

CYTOTOXIC BUTANOLIDES FROM *LITSEA AKOENSIS*

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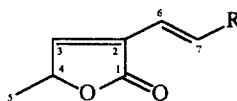
**Key Word Index**—*Litsea akoensis*; Lauraceae; stem bark; akolactones A and B; litseakolides A and B; litsenolide; butanolide;  $\gamma$ -lactone; cytotoxicity.**Abstract**—Four new butanolides, akolactone A, akolactone B, litseakolide A and litseakolide B, along with four known butanolides, litsenolide B<sub>2</sub>, litsenolide C<sub>1</sub>, litsenolide C<sub>2</sub> and hamabiwalactone A were isolated from the stem bark of *Litsea akoensis*. Their structures were elucidated from spectral evidence. These butanolides showed cytotoxic activity against P-388, KB16, A 549 and HT-29 cancer cell lines. © 1998 Elsevier Science Ltd. All rights reserved

## INTRODUCTION

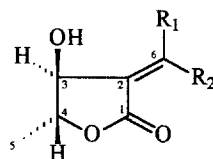
*Litsea akoensis* is an evergreen medium-sized tree, endemic in Taiwan [1]. So far, only an alkaloid, lauriltsine, was reported from the wood of this species [2]. In a series of studies on the anticancer constituents of Formosan plants, we have screened ca 350 species for *in vitro* cytotoxicity and *L. akoensis* was one of the active species. Its stem bark showed significant cytotoxic activity against P-388, KB16, A549 and HT-29 cancer cell lines. Examination of the chloroform-soluble part of the stem bark has led to the isolation of new butanolides (1–4) and four known butanolides (5–8). The known compounds, litsenolide B<sub>2</sub> (5) [3], litsenolide C<sub>1</sub> (6) [3], litsenolide C<sub>2</sub> (7) [3] and hamabiwalactone A (8) [4] were identified by comparisons of their IR, UV, <sup>1</sup>H NMR or <sup>13</sup>C NMR with the corresponding literature data. This paper reports the isolation and structural elucidation of the new butanolides and the cytotoxicity of the isolates.

## RESULTS AND DISCUSSION

Akolactone A (1) was isolated as a colourless oil. The molecular formula was determined to be C<sub>19</sub>H<sub>32</sub>O<sub>2</sub> by EI ([M]<sup>+</sup>, *m/z* 292) and HR mass spectra. It exhibited the absorption band of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone (1750 and 1660 cm<sup>-1</sup>) in the IR spectrum [3]. The UV absorption at 210 nm was similar to that of hamabiwalactone B (9) [4] and also supported the presence of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone skeleton. In the <sup>1</sup>H NMR spectrum (Table 1), the signals of an



- 1 R = —(CH<sub>2</sub>)<sub>11</sub>—CH<sub>3</sub>
- 2 R = —(CH<sub>2</sub>)<sub>8</sub>—CH=CH—CH=CH<sub>2</sub>
- 8 R = —(CH<sub>2</sub>)<sub>8</sub>—C≡CH
- 9 R = —(CH<sub>2</sub>)<sub>8</sub>—CH=CH<sub>2</sub>



- 3 R<sub>1</sub> = —(CH<sub>2</sub>)<sub>9</sub>—CH=CH—CH=CH<sub>2</sub>, R<sub>2</sub> = H
- 4 R<sub>1</sub> = —(CH<sub>2</sub>)<sub>9</sub>—CH=CH—CHO, R<sub>2</sub> = H
- 5 R<sub>1</sub> = —(CH<sub>2</sub>)<sub>9</sub>—C≡CH, R<sub>2</sub> = H
- 6 R<sub>1</sub> = H, R<sub>2</sub> = —(CH<sub>2</sub>)<sub>12</sub>CH<sub>3</sub>
- 7 R<sub>1</sub> = —(CH<sub>2</sub>)<sub>12</sub>CH<sub>3</sub>, R<sub>2</sub> = H
- 10 R<sub>1</sub> = —(CH<sub>2</sub>)<sub>8</sub>—CH=CH—CH<sub>2</sub>CH<sub>3</sub>, R<sub>2</sub> = H

olefinic proton at  $\delta$  7.03 (*d*, *J* = 2.0 Hz), a methine proton at  $\delta$  5.02 (*qd*, *J* = 6.8, 2.0 Hz), a methyl group at  $\delta$  1.42 (*d*, *J* = 6.8 Hz), two other olefinic protons at  $\delta$  6.09 (*d*, *J* = 16.0 Hz),  $\delta$  6.79 (*dt*, *J* = 16.0, 7.2 Hz), which were identical with those of 9 [4], were attributed to H-3, H-4, H-5, H-6 and H-7, respectively.

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Table 1. <sup>1</sup>H NMR data for butanolides 1–8 (400 MHz, CDCl<sub>3</sub>)

H	1	2	3	4	5	6	7	8
3	7.03 (1H, <i>d</i> , <i>J</i> = 2.0 Hz)	7.03 (1H, <i>d</i> , <i>J</i> = 1.6 Hz)	4.55 (1H, <i>br s</i> )	4.55 (1H, <i>br s</i> )	4.55 (1H, <i>br d</i> , <i>J</i> = 6.6 Hz)	4.34	4.54 (1H, <i>br d</i> , <i>J</i> = 6.4 Hz)	7.03 (1H, <i>d</i> , <i>J</i> = 1.6 Hz)
4	5.02 (1H, <i>qd</i> , <i>J</i> = 6.8, 2.0 Hz)	5.02 (1H, <i>br q</i> , <i>J</i> = 6.8 Hz)	4.50 (1H, <i>qd</i> , <i>J</i> = 6.9, 2.0 Hz)	4.49 (1H, <i>qd</i> , <i>J</i> = 6.8, 2.0 Hz)	4.50 (1H, <i>qd</i> , <i>J</i> = 6.6, 2.0 Hz)	(2H, <i>m</i> , H-3 and H-4)	4.50 (1H, <i>qd</i> , <i>J</i> = 6.4, 2.0 Hz)	5.02 (1H, <i>qd</i> , <i>J</i> = 6.8, 1.6 Hz)
5	1.42 (3H, <i>d</i> , <i>J</i> = 6.8 Hz)	1.42 (3H, <i>d</i> , <i>J</i> = 6.8 Hz)	1.35 (3H, <i>d</i> , <i>J</i> = 6.9 Hz)	1.35 (3H, <i>d</i> , <i>J</i> = 6.8 Hz)	1.35 (3H, <i>d</i> , <i>J</i> = 6.6 Hz)	1.39 (3H, <i>d</i> , <i>J</i> = 6.4 Hz)	1.35 (3H, <i>d</i> , <i>J</i> = 6.4 Hz)	1.42 (3H, <i>d</i> , <i>J</i> = 6.8 Hz)
6	6.09 (1H, <i>d</i> , <i>J</i> = 16.0 Hz)	6.09 (1H, <i>dd</i> , <i>J</i> = 15.0, 1.6 Hz)	7.00 (1H, <i>td</i> , <i>J</i> = 7.6, 1.2 Hz)	7.01 (1H, <i>td</i> , <i>J</i> = 8.0, 2.0 Hz)	7.00 (1H, <i>td</i> , <i>J</i> = 8.0, 2.0 Hz)	6.54 (1H, <i>td</i> , <i>J</i> = 7.8, 1.2 Hz)	7.00 (1H, <i>td</i> , <i>J</i> = 7.6, 1.6 Hz)	6.08 (1H, <i>d</i> , <i>J</i> = 16.0 Hz)
7	6.79 (1H, <i>dt</i> , <i>J</i> = 16.0, 7.2 Hz)	6.79 (1H, <i>dt</i> , <i>J</i> = 15.0, 7.2 Hz)	2.40 (2H, <i>m</i> )	2.40 (2H, <i>m</i> )	2.40 (2H, <i>m</i> )	2.76 (2H, <i>m</i> )	2.40 (2H, <i>m</i> )	6.78 (1H, <i>dt</i> , <i>J</i> = 16.0, 7.2 Hz)
8	2.16 (2H, <i>q</i> , <i>J</i> = 7.2 Hz)	2.15 (2H, <i>q</i> , <i>J</i> = 7.0 Hz)	1.51 (2H, <i>m</i> )	1.53 (2H, <i>m</i> )	1.51 (2H, <i>m</i> )	1.47 (2H, <i>m</i> )	1.54 (2H, <i>m</i> )	2.16 (2H, <i>m</i> )
9–14	1.42 (2H, <i>m</i> , H-9)	1.37 (2H, <i>m</i> , H-9)	1.28 (12H, <i>br s</i> , H-9–14)	1.30 (12H, <i>br s</i> , H-9–14)	1.30 (10H, <i>br s</i> , H-9–13) 1.51 (2H, <i>m</i> , H-14)			1.30 (10H, <i>br s</i> , H-9–13) 1.51 (2H, <i>m</i> , H-14)

15		2.06 (2H, <i>q</i> , $J = 7.0$ Hz)	2.07 (2H, <i>q</i> , $J = 7.0$ Hz)	2.33 (2H, <i>m</i> )	2.19 (2H, <i>td</i> , $J = 6.8, 2.4$ Hz)	2.18 (2H, <i>td</i> , $J = 7.2, 2.8$ Hz)
16	1.26 (18H, <i>br s</i> , H-10-18)	5.69 (1H, <i>dt</i> , $J = 15.2, 7.2$ Hz)	5.70 (1H, <i>dt</i> , $J = 15.2, 7.0$ Hz)	6.86 (1H, <i>dt</i> , $J = 15.0, 6.8$ Hz)	—	—
17		6.04 (1H, <i>dd</i> , $J = 15.2, 10.2$ Hz)	6.05 (1H, <i>dd</i> , $J = 15.2, 10.4$ Hz)	6.12 (1H, <i>ddd</i> , $J = 15.0, 7.8$ Hz)	1.26 (20H, <i>br s</i> , H-9-18)	1.25 (20H, <i>br s</i> , H-9-18)
18		6.30 (1H, <i>dt</i> , $J = 17.2, 10.2$ Hz)	6.31 (1H, <i>dt</i> , $J = 17.2, 10.4$ Hz)	9.50 (1H, <i>d</i> , $J = 7.8$ Hz)	—	—
19	0.88 (3H, <i>t</i> , $J = 6.8$ Hz)	4.94 (1H, <i>dd</i> , $J = 10.2, 1.6$ Hz) 5.08 (1H, <i>dd</i> , $J = 17.2, 1.6$ Hz)	4.95 (1H, <i>dd</i> , $J = 10.4, 1.6$ Hz) 5.08 (1H, <i>dd</i> , $J = 17.2, 1.6$ Hz)	—	0.88 (3H, <i>t</i> , $J = 6.8$ Hz)	0.87 (3H, <i>t</i> , $J = 6.8$ Hz)
OH	—	—	* 2.07 (1H)	1.98 (1H, <i>br s</i> )	1.88 (1H, <i>d</i> , $J = 6.6$ Hz)	1.94 (1H, <i>d</i> , $J = 6.4$ Hz)

\* Overlapped with H-15

Table 2.  $^{13}\text{C}$  NMR data for butanolides 1–3 and 5–8 (100 MHz,  $\text{CDCl}_3$ )

C	1	2	3	5	6	7	8
1	171.9	171.9	169.4	169.6	168.0	169.4	171.9
2	129.4	129.4	129.2	129.2	128.8	129.3	129.4
3	146.7	146.7	72.2	72.1	75.6	72.2	146.7
4	76.9	77.0	82.4	82.5	81.2	82.4	76.9
5	19.1	19.1	19.6	19.6	19.1	19.6	19.1
6	118.2	118.2	148.6	148.6	149.3	148.6	118.3
7	138.8	138.8	29.6	29.6	27.8	29.7	138.7
8	33.4	33.4	28.4	28.3	28.8	28.4	33.3
9	28.7	28.7					28.4
10	29.66	29.4	29.39	29.28	29.66	29.7	29.2
11	29.63	29.18	29.31	29.23	29.63	29.64	29.1
12	29.62	29.15	29.30	28.9	29.61	29.62	28.9
13	29.5		29.1	28.6	29.5	29.4	28.7
14	29.4			28.4	29.4	29.3	28.6
15	29.3	32.5	32.5	18.3	29.3		18.3
16	29.2	135.5	135.5	84.7	29.2		84.7
17	31.9	130.8	130.8	68.0	31.9	31.9	68.0
18	22.6	137.3	137.3	—	22.6	22.6	—
19	14.0	114.5	114.5	—	14.0	14.0	—

Thus, a moiety with a *trans*-olefinic group connected to the C-2 position of a 4-methyl-but-2-enolide in **1** was suggested. A dodecyl group connected to the above *trans*-olefinic group was supported by  $^{13}\text{C}$  NMR (Table 2). According to the above observation, the structure of akolactone **A** was represented by the formula **1**, which was further confirmed by COSY, HETCOR and DEPT experiments.

Akolactone **B** (**2**) was also isolated as a colourless oil. The molecular formula was determined to be  $\text{C}_{19}\text{H}_{28}\text{O}_2$  by EI ( $[\text{M}]^+$ ,  $m/z$  288) and HR mass spectrometry. It also exhibited the absorption band of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone ( $1750$  and  $1650\text{ cm}^{-1}$ ) in the IR spectrum. The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra (Tables 1 and 2) of **2** were very similar to those of **1**, except for the presence of the signals of an (*E*)-1, (*E*)-3-butadienyl group [ $\delta$  4.94 (1H, *dd*,  $J = 10.2$ , 1.6 Hz, H-19a),  $\delta$  5.08 (1H, *dd*,  $J = 17.2$ , 1.6 Hz, H-19b),  $\delta$  6.30 (1H, *dt*,  $J = 17.2$ , 10.2 Hz, H-18),  $\delta$  6.04 (1H, *dd*,  $J = 15.2$ , 10.2 Hz, H-17),  $\delta$  5.69 (1H, *dt*,  $J = 15.2$ , 7.2 Hz, H-16)] in **2** instead of an *n*-butyl group in **1**. Therefore, the structure of akolactone **B** was shown to be **2**, which was also further confirmed by COSY, HETCOR and DEPT experiments.

Akolactone **A** (**1**) and akolactone **B** (**2**) both have laevorotatory optical activity and probably show opposite stereochemistry at C-4 with dextrorotatory **8** and **9** [4].

Litseakolide **B** (**3**) was isolated as a colourless oil. The molecular formula was established as  $\text{C}_{19}\text{H}_{30}\text{O}_3$  by EI ( $[\text{M}]^+$ ,  $m/z$  306) and HR mass spectrometry. The IR spectrum showed the absorption bands for a hydroxyl group at  $3450\text{ cm}^{-1}$  and  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone at  $1750$  and  $1680\text{ cm}^{-1}$ .  $^1\text{H}$  NMR analysis showed that compound **3** has the same  $\beta$ -hydroxy- $\gamma$ -methyl- $\alpha,\beta'$ -unsaturated- $\gamma$ -lactone structure, the same

*E*-form geometry of the tri-substituted double bond [ $\delta$  7.00 (1H, *td*,  $J = 7.6$ , 1.2 Hz, H-6)] and the same *trans*-relationship of the substituents at C-3 and C-4 [( $\delta$  4.55 (1H, *br s*) and  $\delta$  4.50 (1H, *qd*,  $J = 6.9$ , 2.0 Hz)] to those of litsenolide **E**<sub>2</sub> (**10**) [4], except for the presence of an (*E*)-1, (*E*)-3-butadienyl group [ $\delta$  4.95 (1H, *dd*,  $J = 10.4$ , 1.6 Hz, H-19a),  $\delta$  5.08 (1H, *dd*,  $J = 17.2$ , 1.6 Hz, H-19b),  $\delta$  6.31 (1H, *dt*,  $J = 17.2$ , 10.4 Hz, H-18),  $\delta$  6.05 (1H, *dd*,  $J = 15.2$ , 10.4 Hz, H-17),  $\delta$  5.70 (1H, *dt*,  $J = 15.2$ , 7.0 Hz, H-16)] attached to C-15 in **3**, instead of an (*E*)-3-butenyl group in **10**. According to the above observations, the structure of litseakolide **B** was represented by the formula **3**, which was further confirmed by COSY and HETCOR experiments. Litseakolide **B** (**3**) has a laevorotatory optical activity and, hence, possesses the (3*S*, 4*R*)-configuration, like litsenolides **A**<sub>1</sub>–**E**<sub>2</sub> [3, 4].

Litseakolide **A** (**4**) was isolated as colourless oil. The HRFAB mass spectrum gave a  $[\text{M} + \text{H}]^+$  ion peak at  $m/z$  309.2073 (Calcd: 309.2067) consistent with a molecular formula of  $\text{C}_{18}\text{H}_{29}\text{O}_4$ . The  $^1\text{H}$  NMR spectrum of **4** revealed the presence of a  $\beta$ -hydroxy- $\gamma$ -methyl- $\alpha,\beta'$ -unsaturated- $\gamma$ -lactone with an *E*-form of geometry of the trisubstituted double bond, like those of litseakolide **B** (**3**). The remaining structure of **4** was clarified to be (2*E*)-dode-2-en-cyl group due to the one aldehyde proton at  $\delta$  9.50 (*d*,  $J = 7.8$  Hz), two *trans*-olefinic protons at  $\delta$  6.12 (*dd*,  $J = 15.0$ , 7.8 Hz, H-17) and  $\delta$  6.86 (*dt*,  $J = 15.0$ , 6.8 Hz, H-16) and nine methylene groups (Table 1). Therefore, the structure of litseakolide **A** shown as to be **4**, which was further confirmed by COSY experiments.

Only one of the compounds isolated in this study, **6** had the trisubstituted double bond with *Z*-stereochemistry and it showed diagnostic resonances for C-7 at  $\delta$  27.8 and C-8 at  $\delta$  28.8, comparable to the *E*-

Table 3. Cytotoxicity of butanolides 1–3 and 5–8

Compound	ED <sub>50</sub> (μg ml <sup>-1</sup> )			
	P-388	KB16	A549	HT-29
Akolactone A (1)	1.36	> 50	2.70	> 50
Akolactone B (2)	0.63	3.73	1.73	1.18
Litseakolide B (3)	0.99	2.83	2.50	1.42
Litsenolide B <sub>2</sub> (5)	1.07	> 50	0.40	0.40
Litsenolide C <sub>1</sub> (6)	0.36	0.69	0.94	1.07
Litsenolide C <sub>2</sub> (7)	0.21	0.87	0.76	0.77
Hamabiwalactone A (8)	1.20	2.92	1.19	5.74
Mithramycin*	0.06	0.08	0.07	0.08

\* Positive control

forms of compounds 3, 5 and 7, with C-7 at  $\delta$  29.6 or 29.7 and C-8 at  $\delta$  28.3 or 28.4 in the <sup>13</sup>C NMR spectrum through HETCOR experiments. This observation is quite different from that in the literature [4], which reported the same chemical shift for C-7 whether in the *Z*- or *E*-form in the trisubstituted double bond.

The cytotoxic activity of seven butanolides were tested *in vitro* against P-388, KB16, A549 and HT-29 cell lines; all showed significant activity (Table 3). In the group of compounds 1, 2 and 8, the long-chain alkyl group connected to C-7 in 1 decreased the cytotoxicity against four tested cell-lines. In another group of compounds with a  $\beta$ -hydroxy- $\gamma$ -methyl- $\alpha,\beta'$ -unsaturated- $\gamma$ -lactone, the geometry of the tri-substituted double bond showed no obvious difference in cytotoxicity between 6 (*Z*-form) and 7 (*E*-form). Furthermore, a terminal triple bond in 5 increased cytotoxic activity against A549 and HT-29 cell lines, but decreased it against P-388 and KB16 cell lines.

#### EXPERIMENTAL

Mps: uncorr. <sup>1</sup>H (400 MHz) and <sup>13</sup>C NMR (100 MHz): CDCl<sub>3</sub>, chemical shifts are given in  $\delta$  with TMS as int. standard. MS: direct inlet system. Optical rotations: CHCl<sub>3</sub>. UV: EtOH. IR: neat. Silica gel of 60–230 and 230–400 mesh (Merck) were used for CC and 60 F-254 for TLC.

#### Plant material

Stem bark of *L. akoensis* Hayata was collected from Wutai, Pingtung Hsien, Taiwan, in August 1996. A voucher specimen is deposited in the Herbarium of the School of Pharmacy, Kaohsiung Medical College, Taiwan, Republic of China.

#### Extraction and separation

Dried stem bark (3.6 kg) was extracted with MeOH and the extract concd under red. pres. The MeOH extract was partitioned between H<sub>2</sub>O–CHCl<sub>3</sub> (1:1) to

afford a CHCl<sub>3</sub>-sol. fr. (fr. A, 95 g). The H<sub>2</sub>O-sol. portion was then partitioned with *n*-BuOH again to obtain a *n*-BuOH-sol. fr. (fr. B, 270 g) and a H<sub>2</sub>O-sol. fr. (fr. C, 230 g). Fraction A (95 g) was chromatographed over silica gel, eluting with CHCl<sub>3</sub>, gradually increasing the polarity with MeOH, to obtain 22 frs (A1–A22). Fraction A4 (2.35 g, CHCl<sub>3</sub>) was resubjected to silica gel CC using *n*-hexane and *n*-hexane–EtOAc mixts to yield 29 frs (A4-1–A4-29). Fraction A4-13 (0.098 g, *n*-hexane–EtOAc, 100:7) was purified by prep. TLC (*n*-hexane–EtOAc, 5:1) to yield 1 (3.3 mg, *R*<sub>f</sub> 0.63). Fraction A9 (2.72 g, CHCl<sub>3</sub>) was rechromatographed on silica gel using *n*-hexane and *n*-hexane–EtOAc mixts to yield 24 frs (A9-1–A9-24). Fraction A9-8 (0.623 g, *n*-hexane–EtOAc, 4:1) was rechromatographed on silica gel and eluted with *n*-hexane and *n*-hexane–EtOAc mixts to yield 17 frs (A9-8-1–A9-8-17). Fraction A9-8-8 (56.4 mg, *n*-hexane–EtOAc, 17:3) was purified by prep. TLC (CH<sub>2</sub>Cl<sub>2</sub>–EtOAc, 20:1) to yield 6 (2.6 mg, *R*<sub>f</sub> 0.5) and 7 (15.2 mg, *R*<sub>f</sub> 0.4). Fraction A9-11 (0.195 g, *n*-hexane–EtOAc, 4:1), was rechromatographed on silica gel and eluted with CH<sub>2</sub>Cl<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub>–EtOAc mixts to obtain 28 frs (A9-11-1–A9-11-28). Fraction A9-11-7 (1.6 mg, CH<sub>2</sub>Cl<sub>2</sub>–EtOAc, 20:1) was purified by prep. TLC (CH<sub>2</sub>Cl<sub>2</sub>–MeOH, 40:1) to yield 4 (0.7 mg, *R*<sub>f</sub> 0.5). Fraction A10 (2.7 g, CHCl<sub>3</sub>) was rechromatographed on silica gel using CH<sub>2</sub>Cl<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub>–EtOAc mixts to yield 25 frs (A10-1–A10-25). Fraction A10-19 (0.155 g, CH<sub>2</sub>Cl<sub>2</sub>–EtOAc, 20:1) was purified by prep. TLC (*n*-hexane–Me<sub>2</sub>CO, 3:1) to yield 3 (4.0 mg, *R*<sub>f</sub> 0.38). Fraction A13 (2.59 g, CHCl<sub>3</sub>) was rechromatographed on silica gel using CH<sub>2</sub>Cl<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub>–EtOAc mixts to yield 26 frs (A13-1–A13-26). Fraction A13-2 (0.056 g, CH<sub>2</sub>Cl<sub>2</sub>) was purified by prep. TLC (*n*-hexane–EtOAc, 10:1) to yield 2 (6.3 mg, *R*<sub>f</sub> 0.42). Fraction A13-3 (0.028 g, CH<sub>2</sub>Cl<sub>2</sub>) was purified by prep. TLC (*n*-hexane–EtOAc, 10:1) to yield 8 (5.2 mg, *R*<sub>f</sub> 0.5). Fraction A13-7 (0.277 g, CH<sub>2</sub>Cl<sub>2</sub>) was purified by prep. TLC (*n*-hexane–EtOAc, 2:1) to yield 5 (15.7 mg, *R*<sub>f</sub> 0.4).

*Akolactone A* (1). Colourless oil.  $[\alpha]_D^{28}$ : –13.2° (ca 0.10, CHCl<sub>3</sub>). UV  $\lambda_{\max}$  nm (log  $\epsilon$ ): 210 (3.19). IR  $\nu_{\max}$

$\text{cm}^{-1}$ : 1750, 1660 ( $\alpha,\beta$ -unsaturated- $\gamma$ -lactone). EIMS  $m/z$  (rel. int.): 292  $[\text{M}]^+$  (10), 179 (10), 152 (11), 137 (66), 123 (38), 111 (17), 107 (26), 105 (19). HRMS:  $\text{C}_{19}\text{H}_{32}\text{O}_2$ . Found: 292.2386. Calcd: 292.2403.  $^1\text{H}$  NMR: Table 1.  $^{13}\text{C}$  NMR: Table 2.

**Akolactone B (2).** Colourless oil.  $[\alpha]_{\text{D}}^{27}$ :  $-10.0^\circ$  (ca 0.10,  $\text{CHCl}_3$ ). UV  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ): 221 (3.21). IR  $\nu_{\text{max}}$   $\text{cm}^{-1}$ : 1750, 1650 ( $\alpha,\beta$ -unsaturated- $\gamma$ -lactone). EIMS  $m/z$  (rel. int.): 288  $[\text{M}]^+$  (1), 191 (13), 177 (9), 161 (9), 147 (19), 145 (14), 137 (23), 133 (32), 131 (16), 121 (35). HRMS:  $\text{C}_{19}\text{H}_{28}\text{O}_2$ . Found: 288.2090. Calcd: 288.2089.  $^1\text{H}$  NMR: Table 1.  $^{13}\text{C}$  NMR: Table 2.

**Litseakolide B (3).** Colourless oil.  $[\alpha]_{\text{D}}^{27}$ :  $-40.0^\circ$  (ca 0.08,  $\text{CHCl}_3$ ). UV  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ): 224 (3.42). IR  $\nu_{\text{max}}$   $\text{cm}^{-1}$ : 3450 (OH), 1750, 1680 ( $\alpha,\beta$ -unsaturated- $\gamma$ -lactone). EIMS  $m/z$  (rel. int.): 306  $[\text{M}]^+$  (0.02), 167 (20), 163 (23), 149 (100), 129 (16), 123 (11), 121 (10), 111 (11), 105 (26). HRMS:  $\text{C}_{19}\text{H}_{30}\text{O}_3$ . Found: 306.2209. Calcd: 306.2195.  $^1\text{H}$  NMR: Table 1.  $^{13}\text{C}$  NMR: Table 2.

**Litseakolide A (4).** Colourless oil. UV  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ): 218 (4.74). EIMS  $m/z$  (rel. int.): 308  $[\text{M}]^+$  (0.08), 149 (17), 137 (11), 133 (10), 125 (16), 123 (15), 121 (14), 111 (23), 109 (18). HRFAB-MS:  $\text{C}_{18}\text{H}_{29}\text{O}_4$ . Found: 309.2073  $[\text{M} + \text{H}]^+$ . Calcd: 309.2067.  $^1\text{H}$  NMR: Table 1.

**Cytotoxicity assay.** Activities of compounds **1–3** and **5–8** against P-338 (mouse lymphocytic

leukaemia), KB16 (human nasopharyngeal carcinoma), A549 (human lung carcinoma) and HT-29 (human colon adenocarcinoma) cells were assayed by the MTT [3-(4,5-dimethylthiazole-2-yl)-2,5-diphenyl tetrazolium bromide] colorimetric method [6, 7].

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