Phytochemistry 52 (1999) 1223-1228

Three plant growth inhibiting saponins from *Duranta repens*

Syuntaro Hiradate^a,*, Hiroshi Yada^b, Tadashi Ishii^c, Naoko Nakajima^a, Mayumi Ohnishi-Kameyama^d, Hajime Sugie^a, Siriporn Zungsontiporn^e, Yoshiharu Fujii^a

^aDepartment of Environmental Biology, National Institute of Agro-Environmental Sciences, 3-1-1 Kan-nondai, Tsukuba, Ibaraki 305-8604, Japan

^bTohoku National Agricultural Experiment Station, Arai, Fukushima 960-2156, Japan

^cForestry and Forest Products Research Institute, PO Box 16, Tsukuba Norin Kenkyu Danchi-nai, Ibaraki 305-8687, Japan

^dNational Food Research Institute, 2-1-2 Kan-nondai, Tsukuba, Ibaraki, 305-8642 Japan

^cDepartment of Agriculture, Jatujak, Bangkok 10900, Thailand

Received 20 January 1999; received in revised form 24 May 1999

Abstract

Three new triterpenoid-type saponins were isolated from leaves of *Duranta repens* using plant growth inhibiting activity against seedlings of *Brassica juncea* var. *cernua* as a guide for fractionation. The structures of these compounds were elucidated as polygalacic acid-3-O- β -D-glucopyranosido-28-O-[α -L-rhamnopyranosyl-(1 \rightarrow 3')- β -D-apiofuranosyl-(1 \rightarrow 4)-[β -D-apiofuranosyl-(1 \rightarrow 3)- β -D-xylopyranosyl-(1 \rightarrow 4)-[β -D-apiofuranosyl-(1 \rightarrow 3)]- α -L-rhamnopyranosyl-(1 \rightarrow 2)- β -L-arabinopyranosido-(durantanin II), and polygalacic acid-3-O- β -D-glucopyranosido-28-O-[α -L-rhamnopyranosyl-(1 \rightarrow 3)- β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 2)- β -L-arabinopyranosido-(durantanin III). © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Duranta repens; Verbenaceae; Golden dew drop; Allelopathy; Plant growth inhibitor; Polygalacic acid; Polygalacic acid glucoside; Polygalacic acid glycoside; Oleanane-type; Triterpenoid saponins; Durantanin I–III

1. Introduction

Duranta repens Linn. (Syn.: Duranta plumieri Jacq) is a tree-type ornamental plant grown in tropical countries. It is generally 2–4 m in height and its large number of leaves create shade year round. Weeds and shrubs growing around it are not healthy, suggesting the presence of water-soluble plant growth inhibitor(s) in the leaves (Zungsontiporn, 1995). This suggests a possible allelopathic effect. The presence of pectolinaringenin and scutellarein (Subramanian & Nair, 1972), lamiid and their esters with cinnamic acid (durantosid II), 4-methoxycinnamic acid (durantosid III), 4-hydroxycinnamic acid and 4-acetoxycinnamic acid have been reported in this plant (Kuo & Kubota, 1976; Rao,

2. Results and discussion

Air-dried leaves of *D. repens* were extracted with 80% MeOH. Successive bioassay-directed liquid-liquid extraction, chromatography on activated charcoal and silica gel columns, solid phase extraction on Bondesil C18 columns, and prep. HPLC on reversed phase columns gave three plant growth inhibitors, compounds 1–3, named durantanin I–III, respectively.

0031-9422/99/\$ - see front matter \odot 1999 Elsevier Science Ltd. All rights reserved. PII: S0031-9422(99)00408-2

Rao & Vijayakumar, 1978; Rimpler & Timm, 1974), although their plant growth inhibitory activities and contributions to allelopathy are not clear. This report describes the isolation of three new candidate allelochemicals, named durantanin I–III, from the leaves of *D. repens* by focusing on the inhibitory effects on plant growth, and determining their structures from chemical and spectral evidences.

^{*} Corresponding author.

Hydrothermolysis (Kim, Higuchi & Komori, 1992) of durantanin I in hot water gave a mixture of compound 4 (prosapogenin) and sugars. The FAB-mass spectrum of compound 4 showed a protonated molecular ion $[M + Na]^+$ at m/z 689, indicating a molecular formula C₃₆H₅₈O₁₁. The broad-band decoupled ¹³C and DEPT NMR spectra of purified compound 4 showed 36 signals due to six methyl, 10 methylene, 12 methine and eight quaternary carbons. The presence of one carboxyl group (C-28, δ 181.3), six tertiary methyl proton groups (δ 0.81–1.39, s, 3H each), one olefinic proton (H-12, δ 5.31, br t, J = 3.2 Hz), and one anomeric carbon (δ 105.5) and proton (δ 4.43, d, J = 7.8 Hz) were indicated by their ¹H and ¹³C NMR spectra. The olean-12-ene triterpene skeleton was inferred from the chemical shifts of C-12 (δ 123.5) and C-13 (δ 145.2) (Fullas et al., 1996). After determining the connectivity between each carbon and proton by ¹³C-¹H COSY NMR spectrum, these C-H units were constructed according to the results of ¹H-¹H COSY, HOHAHA and HMBC NMR spectra, indicating the presence of substituted 2,3,16,23-tetrahydroxyolean-12en-28-oic acid (polygalacic acid). Polygalacic acid had been isolated from *Polygala paenea* L. (Polonski, Pourrat & Seiligmann, 1960) and its chemical structure had been confirmed by spectral evidence (Kubota & Kitatani, 1968). The acid hydrolysate of compound 4 was converted to its TMS ether derivative and analyzed by GC, which showed the presence of a glucosyl residue. The HMBC spectrum showed correlation between the glucose-C-1 (δ 105.5) and polygalacic acid-H-3 (δ 3.61) and between glucose-H-1 (δ 4.43) and polygalacic acid-C-3 (δ 83.9). Based on this evidence, compound 4 was identified as polygalacic acid-3-O-glucoside.

Durantanin I has the molecular formula C₅₈H₉₄O₂₇ (HR FAB MS spectrometry). GC-MS analysis of the acid hydrolysate of durantanin I derivatized to the per-*O*-acetylated alditol showed the presence of one glucose, one arabinose, two rhamnoses and one apiose.

A similar result was obtained from TMS ether derivative of methyl glycoside of acid hydrolysate of durantanin I. Glycosyl-linkage analysis (York, Darvill, McNeil, Stevenson & Albersheim, 1985) using the partially methylated alditol acetate derivative of acid hydrolysate of durantanin I showed the presence of terminal glucopyranose, terminal rhamnopyranose, 3linked apiofuranose, 4-linked rhamnopyranose and 2linked arabinopyranose. From the on-line optical rotation analysis of the hydrolysate of durantanin I by HPLC, the configurations of glucose, rhamnose, apiose and arabinose in durantanin I were determined to be D, L, D and L, respectively. The HMBC spectrum showed correlations between the D-glucopyranose-H-1 (δ 4.43) and polygalacic acid-C-3 (δ 84.0), terminal Lrhamnopyranose-C-1 (δ 102.0) and D-apiofuranose-H-3' (δ 3.45, 3.77), D-apiofuranose-C-1 (δ 111.6) and inner L-rhamnopyranose-H-4 (δ 3.52), inner L-rhamnopyranose-H-1 (δ 4.94) and L-arabinopyranose-C-2 (δ 76.4), and L-arabinopyranose-H-1 (δ 5.72) and polygalacic acid-C-28 (δ 177.1). From the J_{H-H} values of the anomeric protons of D-glucopyranose (7.7 Hz) and Larabinopyranose (2.8 Hz), both configurations were concluded to be β . The configurations of inner and terminal L-rhamnopyranoses were assigned as α from the chemical shifts at C-1 (δ 102.0, Rha-t; δ 101.7, Rha-i), C-3 (δ 72.4, Rha-t; δ 72.5, Rha-i) and C-5 (δ 70.0, Rha-t; δ 69.2, Rha-i), and the ${}^{1}J_{\text{C-H}}$ values (d, J = 168.7 Hz, Rha-t; d, J = 169.3 Hz, Rha-i) of its anomeric carbon signals in the ¹H non-decoupled ¹³C-NMR spectrum (Asada, Ueoka & Furuya, 1989). The configuration of D-apiofuranose was determined to be β on the basis of comparison of chemical shifts at C-1 (δ 111.6) and C-2 (δ 78.3) with previous studies (Asada et al., 1989; Ishii & Yanagisawa, 1999; Snyder & Serianni, 1987). Therefore, the structure of compound 4 was determined to be polygalacic acid-3-O-β-D-glucopyranoside, and the structure of durantanin I was determined to be polygalacic acid-3-O-β-D-glucopyranoside-28-*O*-[α -L-rhamnopyranosyl-(1 \rightarrow 3')- β -Dapiofuranosyl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -L-arabinopyranoside].

The molecular formula of durantanins II and III were $C_{63}H_{102}O_{31}$ and $C_{58}H_{94}O_{27}$ (HR FAB MS spectrometry), respectively. The hydrothermolysis of durantanins II and III each gave compound II. The sugar analyses using per-II-acetylated alditol and methyl TMS derivatives showed the presence of one glucose, two rhamnoses, one xylose, one apiose and one arabinose in durantanin II, and one glucose, two rhamnoses, one xylose and one arabinose in durantanin III. The on-line optical rotation analysis showed the configurations of glucose, rhamnose, xylose, apiose and arabinose in durantanins II and IIII are III. D, III and IIII are III are III.

Glycosyl-linkage analysis of durantanin II indicated

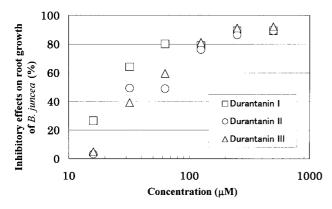


Fig. 1. Inhibitory effects of durantanin I-III on root growth of B. juncea.

the presence of terminal D-glucopyranose, terminal Lrhamnopyranose, terminal D-apiofuranose, 3-linked Dxylopyranose, 3,4-di-linked L-rhamnopyranose and 2linked L-arabinopyranose. The HMBC spectrum of durantanin II showed correlations between D-glucopyranose-H-1 (δ 4.43) and polygalacic acid-C-3 (δ 84.0), terminal L-rhamnopyranose-H-1 (δ 5.12) and D-xylopyranose-C-3 (δ 84.0), D-xylopyranose-H-1 (δ 4.65) and inner L-rhamnopyranose-C-4 (δ 78.3), terminal Dapiofuranose-H-1 (δ 5.24) and inner L-rhamnopyranose-C-3 (δ 81.9), inner L-rhamnopyranose-H-1 (δ 4.94) and L-arabinopyranose-C-2 (δ 76.1), and L-arabinopyranose-H-1 (δ 5.68) and polygalacic acid-C-28 (δ 177.1). The configuration of D-xylopyranose was assigned as β because of the J_{H-H} value (7.7 Hz) between H-1 and H-2. The other anomeric configurations of sugar moieties in durantanin II were the same as those of durantanin I. Therefore, the structure of durantanin II was determined to be polygalacic acid-3-O-β-D-glucopyranosido-28-O-[α-L-rhamnopyranosyl- $(1 \rightarrow 3)$ - β -D-xylopyranosyl- $(1 \rightarrow 4)$ - $[\beta$ -D-apiofuranosyl- $(1 \rightarrow 3)$]- α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -L-arabinopyranoside].

Durantanin III was shown to have terminal D-glucopyranose, terminal L-rhamnopyranose, 3-linked D-xylopyranose, 4-linked L-rhamnopyranose and 2-linked Larabinopyranose in glycosyl-linkage analyses. The connectivities between D-glucopyranose-H-1 (δ 4.43) and polygalacic acid-C-3 (δ 84.0), terminal L-rhamnopyranose-H-1 (δ 5.14) and D-xylopyranose-C-3 (δ 84.2), Dxylopyranose-H-1 (δ 4.53) and L-rhamnopyranose-C-4 (δ 78.5), L-rhamnopyranose-H-1 (δ 5.03) and L-arabinopyranose-C-2 (δ 75.8), and L-arabinopyranose-H-1 (δ 5.62) and polygalacic acid-C-28 (δ 177.1) were observed in HMBC. The configurations of sugar residues in durantanin III were same as those of durantanin II. Therefore, the structure of durantanin III is acid-3-O-β-D-glucopyranosido-28-O-[αpolygalacic

Table 1 ¹³C NMR spectral data for durantanin I–III (compounds 1–3) and compound 4 (150.8 MHz, CD₃OD)

C	Durantanin I	Durantanin II	Durantanin III	Compound 4 ^a
1	44.5	44.5	44.5	44.4
2	71.2	71.2	71.2	71.2
3	84.0	84.0	84.0	83.9
4	43.2	43.2	43.2	43.1
5	48.3	48.3	48.3	48.2
6	18.7	18.7	18.8	18.6
7	33.8	33.8	33.7	33.7
8	41.0	40.9	40.9	40.7
9	48.5	48.5	48.5	48.4
10	37.6	37.6	37.6	37.5
11	24.7	24.7	24.7	24.6
12	124.0	123.9	123.9	123.5
13	144.8	144.8	144.8	145.2
14	43.0	42.9	43.0	42.9
15	36.4	36.3	36.4	36.1
16	74.7	74.8	74.6	75.3
17	50.4	50.4	50.4	49.6
18	42.1	42.2	42.2	42.1
19	47.6	47.6	47.7	47.7
20	31.4	31.4	31.4	31.4
21	36.4	36.4	36.5	36.6
22	31.9	31.8	32.0	32.7
23	65.7	65.8	65.9	65.6
24	14.8	14.9	14.9	14.7
25	18.1	18.1	18.0	17.8
26	17.7	17.7	17.7	17.5
27	27.5	27.4	27.4	27.3
28	177.1	177.1	177.1	181.2
29	33.4	33.4	33.4	33.4
30	25.3	25.3	25.2	24.9

^a Compound 4: prosapogenin of durantanin I-III.

L-rhamnopyranosyl- $(1 \rightarrow 3)$ - β -D-xylopyranosyl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -L-arabinopyranoside].

Inhibitory effects of durantanin **I–III** on root growth of seedlings of *B. juncea* were found to have the same order (Fig. 1), and 50% of the root growth was inhibited around 50 μ M of durantanin **I–III**.

3. Experimental

3.1. General

Mps: uncorr; NMR: JEOL JNM α -600 spectrometer, CD₃OD with TMS as int. ref; MS: JEOL SX102A spectrometer, direct inlet system, glycerol was used as a matrix; IR: KBr disc; CC: activated charcoal (Charcoal, Activated, 031-02135, Wako Pure Chem. Ind., Ltd), silica gel (Wakogel C-200, 74–149 μ m, Wako Pure Chem. Ind., Ltd); solid phase extraction: chemically modified silica gel (Analytichem Bondesil C18, 40 μ m, preparative grade, varian); prep. HPLC:

Table 2 ¹³C NMR spectral data for the sugar moieties of durantanin I–III (compounds 1–3) and compound 4 (150.8 MHz, CD₃OD)

C	Durantanin I	Durantanin II	Durantanin III	Compound 4 ^a
C-3- <i>O</i> -β-D-Glc				
1	105.5	105.5	105.5	105.5
2	75.4	75.4	75.4	75.4
3	78.3	78.3	78.3	78.2
4	71.2	71.2	71.2	71.1
5	77.8	77.8	77.8	77.7
6	62.4	62.4	62.4	62.3
α -L-Rha- t^a				
1	102.0	102.7	102.6	
2	72.1	72.4	72.3	
3	72.4	72.3	72.3	
4	74.0	74.0	74.1	
5	70.0	70.1	70.1	
6	18.3	17.9	17.9	
β-D-Xyl	10.0	27.5	1,	
1		105.0	106.6	
2		76.0	76.3	
3		84.0	84.2	
4		70.2	69.9	
5		67.0	67.2	
β-D-Арі		07.0	07.2	
1	111.6	112.1		
2	78.3	78.2		
3	79.5	80.1		
3'	70.6	64.9		
4	75.1	74.9		
α-L-Rha-i ^b	/3.1	74.9		
1	101.7	101.2	101.4	
2	72.7	72.0	72.2	
3	72.7	81.9	72.4	
	80.3	78.3	83.5	
4 5	69.2	69.2	69.1	
6	18.5	18.3	18.2	
C-28- <i>O</i> -β-L-Ara	02.7	02.9	04.1	
1	93.7	93.8	94.1	
2	76.4	76.1	75.8	
3	70.2	70.2	71.2	
4	66.3	66.3	67.1	
5	62.8	62.9	63.8	

 $^{^{}a}$ α -L-Rha-t: terminal α -L-rhamnopyranose.

Shim-pack PREP-ODS(H) (20 mm \times 25 cm, Shimadzu).

3.2. Extraction and separation

Air-dried leaves of *D. repens* (from Bangkok, Thailand, 200 g) were extracted with 80% MeOH. The resulting water suspension of the MeOH extract was divided into frs soluble in hexane (7 g), acidic EtOAc (21 g), acidic *n*-BuOH (44 g), and water (23 g). The *n*-BuOH fr. was chromatographed on an activated charcoal column and eluted with H₂O, MeOH, Me₂CO and EtOAc, successively. The MeOH eluate (11 g) was

sepd into 11 frs on a column of silica gel (CHCl₃, successively and finally MeOH). The 40% MeOH/CHCl₃ eluate (3 g) was sepd into eight frs by solid phase extraction (successively from 40% aq. MeOH to pure MeOH). The 50 and 55% aq. MeOH frs were chromatographed by prep. HPLC [Shim-pack PREP-ODS(H), 25 cm × 20 mm, eluted with 50% aq. MeOH] to obtain active durantanin I (0.42 g), II (2.01 g) and III (0.37 g).

3.2.1. Durantanin I

White powder, mp 216–220°, $[\alpha]_D^{22}$ –57.4° (MeOH; *c* 1.21). HR FAB MS: m/z 1223.6053, $C_{58}H_{94}O_{27}+H$

 $^{^{}b}$ α -L-Rha-i: inner α -L-rhamnopyranose.

requires 1223.6061. IRv_{max} (KBr) cm⁻¹: 3402, 2936, 1736, 1720, 1638, 1052. ¹H NMR (600 MHz, CD₃OD) δ : 0.80, 0.88, 0.94, 0.99, 1.30, 1.38 (all 3H, s), 1.25 (3H, d, J = 6.2 Hz, H-6 of Rha-i), 1.27 (3H, d, J = 6.2 Hz, H-6 of Rha-t), 4.43 (1H, d, J = 7.7 Hz, H-1 of Glc), 4.71 (1H, d, J = 1.5 Hz, H-1 of Rha-t), 4.94 (1H, d, J = 1.5 Hz, H-1 of Rha-i), 5.34 (1H, d, J = 2.4 Hz, H-1 of Api), 5.72 (1H, d, J = 2.8 Hz, H-1 of Ara). ¹³C NMR: Tables 1 and 2.

3.2.2. Durantanin II

White powder, mp 225–227°, $[\alpha]_D^{22}$ –62.8° (MeOH; c 1.22). HR FAB MS: m/z 1355.6487, $C_{63}H_{102}O_{31}+H$ requires 1355.6484. IRv_{max} (KBr) cm⁻¹: 3437, 2935, 1736, 1720, 1638, 1048. ¹H NMR (600 MHz, CD₃OD) δ : 0.80, 0.88, 0.95, 0.98, 1.30, 1.37 (all 3H, s), 1.24 (3H, d, J = 5.6 Hz, H-6 of Rha-t), 1.26 (3H, d, J = 5.9 Hz, H-6 of Rha-i), 4.43 (1H, d, J = 7.7 Hz, H-1 of Glc), 4.65 (1H, d, J = 7.7 Hz, H-1 of Xyl), 4.95 (1H, d, J = 1.8 Hz, H-1 of Rha-i), 5.12 (1H, d, J = 1.5 Hz, H-1 of Rha-t), 5.24 (1H, d, J = 3.7 Hz, H-1 of Api), 5.68 (1H, d, J = 2.9 Hz, H-1 of Ara). ¹³C NMR: Tables 1 and 2.

3.2.3. Durantanin III

White powder, mp 223–225°, $[\alpha]_D^{22}$ –45.6° (MeOH; c 1.37). HR FAB MS: m/z 1223.6104, $C_{58}H_{94}O_{27}+H$ requires 1223.6061. IRv_{max} (KBr) cm⁻¹: 3437, 2933, 1736, 1720, 1638, 1047. ¹H NMR (600 MHz, CD₃OD) δ : 0.79, 0.88, 0.96, 0.97, 1.30, 1.38 (all 3H, s), 1.25 (3H, d, J = 6.2 Hz, H-6 of Rha-t), 1.28 (3H, d, J = 6.2 Hz, H-6 of Rha-i), 4.44 (1H, d, J = 7.7 Hz, H-1 of Glc), 4.53 (1H, d, J = 7.7 Hz, H-1 of Syl), 5.03 (1H, d, J = 1.5 Hz, H-1 of Rha-i), 5.14 (1H, d, J = 1.5 Hz, H-1 of Rha-t), 5.62 (1H, d, J = 3.7 Hz, H-1 of Ara). ¹³C NMR: Tables 1 and 2.

3.2.4. Compound 4

White powder, mp 226–230°, $[\alpha]_D^{22} + 17.1^\circ$ (MeOH; c 0.053). HR FAB MS: m/z 689.3868, $C_{36}H_{58}O_{11} + Na$ requires 689.3877. IRv_{max} (KBr) cm⁻¹: 3432, 2927, 1701, 1686, 1042. ¹H NMR (600 MHz, CD₃OD) δ : 0.81, 0.88, 0.94, 0.97, 1.29, 1.39 (all 3H, s), 4.45 (1H, d, J = 7.7 Hz, H-1 of Glc). ¹³C NMR: Tables 1 and 2.

3.3. Hydrothermolysis

A solution of durantanin I, II or III (10 mg) in 1 ml of H_2O was allowed to stand for 2 days at 105° . The soln was subjected to sept on solid phase extraction (H_2O and 40, 55, 80 and 100% aq. MeOH). The 80% aq. MeOH eluate gave compound 4.

3.4. Acid hydrolysis

Dry powder of durantanin I–III (1 mg) was mixed with 0.25 ml of 2 M TFA and allowed to stand at 121° for 1 h.

3.5. On-line optical rotation analysis by HPLC

Acid hydrolysate of durantanin **I–III** and standard sugar samples were analyzed using an HPLC system (Waters 626 LC System) equipped with a Shodex Asahipack NH2P-50 4E column (4.6 mm i.d. \times 250 mm, H₃PO₄:water:acetonitrile = 1:19:80, 1 ml/min, 40°) and optical rotational detector (Shodex OR2).

3.6. Measurement of activity

Test solutions were placed into test tubes and dried completely under vacuum at 40° . One milliliter of hot agar aq. solution ($40-50^{\circ}$, 0.5%) was added to the test tube, and the sample was dissolved or suspended. After solidification of the agar solution, seeds of *Brassica juncea* var. *cernua* were placed on it. After incubation at 25° in light conditions for 5 days, the inhibitory effect on elongation growth was observed by measuring the length of hypocotyls and roots.

Acknowledgements

This work was supported in part by a grant from the Ministry of Agriculture, Forestry, and Fisheries of Japan. This work is part of the Ministry's project, Development of Novel Weed Control Technology by Applying Metabolic Genes in Plants. The authors acknowledge Dr Z. Iqbal's critical reading of the manuscript.

References

Asada, Y., Ueoka, T., & Furuya, T. (1989). Chem. Pharm. Bull., 37, 2139.

Fullas, F., Wani, M. C., Wall, M. E., Tucker, J. C., Beecher, C. W. W., & Kinghorn, A. D. (1996). *Phytochemistry*, 43, 1303.

Ishii, T., & Yanagisawa, M. (1999). Carbohydr. Res. in press.

Kim, Y. C., Higuchi, R., & Komori, T. (1992). *Liebigs Ann. Chem.*, 1992, 453.

Kubota, T., & Kitatani, H. (1968). Chem. Comm., 739, 1005.

Kuo, Y. H., & Kubota, T. (1976). Experientia, 32, 968.

Polonski, J., Pourrat, H., & Seiligmann, J. (1960). *Compt. Rend.*, 251, 2374.

Rao, B., Rao, T. N., & Vijayakumar, E. K. S. (1978). *Indian J. Chem.*, 16B, 844.

Rimpler, R., & Timm, H. (1974). Z. Naturforsch., 29c, 111.

Snyder, J. R., & Serianni, A. S. (1987). Carbohydr. Res., 166, 85.Subramanian, S. S., & Nair, A. G. R. (1972). Phytochemistry, 11, 3095.

York, W. S., Darvill, A. G., McNeil, M., Stevenson, T. T., &

Albersheim, P. (1985). *Methods in enzymology* (Vol. 118, p. 3). Academic Press, Inc, New York.

Zungsontiporn, S. (1995). Ph. D. diss. Tokyo Univ. of Agric. Technol., Fuchu, Japan.