<sub>3</sub>, C<sub>2</sub> and C<sub>5</sub>*

The polysaccharides (1 mg in 0.5 ml of 50 mM NaOH) were applied to a column of Sephacryl S-300 (1.2 × 36 cm) and eluted with 50 mM NaOH. Elution of material was monitored by the PhOH-H<sub>2</sub>SO<sub>4</sub> method at 480 nm and the *M<sub>s</sub>* estimated using a calibration curve prepared using standard dextrans with *M<sub>s</sub>* of 266000, 72000, 40000 and 17000 (Sigma).

#### *Methylation analysis of fractions S<sub>3</sub>, C<sub>2</sub> and C<sub>5</sub>*

Samples of polysaccharides (20 mg, dried over P<sub>2</sub>O<sub>5</sub> in an Abderhalden apparatus) were dissolved in DMSO (2 ml) and powdered NaOH (80 mg) added. After vigorous stirring for 10 min at 25°, MeI (0.5 ml) was added and again stirred for 10 min. The mixts were neutralized with HOAc and extracted with CHCl<sub>3</sub> [36]. This procedure was repeated for each sample and the per-*O*-methylated polysaccharides shown to be free of OH groups by IR spectroscopy. Each methylated polysaccharide was hydrolysed with dil. H<sub>2</sub>SO<sub>4</sub>, as described above, and analyzed by GC and GC-MS.

**Acknowledgements**—The authors would like to thank Prof. O. A. Guimarães (Departamento de Botânica, UFPR) for the identification of the two species and EMBRAPA, CPATU (Belém) for supply of *S. amazonicum* seeds. Thanks are due to Renato Boichichio for help in the preparation of this manuscript, Dr Alain Heyraud (CERMAV-CNRS) for HPLC analysis and to CNPq and PRONEX for financial assistance.

#### REFERENCES

- Dea, I. C. M. and Morrison, A., *Advances in Carbohydrate Chemistry and Biochemistry*, 1975, **31**, 241.
- Reid, J. S. G. in *Biochemistry of Storage Carbohydrates in Green Plants*, eds. P. M. Dey and R. A. Dixon. Academic Press, London 1985, pp. 265–288.
- Reid, J. S. G. and Edwards, M. E. in *Food Polysaccharides and their Applications*, eds. A. M. Stephen and M. Dekker. Inc. New York, 1995, pp. 155–186.
- Buckeridge, M. S., Panegassi, V. R., Rocha, D. C. and Dietrich, S. M. C., *Phytochemistry*, 1995, **38**(4), 871.
- Dea, I. M. C., Clark, A. H. and McCleary, B. V., *Carbohydrate Research*, 1986, **147**, 275.
- Rodrigues, W. A. Correção ortográfica no nome científico do guapuruvu (*Schizolobium parahybae* (Vell.) S. F. Blake) in *XLVIII Congresso Nacional de Botânica*, (Crato-Ce), 1997, pp. 304.
- Ernani, P. E. R. in *Espécies Florestais Brasileiras*, ed. Embrapa-SPI. Colombo, Brazil, 1994, pp. 470–475.
- Rizzini, T. in *Árvores e Madeiras Úteis do Brasil*. Edgar Blucher ed. São Paulo, 1987, pp. 133–136.
- Hooker, J. D. and Jackson, D., *Index Kewensis*. Clarendon Press., Oxford, 1960.
- Ganter, J. L. M. S., Zawadzki-Baggio, S. F., Leitner, S. C. S., Sierakowski, M. R. and Reicher, F., *Journal of Carbohydrate Chemistry*, 1993, **12**(6), 753.
- Ganter, J. L. M. S., Heyraud, A., Petkowicz, C. L. O., Rinaudo, M. and Reicher, F., *International Journal of Biological Macromolecules*, 1995, **17**(1), 13.
- Zawadzki-Baggio, S. F., Sierakowski, M. R., Correa, J. B. C. and Reicher, F., *Carbohydrate Research*, 1992, **233**, 265.
- Joseleau, J. P., Chambat, G. and Lanvers, M., *Carbohydrate Research*, 1983, **122**, 107.
- Capek, P., Toman, R., Kardosova, A. and Rosik, J., *Carbohydrate Research*, 1983, **117**, 133.
- Muralikrishna, G. and Tharanathan, R. N., *Food Chemistry*, 1986, **22**, 245.
- Rae, A. L., Harris, P. J., Bacic, A. and Clarke, A. E., *Planta*, 1985, **166**, 128.
- Joseleau, J. P., Chambat, G., Vignon, M. and Barnoud, F., *Carbohydrate Research*, 1977, **58**, 165.
- Tomoda, M. M., Takahashi, M. and Nakatsuka, S., *Chem. Pharm. Bull.*, 1973, **21**(4), 707.
- Eriksson, I., Andersson, R., Westerlund, E., Andersson, R. and Aman, P., *Carbohydrate Research*, 1996, **281**, 161.
- Stevens, B. J. H. and Selvendran, R. R., *Phytochemistry*, 1980, **19**, 559.
- Kasapis, S., Morris, E. R. and Norton, I. T. in *Gums and Stabilizers for the Food Industry*, eds. G. O. Phillips, P. A. Williams and D. J. Wedlock. IRL Press, Oxford, 1992, pp. 419.
- Churms, S. C., Merrifield, E. H., Stephen, A. M., Walwyn, D. R., Polson, A., van der Merwe, K. J., Spies, H. S. C. and Costa, N., *Carbohydrate Research*, 1983, **113**, 339.

23. Radha, A. and Chandrasekaran, R., *Carbohydrate Research*, 1997, **298**, 105.
24. *Mem. N.Y. Bot. Gard., Bronx*, **35**, 1982, 40.
25. Tavares, G. A. Ms. Thesis, 1994, Universidade Federal do Paraná.
26. Bociek, S. M., Izzard, M. J., Morrison, A. and Welti, D., *Carbohydrate Research*, 1981, **93**, 279.
27. Manzi, A. E., Cerezo, A. S. and Shoolery, J. N., *Carbohydrate Research*, 1986, **148**, 189.
28. Whistler, R. L. and Feather, M. S., *Methods in Carbohydrate Chemistry* 1965, **5**(144), 353.
29. Swamy, N. R. and Salimath, P. V., *Phytochemistry*, 1991, **30**, 263.
30. Rees, D. A. and Steele, I. W., *Biochemistry*, 1966, **5**, 3108.
31. Backinowski, L. V., Nepogodev, S. A. and Kochetkov, N. K., *Carbohydrate Research*, 1985, **137**, C1.
32. Gorin, P. A. J. and Mazurek, M., *Can. J. Chem.* 1975, **53**(8), 1212.
33. Trevelyan, W. E., Procter, D. P. and Harrison, J. S., *Nature*, 1950, **166**, 444.
34. Dubois, M., Gilles, K. A., Hamilton, J. K., Rebers, P. A. and Smith, F., *Analytical Chemistry*, 1956, **28**, 350.
35. Blumenkrantz, N. and Asboe-Hansen, G., *Analytical Biochemistry*, 1973, **54**, 484.
36. Ciucanu, I. and Kerek, F., *Carbohydrate Research*, 1984, **131**, 209.