Phytochemistry 52 (1999) 357-360

Isobutylamides from the fruit of Zanthoxylum integrifoliolum

Ih-Sheng Chen*, Tzu-Li Chen, Wei-Yu Lin, Ian-Lih Tsai, Yu-Chang Chen

Graduate institute of Natural Products, Kaohsiung Medical College, Kaohsiung, Taiwan

Received 21 July 1998; received in revised form 29 January 1999; accepted 29 January 1999

Abstract

Investigation of the fruit of *Zanthoxylum integrifoliolum* led to the isolation of three new isobutylamides, lanyuamide I–III and six known isobutylamides, tetrahydrobungeanool, γ -sanshoöl, hydroxy γ -sanshoöl, mixture of (2E,4E,8Z,11E)- and (2E,4E,8Z,11Z)-2'-hydroxy-N-isobutyl-2,4,8,11-tetradecatetraenamide and hazaleamide which was mixed with lanyuamide III. These amides were all with a (2E,4E)-dienamide moiety and their structures were elucidated on the basis of spectral analyses. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Zanthoxylum integrifoliolum; Rutaceae; Fruit; Alkaloids; Amides; Isobutylamides; Lanyuamide

1. Introduction

Zanthoxylum integrifoliolum (Merr.) (Rutaceae) is a large evergreen tree that grows only in the northern Philippines and on Lanyu Island in Taiwan province (Chang & Hartley, 1993). Its bark is used by Ya-Mei aborigines as a folk medicine to treat snake-bites. The fruit possesses a pungent taste, but has not been utilized as a substitute for Pericarpi Zanthoxyli. The chemical constituents of the bark (Chua, Maglaya, & Santos, 1970; Jen, Tsai, Horng, & Chen. 1993) and root wood (Ishii, Chen, Akaike, Ishikawa, & Lu, 1982) of this plant have been studied. In a preliminary study, three indolopyridoquinazoline alkaloids with anti-platelet aggregation activity were obtained from a small amount of the fruit (Sheen, Tsai, Teng, Ko, & Chen, 1996). Further examination of the chemical constituents and anti-platelet aggregation principles obtained from large amounts of fruit has resulted in the isolation of nine isobutylamides, including three new compounds: lanyuamide I-III (1-3) and six known amides, tetrahydrobungeanool (4) (Xiong, Shi, Yamamoto, & Mizuno, 1997), γ-sanshoöl

2. Results and discussion

Nine amides (1–9) were obtained as colorless oils by chromatography on a silica gel column. The UV spectra of each amide showed maximal absorption near 259 nm, indicating the presence of a conjugated system related to sorbic acid isobutylamides (Eisner, Elvidge, & Linstead, 1953; Crombie, 1955). The IR spectra of each amide showed characteristic bands for amidoamino and amidocarbonyl groups, indicating the presence of a *trans-2-trans-4*-dienamide skeleton (Crombie, 1955; Dhar & Atal, 1967).

Lanyuamide I (1) showed an additional ketone absorption at 1700 cm⁻¹ in its IR spectrum. The ¹H NMR spectrum showed an *N*-isobutyl group [δ 0.92 (6H, d, J = 6.8 Hz, H-3', 4'), 1.79 (1H, m, H-2'), 3.16

0031-9422/99/\$ - see front matter \odot 1999 Elsevier Science Ltd. All rights reserved. PII: S0031-9422(99)00175-2

^{(5) (}Yasuda, Takeya, & Itokawa, 1981, 1982), hydroxy γ-sanshoöl (6) (Yasuda et al., 1981, 1982), mixture of (2*E*,4*E*,8*Z*,11*E*)- and (2*E*,4*E*,8*Z*,11*Z*)-2'-hydroxy-*N*-isobutyl-2,4,8,11-tetradecatetraenamide (7 and 8) (Mizutani et al., 1988) and hazaleamide (9) (Shibuya, Takeda, Zhang, Tong, & Kitagawa, 1992) which was mixed with 3. Here, we describe, for the first time, the structural elucidation of these new amides.

^{*} Corresponding author.

(2H, t, J = 6.8 Hz, H1'), 5.47 (1H, br s, NH)], fourprotons on a trans-2-trans-4-dienamide [δ 5.75 (1H, d, J = 14.9 Hz, H-2, 6.02 (1H, dt, J = 15.1, 6.9 Hz, H-5), 6.16 (1H, dd, J = 15.1, 10.6 Hz, H-4), 7.16 (1H, dd, J = 14.9, 10.6 Hz, H-3)], two methylenes [δ 2.42 (2H, q, J = 6.9 Hz, H-6) and 2.53 (2H, t, J = 6.9 Hz,H-7)] and an *n*-hexyl group [δ 0.88 (3H, t, J = 6.8 Hz, H-14), 1.25 (6H, br s, H-11-13), 1.56 (2H, quint, J = 7.1 Hz, H-10), 2.39 (2H, t, J = 7.1 Hz, H-9)] connected to a keto group. The molecular formula of 1 was established as $C_{18}H_{31}NO_2$ by El ([M]⁺, m/z 293) and HR-mass spectrometry and the position of the keto group at C-8 was supported by prominent mass fragments at m/z 113 $[C_7H_{13}O]^+$ $[C_{11}H_{18}ON]^+$. According to the above data, the structure of 1 was predicted to be a (2E,4E)-8-keto-N-isobutyl-2,4-tetradecadienamide, which was

8 R= OH, 9 R= H

Table 1 ¹³C NMR spectral data of compounds 1–9 (100 MHz, CDCl₃)

Carbon	1	2	3	4	5	6	7	8	9
1	166.1	166.2	166.3	167.5	166.3	167.3	167.4	167.4	166.3
2	122.5	122.1	122.1	121.1	122.3	121.6	121.5	121.5	122.1
3	140.6	141.0	141.1	142.0	140.8	141.7	141.8	141.8	141.1
4	129.1	128.7	128.6	128.1	128.7	128.7	128.5	128.6	128.6
5	140.7	141.9	142.0	143.8	141.6	142.3	142.7	142.6	142.0
6	23.8	32.8	32.9	32.9	32.9	33.0	32.9	32.9	32.9
7	41.5 ^a	26.4	26.4	28.8	27.0	27.0	26.4	26.5	26.5
8	209.9	128.8	127.0	29.2	129.8	129.8	127.1	127.1	127.0
9	43.0^{a}	129.5	128.8	29.3	129.4	129.5	128.7	129.1 ^a	128.8 ^a
10	28.9	21.7	30.4	29.4	125.3	125.3	30.4	25.6	25.6
11	27.0	42.0	132.6	29.5	133.3	133.4	132.6	132.1	132.0
12	31.6	211.1	129.0	31.9	131.8	131.8	128.8	128.5 ^a	129.0^{a}
13	22.5	36.0	25.6	22.7	129.9	130.0	25.5	20.6	20.5
14	13.9	7.76	13.8	14.1	18.3	18.3	13.8	14.2	14.2
1'	46.9	46.8	46.9	50.5	46.9	50.5	50.5	50.5	46.9
2'	28.6	28.6	28.6	71.1	28.6	71.1	71.7	71.1	28.6
3′	20.1	20.1	20.1	27.3	20.1	27.3	27.3	27.3	20.1
4'	20.1	20.1	20.1	27.3	20.1	27.3	27.3	27.3	20.1

^a Assignments may be reversed in each column.

confirmed by COSY, HETCOR and ¹³C NMR spectral analyses (Table 1).

Lanyuamide II (2) showed an $[M+1]^+$ ion at m/z292 by FAB mass spectroscopy, which is 2 amu less than that of 1 and suggests one more disubstituted double bond in 2.The ¹H NMR spectrum of 2 indicated that the two allylic methylenes at δ 2.19 (4H, m, H-6 and H-7) were coupled with H-5 of a 1,3-dienamide and one of the olefinic protons (2H, m), which also coupled with the third allylic methylene at δ 2.30 (2H, q, J = 6.8 Hz, H-10). The additional double bond was assigned at C-8 and C-9 by a subsequent COSY experiment. The geometry of the C-8 and C-9 double bond was determined to be the Z-form by the chemical shift of allylic C-7 at δ 26.4 ppm and C-10 at δ 21.7 ppm (Xiong et al., 1997). Two methylene groups, [(δ 2.41, 2H, q, J = 7.2 Hz, H-13) and 2.44 (2H, t, J = 7.6 Hz)], neighboring a keto group, were found in the ¹H NMR spectrum. The former was only coupled with a primary methyl group at δ 1.05 (3H, t, J = 7.2 Hz, H-14) and the latter was coupled with H-10 at δ 2.30 (2H, q, J = 6.8 Hz, H-11). Thus, this keto group was unambiguously assigned at C-12, which appeared at δ 211.1 ppm in the ¹³C NMR spectrum and showed maximal absorption at 1720 cm⁻¹ in the IR spectrum. According to the above data, the structure of 2 was elucidated as (2E,4E,8Z)-12-keto-N-isobutyl-2,4,8-tetradecatrienamide, and fully confirmed by COSY, HETCOR, DEPT and 13C NMR spectral analyses (Table 1).

Lanyuamide III (3) was isolated as an oil, admixed with a structurally-similar amide in a 1:3 ratio. The mixture showed a molecular ion at m/z 275 and a molecular formula as $C_{18}H_{29}NO$ by El- and HR mass

spectrometry. All major signals in the ¹H NMR spectrum were completely identical with those of hazaleamide (9) (Shibuya et al., 1992). However, the chemical shifts of minor signals at δ 0.93 (6H, d, J = 6.8 Hz), 0.96 (3H, t, J = 7.4 Hz), 2.00 (2H, m), 2.74 (2H, t, J = 6.0 Hz) and 5.31 (2H, m) of 3 were all close to H-3' and H-4', H-14, H-13, H-10 and H-11, H-12 of 9. Thus, these protons in 3 could be assigned to the same location as those in 9. The other signals of 3 completely overlapped with the respective signals of 9 as measured by their integration. From the above analyses of the ¹H NMR spectrum, the major 9 and the minor 3 components of the mixture were deduced to be a pair of geometric isomers which differed only at one double bond. Thus, the presence of allylic carbons, C-10 at δ 30.4 and C-13 at δ 25.6 as minor signals in 3 substantially differed from C-10 at δ 25.6 and C-13 at δ 20.5 in **9**. This fact supports the geometry of C-11 and C-12 in 3 to be in the E-form (Xiong et al., 1997). From the above data, the structure of 3 was elucidated (2E,4E,8Z,11E)-N-isobutyl-tetradecatetraenamide, which was further confirmed by COSY, HETCOR, DEPT and ¹³C NMR spectral analyses (Table 1).

3. Experimental

M.p.'s are uncorr. 1 H (400 MHz) and 13 C (100 MHz) NMR were taken in CDCl₃. Chemical shifts were given in δ with TMS as int. standard. MS were measured using a direct inlet system. IR spectra were determined neat, and UV spectra were measured with EtOH as solvent. Silica gel (60–230 mesh, 230–400 mesh) (Merck) was used for CC and silica gel 60 F-254 for TLC.

3.1. Plant material

Fruits of *Z. integrifoliolum* were collected at Lanyu Island, Tai-tung County, Taiwan, on August 12th, 1995. A voucher specimen was deposited in the Herbarium of the School of Pharmacy, Kaohsiung Medical College, Kaohsiung, Taiwan.

3.2. Extraction and isolation

Dried fruits (16.5 kg) were brayed, extracted with MeOH and concd in vacuo to leave a brownish fluid. The MeOH ext. was partitioned with CHCl₃–H₂O (1:1). The H₂O soluble fr. was partitioned between H₂O: *n*-BuOH (1:1) to afford a H₂O fr. (620 g) and *n*-BuOH fr. (130 g). The CHCl₃ soluble fr. was extracted with 90% MeOH: *n*-hexane (1:1), yielding an *n*-hexane fr. (420 g). The 90% MeOH ext. was first treated with CHCl₃ to produce yellowish crystal I (2.54 g), then a second batch of yellowish crystal II (0.58 g) was

obtained from the filtrate. The filtrate was concd under red. pres. to obtain a CHCl3 fr. (220 g). Part of the CHCl₃ soluble fr. (99 g) was chromatographed over silica gel (2,500 g), eluting with CH₂Cl₂ gradually enriched with EtOAc to give 12 frs (C1–C12) [Cl (3.75 g, CH₂Cl₂), C2 (0.88 g, CH₂Cl₂), C3 (2.92 g, CH₂Cl₂), C4 (11.4 g, CH₂Cl₂), C5 (1.63 g, CH₂Cl₂), C6 (0.15 g, CH₂Cl₂-EtOAc, 8:2), C7 (1.63 g, EtOAc), C8 (19.1 g, EtOAc), C9 (18.1 g, EtOAc), C10 (1.36 g, CHCl₃-MeOH, 19:1), C11 (8.0 g, CHCl₃-MeOH, 9:1), C12 (5.1 g, CHCl₃-MeOH, 8:2)]. Fr. C5 (1.63 g) was rechromatographed on silica gel (49 g) using n-hexane-acetone (2:1) to yield 12 frs (C5-1-C5-12). Fr. C5-1 (64.5 mg) was purified by prep. TLC (n-hexaneacetone, 5:1) to yield a mixture of 3 and 9 (23.5 mg). Fr. C5-2 (40 mg) was purified with prep. TLC (CH₂Cl₂: EtOAc, 15:1) to afford **5** (24.7 mg). Fr. C8 (19.1 g) was rechromatographed on silica gel (480 g) and eluted with CHCl3-MeOH mixts. to yield 13 frs (C8-1-C8-13). Fr. C8-5 (4.1 g) was separated by silica gel CC (125 g) and eluted with CHCl₃-acetone to obtain 10 frs (C8-5-1-C8-5-10). Fr. C8-5-1 (68 mg) was rechromatographed by silica gel (2 g), eluting with *n*-hexane–EtOAc (1:1), gradually enriched with EtOAc, to obtain 7 frs (C8-5-1-1-C8-5-1-7). Then, fr. C8-5-1-2 (44.8 mg) was purified by prep. TLC (n-hexane-EtOAc, 1:1) to give 2 (27.8 mg). Fr. C8-5-2 (340 mg) was rechromatographed on silica gel (40 g) and eluted with CHCl₃-acetone to afford 6 frs (C8-5-2-1-C8-5-2-6). Fr. C8-5-2-3 (77.3 mg) was rechromatographed on silica gel (2.3 g) and eluted with *n*-hexane– EtOAc mixts to yield 9 frs (C8-5-2-3-1–C8-5-2-3-9). Fr. C8-5-2-3-2 (47 mg) was further purified by prep. TLC (n-hexane–EtOAc, 1:1) to obtain 1 (30 mg). Part (2.01 g) of fr. C9 (18.06 g) was rechromatographed on silica gel (60 g) and eluted with CH₂Cl₂-MeOH mixts and 13 frs (C9-1–C9-13) were collected. Part (195 mg) of fr. C9-4 (1.12 g) was purified by prep. TLC (n-hexane-acetone, 2:1) to afford 4 (21.1 mg), 6 (77.8 mg) and a mixture of 7 and 8 (37 mg).

3.3. Lanyuamide I (1)

Colorless oil. UV $\lambda_{\rm max}^{\rm EtOH}$ nm (log ε) 259 (4.00). IR $\nu_{\rm max}^{\rm Neat}$ cm⁻¹ 3300 (NH), 1700 (CO), 1650 (C=C), 1625 (CO). EI-MS m/z (rel. int.): 293 [M]⁺ (18), 180 (97), 165 (82), 152 (100), 113 (63), 107 (36), 81 (38), 79 (48), 57 (54), 43 (58), 41 (47). HR-MS: $C_{18}H_{31}NO_2$, found: 293.2350, calcd: 293.2355. ¹H NMR: δ 0.88 (3H, t, J=6.8 Hz, H-14), 0.92 (6H, d, J=6.8 Hz, H-3′, 4′), 1.25 (6H, br s, H-11–13), 1.56 (2H, quint, J=7.1 Hz, H-10), 1.79 (1H, m, H-2′), 2.39 (2H, t, J=7.1 Hz, H-9), 2.42 (2H, q, J=6.9 Hz, H-6), 2.53 (2H, t, J=6.9 Hz, H-7), 3.16 (2H, t, J=6.8 Hz, H-1′), 5.47 (1H, br s, NH, exchangeable with D_2O), 5.76 (1H, d, J=14.9 Hz, H-2), 6.02 (1H, dt, J=15.1, 6.9 Hz, H-5), 6.16 (1

H, dd, J = 15.1, 10.6 Hz, H-4), 7.16 (1 H, dd, J = 14.9, 10.6 Hz, H-3). ¹³C NMR spectral analyses: see Table 1.

3.4. Lanyuamide II (2)

Colorless oil. UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ε): 259 (3.58). IR $\nu_{\text{max}}^{\text{Neat}}$ cm⁻¹: 3320 (NH), 1720 (CO), 1648 (C=C), 1630 (CO). FAB-MS (positive): 292 [M+H]⁺. ¹H NMR: δ 0.93 (6H, d, J=6.8 Hz, H-3', 4'), 1.05 (3H, t, J=7.2 Hz, H-14), 1.79 (1 H, m, H-2'), 2.19 (4H, m, H-6, H-7), 2.30 (2H, q, J=6.8 Hz, H-10), 2.41 (2H, q, J=7.2 Hz, H-13), 2.44 (2H, t, J=7.6 Hz, H-11), 3.16 (2H, t, J=6.8 Hz, H-1'), 5.35 (2H, m, H-8, H-9), 5.59 (1H, br s, NH, exchangeable with D₂O), 5.77 (1H, d, J=15.2 Hz, H-2), 6.05 (1H, dt, J=15.0, 6.4 Hz, H-5), 6.15 (1H, dd, J=15.0, 10.6 Hz, H-4), 7.16 (1H, dd, J=15.2, 10.6 Hz, H-3). ¹³C NMR spectral analyses: see Table 1.

3.5. Lanyuamide III (3)

Colorless oil. UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ε): 259 (4.23). IR $\nu_{\text{max}}^{\text{Neat}}$ cm⁻¹: 3300 (NH), 1650 (C=C), 1625 (CO). EI-MS m/z (rel. int.): 275 [M]⁺ (1), 154 (13), 152 (11), 95 (13), 94 (12), 81 (19), 79 (20), 69 (16), 68 (15), 67 (79), 66 (30), 65 (13), 57 (53), 55 (57), 44 (18). HR-MS: C₁₈H₂₉NO, found: 275.2255, calcd: 275.2234. ¹H NMR: 0.93 (6H, d, J=6,8 Hz, H-3' 4'), 0.96 (3H, t, J=7.4 Hz, H-14), 1.79 (1H, m, H-2'), 2.00 (2H, m, H-13), 2.21 (4H, m, H-6, H-7), 2.74 (2H, t, J=6.0 Hz, H-10), 3.17 (2H, t, J=6.8 Hz, H-1'), 5.31 (2H, m, H-11, H-12), 5.38 (2H, m, H-8, H-9), 5.48 (1 H, br s, NH, exchangeable with D₂O), 5.76 (1H, d, J=15.2 Hz, H-2), 6.06 (1H, dt, J=15.2, 6.4 Hz, H-5), 6,15 (1H, dd, J=15.2, 10.2 Hz, H-4), 7.18 (1H, dd,

J=15.2, 10.2 Hz, H-3). ¹³C NMR spectral analyses: see Table 1.

Acknowledgements

This work was financially supported by the National Science Council of the Republic of China (NSC 85-2331-B-037-074-M25).

References

- Chang, C. E. & Hartley, T. G. (1993). *Rutaceae in flora of Taiwan* (2nd ed., Vol. III, p. 538). Taipei: Editorial Committee of the Flora of Taiwan.
- Chua, M. T., Maglaya, A., & Santos, A. C. (1970). Philippine Journal of Sciences, 99, 205.
- Crombie, L. (1955). Journal of the Chemical Society, 999.
- Dhar, K. L., & Atal, C. K. (1967). *Indian Journal of Chemistry*, 5, 588.
- Eisner, U., Elvidge, J., & Linstead, R. I. (1953). Journal of the Chemical Society, 1372.
- Ishii, H., Chen, I. S., Akaike, M., Ishikawa, T., & Lu, S. T. (1982).
 Yakugaku Zasshi, 102, 182.
- Jen, C. M., Tsai, I. L., Horng, D. J., & Chen, I. S. (1993). Journal of Natural Products, 56, 2019.
- Mizutani, K., Fukunaga, Y., Tanaka, O., Takasugi, N., Saruwatari, Y. I., Fuwa, T., Yamauchi, T., Wang, J., Jia, M. R., Li, F. Y., & Ling, Y. K. (1988). Chemical and Pharmaceutical Bulletin, 36, 2362
- Sheen, W. S., Tsai, I. L., Teng, C. M., Ko, F. N., & Chen, I. S. (1996). Planta Medica, 62, 175.
- Shibuya, H., Takeda, Y., Zhang, R. S., Tong, R. X., & Kitagawa, I. (1992). *Chemical and Pharmaceutical Bulletin*, 40, 2325.
- Xiong, Q., Shi, D., Yamamoto, H., & Mizuno, M. (1997).*Phytochemistry*, 46, 1123.
- Yasuda, I., Takeya, K., & Itokawa, H. (1981). Chemical and Pharmaceutical Bulletin, 29, 1791.
- Yasuda, I., Takeya, K., & Itokawa, H. (1982). *Phytochemistry*, 21, 1295.