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# Crotonadiol, a labdane diterpenoid from the stem bark of Croton zambesicus

Bonaventure T. Ngadjui<sup>a,\*</sup>, Gabriel G. Folefoc<sup>a</sup>, Felix Keumedjio<sup>a</sup>, Etienne Dongo<sup>a</sup>, Beiban L. Sondengam<sup>a</sup>, Joseph D. Connolly<sup>b</sup>

<sup>a</sup>Department of Organic Chemistry, University of Yaoundé-1, P.O. Box 812, Yaoundé, Cameroon <sup>b</sup>Department of Chemistry, University of Glasgow, Glasgow G12 800, UK

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#### Abstract

A new labdane diterpenoid, crotonadiol, was isolated from an extract of the stem bark of *Croton zambesicus* together with the known clerodane crotocorylifuran and two trachylobanes:  $7\beta$ -acetoxytrachyloban-18-oic acid and trachyloban- $7\beta$ ,18-diol. Lupeol,  $\beta$ -sitosterol and its 3- $\beta$ -glucopyranosyl derivative were also isolated. The structure of crotonadiol was determined as 8(17),13-labdadiene- $6\alpha$ ,15-diol, by spectral analysis. © 1999 Elsevier Science Ltd. All rights reserved.

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#### 1. Introduction

Croton zambesicus Muell. Arg. (Syn. C. amabilis Muell. Arg.) is a shrub or small tree reaching 10 m in height and widespread in tropical west and Central Africa (Hutchinson & Dalziel, 1958). The stems are used for house posts in parts of West Africa. The roots are used as an aperient. The leaf decoction is used as a wash for fevers in Sierra Leone and Nigeria and internally for dysentery and convulsions (Irvine, 1961). Alkaloids (Shamma, Shine, & Dudock, 1967; Stuart & Byfield, 1971) and diterpenoids (Burke, Chan, Keith, Blount, & Manchand, 1981; Craveiro, Afianio, & Edilberto, 1982) have mainly been recorded from this genus. No previous phytochemical and pharmacological studies have been reported on C. zambesicus. As part of our continuing studies on Cameroonian plants of medicinal interest, we have examined the extract of the stem bark of C. zambesicus. This paper reports the isolation and structural elucidation of a new labdane diterpenoid, crotonadiol (1), together with the previously known clerodane, crotocorylifuran (2) (Burke et al., 1976; Burke et al., 1981; Tchissambou, Chiaroni, Riche, & Khuong-Huu, 1990), the trachylobanes,  $7\beta$ -acetoxytrachyloban-18-oic acid (3) (Hasan, Healey, & Waterman, 1982) and trachyloban- $7\beta$ , 18-diol (4) (Gonzalez, Breton, Fraga, & Luis, 1971;

R<sup>1</sup> R<sup>2</sup> COOH Ac

4 CH₂OH H

<sup>\*</sup>Corresponding author. Fax: +23-53-88; e-mail: bngadjui@uycdc.uninet.cm.

Gonzalez, Fraga, Hernandez, & Luis, 1973), the pentacyclic triterpenoid, lupeol (Ayafor, Ngadjui, Sondengam, & Tsamo, 1984) and sitosterol and its glucoside.

#### 2. Results and discussion

Crotonadiol (1) was isolated from a chloroform-methanol (1:1) extract of dried powdered stem bark, as described in Section 3. Its molecular formula was determined as C<sub>20</sub>H<sub>34</sub>O<sub>2</sub> from NMR and EI-mass spectroscopy measurements. Crotonadiol showed IR absorption bands for hydroxy groups at  $v_{\text{max}}$  3500–3400 cm<sup>-1</sup> and an exomethylene group at  $v_{\text{max}}$  1630, 890 cm<sup>-1</sup>. The NMR spectra of 1 showed typical characteristics of a labdan-8(17), 13-diene skeleton [ $\delta_{\rm H}$  4.61 and 4.90 (1H each, br s, 2H-17), 5.38 (1H, t, J = 6.9 Hz, H-14),  $\delta_C$  108.2 (t, C-17), 145.5 (s, C-8), 123.2 (d, C-14), 140.3(s, C-13)]. The <sup>1</sup>H NMR spectrum of crotonadiol also indicated the presence of four tertiary methyl groups at 0.72, 1.02, 1.19 and 1.72 (3H each, s). The *E*-geometry of the side chain was established by the chemical shift of the methyl carbon at C-13 ( $\delta_c$  16.3). The chemical shift ( $\delta_{Ca}$  4.15) of the methylene group at C-15 which was coupled to the proton at C-14 ( $\delta$  5.38, t, J=6.9 Hz) indicated that this methylene group was hydroxylated. Furthermore the <sup>1</sup>H NMR spectrum of crotonadiol displayed a broad triplet doublet signal  $[\delta 3.83 (1H, J=4.9, 10.7 Hz)]$  due to an oxymethine which was coupled to both a proton at 1.14 (d, J=10.7Hz) and methylene protons at 2.04 (1H, br t, J = 12.0 Hz) and 2.68 (1H, dd, J = 4.9, 12.1 Hz). Therefore, this second hydroxy group should be attached to C-6 or C-11. However, the latter alternative position for the hydroxy group was discarded from HMBC and NOESY since the chemical shift ( $\delta_{Ca}$  1.7) of the proton at C-9 was observed at a similar position to those in a similar compound (Iwagawa, Yaguchi, Hase, Okubo, & Kim, 1992). The trans-relationship between the protons at C-5 and C-6 was deduced from the diaxial coupling constant (J=10.7)Hz). Thus crotonadiol was identified as labda-8(17), 13Edien-6α, 15-diol. This structure was confirmed by both the <sup>13</sup>C NMR spectrum and the EI mass spectrum. The EI mass spectrum showed fragments at m/z 288, 273 and 255 corresponding to [M-H<sub>2</sub>O]<sup>+</sup>, [M-H<sub>2</sub>O–Me]<sup>+</sup> [M-2H<sub>2</sub>O-Me]<sup>+</sup>, respectively. The <sup>13</sup>C NMR (Tables 1, 2 and 3) was fully assigned using DEPT spectra and by comparison of measured values with those reported for similar compound (Iwagawa et al., 1992). Crotonadiol (1) is reported here for the first time while its 6.15-di-Oglucopyranoside (gomojoside H) has been isolated from Viburnum suspensum (Caprifoliaceae) (Iwagawa et al., 1992).

### 3. Experimental

## 3.1. General

M.p.'s uncorr; UV-visible: MeOH solution; IR: KBr disk or CHCl<sub>3</sub> solution; EIMS direct inlet 70 eV. <sup>1</sup>H

Table 1 <sup>13</sup>C NMR spectral data of **1** and **1a** in CDCl<sub>3</sub> at 90 MHz

C	1	1a	
1	39.3 (t)	39.1 (t)	
2	19.1 (t)	19.0 (t)	
3	43.7 (t)	44.2 (t)	
4	33.9 (s)	33.5 (s)	
5	60.5 (d)	57.5 (d)	
6	71.7 (d)	73.2 (d)	
7	49.2 (t)	43.5 (t)	
8	145.5 (s)	144.2 (s)	
9	55.5 (d)	55.2 (d)	
10	39.4 (s)	39.6 (s)	
11	22.1 (t)	21.9 (t)	
12	38.4 (t)	38.3 (t)	
13	140.3 (s)	142.7 (s)	
14	123.2 (d)	118.2 (d)	
15	59.4 (t)	61.3 (t)	
16	16.3 (q)	16.5 (q)	
17	108.2 (t)	109.3 (t)	
18	36.6 (q)	36.1 (q)	
19	22.4 (q)	22.4 (q)	
20	16.1 (q)	16.0 (q)	
COMe		21.1 (q)	
COMe		21.8 (q)	
<i>CO</i> Me		170.1 (s)	
<i>CO</i> Me		171.1(s)	

NMR (360 MHz) and <sup>13</sup>C NMR (90 MHz) recorded at room temp., residual solvent peaks as internal reference. HMBC, HMQC and NOESY experiments were performed with gradient enhancements.

#### 3.2. Plant material

The stem bark of *Croton zambesicus* was collected at Eloundem mountain, Yaoundé, in the Central Province of Cameroon. A voucher specimen (No. 8204/SRFCAM) for the collection is deposited at the National Herbarium, Yaounde, Cameroon.

#### 3.3. Extraction, isolation and characterization

The air-dried powdered plant material (2 kg) was macerated in  $CH_2Cl_2$ –MeOH (1:1). Removal of the solvent under red. pres. yielded a dark brown extract (60 g). Part (50 g) of this extract was subjected to repeated chromatographic fractionations on a silica gel column eluted with hexane followed by a hexane–EtOAc gradient. Sitosterol (40 mg), lupeol (35 mg) and crotocorylifuran (2; 15 mg) were obtained from frs eluted by hexane–EtOAc 19:1, 9:1 and 17:3, respectively;  $7\beta$ -acetoxytrachyloban-18-oic acid (3; 20 mg), trachyloban- $7\beta$ , 18-diol (4; 15 mg), crotonadiol (1; 20 mg) together with sitosterol glucoside (150 mg), were obtained from frs eluted by hexane–EtOAc (1:1). Known compounds were identified by comparison (m.p.,  $^1$ H,  $^{13}$ C NMR) with

Table 2 <sup>1</sup> *J* (from HMQC) <sup>2</sup> *J* and <sup>3</sup> *J* gradient HMBC correlations for compound 1

Proton	Position	$^{1}J$ correlated carbon	$^{2}J$ , $^{3}J$ correlated carbons
5.38	14	123.2	16.3 (C-16)
4.90	17	108.2	49.2 (C-7); 55.5 (C-9)
4.61	17	108.2	49.2 (C-7); 55.5 (C-9)
4.15	15	59.4	123.2 (C-14); 140.3 (C-13)
3.83	6	71.7	
2.68	7	49.2	55.5 (C-9); 60.5 (C-5); 71.7 (C-6); 108.2 (c-17); 145.5 (C-8)
2.04	7	49.2	71.7 (C-6); 108.2 (C-17); 145.5 (C-8)
1.83	1	39.3	16.1 (C-20)
1.72	Me-16	16.3	38.4 (C-12); 123.2 (C-14); 140.3 (C-13)
1.64	9	55.5	16.3 (C-16); 39.4 (C-10); 22.1 (C-11); 38.4 (C-12)
1.53	2	19.1	43.7 (C-3); 33.9 (C-4)
1.48	2	19.1	33.9 (C-4)
1.19	Me-18	36.6	22.4 (C-19); 33.9 (C4); 43.7 (C-3); 60.5 (C-5)
1.14	5	60.5	22.4 (C-19); 33.9 (C4); 36.6 (C-18)
1.02	Me-19	22.4	33.9 (C4); 43.7 (C-3); 60.5 (C-5); 36.6 (C-18)
0.72	Me-20	16.1	39.3 (C-1); 39.4 (C-10); 55.5 (C-9); 60.5 (C-5)

Table 3 Volume integrated NOESY correlations observed for H-14, 2H-17, 2H-15, H-6 $\beta$ , H-7eq, H-7ax, 2H-12, H-5, Me-16, Me-18, Me-19 and Me-20 in crotonadiol (1)

From	to
H-5	H-9 (3.9) H-7a (1.5)
$H-6\beta$	H-7eq (1.6); Me-20 (1.5); Me-19 (2.2)
H-7ax	H-7eq (5.4); H-5 (0.7) H-9 (0.7)
H-7eq	H-7ax (6.2); H-6 $\beta$ (1.7), H-17a (2.1)
H-12a	H-12b (4.0)
H-12b	H-12a (3.7); H-9 (1.4); H-14 (0.9)
H-14	2H-15 (2.7)
H-15a	H-14 (2.0); Me-16 (2.0)
H-15b	H-14 (2.0); Me-16 (2.0)
H-17a	H-7eq (4.0); H-17b (13.0)
H-17b	H-17a (4.0); H-12 (1.3); H-11 (3.3)
Me-16	H-1 (3.3)
Me-18	Me-19 (3.1); H-3ax (1.2)
Me-19	Me-20 (7.1); H-6β (5.2); Me-18 (2.8); H-3eq (1.3); H-2ax (1-4)
Me-20	Me-19 (6.8); H-2ax (1.8); H-1eq (0.8)

The volme of integration (%) is given in parentheses.

authentic samples or published information. Sitosterol glucoside (60 mg), which was insoluble in the usual organic solvents, was acetylated using boiling  $Ac_2O$  (6 ml) for 2 h. The reaction mixture was evaporated in a petri dish to leave a residue which was chromatographed by CC (hexane–EtOAc, 3:2) to give white platelets of tetraacetate of sitosterol-3- $\beta$ -D-glucopyranoside (55 mg, 80%) m.p. 166°C.

# 3.4. *Crotonadiol* (1)

Colourless oil;  $[\alpha]_D^{25}$  – 28°C (CHCl<sub>3</sub>, c 0.12), UV  $\lambda_{\max}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 210 (4.10); IR  $\nu_{\max}^{\text{CHCl}_3}$  cm $^{-1}$ : 3500–3400

(OH), 1630, 1530, 1420, 1200, 1100, 930, 890, 860, EIMS m/z (rel.int.): 306 ([M]<sup>+</sup>, 10), 288 ([M-H<sub>2</sub>O]<sup>+</sup>, 25), 273 ([M-H<sub>2</sub>O-Me]<sup>+</sup>, 40), 255 ([M-Me-2H<sub>2</sub>O]<sup>+</sup>, 35), 243 (15), 203 (18), 190 (20), 187 (17), 153 (35), 109 (65), 69 (100); <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>): δ 0.72 (3H, s, 3H-20), 1.02 (3H, s, 3H-19), 1.14 (1H, d, J=10.7 Hz, H-5), 1.19 (3H, s, 3H-18), 1.48 (1H, m, H-2a), 1.53 (1H, m, H-2b), 1.64 (1H, br d, J=11.8, H-9), 1.72 (3H, s, 3H-16), 1.83 (1H, m, H-1), 1.92 (1H, br dt, J=8.0, 14.0 Hz, H-12a), 2.04 (1H, br t, J=12.0 Hz, H<sub>α</sub>-7), 2.17 (1H, m, H-12b), 2.68 (1H, dd, J=4.9, 12.1 Hz, H<sub>β</sub>-7), 3.83 (1H, br dt, J=4.9, 10.7 Hz, H-6), 4.15 (2H, d, J=6.9 Hz, 2H-15), 4.61 (1H, br s, H-17a), 4.90 (1H, br s, H-17b) and 5.38 (1H, br t, J=6.9 Hz, H-14). <sup>13</sup>C NMR (90 MHz, CDCl<sub>3</sub>): Table 1.

## 3.5. Acetylation of crotonadiol (1)

Compound 1 (12 mg) in CH<sub>2</sub>Cl<sub>2</sub> (2 ml) was treated with Ac<sub>2</sub>O (2 ml) in the presence of a catalytic amount of DMAP for 2 h. The reaction was monitored by TLC. The reaction mixture was mixed with celite (3 g) and evaporated into dryness in vacuo and the powder obtained was introduced on to a silica gel column and eluted with hexane-EtOAc (7:3) to give the diacetate 1a (10 mg, 70%): colourless oil; ÚV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\varepsilon$ ): 211 (3.80); IR  $v_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1700 (C=0), 1640 (C=C), 1580, 1500, 1450, 1400, 1100, 880. EIMS *m/z* (rel. int.): 390 ([M]<sup>+</sup>, 15). <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>):  $\delta$  0.74 (3H, s, 3H-20), 0.87 (3H, s, 3H-19) 1.01 (3H, s, 3H-18), 1.39 br dt, J = 7.8, 13.7 Hz, H-12a); 2.03 (3H, s,  $CH_3CO$ ), 2.05 (3.14, s, CH<sub>3</sub>CO), 2.20 (1H, m, H-12b), 2.69 (1H, dd, J = 5.0, 12.1 Hz, H<sub>8</sub>-7), 4.58 (2H, br d, J = 6.7 Hz 2H-15), 4.62 (1H, br s, H-17a), 4.93 (1H, br s, H-17b), 5.03 (1H, dt, J=5.0, 10.2 Hz H-6), 5.30 (1H, br t, J=6.7 Hz H-14), <sup>13</sup>C NMR (90 MHz, CDCl<sub>3</sub>): Table 1.

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## References

- Hutchinson, L. J. & Dalziel J. M. (1958). Revised by R.W.J. Keay (Ed.), Flora of West Tropical Africa (2nd ed., Vol. 1, Part 2, p. 393). London: White Frias Press.
- Irvine, F.R. (1961). Woody plants of Ghana (p. 221). London: Oxford University Press.
- Stuart, L., & Byfield, D. Y. (1971). Phytochemistry, 10, 460.
- Shamma, M., Shine, R. J., & Dudock, B. S. (1967). *Tetrahedron*, 23, 2887.

- Craveiro, A., Afianio, S., & Edilberto, R. (1982). *Phytochemistry*, 21, 2571.
- Burke, A. B., Chan, W. R., Keith, P. D., Blount, J. F., & Manchand, P. S. (1981). J. Chem. Soc. Perkin Trans., 1(10), 2666.
- Tchissambou, L., Chiaroni, A., Riche, L., & Khuong-Huu, F. (1990). Tetrahedron, 46, 5199.
- Burke, B. A., Chan, W. R., Prince, E. C., Manchand, P. S., Eickman, N., & Clardy, J. (1976). *Tetrahedron*, 1881, 32.
- Hasan, C. M., Healey, T. M., & Waterman, P. G. (1982). *Phytochemistry*, 21, 177.
- Gonzalez, A. G., Fraga, B. M., Hernandez, M. G., & Luis, J. G. (1973). *Phytochemistry*, 12, 1113.
- Gonzalez, A. G., Breton, J. L., Fraga, B. M., Luis, J. G. (1971). Tetrahedron Letters, 3097.
- Ayafor, J. F., Ngadjui, B. T., Sondengam, B. L., & Tsamo, E. (1984). *Planta Medica*, 50, 205.
- Iwagawa, T., Yaguchi, S., Hase, T., Okubo, T., & Kim, M. (1992). Phytochemistry, 31, 1311.