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EUSCAPHOLIDE AND ITS GLUCOSIDE FROM LEAVES OF EUSCAPHIS JAPONICA

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Key Word Index—*Euscaphis japonica*; Staphyleaceae; tetraketide; euscapholide; euscapholide glucoside.

Abstract—A new tetraketide, euscapholide and its glucoside were isolated from the leaves of *Euscaphis japonica*. The structures were elucidated by spectroscopic and chemical analyses. © 1998 Elsevier Science Ltd. All rights reserved

INTRODUCTION

Flavonol glycosides, anthocyanins and compounds positive to Ehrlich reagent have been isolated from the capsule of *Euscaphis japonica* (Thunb.) Kanitz [1, 2]. In the course of our studies on the constituents of subtropical plants, we examined the constituents of the leaves of the title plant harvested in Okinawa Prefecture, Japan and isolated a tetraketide, euscapholide (1), and its glucoside (3). This paper describes the isolation and structural elucidation of the new compounds.

RESULTS AND DISCUSSION

Euscapholide (1) and its glucoside (3) were isolated from the *n*-BuOH-soluble fraction of the methanolic extract obtained from the leaves of *E. japonica*.

Euscapholide (1) was found to have the molecular formula $C_8H_{12}O_3$ by HR-EI-mass spectrometry. The IR spectrum showed the presence of a hydroxyl group (3420 cm⁻¹) and a δ -lactone (1700 cm⁻¹). The ¹H NMR spectrum showed the presence of a secondary methyl group [δ 1.26 (H_h)] on one of the carbon atoms (C-8) having an oxygen atom, two methylene groups [δ 1.77 (H_g), 2.03 (H_f) and 2.42

 (H_{e2})], a methine on C-8 [δ 4.08 (H_d)], a methine on the carbon having an acyloxy group $[\delta 4.65 (H_c)]$ and a conjugated disubstituted double bond with Zgeometry [δ 6.02 (H_b) and 6.92 (H_a). The ¹³C NMR spectrum (Table 1) contained the signals for above mentioned functional groups. Acetylation of 1 with acetic anhydride and pyridine gave the monoacetate **2** [δ 2.05 (3H, s) and 5.13 (1H, m), H_d]. The planar structure was deduced as 1 by analysis of the ¹H-¹H COSY spectrum. Namely, the cross peaks were followed as $H_b \rightarrow H_a \rightarrow H_c \rightarrow H_c \rightarrow H_f, H_g \rightarrow H_d \rightarrow H_h$. The absolute stereochemistry at C-6 was elucidated as R from the fact that 1 showed the opposite Cotton effect $[\Delta\epsilon_{255} + 3.0]$ to that of (S)-5,6-dihydro-6hydroxymethyl-2H-pyran-2-one(7) [3]. asymmetric centre at C-8 was determined to have S-configuration by a modified Mosher's method (Fig. 1) [4]. Thus, the structure of euscapholide was elucidated as 1.

Euscapholide glucoside (3) was assigned the molecular formula $C_{14}H_{22}O_8$ by negative ion HR-FAB-mass spectrometry. The ¹H and ¹³C NMR spectra were very similar to those of euscapholide (1), except for the appearance of the signals due to β -glucopyranosyl moiety (Table 1 and Experimental). Thus, compound 3 was presumed to be β -glucopyranoside of 1. Compound 3 gave the tetraacetate 4 [δ _H 2.00, 2.02, 2.03 and 2.09 (each 3H, s)] on acetylation with acetic anhydride and pyridine and euscapholide (1), [α]_D+116.4° (MeOH), upon enzy-

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1 R=H

2 R=COMe

3 R=G1c

4 R=Glc(COMe)₄

5 R=(R)-MTPA

6 R=(S)-MTPA

matic hydrolysis. Thus, the structure of euscapholide glucoside was elucidated as 3.

EXPERIMENTAL

General

NMR: 1 H (400 MHz) and 13 C (100 MHz) with TMS as int. standard; FABMS: PEG-400 as matrix; CC: Dianion HP-20 (Mitsubishi Kasei) and silica gel 60 (230–400 mesh, Merck); TLC: silica gel 60 F_{254} (0.25 mm and 0.5 mm in thickness); prep. HPLC: Cosmosil 10 C_{18} , 20×250 mm; solvent: MeOH-H₂O 1:4, 6 ml min⁻¹, detection 210 nm.

Plant material

The plant material was collected at Kunigamison, Okinawa Prefecture, Japan in July 1994 and identified as *Euscaphis japonica* (Thunb.) Kanitz by one of the authors (A. T.). A voucher specimen (9407-4) was deposited in the Herbarium of Institute of Pharmaceutical Sciences, Hiroshima University School of Medicine.

Table 1. ¹³C NMR data for euscapholide (1) and its glucoside (3), (CD₃OD, 100 MHz)

C	1	3
2	166.9 (164.2)*	167.1
3	121.3 (121.2)	121.3
4	148.3 (145.4)	148.3
5	30.2 (29.3)	30.4
6	77.5 (76.8)	77.1
7	44.7 (43.5)	42.6
8	64.8 (65.0)	74.6
9	23.6 (23.6)	22.2
1'		104.2
2'		75.2
3'		78.2
4'		71.6
5'		77.9
6′		62.8

^{*} Chemical shifts in parentheses are for chloroform-d.

Isolation

Dried leaves (2.9 kg) of E. japonica were extracted (x2) with MeOH (451) at room temp for three weeks. The combined methanolic extract was concentrated in vacuo. The residue was dissolved in 90% aq. MeOH (2.11) and the solution washed with *n*-hexane (11×3) . The aq. MeOH layer was concentrated in vacuo and the residue was suspended in H₂O (11). The suspension was extracted with EtOAc (11×3) and n-BuOH (11×3), successively. The n-BuOH extract was concentrated in vacuo to give a residue (77 g) which was chromatographed on Dianion HP-20 (a highly-porous synthetic resin $[\Phi = 78 \text{ mm}, L = 536 \text{ mm}]$ with stepwise increases of MeOH content in H₂O [0 (41), 10 (51), 30 (61), 40 (6.51), 50 (6.51), 70 (6.51) and 100 (61)]. Frs. of 500 ml being collected.

Frs 14–17 were combined and concentrated *in vacuo* to give a residue (10.7 g) which was chromatographed over silica gel (CHCl₃–MeOH with increasing amounts of MeOH). The fractions which contained a spot with R_f 0.35 on TLC (CHCl₃–MeOH–H₂O 15:6:1) were collected and the residue (418.7 mg) was purified by prep. HPLC to give euscapholide glucoside (3) (128.4 mg).

Frs 18–23 were combined and concentrated *in vacuo* to give a residue (12.2 g), which was repeatedly chromatographed over silica gel (CHCl₃–MeOH, Et₂O and finally CHCl₃–MeOH) to give euscapholide (1) (1.21 g), R_f 0.27 (Et₂O) and 0.29 (CHCl₃–Me₂CO 4:1).

Euscapholide (1)

Colourless oil, $[\alpha]_{0}^{30} + 115.5^{\circ}$ (MeOH; c 1.52). IR v_{max} (CHCl₃) cm⁻¹; 3420, 1700, 1230, 1200, 1050, 1025; UV λ_{max} (MeOH) nm (log ϵ): 209 (3.89); 1 H NMR (CDCl₃): δ 1.26 (3H, d, J = 6.3 Hz, 9-H₃), 1.77 (1H, ddd, J = 14.6, 5.4 and 4.4 Hz, 7-H₁), 2.03 (1H, dt, J = 14.6 and 7.8 Hz, 7-H₁), 2.42 (2H, m, 5-H₂), 4.08 (1H, m, 8-H), 4.65 (1H, m, 6-H), 6.02 (1H, dt, J = 10.1 and 1.7 Hz, 3-H) and 6.92 (1H, ddd, J = 10.1, 5.4 and 3.4 Hz, 4-H); 13 C NMR:

- (5) R=(R)-MTPA
- (6) R=(S)-MTPA

 $\triangle \delta$ values in Hz $(\delta S - \delta R, 400 \text{MHz})$

Fig. 1.

Table 1; CD (MeOH, $c = 1.41 \times 10^{-4} \text{ M}$): $\Delta \epsilon$ (nm) + 3.0 (255); HR-EIMS $m/z = 156.0789 \text{ [M]}^+$ (C₈H₁₂O₃ requires 156.0786).

Euspholide glucoside (3)

Amorphous powder, [α]_D + 36.4° (MeOH, c 2.47). UV $v_{\rm max}$ (MeOH) nm (log ϵ): 209 (3.98); IR $v_{\rm max}$ (dry film) cm⁻¹: 3387, 1703; ¹H NMR (CD₃OD): δ 1.33 (3H, d, J = 5.9 Hz, 9-H₃), 1.81 (1H, dt, J = 14.2 and 5.9 Hz, 7-H₁), 2.12 (1H, dt, J = 14.2 and 7.1 Hz, 7-H₁), 2.39 (1H, ddt, J = 18.6, 11.2 and 2.7 Hz, 5-H₁), 2.57 (1H, br.dt, J = 18.6 and 4.9 Hz, 5-H₁), 3.14 (1H, dd, J = 8.3 and 7.8 Hz, H-2'), 3.66 (1H, dd, J = 11.5 and 4.9 Hz, 6'-H₁), 3.85 (1H, br.d, J = 11.5 Hz, 6'-H₁), 4.03 (1H, m, 8-H), 4.32 (1H, d, J = 7.8 Hz, 1'-H), 4.74 (1H, m, 6-H), 5.97 (1H, dd, J = 9.8 and 1.5 Hz, 3-H), 7.04 (1H, ddd, J = 9.8, 5.9 and 2.4 Hz, 4-H); ¹³C NMR: Table 1; HR-FABMS (negative) m/z: 317.1229 [M-H]⁻ (C₁₄H₂₁O₈ requires 317.1237).

Euscapholide monoacetate (2)

Euscapholide (1) (51.5 mg) was dissolved in a mixture of Ac₂O (0.5 ml) and pyridine (0.5 ml) and the mixture was left for 24 hr at room temp. Excess MeOH was added and the solution was concentrated in vacuo. The residue was purified by silica gel CC (CHCl₃-Me₂CO) to give the monoacetate 3 (55.9 mg) as a colourless oil. IR v_{max} (CHCl₃) cm⁻¹: 1710, 1630, 1370, 1220, 1040; 1 H NMR (CDCl₃): δ 1.31 (3H, d, J = 6.4 Hz, 9-H₃), 1.85 (1H, ddd, $J = 14.6, 6.3 \text{ and } 4.9 \text{ Hz}, 7-\text{H}_1$, 2.05 (3H, s, OAc), 2.19 (1H, ddd, J = 14.6, 8.1 and 6.6 Hz, 7-H₁), 2.33 $(1H, ddt, J = 18.8, 11.7 \text{ and } 2.4 \text{ Hz}, 5-H_1), 2.47$ $(1H, dt, J = 18.8 \text{ and } 4.7 \text{ Hz}, 5-H_1), 4.53 (1H, m, 6-$ H), 5.13 (1H, m, 8-H), 6.02 (1H, dd, J = 9.8 and 1.5 Hz, 3-H), 6.90 (1H, ddd, J = 9.8, 5.9 and 2.4 Hz, 4-H); HR-EIMS m/z 198.0866 [M] (Calculated for $C_{10}H_{14}O_4$:198.0892).

(R)-MTPA ester (5) from euscapholide (1)

Euscapholide (1) (22.3 mg) was dissolved in CH_2Cl_2 (1 ml). (*R*)- α -methoxy- α -trifluoromethylphenyl acetic acid (MTPA) (146.7 mg), dimethylaminopyridine (42.3 mg) and dicyclohexylcarbodii-

mide (DCC) (126.0 mg) were added successively to the solution and the mixture was stirred for 5 min under ice cooling. The mixture was diluted with EtOAc (15 ml) and the solution was washed with 5% HCl, satd NaHCO₃ and satd NaCl successively. The dried EtOAc extract was concentrated in vacuo to give a residue which was purified by prep. TLC (CHCl₃-Me₂CO 9:1) to give the (R)-MTPA ester 5 (40.3 mg). ¹H NMR (CDCl₃): δ 1.44 (3H, d, $J = 6.4 \text{ Hz}, 9-\text{H}_3$), 1.88 (1H, ddd, J = 14.2, 6.4 and $4.9 \text{ Hz}, 7-\text{H}_1$), $2.19 (3H, m, 4-\text{H}_2, 7-\text{H}_1)$, $3.56 (3H, m, 4-\text{H}_2, 7-\text{H}_1)$ d, J = 1.0 Hz, OMe), 4.30 (1H, m, 6-H), 5.34 (1H, m, 8-H), 5.97 (1H, br.d, J = 9.8 Hz, 3-H), 6.77 (1H, m, 4-H), 7.38 (3H, m), 7.52 (2H, m); HR-FABMS 371.1087 (Calculated (negative): m/z $C_{18}H_{18}O_5F_3$: 371,1107).

(S)-MTPA ester (6) from euscapholide (1)

Euscapholide (1) (21.8 mg) was esterified with (*S*)-MTPA as described above to give the (*S*)-MTPA ester **6** (37.8 mg). 1 H NMR (CDCl₃): δ 1.36 (3H, d, J = 6.3 Hz, 9-H₃), 1.93 (1H, dt, J = 14.7 and 5.9 Hz, 7-H₁), 2.26 (1H, dt, J = 14.7 and 7.1 Ht, 7-H₁), 2.32 (2H, m, 5-H₂), 3.51 (3H, d, J = 1.0 Hz, OMe), 4.46 (1H, m, 6-H), 5.36 (1H, m, 8-H), 6.02 (1H, br.d, J = 9.8 Hz, 3-H), 6.84 (1H, m, 4-H), 7.41 (3H, m), 7.51 (2H, m); HR-FABMS (negative): m/z 371.1136 [M-H]⁻ (Calculated for $C_{18}H_{18}O_5F_3$; 371.1107).

Euscapholide glucoside tetraacetate (4)

Compound **3** (5.7 mg) was acetylated as above to give the tetraacetate **4** (7.7 mg) as an amorphous powder. IR v_{max} (CHCl₃) cm⁻¹: 1740, 1370, 1220, 1040; ¹H NMR (CDCl₃): δ 1.32 (3H, d, J = 6.4 Hz, 9-H₃), 1.73 (1H, ddd, J = 14.2, 6.8 and 5.4 Hz, 7-H₁), 2.00, 2.02, 2.03, 2.09 (each 3H, s, 4 × OAc), 2.37 (2H, m, 5-H₂), 3.71 (1H, m, 5'-H), 4.05 (1H, m, 8-H), 4.14 (1H, dd, J = 12.2 and 2.5 Hz, 6'-H₁), 4.25 (1H, dd, J = 12.2 and 5,4 Hz, 6'-H₁), 4.49 (1H, m, 6-H), 4.58 (1H, d, d, d = 7.8 Hz, 1'-H), 4.95 (1H, dd, d = 9.8 and 7.8 Hz, 2'-H), 5.05 (1H, d, d = 9.8 Hz, 4'-H), 5.19 (1H, d, d = 9.8 Hz, 3'-H), 6.04 (1H, dd, d = 9.8 and 2.0 Hz, 3-H), 6.89 (1H,

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m, 4-H); HR-FABMS (negative): m/z 485.1657 [M-H]⁻ (Calculated for $C_{22}H_{29}O_{12}$: 485.1659).

Enzymatic hydrolysis of euscapholide glucoside (3)

Compound 3 (20.1 mg) was dissolved in EtOH- $\rm H_2O$ (1:9, 10 ml) and β -glucosidase (ex almond) (purchased from Toyobo Co. Ltd., Osaka, Japan) (17.6 mg) was added. The mixture was incubated at 37° for 24 hr. The solution was extracted with EtOAc (10 ml × 3). The EtOAc extracts were washed with sat. NaCl aq. solution, dried and evaporated in vacuo to give the aglucone (7.2 mg) as a colourless oil, $[\alpha]_D^{21} + 116.4^\circ$ (MeOH, c 0.36), HR-EIMS: m/z 156.0736 [M] $^+$ (Calculated for $\rm C_8H_{12}O_3$: 156.0786). The compound was identified with the authentic sample of euscapholide (1) by comparisons of IR and $^1\rm H$ and $^{13}\rm C$ NMR spectra.

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