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Phenolic compounds from Anomianthus dulcis

Andrea Sinz^a, Rudolf Matusch^{a,*}, Floris van Elsäcker, Thawatchai Santisuk^b, Suttiporn Chaichana^c, Vichai Reutrakul^c

^aInstitut für Pharmazeutische Chemie, Philipps-Universität Marburg, Marbacher Weg 6, 35032 Marburg, Germany ^bThe Herbarium, Royal Forest Department, Bangkok 10900, Thailand ^cDepartment of Chemistry, Faculty of Science, Mahidol University, Rama 6 Road, Bangkok 10400, Thailand

Received 6 August 1998

Abstract

p-Coumaroyl-β-phenethylamine (1) and 2′,3′-dihydroxy-4′,6′-dimethoxydihydrochalcone (2) were isolated by preparative HPLC from a methanolic extract of *Anomianthus dulcis* leaves. This is the first time that these compounds were isolated from a natural source. In addition, the known compounds chrysin (3), pinocembrin (4), 5,7-dimethoxy-8-hydroxyflavanone (5) and 2′,3′-dihydroxy-4′,6′-dimethoxychalcone (6) were isolated for the first time from the genus *Anomianthus*. The isolated compounds were identified by ¹H and ¹³C NMR, MS, UV and IR. © 1999 Elsevier Science Ltd. All rights reserved.

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1. Introduction

The Annonaceae are a large family of aromatic trees, shrubs or climbers (ca. 120 genera and more than 2000 species), which are widely distributed in tropical and subtropical regions (Heywood, 1978).

The climber Anomianthus dulcis (Dun.) Sinclair is a member of the Annonaceae family that grows in many parts of South East Asia. Recently, we reported the isolation of alkaloids from the leaves and stem (Sinz et al., 1998b), as well as the isolation of the Annonaceous acetogenin squamocin from the stem of Anomianthus dulcis (Sinz, Matusch, Santisuk, Chaichana, & Reutrakul, 1998a). Our present phytochemical investigations of this species reveal the presence of several phenolic constituents in the leaves of Anomianthus dulcis.

2. Results and discussion

A crude methanolic extract from the leaves of *Anomianthus dulcis* was evaporated and dissolved in water. The suspension was acidified to pH 1.5 and successively extracted with *n*-hexane and methylene chloride. Alkalization of the water phase to pH 10.0 and extraction

with methylene chloride lead to the isolation of the alkaloids as recently reported (Sinz et al., 1998b). From the fraction extracted with methylene chloride at pH 1.5 compounds **1–6** were isolated by preparative HPLC. To avoid the isolation of artifacts, analytical HPLC was performed before and after acid and base treatment. No difference in the chromatograms was observed.

Compound 1 gave a UV spectrum with strong absorption maxima at $\lambda = 300$, 277, 223 and 217 nm. The EImass spectrum of 1 exhibited a [M]⁺ peak at m/z = 267 and two characteristic fragment ions at m/z = 148 and 120. High resolution mass spectrometry revealed the [M]⁺ peak at m/z = 267.1259 suggesting that compound 1 possesses the molecular formula $C_{17}H_{17}NO_2$. The fragment ion at m/z = 120.0600 was assigned to the molecular formula C_8H_8O which constitutes a fragment formed by the cleavage of the C-8′ carbonyl bond, while the fragment ion at m/z = 148.0768 was assigned to the molecular formula $C_9H_{10}NO$.

The ¹H NMR spectrum of **1** showed two signals at δ 7.02 (d, 2H, J=8.7 Hz) and 6.69 (d, 2H, J=8.7 Hz) that are due to the C-2′ and C-6′ and the C-3′ and C-5′ protons, respectively. The C-3 and C-5 protons appeared at δ 7.40 (2H, dd, J=7.4 Hz, J=6.8 Hz), while the C-2 and C-6 protons exhibited a dd at δ 7.55 (2H, J=7.4 Hz, J=1.7 Hz). A tt at δ 7.38 (1H, J=6.8 Hz, J=1.7 Hz) was assigned to the C-4 proton. The downfield doublet at δ 7.41 (J=15.7 Hz) was assigned to the olefinic proton

^{*}Corresponding author. Tel.: +49-6421-285-811; fax: +49-6421-287-052; e-mail: sinz@mailer.uni-marburg.de.

HO
$$\frac{2^{2}}{5^{2}}$$
 $\frac{7}{8^{2}}$ $\frac{8}{7}$ $\frac{4}{1}$ $\frac{4}{2}$ $\frac{4}{3}$

2

at C-7' which showed *trans*-coupling with the olefinic proton at C-8' that appeared as doublet at δ 6.62 (J=15.7 Hz). The C-7 protons showed a triplet at δ 2.67 (J=7.1 Hz). At room temperature, the presence of two amide rotamers in 1 was confirmed by the C-8 protons which exhibited two triplets (J=7.1 Hz) at δ 3.36 for the Z rotamer and at δ 3.34 for the E rotamer, respectively. Integration of the signals revealed an approximate 1:1 distribution of the two rotamers.

The ¹³C NMR spectrum of **1** showed 13 carbon resonances (Table 1). The assignments of the carbon atoms were made by means of the DEPT pulse sequence, HMQC and gated decoupling experiments. Paprazine, a similar alkaloid from *Fumaria indica* (Atta-ur-Rahman, Bhatti, Akhtar, & Choudhary, 1992) differs from **1** only by the presence of an additional hydroxyl group at C-4.

Chrysin (3) (Shen, Chang, & Ho, 1993), pinocembrin (4) (Wagner & Chari, 1976) and 5,7-dimethoxy-8-hydroxyflavanone (5) (Bhaskar & Seshadri, 1974; Vieira, De Alvarenga, Gottlieb, & Gottlieb, 1980) were unambiguously identified by comparison with reported data.

Compound 6, obtained as orange crystals, revealed a strong UV absorption maximum at $\lambda = 342$ nm which is

characteristic of a chalcone structure. The spectral data were in close agreement with those of 2',3'-dihydroxy-4',6'-dimethoxychalcone (Bhaskar & Seshadri, 1974; Vieira et al., 1980; Ichino, Tanaka, & Ito, 1988).

Compound 2, obtained as colorless crystals, showed UV maxima at $\lambda = 350$, 291 and 242 nm. Treatment with AlCl₃ exhibited a bathochromic shift that was stable in acid solution. The ¹³C NMR spectrum of 2 Table 1 appeared similar to that of 6. The proton NMR spectrum of 2 showed a characteristic downfield signal for the C-2'-OH group at δ 13.8 which is caused by a hydrogen bridge formation with the carbonyl oxygen. In addition, the ¹H NMR spectrum of 2 showed signals corresponding to the protons of the unsubstituted B ring and a singlet appeared at δ 6.06 for the C-5′-proton. However, instead of signals corresponding to the trans-double bond present in 6, two triplets at δ 3.34 and δ 3.00 (J=8.0 Hz) were observed suggesting the presence of a saturated bond between C- α and C- β . This assignment was confirmed by a DEPT NMR spectrum. Therefore, the structure of compound 2 was determined to be 2',3'-dihydroxy-4',6'dimethoxydihydrochalcone. The structure was verified by the IR, UV and MS data which correspond with data

Table 1 ¹³C NMR spectral data of compounds 1 and 2, 100 MHz

Carbon	1 (DMSO- <i>d</i> ₆)	2 (CDCl ₃)
1	135.5	135.5
2	128.0	128.5
3	129.4	129.0
4	129.9	130.3
5	129.4	129.0
6	128.0	128.5
7	34.9	
8	41.2	
C-α		45.9
С-β		30.6
1'	130.1	106.4
2'	130.0	152.7
3'	115.6	128.5
4'	156.2	155.8
5'	115.6	87.7
6'	130.0	152.1
7′	138.9	
8'	122.9	
C=O	165.3	193.3
OCH ₃		56.3, 56.2

reported by Bhaskar and Seshadri who obtained 2 from 6 by catalytic hydrogenation (Bhaskar & Seshadri, 1974).

3. Experimental

3.1. General

M.p.'s: uncorrected; NMR spectra: Jeol GX 400 Delta spectrometer using TMS as internal standard; UV spectra: UV-VIS photometer 2101 PC (Shimadzu); IR spectra: Nicolet 510 P FT-IR spectrometer; EI-mass spectra: 70 eV; prep. HPLC: Waters prep LC 4000 system with 990 Diode Array Detector.

3.2. Plant material

Leaves were collected in July 1995 at Mae Rim-Samoeng road (Thailand). The identity of the plant material was confirmed by Dr. P. Kessler, Rijk-sherbarium Leiden (L), where a voucher specimen has been deposited. Voucher specimen: Coll. A. Nuntasan s.n., Thailand, Mae Rim-Samoeng Road, Chiangmai, 7 July 1995 (L 997.104 782).

3.3. Extraction and isolation

The plant material was dried (4 kg d.w.), crushed and percolated at 30°C with MeOH. After evaporation of MeOH, the residue (687 g) was dissolved in $\rm H_2O$, acidified with 1% HCl (pH 1.5) and filtered. The solution was extracted with *n*-hexane (3 × 500 ml) and $\rm CH_2Cl_2$ (3 × 500

ml). The CH₂Cl₂ phase, evaporated to dryness, afforded 9.85 g of a gummy material. This was fractionated by prep. HPLC (250×25 mm, RPselect B, 15 µm (Merck), 32 ml min⁻¹, MeOH–H₂O (6:4)) into 8 frs (fr. 1–8). Fr. 2 consisted of 5,7-dimethoxy-8-hydroxyflavanone (**5**)(1.169 g), fr. 3 was separated further by prep. HPLC (250×25 mm, RPselect B, 15 µm (Merck), 32 ml min⁻¹, MeOH–H₂O (1:1)) to yield 78 mg of compound **1**, fr. 5 consisted of 297 mg pinocembrin (**4**), fr. 6 was separated by prep. HPLC (250×25 mm, DIOL, 7 µm (Merck), 40 ml min⁻¹ *tert*-butylmethylether–hexane (4:6), to give 149 mg of 2',3'-dihydroxy-4',6'-dimethoxychalcone (**6**) and 140 mg of 2',3'-dihydroxy-4',6'-dimethoxydihydrochalcone (**2**). Fr. 7 consisted of 205 mg chrysin (**3**).

3.4. p-Coumaroyl- β -phenethylamine (1)

Colorless crystals (78 mg), m.p.: 146-148°C, UV $\lambda^{\text{MeOH}}_{\text{max}}$ nm (log ϵ): 300 sh (4.12), 277 (4.51), 223 (4.47), 217 (4.49). IR (KBr, cm⁻¹): 3278, 1668, 1626, 1535. ¹H NMR (400 MHz, DMSO- d_6): δ 9.12 (1H, s, OH), 7.55 (2H, dd, J=7.4, 1.7, H-2, H-6), 7.41 (1H, d, J=15.7, H-7'), 7.40 (2H, dd, J=7.4, 6.8, H-3, H-5), 7.38 (1H, tt, J=6.8, 1.7, H-4), 7.02 (2H, d, J=8.7, H-2', H-6'), 6.69 (2H, d, J=8.7, H-3', H-5'), 6.62 (1H, d, J=15.7, H-8'),5.74 (1H, s, NH), 3.36 (Z rotamer, t, J=7.1, H-8), 3.34 (E rotamer, t, J=7.1, H-8), 2.67 (1H, t, J=7.1, H-7). 13 C NMR (100 MHz, DMSO- d_6): Table 1. EI-MS (70 eV): m/z (rel. int.) 267 [M]⁺ (4), 148 (58), 132 (10), 131 (97), 121 (10), 120 (100), 107 (18), 103 (47), 77 (32). HR-MS: found: 267.1259; calculated for $C_{17}H_{17}NO_2$: 267.1259. Fragment C₈H₈O: found: 120.0600; calculated: 120.0575. Fragment C₉H₁₀NO: found: 148.0768; calculated: 148.0762.

3.5. 2',3'-Dihydroxy-4',6'-dimethoxydihydrochalcone (2)

Colorless crystals (140 mg), m.p.: 134–136°C, UV $\lambda^{\text{MeOH}}_{\text{max}}$ nm (log ε): 350 (3.59), 291 (4.27), 242 (4.07); +AlCl₃: 414, 318, 252; +AlCl₃/HCl: 401, 367, 316, 246; +NaOMe: 372, 298 sh, 262; +NaOAc: 355, 298, 247; +NaOAc/H₃BO₃: 351, 313 sh, 293, 245. IR (KBr, cm⁻¹): 1635 (C=O). ¹H NMR (400 MHz, CDCl₃): δ 13.84 (1H, brs, OH-2'), 7.30–7.22 (5H, m, H-2, H-3, H-4, H-5, H-6), 6.06 (1H, s, H-5'), 4.04 (3H, s, OCH₃), 4.00 (3H, s, OCH₃), 3.34 (2H, t, J=8.0, H- α), 3.00 (2H, t, J=8.0, H- β). ¹³C NMR (100 MHz, CDCl₃): Table 1. EI-MS (70 eV): m/z (rel. int.) 303 [M+H]⁺ (6), 302 [M]⁺ (32), 198 (10), 197 (100), 182 (17), 170 (27), 167 (4), 153 (4), 139 (4), 104 (5), 91 (24), 77 (6), 69 (10). HR-MS: found: 302.1142, calculated for C₁₇H₁₈O₅: 302.1154.

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