



CRITHMIFOLIDE: A SESQUITERPENE LACTONE FROM ACHILLEA CRITHMIFOLIA

MILKA N. TODOROVA,* MIGLENA M. MARKOVA and ELENA T. TSANKOVA

Institute of Organic Chemistry with Centre of Phytochemistry, Bulgarian Academy of Sciences,
1113 Sofia, Bulgaria

(Received in revised form 16 February 1998)

Key Word Index—*Achillea crithmifolia*; Asteraceae; sesquiterpene lactones; 1,2-*seco*-guaianolide hemiacetal.

Abstract—Crithmifolide, a novel 1,2-*seco*-guaianolide hemiacetal has been isolated from the aerial parts of *Achillea crithmifolia*, in addition to 11 known sesquiterpene lactones. The structure of the new compound has been elucidated by spectroscopic methods. © 1998 Published by Elsevier Science Ltd. All rights reserved

INTRODUCTION

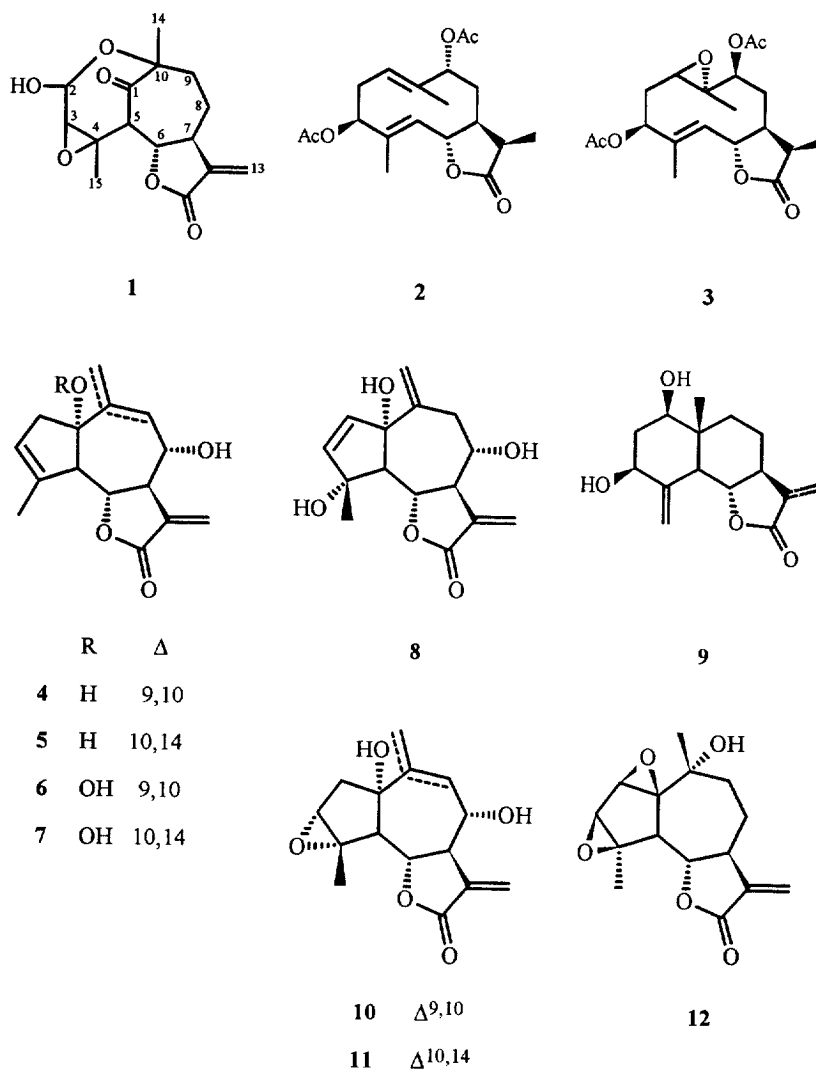
In our search for biologically active compounds from traditional medicinal plants, we have undertaken an investigation of *Achillea crithmifolia* Wald. & Kit. The genus *Achillea*, well-known for the medicinal properties of some members, has received much attention. However, a survey of the literature revealed only a small number of papers dealing with the chemical constituents of *A. crithmifolia* [1–7]. This might be due to the restricted distribution of the species to the Balkan peninsula, extending northwards to south-east Slovakia [8]. We now report on the isolation and structural elucidation of a new sesquiterpene lactone, named crithmifolide (**1**) from the aerial parts of *A. crithmifolia*. In addition, 11 known lactones, **2–12**, were also identified.

RESULTS AND DISCUSSION

The chloroform extract of the plant material afforded one new (**1**) and 11 known sesquiterpene lactones (**2–12**). The latter were identified by spectral data and TLC comparison using authentic samples as references. Crithmifolide (**1**) was obtained as a colourless gum. Its molecular formula, $C_{15}H_{18}O_6$, was deduced by CIMS ($[M + NH_4]^+$ m/z 312). The 1H and ^{13}C NMR spectra of **1** (Table 1) showed it had one exomethylene double bond and two carbonyl groups, thus it

possessed four rings based on the calculation of the degrees of unsaturation ($n = 7$). An α -methylene- γ -lactone ring was indicated by the carbon signals at δ 169.4, 137.7 and 121.1, and also by the characteristic 1H NMR signals for H-13 (δ 5.52 and 6.27). The HMQC and 1H - 1H COSY spectra established the connectivity of C-5/C-9 and the coupling patterns of the signals due to H-5, H-6 and H-7 allowed the assignment of the relative stereochemistry to give the structural fragment A (Fig. 1). The second ring in the structure of **1** was assumed to be an oxirane bearing a methyl group (δ 1.58) and a proton, which appeared as a sharp singlet at δ 3.20. Furthermore, the NMR data indicated the presence of a quaternary carbon carrying an oxygen function (δ 83.4) and a methyl group (δ 1.38 s, 20.4 q), and a carbonyl group (δ 207.5). The 1H NMR spectrum displayed two broad signals at δ 2.90 and 5.50, attributable to a OH proton and a carbinol proton, as the former signal disappeared and the latter changed to a sharp singlet in a D_2O exchange experiment. These spectral features suggested the presence of four additional structural units, B–E (Fig. 1), which together with fragment A accounted for the Mr of crithmifolide (**1**). The connectivities of these structural fragments leading to the tricyclic partial structure F were determined by detailed analysis of the HMBC spectrum of **1** (Table 1). The carbonyl carbon at δ 207.5 exhibited a long-range C–H correlation with H-5 and H-9b, while H-9a, H-8 and H-5 showed correlation with the quaternary carbon at δ 83.4. Therefore, the presence of a seven-membered ring formed by the structural units

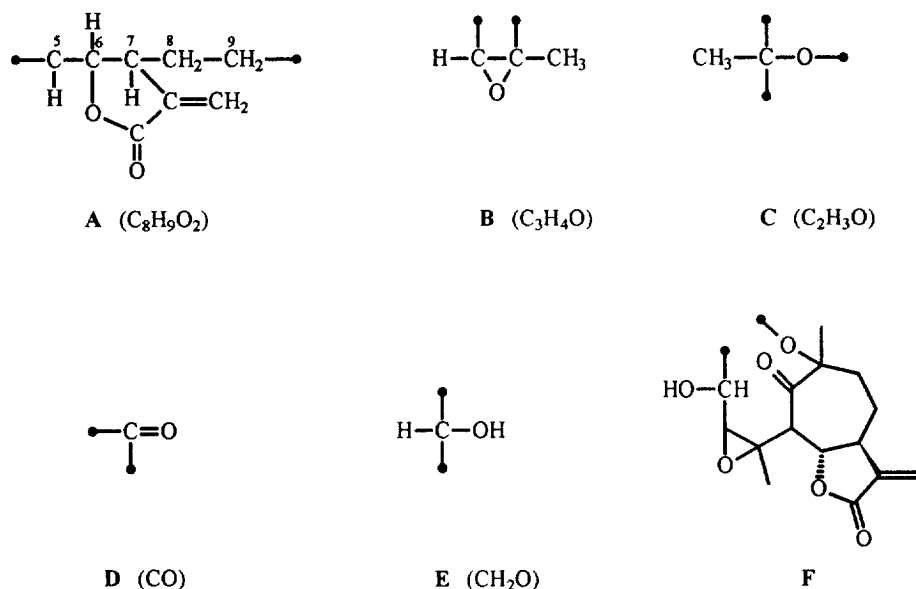
*Author to whom correspondence should be addressed.



A, C and D was unequivocally established. Further, the connectivity C-2/C-5 was indicated by the observed correlations of the carbon at δ 90.1 with

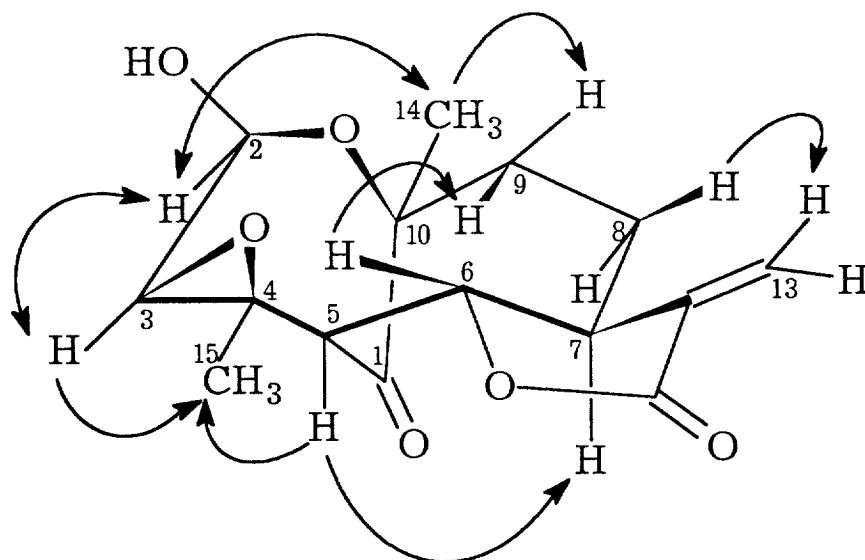
H-3 and the carbon at δ 64.9 with H-5 and H-15. The number of degrees of unsaturation for **1** meant that the only logical connectivity between C-2 and

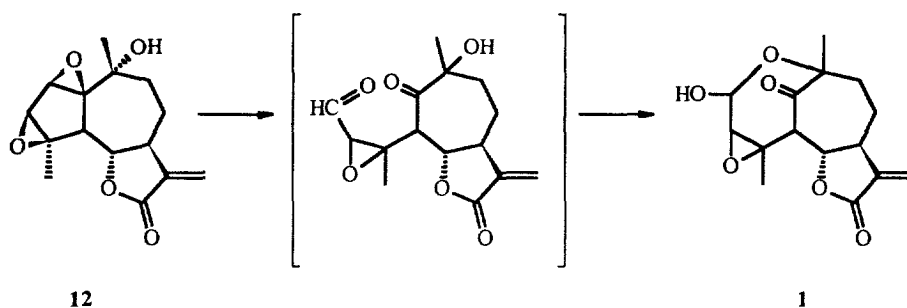
* ^{13}C multiplicities were assigned from DEPT and HMQC spectra.

Fig. 1. Structural fragments of crithmifolide (**1**).

C-10 in F was by an ether-type linkage. The presence of the oxygen bridge was supported by the chemical shifts of C-2 (δ 90.1) and the hemiacetal proton H-2 (δ 5.50). Thus, the structure of crithmifolide (**1**) was unambiguously determined as a sesquiterpene hemiacetal of the 1,2-*seco*-guaianolide type. The structure of **1** contained seven stereogenic centres. These were assigned by NOE difference measurements and analysis of the 1H - 1H coupling constants. The NOE connectivity pattern and the preferred conformation of crithmifolide (**1**) are depicted in Fig. 2. As mentioned above, the *trans*-closure of the lactone ring followed from the observed couplings $J_{6,7}$ (9.8 Hz) and $J_{7,13}$ (3.1 and 3.4 Hz). Irradiation of H-6 resulted in enhancement

of only H-9a (δ 1.65), thus placing the two protons on the same β -face of the seven-membered ring. NOE interactions between H-5, H-7 and H-15; H-3, H-15 and H-2, as well as between H-14, H-2 and H-9b indicated their *syn*- α -orientation. The Dreiding model showed that the observed NOEs require the seven-membered ring to adopt a boat-like conformation with the carbonyl group, the epoxide ring and the ether bridge lying out of the plane. This conformation demonstrated the spatial proximity of the two oxygen atoms to H-6, which caused the latter's downfield shift to δ 4.94, and also explained the relative downfield shift of H-5 (δ 3.35), which is obviously affected by the anisotropy of the *syn*-oriented carbonyl group. Moreover, a

Fig. 2. Observed NOE of crithmifolide (**1**).



Scheme 1.

dihedral angle of ca 90° between H-2 and H-3 explained the lack of vicinal coupling in the ^1H NMR spectrum, and established the relative stereochemistry at C-2 as depicted in Fig. 2.

Crithmifolide (**1**) is most probably derived from a guaianolide precursor, and the lactone **12** is a likely candidate. As shown in Scheme 1, oxidative cleavage of the five-membered ring in **12** followed by intramolecular formation of the hemiacetal would give **1**.

EXPERIMENTAL

Plant material

The aerial part of *A. crithmifolia* were collected from Pirin mountain (south-western Bulgaria) in July 1996. The plant material was identified by Dr. D. Peev from the Institute of Botany, Bulgarian Academy of Sciences and a voucher specimen (SOM 153315) was deposited in the Herbarium of the same Institute.

Extraction and isolation

The air-dried plant material (130 g) was ground and extracted with CHCl_3 to give, after evaporation of the solvent under red. pres., a brownish gum (10 g). This was then separated into 7 frs. by vacuum-liquid chromatography on silica gel using CH_2Cl_2 – Me_2CO mixtures as eluents. The lactone-containing frs. **4**–**6** (IR control) were subjected to repeated chromatography to yield **1** (5 mg) and the sesquiterpene lactones **2** (3 mg) [9], **3** (4 mg) [3], **4** (7 mg) and **5** (8 mg) [10], **6** (3 mg) and **7** (4 mg) [11], **8** (5 mg) [6], **9** (5 mg) [10], **10** (4 mg) and **11** (3 mg) [12], and **12** (5 mg) [13].

Crithmifolide (**1**)

CIMS (NH_3): m/z (rel. int.): 312 [$\text{M} + \text{NH}_4$] $^+$ (100), 296 [$\text{M} + \text{NH}_4 - 16$] $^+$ (5), 295 [$\text{M} + \text{NH}_4 - 17$] $^-$ (3), 294 [M] $^+$ (6); EIMS (70 eV) m/z (rel. int.): 294 [M] $^+$ (5), 235 (10), 219 (23), 205 (25), 189 (35), 164 (44), 121 (62), 109 (92), 95 (95), 85 (100), 53 (85); NMR: in Table 1.

Acknowledgements—The authors gratefully acknowledge the assistance of Dr. J. Platzek with the MS data and Ms. R. Taskova for supplying the plant material. This work was partially supported by the Bulgarian National Research Foundation, Project X-513.

REFERENCES

1. Valant, K., *Naturwissenschaften*, 1978, **65**, 437.
2. Greger, H., Grenz, M. and Bohlmann, F., *Phytochemistry*, 1981, **20**, 2579.
3. Miloslavijevic, S., Aliancic, I., Macura, S., Maliucovic, D. and Stefanovic, M., *Phytochemistry*, 1991, **30**, 3464.
4. Tsakou, T., Loukis, A. and Argyriadon, N., *Journal of Essential Oil Research*, 1993, **5**, 345.
5. Maffei, M., Mucciarelli, M. and Seannerini, S., *Biochemical Systematics and Ecology*, 1994, **22**, 679.
6. Miloslavijevic, S., Macura, S. and Stefanovic, M., *Journal of Natural Products*, 1994, **57**, 64.
7. Tsakou, O., Skaltsa, H. and Harvala, C., *Scientia Pharmaceutica*, 1996, **64**, 197. *Chem. Abstr.*, 1996, **125**, 190661m.
8. Richardson, I. B. K., in *Flora Europae*, Vol. 4, ed. T. G. Tutin, V. H. Heywood, N. A. Burges, D. M. Moore, D. H. Valentine, S. M. Walters and D. A. Webb. Cambridge University Press, Cambridge, UK, 1976, p. 159.
9. Goren, N., Oksuz, S. and Ulubelen, A., *Phytochemistry*, 1988, **27**, 2346.
10. Irwin, M. A. and Geissman, T. A., *Phytochemistry*, 1973, **12**, 863.
11. Bohlmann, F., Knoll, K. H., Robinson, H. and King, R. M., *Phytochemistry*, 1980, **19**, 599.
12. Todorova, M., Krasteva, M., Markova, M., Tsankova, E., Taskova, R. and Peev, D., in press.
13. Bohlmann, F. and Zdero, C., *Phytochemistry*, 1982, **21**, 2543.