



## Japonenyne, halogenated C<sub>15</sub> acetogenins from *Laurencia japonensis*

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### Abstract

Two brominated C<sub>15</sub> nonterpenoid compounds, japonenyne-A and -B, with a novel 2,7-dioxabicyclo[4.3.0]nonane skeleton have been isolated from the red alga *Laurencia japonensis* Abe et Masuda. The structures of these metabolites were deduced from spectral evidence. © 1999 Elsevier Science Ltd. All rights reserved.

**Keywords:** *Laurencia japonensis*; Rhodomelaceae; Red alga; Acetogenin; Halogenated compound; Chemotaxonomy

### 1. Introduction

In continuing our chemotaxonomical studies on Japanese species of the red algal genus *Laurencia*, we have recently reported that *L. japonensis* (Abe & Masuda, 1998), collected from several locations in Japanese waters, contains aplysiadiol (**1**) (diterpenoid) as a major metabolite along with 2,10-dibromo-3-chloro- $\alpha$ -chamigrene (**2**) (sesquiterpenoid), a set of halogenated compounds characteristic to this species (Takahashi, Suzuki, Abe, & Masuda, 1998).

Further investigation of the methanol extracts of this species led to the isolation of two novel halogenated C<sub>15</sub> nonterpenoids, designated as japonenyne-A (**3**) and -B (**4**), along with japonenyne-C (**5**). Japonenyne-A (**3**) and -B (**4**) consist of 2,7-dioxabicyclo[4.3.0]nonane skeleton with the same molecular formula of C<sub>15</sub>H<sub>18</sub>Br<sub>2</sub>O<sub>2</sub> as that of laurenenyne-A (**6**) and -B (**7**), which have previously been obtained from this species collected at Kamishima, Mie Prefecture, Japan (Suzuki, Matsuo, & Masuda, 1993; Takahashi et al., 1998). We wish to report herein the isolation and structural elucidation of these novel halogenated metabolites.

### 2. Results and discussion

Chromatographic separation of the methanol extracts obtained from *L. japonensis* collected at several locations

in Japanese waters has led to the isolation of three halogenated C<sub>15</sub> nonterpenoids, **3–5**, together with aplysiadiol (**1**), which has previously been isolated from a Japanese sea hare *Aplysia kurodai* (Ojika, Yoshida, Okumura, Ikeda, & Yamada, 1990), and 2,10-dibromo-3-chloro- $\alpha$ -chamigrene (**2**), which has already been found in other *Laurencia* species (Howard & Fenical, 1975; Suzuki, Furusaki, & Kurosawa, 1979; Suzuki, Segawa, Kikuchi, Suzuki, & Kurosawa, 1985). These C<sub>15</sub> nonterpenoids, which exist as colorless compounds when developed on Si gel TLC plate, turned to characteristic dark red-colored spots when sprayed with 5% phosphomolybdic acid in ethanol and subjected to heating. These compounds were also observed to be very unstable and decomposed even when stored at –18°C for just a few days.

Japonenyne-A (**3**), colorless oil, [ $\alpha$ ]<sub>D</sub><sup>21</sup> +40.2° (*c* 1.39; CH<sub>2</sub>Cl<sub>2</sub>), was shown to have the molecular formula of C<sub>15</sub>H<sub>18</sub>Br<sub>2</sub>O<sub>2</sub> by a high-resolution mass spectrum (*m/z* 389.9673 [M],  $\Delta$ +2.0 mmu). The IR spectrum showed the presence of terminal acetylenic ( $\nu_{\max}$  3310 cm<sup>–1</sup>) and vinyl ethereal (1690 and 1205 cm<sup>–1</sup>) functionalities. The presence of a 2-penten-4-ynyl moiety, which is frequently encountered in *Laurencia*'s C<sub>15</sub> acetogenin, was readily recognized by the <sup>1</sup>H NMR spectrum (Table 1) [ $\delta_{\text{H}}$  2.85 (1H, d, *J*=1.5 Hz), 5.57 (1H, dddd, *J*=16.1, 1.5, 1.5 and 1.5 Hz) and 6.19 (1H, ddd, *J*=16.1, 6.4 and 6.4 Hz)]. The magnitude of coupling constants (*J*<sub>3,4</sub>=16.1 Hz) for the H-3 and H-4 as well as the chemical shift value ( $\delta_{\text{H}}$  2.85) of the acetylenic proton indicated the geometry of the double bond at C-3 to be *E* (Suzuki & Kurosawa, 1987).

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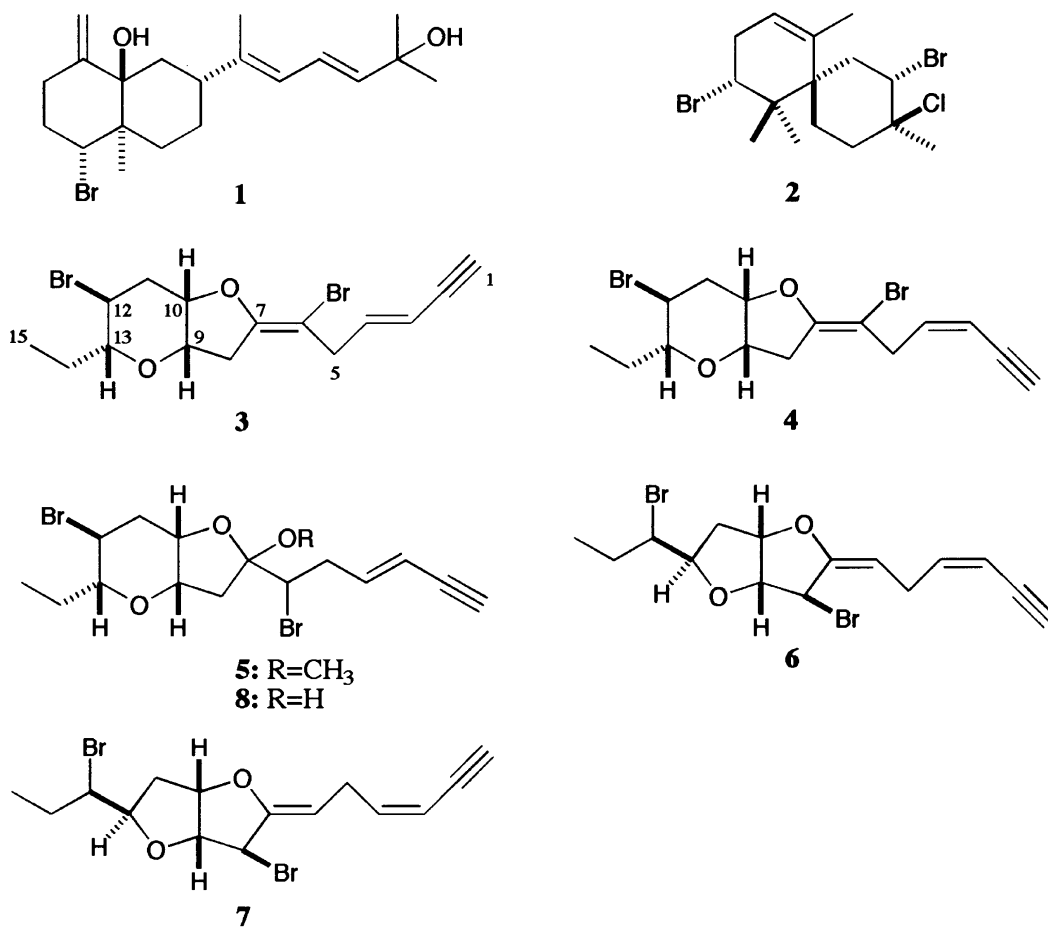
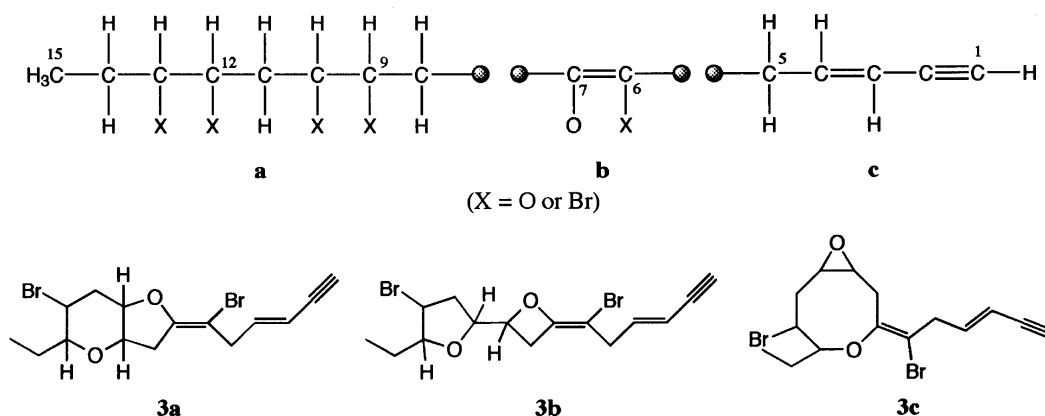


Table 1

<sup>13</sup>C (100 MHz, DEPT) and <sup>1</sup>H NMR (400 MHz) and HMBC data<sup>a</sup> for japonenyne-A (**3**)

C <sup>b</sup>	<sup>13</sup> C (δ)	<sup>1</sup> H (δ)	<i>J</i> (Hz)	Long-range correlations
1	<sup>c</sup>	2.85	d (1.5)	
2	83.49			H-4
3	110.19	5.57	dddd (16.1, 1.5, 1.5, 1.5)	H-1, H <sub>2</sub> -5
4	141.64	6.19	ddd (16.1, 6.4, 6.4)	H <sub>2</sub> -5
5	38.90	3.13	m	H-4
6	88.44			H-4, H <sub>2</sub> -5, H <sub>β</sub> -8
7	152.58			H <sub>2</sub> -5, H <sub>α</sub> -8, H-9
8	36.76	2.69	d (16.6 (H <sub>α</sub> ))	
		2.62	dd (16.6, 3.9 (H <sub>β</sub> ))	
9	75.68	4.26	dd (3.9, 2.4)	H <sub>α</sub> -8, H <sub>α</sub> -11
10	89.91	4.21	ddd (3.9, 2.4, 2.4)	H <sub>α</sub> -8, H-9, H <sub>α</sub> -11
11	37.33	2.94	ddd (14.6, 4.4, 2.4 (H <sub>α</sub> ))	H-13
		2.22	ddd (14.6, 12.2, 3.9 (H <sub>β</sub> ))	
12	46.38	3.97	ddd (12.2, 9.3, 4.4)	H <sub>2</sub> -11
13	80.96	3.31	ddd (9.3, 8.3, 2.4)	H-12, H <sub>3</sub> -15
14	26.13	2.01	ddq (14.6, 2.4, 7.3 (H <sub>α</sub> ))	H <sub>3</sub> -15
		1.54	ddq (14.6, 8.3, 7.3 (H <sub>β</sub> ))	
15	9.22	0.94	t (7.3)	H <sub>α</sub> -14

<sup>a</sup> Measured in chloroform-*d*<sub>1</sub>.<sup>b</sup> All assignments are based on the results of the HSQC spectrum.<sup>c</sup> This carbon was not observed in the <sup>13</sup>C NMR spectrum.

Fig. 1. Partial structural units and possible structures for **3**.

Moreover, japonenyne-B (**4**), colorless oil,  $[\alpha]_D^{21} + 24.4^\circ$  ( $c$  0.61;  $\text{CH}_2\text{Cl}_2$ ), was an isomeric metabolite of japonenyne-A (**3**). The  $^1\text{H}$  and  $^{13}\text{C}$  NMR signals (Table 2) on the carbons from C-8 to C-15 in both compounds **4** and **3** were virtually identical. Comparison of the spectral data of **4** and **3** suggested that japonenyne-B is a geometric isomer with *Z*-configuration at C-3.

Detailed analysis of the  $^1\text{H}$  and  $^{13}\text{C}$  NMR,  $^1\text{H}$ - $^1\text{H}$  COSY and HSQC spectra of **3** revealed the presence of partial structural units **a** and **c** (Figure 1). In the  $^{13}\text{C}$  NMR spectrum, the chemical shift ( $\delta_{\text{C}}$  46.38) of the methine carbon at C-12 in the unit **a** indicated that a bromine atom is attached to this carbon. Furthermore, since the IR spectrum showed no absorption ascribable to an hydroxyl group, the remaining substituents at C-9, C-10

and C-13 were verified to be ethereal oxygen atoms based upon the chemical shift values of the pertinent carbons at  $\delta_{\text{C}}$  75.68, 89.91 and 80.96, respectively. As described above, in the IR spectrum the intense absorption at  $\nu_{\text{max}}$  1690 and  $1205\text{ cm}^{-1}$  suggested the presence of a tri-substituted vinyl ether grouping (partial unit **b**) which was further supported by the signals of two quaternary carbons at  $\delta_{\text{C}}$  152.58 and 88.44 in the  $^{13}\text{C}$  NMR spectrum. Since the  $\text{C}_{15}$  nonterpenoid cyclic ethers of *Laurencia* species seem to be biosynthesized from straight-chain precursors, viz. laurediols from *L. nipponica* (Kikuchi, Suzuki, Kurosawa, & Suzuki, 1991), the unit **b** necessitates being linked to C-8 at one end and to C-5 at the other end. Judging from the chemical shift value of  $\delta_{\text{C}}$  88.44, the fourth substituent of the unit **b** must obviously

Table 2  
 $^{13}\text{C}$  (100 MHz, DEPT) and  $^1\text{H}$  NMR (400 MHz) and HMBC data<sup>a</sup> for japonenyne-B (**4**)

C <sup>b</sup>	$^{13}\text{C}$ ( $\delta$ )	$^1\text{H}$ ( $\delta$ )	$J$ (Hz)	Long-range correlations
1	80.20	3.12	d (2.0)	
2	82.14			H-5
3	109.35	5.56	br dd (10.7, 2.0)	H <sub>2</sub> -5
4	141.80	6.07	ddd (10.7, 7.3, 6.8)	H <sub>2</sub> -5
5	36.44	3.43	dd (16.1, 7.3)	
		3.30	dd (16.1, 6.8)	
6	89.73			H <sub>2</sub> -5, H <sub>2</sub> -8
7	152.07			H <sub>2</sub> -5, H <sub>2</sub> -8
8	37.03	2.81	d (16.6 (H <sub>2</sub> ))	
		2.68	dd (16.6, 4.4 (H <sub>2</sub> ))	
9	75.84	4.25	dd (4.4, 2.0)	H <sub>2</sub> -8
10	80.92	4.19	ddd (3.9, 2.4, 2.0)	H <sub>2</sub> -8, H-9
11	37.46	2.93	ddd (14.7, 4.4, 2.4 (H <sub>2</sub> ))	
		2.21	ddd (14.7, 11.7, 3.9 (H <sub>2</sub> ))	
12	46.53	3.97	ddd (11.7, 10.3, 4.4)	H <sub>2</sub> -11
13	81.04	3.30	ddd (10.3, 8.3, 2.4)	H <sub>2</sub> -11, H <sub>2</sub> -14, H <sub>3</sub> -15
14	26.22	2.01	ddq (14.6, 2.4, 7.3 (H <sub>2</sub> ))	H <sub>3</sub> -15
		1.52	ddq (14.6, 8.3, 7.3 (H <sub>2</sub> ))	
15	9.25	0.94	t (7.3)	H <sub>2</sub> -14

<sup>a</sup> Measured in chloroform- $d_1$ .

<sup>b</sup> All assignments are based on the results of the HSQC spectrum.

not be oxygen but bromine. Then in order to prove the planar structure, we measured  $^1\text{H}$ -detected heteronuclear multiple-bond  $^1\text{H}$ - $^{13}\text{C}$  correlation spectrum (HMBC). In the HMBC spectrum Table 1, the H-4 ( $\delta_{\text{H}}$  6.19) showed a cross peak to the quaternary carbon at  $\delta_{\text{C}}$  88.44 and the H-9 ( $\delta_{\text{H}}$  4.26) showed a cross peak to the quaternary carbon at  $\delta_{\text{C}}$  152.58, thus revealing that the carbon of  $\delta_{\text{C}}$  88.44 (vinyl bromide carbon) is linked to C-5 and hence that of  $\delta_{\text{C}}$  152.58 to C-7. Furthermore, since japonenynne-A (**3**) has six degrees of unsaturation, **3** has to contain two oxide rings, leading to three possible planar structures **3a**, **3b** and **3c** Fig. 1 assignable for japonenynne-A. The structure **3c**, however, could easily be ruled out because the  $^1\text{H}$  NMR spectrum exhibited no signals due to a 1,2-disubstituted oxirane ring (Tori, Komeno, & Nakagawa, 1964). Furthermore, in the  $^1\text{H}$  NMR spectrum the protons on the carbons from C-9 to C-13 showed coupling constants of  $J_{9,10}=2.4$  Hz,  $J_{10,11\alpha}=2.4$  Hz,  $J_{10,11\beta}=3.9$  Hz,  $J_{11\alpha,12}=4.4$  Hz,  $J_{11\beta,12}=12.2$  Hz and  $J_{12,13}=9.8$  Hz, which are typical equatorial–equatorial, equatorial–axial and axial–axial coupling constants due to the protons not on a tetrahydrofuran ring but on a tetrahydropyran ring with a chair-like conformation. Thus japonenynne-A should have the planar formula **3a**. The geometry of the double bond between C-6 and C-7 in **3** was defined by the NOESY correlation spectrum. As shown in Fig. 2, the correlation between  $\text{H}_2$ -5 and  $\text{H}_\alpha$ -8 indicated that the configuration of C-6 and C-7 is *Z*. Furthermore, the relative stereochemistry was also defined by the NOESY correlation spectrum. The NOEs between  $\text{H}_\beta$ -8/H-10, H-9/ $\text{H}_\beta$ -11, H-9/H-13, H-10/ $\text{H}_\beta$ -11,  $\text{H}_\beta$ -11/H-13,  $\text{H}_\alpha$ -11/H-12 and H-12/ $\text{H}_3$ -15 showed that the relative configuration between H-9, H-10 and H-13 are all *cis*. Likewise the gross structure of japonenynne-B was confirmed with the aid of the 2-D NMR spectra, such as  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC and HMBC Table 2. Moreover, the relative stereochemistry was also determined by the NOESY spectrum

Fig. 2 which was very similar to that of **3**. Therefore, the structures of japonenynne-A and -B should be shown as formulas **3** and **4**, respectively, each of which includes the relative configuration of  $9S^*$ ,  $10S^*$ ,  $12S^*$  and  $13R^*$ .

Many halogenated  $\text{C}_{15}$  acetogenins isolated from various *Laurencia* species are suggested to arise from (6*S*,7*S*)- or (6*R*,7*R*)-laurediol (Kikuchi et al., 1991). Japonenynne-A (**3**) may be biosynthesized from a bromohydrin **8** through an enzymatic E2-type dehydration. A bromohydrin **8** would be derived from (6*S*,7*S*)- or (6*R*,7*R*)- (3*E*)-laurediol via a bromonium ion-catalyzed cyclization and have a *threo* configuration at C-6 and C-7. The presence of a plausible precursor **8** should strongly be supported by the isolation of japonenynne-C (**5**), whose structure was determined by spectral analyses using extensive 2-D NMR such as  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC and HMBC as in the case of **3** and **4**. The stereochemistry at C-6 and C-7, remains unsettled because compound **5** decomposed while measuring the NOESY spectrum. However, compound **5** was obtained, not as a mixture of diastereomers, but as a single compound, based on the single  $^1\text{H}$  and  $^{13}\text{C}$  NMR signals due to the bromomethine at C-6 and methoxyl methyl group, it may nevertheless be an artefact formed during extraction. Japonenyne represent the first example of halogenated  $\text{C}_{15}$  acetogenins found from various species of the red algal genus *Laurencia* possessing a 2,7-dioxabicyclo[4.3.0]nonane skeleton and a bromine atom substituent at C-6.

### 3. Experimental

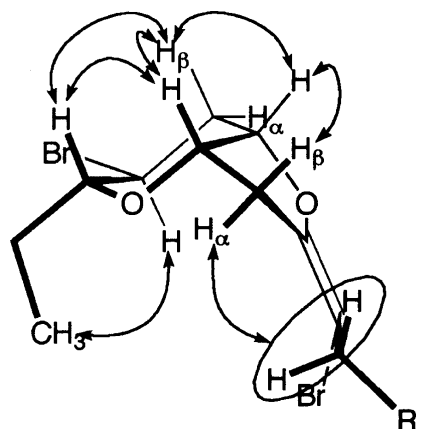
#### 3.1. General

The  $^1\text{H}$  NMR: 400 MHz and  $^{13}\text{C}$  NMR: 100 MHz,  $\text{CDCl}_3$  or  $\text{C}_6\text{D}_6$ , TMS as int. standard; LR-MS and HR-MS: 70 eV; CC: silica gel (Merck, Kieselgel 60, 70–230 mesh); prep. TLC: silica gel plate (Merck, Kieselgel 60  $\text{F}_{254\text{S}}$ ).

#### 3.2. Isolation

As recently described (Takahashi et al., 1998), the methanol extract (691 mg) of the Chinzei sample (SAP 062630) was fractionated by column chromatography on Si gel with a step gradient (hexane and EtOAc). The fraction (288 mg) eluted with hexane–EtOAc (9:1) was further subjected to prep. TLC with toluene to give japonenynne-A (**3**) (2.1 mg, 0.3%) and japonenynne-C (**5**) (16.6 mg, 2.4%) along with anhydroaplysiadiol (Takahashi et al., 1998).

Japonenynne-B (**4**) was obtained in 0.2% yield of the extract from Toyooka sample (SAP 062614) by combination of column and thin-layer chromatography in the same manner as described above. Japonenynne-A (**3**) has also been isolated from Toyooka and Heki sample



**3:** R = *trans*-2-penten-4-ynyl moiety  
**4:** R = *cis*-2-penten-4-ynyl moiety

Fig. 2. NOEs from NOESY spectra of japonenynne-A (**3**) and -B (**4**).

(SAP 062611) and japonenyne-B (**4**) from Iwami sample (SAP 062613) as the minor metabolite.

### 3.3. Japonenyne-A (**3**)

Colorless oil;  $[\alpha]_D^{21} +40.2^\circ$  ( $c$  1.39;  $\text{CH}_2\text{Cl}_2$ ); IR  $\nu_{\text{max}}$  ( $\text{CH}_2\text{Cl}_2$ )  $\text{cm}^{-1}$ : 3310, 2940, 1690, 1420, 1205, 1180, 1130, 1100, 1040 and 960;  $^1\text{H}$  and  $^{13}\text{C}$  NMR Table 1; EI–LR–MS  $m/z$  (rel. int.): 392 (5.5), 390 (10.7), 388 (5.6)  $[\text{M}]^+$ , 311 (15.3), 309 (14.8)  $[\text{M}-\text{Br}]^+$ , 281 (10.8), 279 (10.2), 267 (17.8), 229 (17.5)  $[\text{M}-2\text{Br}]^+$ , 191 (22.9), 189 (23.5), 147 (57.3), 127 (37.7), 109 (73.7), 105 (100), 81 (56.4), 77 (65.8) and 41 (25.0); EI–HR–MS  $m/z$ : 389.9673. Calcd for  $\text{C}_{15}\text{H}_{18}^{79}\text{Br}^{81}\text{BrO}_2$ , 389.9652  $[\text{M}]$ .

### 3.4. Japonenyne-B (**4**)

Colorless oil,  $[\alpha]_D^{21} +24.4^\circ$  ( $c$  0.61;  $\text{CH}_2\text{Cl}_2$ ); IR  $\nu_{\text{max}}$  ( $\text{CH}_2\text{Cl}_2$ )  $\text{cm}^{-1}$ : 3310, 2980, 1690, 1420, 1205, 1180, 1130, 1100, 1040, 950 and 920;  $^1\text{H}$  NMR Table 2; EI–LR–MS  $m/z$  (rel. int.): 392 (0.3), 390 (0.7), 388 (0.4)  $[\text{M}]^+$ , 311 (1.7), 309 (1.8)  $[\text{M}-\text{Br}]^+$ , 310 (2.4), 308 (2.1)  $[\text{M}-\text{HBr}]^+$ , 281 (10.8), 279 (10.6), 253 (1.4), 251 (1.5), 239 (3.5), 273 (3.3), 216 (39.2), 201 (100), 159 (42.5), 145 (46.7), 115 (31.0), 82 (70.0) and 80 (68.0); EI–HR–MS  $m/z$ : 389.9672. Calcd for  $\text{C}_{15}\text{H}_{18}^{79}\text{Br}^{81}\text{BrO}_2$ , 389.9652  $[\text{M}]$ .

### 3.5. Japonenyne-C (**5**)

Colorless oil;  $[\alpha]_D^{26} -35.5^\circ$  ( $c$  0.06;  $\text{CHCl}_3$ ); IR  $\nu_{\text{max}}$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 3330, 2940, 1545, 1460, 1435, 1210, 1110, 1040, 815 and 730;  $^1\text{H}$  NMR ( $\text{C}_6\text{D}_6$ ):  $\delta$  0.91 (3H, t,  $J=7.3$  Hz;  $\text{H}_3$ -15), 1.56 (1H, ddq,  $J=14.2, 6.8, 7.3$  Hz;  $\text{H}_b$ -14), 1.74 (1H, ddd,  $J=14.7, 11.7, 3.9$  Hz;  $\text{H}_\beta$ -11), 1.84 (1H, dd,  $J=14.7, 5.4$  Hz;  $\text{H}_\beta$ -8), 1.91 (1H, ddq,  $J=14.2, 2.9, 7.3$  Hz;  $\text{H}_a$ -14), 2.37 (1H, d,  $J=14.7$  Hz;  $\text{H}_x$ -8), 2.48 (1H, m;  $\text{H}_b$ -5), 2.52 (1H, ddd,  $J=14.7, 4.4, 2.4$  Hz;  $\text{H}_x$ -11), 2.56 (1H, d,  $J=1.5$  Hz; H-1), 2.80 (1H, m;  $\text{H}_a$ -5), 2.81 (3H, s; OMe), 3.01 (1H, ddd,  $J=9.8, 6.8, 2.9$  Hz; H-13), 3.37 (1H, ddd,  $J=3.9, 2.4, 2.0$  Hz; H-10), 3.42 (1H, dd,

$J=5.4, 2.0$  Hz; H-9), 3.96 (1H, ddd,  $J=11.7, 9.8, 4.4$  Hz; H-12), 4.17 (1H, dd,  $J=10.3, 3.4$  Hz; H-6), 5.45 (1H, dddd,  $J=16.1, 1.9, 1.5, 1.5$  Hz; H-3) and 6.32 (1H, ddd,  $J=16.1, 7.3, 7.3$  Hz; H-4);  $^{13}\text{C}$  NMR ( $\text{C}_6\text{D}_6$ ):  $\text{CH}_3$ ;  $\delta$  8.80 (C-15), 48.50 (OMe),  $\text{CH}_2$ ;  $\delta$  26.50 (C-14), 36.58 (C-5), 37.73 (C-11), 40.64 (C-8), CH;  $\delta$  47.19 (C-12), 56.02 (C-6), 76.76 (C-9), 77.96 (C-1), 79.24 (C-10), 80.62 (C-13), 111.72 (C-3), 143.20 (C-4) and C;  $\delta$  109.58 (C-7); EI–LR–MS  $m/z$  (rel. int.): 424 (0.4), 422 (0.8), 420 (0.4)  $[\text{M}]^+$ , 393 (1.4), 391 (2.5), 389 (1.3)  $[\text{M}-\text{OMe}]^+$ , 343 (3.3), 341 (3.5)  $[\text{M}-\text{Br}]^+$ , 311 (25.5), 309 (25.0), 265 (33.0), 263 (33.2), 229 (10.7), 191 (16.6), 189 (17.8), 183 (36.5), 149 (35.0), 109 (100), 105 (86.9), 81 (66.3), 69 (54.3), 57 (76.6) and 41 (87.5); EI–HR–MS  $m/z$ : 423.9890. Calcd for  $\text{C}_{16}\text{H}_{22}^{79}\text{Br}^{81}\text{BrO}_3$ , 423.9894  $[\text{M}]$ .

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## References

- Abe, T., & Masuda, M. (1998). *European Journal of Phycology*, 33, 17.
- Howard, B. M., & Fenical, W. (1975). *Tetrahedron Letters*, 0, 1687.
- Kikuchi, H., Suzuki, T., Kurosawa, E., & Suzuki, M. (1991). *Bulletin of the Chemical Society of Japan*, 64, 1763.
- Ojika, M., Yoshida, M., Okumura, M., Ikeda, S., & Yamada, K. (1990). *Journal of Natural Products*, 53, 1619.
- Suzuki, M., Furusaki, A., & Kurosawa, E. (1979). *Tetrahedron*, 35, 823.
- Suzuki, M., & Kurosawa, E. (1987). *Bulletin of the Chemical Society of Japan*, 60, 3791.
- Suzuki, M., Matsuo, Y., & Masuda, M. (1993). *Tetrahedron*, 49, 2033.
- Suzuki, M., Segawa, M., Kikuchi, H., Suzuki, T., & Kurosawa, E. (1985). *Phytochemistry*, 24, 2011.
- Takahashi, Y., Suzuki, M., Abe, T., & Masuda, M. (1998). *Phytochemistry*, 48, 987.
- Tori, K., Komeno, T., & Nakagawa, T. (1964). *The Journal of Organic Chemistry*, 29, 1136.