



## Bisquinolinone alkaloids from *Melicope ptelefolia*

Christine Kamperdick<sup>a</sup>, Nguyen Hong Van<sup>b</sup>, Tran Van Sung<sup>b</sup>, Günter Adam<sup>a,\*</sup>

<sup>a</sup>Institute of Plant Biochemistry, Weinberg 3, 06120 Halle/Saale, Germany

<sup>b</sup>Institute of Chemistry, National Centre for Natural Science and Technology of Vietnam, Nghia Do, Tu Liem, Hanoi, Viet Nam

Received 15 June 1998

### Abstract

The investigation of the leaves of *Melicope ptelefolia* (Rutaceae) afforded, besides *N*-methylflindersine, two new bisquinolinone alkaloids named melicobisquinolinone A and B. Their structures were established by MS and NMR spectroscopy, especially NOE and HMBC experiments. © 1998 Published by Elsevier Science Ltd. All rights reserved.

**Keywords:** *Melicope ptelefolia*; Rutaceae; Bisquinolinone alkaloids; Melicobisquinolinones A and B

### 1. Introduction

In continuation of our phytochemical studies on the Vietnamese medicinal plant *Melicope ptelefolia* (Champ. ex Benth.) Hartley [= *Evodia leptota* (Spreng.) Merr.], which afforded a series of 2,2-dimethyl-2*H*-1-benzopyrans (Kamperdick, Van, Sung, & Adam, 1997) and two benzopyran dimers (Kamperdick, Van, Sung, & Adam, 1998), we now report the isolation and structure elucidation of the two new bisquinolinone alkaloids, melicobisquinolinone A and B, from this plant.

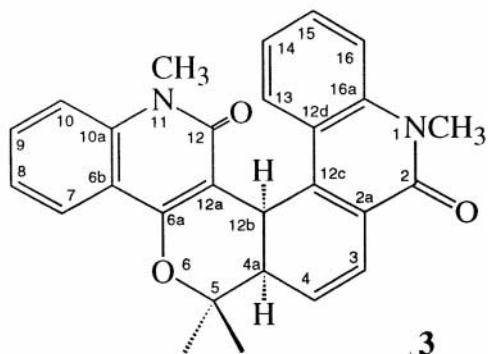
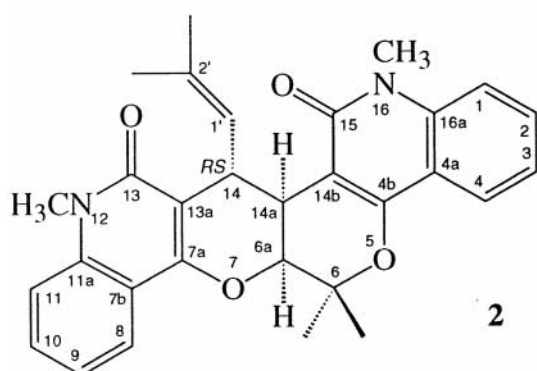
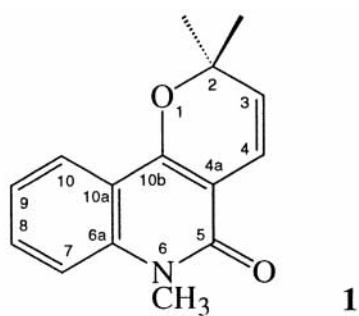
### 2. Results and discussion

The *n*-hexane extract of the leaves upon silica gel chromatography yielded large amounts of an alkaloid of *m/z* 241 ( $[M]^+$ ,  $C_{15}H_{15}NO_2$ ), identified as *N*-methylflindersine (**1**) from its MS and NMR spectra (see Section 3).

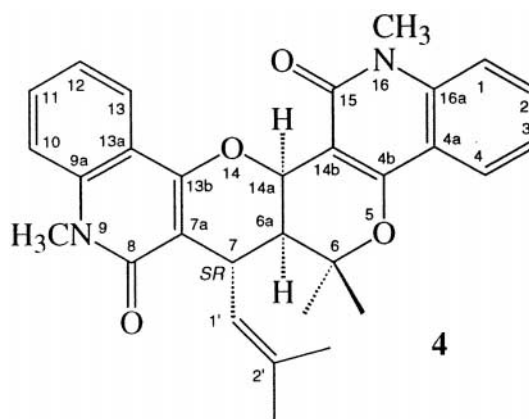
In the same manner chromatography of the EtOAc extract afforded two minor alkaloids **2** and **3**. The HRMS of compound **2** shows the molecular ion as base peak at *m/z* 482.2188 with the elemental composition  $C_{30}H_{30}N_2O_4$ , corresponding exactly to two molecules of **1**. Other prominent peaks at *m/z* 241, 227

and 226 occur also in the MS of **1** representing there the  $[M]^+$ ,  $[M-CH_2]^+$  and  $[M-Me]^+$ -ions. The  $^{13}C$  NMR spectrum of **2** shows eight pairs of carbon signals which correspond in their chemical shifts to the quinolinone moiety of **1** (positions 5–10a). The presence of two quinolinone units is also confirmed by the C–H long-range correlations (Table 1), suggesting **2** as a dimer of **1**. Furthermore, the NMR spectra display a trisubstituted double bond, three aliphatic methine groups and four quaternary methyl groups. The sequence of the CH-groups follows from the  $^3J_{HH}$  coupling constants [ $\delta_H$  4.53 (d,  $J = 4.1$  Hz), 3.25 (dd,  $J = 4.1$  and 1.9 Hz), 5.58 dd ( $J = 9.5$  and 1.9 Hz), 5.17 (d,  $J = 9.5$  Hz)]. From these data together with the  $^1J_{CH}$  correlations from the HMQC experiment, the subunit –OCH–CH–CH=CH– is established. The connection of this subunit with the two quinolinone moieties was deduced from the C–H long-range correlations (Table 1, Fig. 1) and establishes a six-cyclic constitution. The *cis*-junction of the two dihydropyran rings was obtained from the relatively small  $^3J_{HH}$  coupling of 4.1 Hz between H-6a ( $\delta$  4.53) and H-14a ( $\delta$  3.25) and the NOE effect (Table 1) between these protons. According to this, the 6 $\alpha$ -methyl group shows interactions to both H-6a and H-14a. The  $\alpha$ -orientation of the isobutenyl side chain is revealed by the NOE effect between H-6a and the olefinic proton H-1' ( $\delta$  5.17). The assignments of the (*E*)- and (*Z*)-2'-methyl groups were done according to the NOE inter-

\* Corresponding author.



actions between H-1 and the protons at  $\delta$  1.78 [2-Me (*E*)] as well as between H-14 ( $\delta$  5.58) and the protons at  $\delta$  2.25 [2-Me (*Z*)]. Further NOE effects of H-6a to both the 6 $\alpha$ - and 6 $\beta$ -methyl groups, which are of equal size, indicate a half-chair or a twisted conformation for the dihydropyran ring with a staggered arrangement of the substituents and excludes a half-boat conformation. The second pyran ring is also supposed to take a half-chair or twisted conformation with a dihedral angle of about 90° between H-14 and H-14a (according to the Dreiding model), which explains the small  $^3J_{HH}$  coupling of 1.9 Hz. Such a conformation is confirmed also by comparison with NMR data of paraensidimerine D (**4**), another dimer of **1**, which was isolated from *Euxylophora paraënsis* (Rutaceae). Especially the  $^3J_{HH}$  couplings of **4** are very similar to those of **2**. From the X-ray analysis of **4**, the twisted conformation of the dihydropyran rings and a dihedral



angle of 86.9° between H-6a and H-7 (which correspond to H-14 and H-14a in **2**) were obtained (Jurd, Wong, & Benson, 1982).

Paraensidimerine D (**4**) is supposed to be biosynthetically formed by a hetero Diels-Alder reaction of *N*-methylflindersine (**1**) with a C-prenylquinolinone precursor **1a** (Jurd et al., 1982; Ngadjui et al., 1989), arising formally from **1** by a retro Diels-Alder reaction. Thus, alkaloid **2** may be formed in a similar manner, however with a different arrangement of the educts (Fig. 2).

A striking feature in the NMR data of **2** is the strong deshielding of the aliphatic H-14 ( $\delta$  5.58) and the shielding of the connected carbon C-14 ( $\delta$  28.7). The two amide groups with a partially negative charged oxygen and a partially positive charged nitrogen cause electric fields. This C-H bond is situated close and parallel to both electric fields, which push the electrons from the proton H-14a towards the carbon.

The HRMS of alkaloid **3** gives the molecular ion at  $m/z$  424.1768 indicating the elemental composition  $C_{27}H_{24}N_2O_3$ . The base peak at  $m/z$  365 is formed by loss of  $-OC_3H_7$ . A fragment at  $m/z$  226 [*N*-methylflindersine-Me]<sup>+</sup> suggests that **3** also contains a *N*-methylflindersine unit. As in **2** the NMR spectra of **3** shows two quinolinone units. Additionally, the NMR spectra exhibited one *cis*-disubstituted double bond ( $\delta_H$  6.06 and 6.98,  $J$  = 9.7 Hz,  $\delta_C$  = 126.6 and 126.0), two aliphatic CH-groups ( $\delta_H$  2.77/ $\delta_C$  41.8 and  $\delta_H$  4.69/ $\delta_C$  30.0) and a quaternary carbon ( $\delta_C$  77.2) bearing one oxygen atom and two methyl groups ( $\delta_C$  25.8 and 25.5,  $\delta_H$  1.74 and 1.56). Analysis of the C-H long-range correlations from the HMBC experiment (Table 2 and Fig. 1) gives the connection of the two quinolinone units with these functional groups and the carbon assignments, indicating also for **3** a six-cyclic ring-constitution. The *cis*-junction of C-4a and C-12b follows from the NOE effect between H-4a ( $\delta$  2.77) and H-12b ( $\delta$  4.69). The differentiation of the 5 $\alpha$ - and

Table 1.  $^{13}\text{C}$  NMR and  $^1\text{H}$  NMR spectral data of melicobisquinolinone A (**2**) in  $\text{CDCl}_3$  (75/300 MHz)

	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J in Hz)	C–H long-range correlations	NOE-effects
1	113.7 <sup>a</sup>	7.22 <sup>a</sup> m	H-3 (w)	
2	130.5	7.45 m	H-4 (w)	
3	121.2	7.16 <sup>b</sup> m	H-1	
4	123.3	7.93 dd (8.1/1.5)	H-2	
4a	115.8 <sup>b</sup>	–	H-1, H-3	
4b	156.3	–		
6	77.9	–	H-6a, 6 $\alpha$ -Me, 6 $\beta$ -Me	
6a	73.7	4.53 d (4.1)	H-14, 6 $\alpha$ -Me, 6 $\beta$ -Me	H-14a, H-1', 6 $\beta$ -Me
7a	153.9	–	H-14	
7b	116.0 <sup>b</sup>	–	H-9, H-11	
8	122.6	7.89 dd (8.1/1.5)	H-10	
9	121.2	7.13 <sup>b</sup> m	H-11	
10	129.9	7.45 m	H-8	
11	113.6 <sup>a</sup>	7.21 <sup>a</sup>	H-10	
11a	138.8	–	H-8, H-10, 12-Me	
13	162.54 <sup>c</sup>	–	H-14 (w), 12-Me	
13a	108.5	–	H-14	
14	28.7	5.58 dd (9.5/1.9)		H-14a, 2'-Me (Z)
14a	34.5	3.25 dd (4.1/1.9)		H-6a, H-14, H-1', 6 $\alpha$ -Me
14b	104.0	–	H-6a, H-14	
15	162.45 <sup>c</sup>	–	16-Me	
16a	139.1	–	H-2, H-4, 16-Me	
1'	126.5	5.17 dm (9.5)	H-14, 2'-Me (E), 2'-Me (Z)	H-6a, 2'-Me (E)
2'	135.5	–	H-14, 2'-Me (E), 2'-Me (Z)	
6 $\alpha$ -Me	22.8	1.38 s	6 $\beta$ -Me	H-6a, H-14a, 6 $\beta$ -Me
6 $\beta$ -Me	25.2	1.93 s	6 $\alpha$ -Me	H-8, H-6a, 6 $\alpha$ -Me
12-Me	29.24 <sup>d</sup>	3.61 <sup>c</sup> s		
16-Me	29.17 <sup>d</sup>	3.60 <sup>c</sup> s		
2'-Me (E)	26.1	1.78 d (1.1)	2'-Me (Z)	H-1', 2'-Me (Z)
2'-Me (Z)	18.8	2.25 d (1.1)	2'-Me (E)	H-14, H-14a (w)

<sup>a</sup>, <sup>b</sup>, <sup>c</sup>, <sup>d</sup> Assignments interchangeable.

w = weak.

the 5 $\beta$ -methyl group is enabled by the NOE interaction between H-12b ( $\delta$  4.69,  $\alpha$ ) and the methyl protons at  $\delta$  1.56, which thus belongs to the  $\alpha$ -methyl group. Accordingly, the 5 $\beta$ -methyl group at  $\delta$  1.74 shows an NOE interaction to H-4 ( $\delta$  6.06). Like in **2**, the nearly equal-sized NOE effects of both methyl groups at C-5 ( $\delta_{\text{H}}$  1.56 and 1.74) indicate the expected half-chair or twisted conformation of the dihydropyran ring. The residual NOE effects (Table 2) confirm the configuration of **3**.

Because both alkaloids **2** and **3** show no Cotton effect in the CD spectra, they are suggested to be racemic.

*N*-methylflindersine is a characteristic constituent of plants from the Rutaceae and Meliaceae family. It is known to possess insect growth inhibitory, antifeedant and fungistatic activities (Hegnauer, 1990). The alkaloids **1–3** were tested for fungitoxic activity against *Cladosporium cucumerinum*. The lowest amount necessary to inhibit mycel growth, was 12.5 nmol of **1** and 3 nmol of **3**. Alkaloid **2** showed no activity, which reflects, compared with the found high activity of **3**, a dramatic structural influence of both dimer types.

### 3. Experimental

EI-MS: AMD 402, 70 eV. NMR: 1D: Varian Gemini 300, 2D: Varian Unity 500. CC: silica gel 60, 70–200 and 230–400 mesh ASTM (Merck); TLC: precoated silica gel plates 60 F<sub>254</sub>, thickness 1 mm (Merck).

#### 3.1. Plant material

Leaves and branches of *Melicope ptelefolia* were collected in Lao cai, North Vietnam, in June 1994 and identified by Dr T.D. Dai. A voucher specimen (No. 382) is deposited in the Institute of Ecology and Natural Resources, National Centre for Natural Science and Technology, Hanoi, Vietnam.

#### 3.2. Extraction and isolation

Dried leaves (500 g) were extracted 3 $\times$  with 80% aq. MeOH at room temp. and the organic solvent removed under red. pres. The aq. residue was extracted 3 $\times$  with *n*-hexane and 3 $\times$  with EtOAc, giving 17.3 g

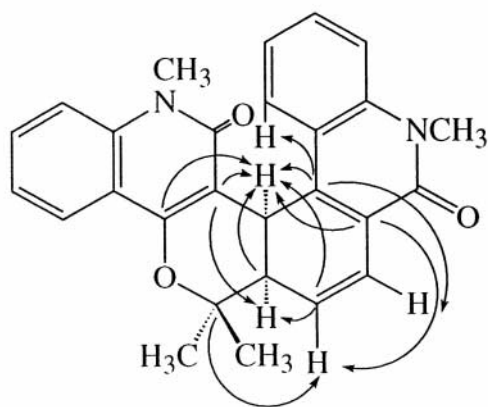
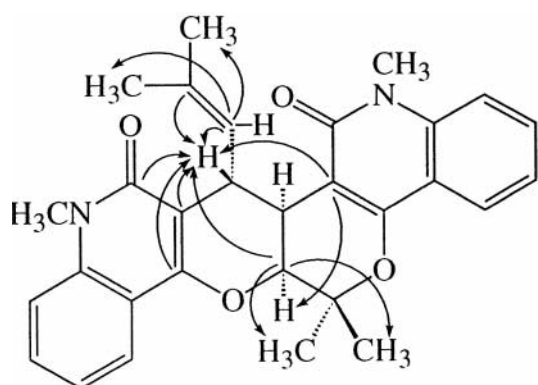


Fig. 1.

*n*-hexane extract and 15.8 g EtOAc extract. The *n*-hexane extract was sepd by chromatography on silica gel (200 g, 70–200 mesh) with increasing amounts of EtOAc in *n*-hexane as eluent (2–100% EtOAc, 237 frs, each 20 ml). Frs 205–236 (2.43 g) afforded 1.35 g of **1** by chromatography on silica gel 60 (230–400 mesh) using *n*-hexane–EtOAc (1:1).

The EtOAc extract was fractionated on silica gel 60 (250 g, 70–200 mesh) with an eluent of increasing polarity (10–100% CHCl<sub>3</sub> in *n*-hexane, followed by 10–50% MeOH in CHCl<sub>3</sub>) giving 400 frs with 20 ml. Frs 82–112 (1.51 g) yielded upon further chromatography 0.60 g of **1**. Frs 112–121 (1.59 g) were chro-

matographed by CC on silica gel 60 (150 g, 230–400 mesh) using *n*-hexane–EtOAc (4:1). Frs 22–30 (170 mg) were further purified by CC on silica gel 60 (230–400 mesh) eluting with *n*-hexane–EtOAc (1:1) followed by preparative TLC using *n*-hexane–CHCl<sub>3</sub>–diethylamine (1:4:1) to afford 4.9 mg of **3**. Preparative TLC of frs 31–40 (30 mg) using *n*-hexane–EtOAc (1:9) yielded 7.2 mg of **2**.

The bioassay with *Cladosporium cucumerinum* was done according to Gottstein, Groß, and Lehmann (1984).

### 3.2.1. *N*-Methylflindersine (**1**)

Crystals, m.p. 82–84. UV  $\lambda_{\max}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 348 (4.17), 226 (4.62). IR  $\nu_{\max}^{\text{KBr}}$  (cm<sup>-1</sup>): 3055, 2971, 2934, 1642, 1623, 1569, 1507, 1461, 1417, 1359, 1325, 1148, 1123, 1091, 895, 747, 720. MS  $m/z$  (rel. int.): 241 [M]<sup>+</sup> (13), 227 [M–CH<sub>2</sub>]<sup>+</sup> (15), 226 [M–Me]<sup>+</sup> (100), 211 [M–2Me]<sup>+</sup> (1.4), 200 (1.8), 183 (3.2), 113 (3.7). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 7.97 (1 H, dd,  $J$  = 8.2 and 1.5 Hz, H-10), 7.55 (1 H, ddd,  $J$  = 8.6, 7.1 and 1.5 Hz, H-8), 7.32 (1 H, d,  $J$  = 8.6 Hz, H-7), 7.23 (1 H, ddd,  $J$  = 8.2, 7.1 and 1.0 Hz, H-9), 6.76 (1 H, d,  $J$  = 9.8 Hz, H-4), 5.54 (1 H, d,  $J$  = 9.8 Hz, H-3), 3.70 (3 H, s, N–Me), 1.52 (6 H, s, 2-Me<sub>2</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$ : 161.0 (C-5), 155.2 (C-10b), 139.4 (C-6a), 130.8 (C-8), 126.3 (C-3), 123.1 (C-10), 121.7 (C-9), 118.0 (C-4), 116.1 (C-10a), 114.0 (C-7), 105.9 (C-4a), 78.7 (C-2), 29.2 (NMe), 28.2 (2-Me<sub>2</sub>). The proton shifts are in agreement with reference data (Hifnawy, Vaquette, Sévenet, Pouset, & Cavé, 1977). As no carbon shifts of **1** could be found in the literature, these data are added here. Carbons were assigned by comparison with **2** and **3**.

### 3.2.2. Melicobisquinolinone A (**2**)

Amorphous. UV  $\lambda_{\max}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 318 (4.13), 274 (4.14), 226 (4.92). IR  $\nu_{\max}^{\text{CHCl}_3}$  (cm<sup>-1</sup>): 3002, 2929, 2856, 1646, 1625, 1595, 1577, 1503, 1465, 1389, 1331, 1190, 1164, 1120, 1094. MS  $m/z$  (rel. int.): 482 [M]<sup>+</sup> (100), 467 [M–Me]<sup>+</sup> (5), 439 (23), 427 (33), 308 (5), 294 (6), 264 (6), 252 (15), 242 [N-methylflindersine + H]<sup>+</sup> (51), 241 [N-methylflindersine]<sup>+</sup> (65), 227 [N-methylflinder-

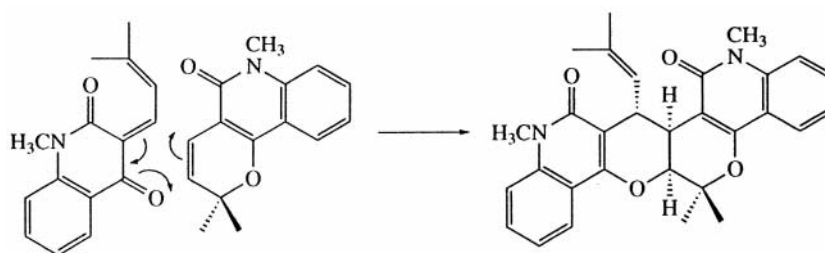


Fig. 2.

Table 2.  $^{13}\text{C}$  NMR and  $^1\text{H}$  NMR spectral data of melicobisquinolinone B (**3**) in  $\text{CDCl}_3$  (75/300 MHz)

	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J in Hz)	CH long-range correlation	NOE-effects
2	160.0	—	1-Me	
2a	123.3	—	H-4, H-12b	
3	126.0	6.98 dd (9.7/2.7)	H-4	H-4
4	126.62 <sup>a</sup>	6.06 d (9.7)	H-4a (w), H-12b	H-3, H-4a, 5 $\beta$ -Me
4a	41.8	2.77 dd (5.5/2.7)	H-3 (w), H-4, H-12b, 5 $\alpha$ -Me, 5 $\beta$ -Me	H-4, H-12b, 5 $\alpha$ -Me, 5 $\beta$ -Me
5	77.2	—	5 $\alpha$ -Me, 5 $\beta$ -Me, H-4 (w)	
6a	157.6	—	H-7, H-12b,	
6b	116.0	—	H-8, H-10	
7	123.2	7.99 dd (8.6/1.5)	H-9	H-8
8	121.3	7.20 t (7.3)	H-10	
9	130.6	7.51 m	H-7	
10	113.6	7.21 d (7.6)	H-8	
10a	138.9	—	H-7, H-9, 11-Me	
12	161.9	—	11-Me	
12a	104.2	—	H-4a (w), H-12b	
12b	30.0	4.69 d (5.5)	H-4 (w)	H-4a, H-13, 5 $\alpha$ -Me
12c	142.0	—	H-3 (w), H-12b, H-13	
12d	123.0	—	H-12b, H-14, H-16	
13	126.56 <sup>a</sup>	8.28 d (7.9)	H-15	H-12b, H-14
14	121.6	7.35 m	H-16	
15	129.1	7.54 m	H-13, H-16 (w)	
16	114.6	7.40 d (7.9)	H-14 (w)	
16a	138.5	—	H-13, H-15, 1-Me	
1-Me	29.8	3.75 s		H-16
5 $\alpha$ -Me	25.5	1.56 s	5 $\beta$ -Me	H-4a, H-7 (w), H-12b, 5 $\beta$ -Me
5 $\beta$ -Me	25.8	1.74 s	5 $\alpha$ -Me	H-4, H-4a, H-7 (w), 5 $\alpha$ -Me
11-Me	29.4	3.43 s		H-10

<sup>a</sup>Assignment interchangeable.

w = weak.

sine- $\text{CH}_2$ )<sup>+</sup> (21), 226 [*N*-methylflindersine-Me]<sup>+</sup> (97), 176 (8), 134 (11). HRMS 482.2188 [M]<sup>+</sup> ( $\text{C}_{30}\text{H}_{30}\text{N}_2\text{O}_4$  requires 482.2206).

### 3.2.3. Melicobisquinolinone B (**3**)

Amorphous. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 322 (4.08), 229 (4.80). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  ( $\text{cm}^{-1}$ ): 3000, 2929, 2855, 1642, 1620, 1592, 1502, 1464, 1385, 1328, 1158, 1119, 1015. MS  $m/z$  (rel. int.): 424 [M]<sup>+</sup> (43), 366 [M-C<sub>3</sub>H<sub>6</sub>O]<sup>+</sup> (43), 365 [M-C<sub>3</sub>H<sub>7</sub>O]<sup>+</sup> (100), 351 (23), 349 (16), 337 (15), 226 [*N*-methylflindersine-Me]<sup>+</sup> (11), 192 (15). HRMS 424.1768 [M]<sup>+</sup> ( $\text{C}_{27}\text{H}_{24}\text{N}_2\text{O}_3$  requires 424.1787).

### Acknowledgements

We thank the Bundesministerium für Bildung, Wissenschaft, Forschung und Technologie, Bonn, for financial support, Dr. J. Schmidt, Halle, for mass spec-

tra, as well as Dr. T.D. Dai, Hanoi, for the identification of the plant material. NHV is indebted to the Volkswagen-Stiftung, Hannover, for a grant.

### References

- Gottstein, D., Groß, D., & Lehmann, H. (1984). *Zeitschrift für die gesamte Hygiene und ihre Grenzgebiete*, 30, 620.
- Hegnauer, R., *Chemotaxonomie der Pflanzen*, Band IX. Birkhäuser, Basel, 1990, pp. 38, 449.
- Hifnawy, M. S., Vaquette, J., Sévenet, T., Pousset, J. -L., & Cavé, A. (1977). *Phytochemistry*, 16, 1035.
- Jurd, L., Wong, R. Y., & Benson, M. (1982). *Australian Journal of Chemistry*, 35, 2505.
- Kamperdick, C., Van, N. H., Sung, T. V., & Adam, G. (1997). *Phytochemistry*, 45, 1049.
- Kamperdick, C., Van, N.H., Sung, T.V., & Adam, G. (1998). *Phytochemistry*, 48, 1055.
- Ngadjui, B. T., Ayafor, J. F., Mitaku, S., Skaltsounis, A. -L., Tillequin, F., & Koch, M. (1989). *Journal of Natural Products*, 52, 300.