Phytochemistry 52 (1999) 1657-1659

Aryl naphthalenes norlignans from Vitex rotundifolia

Kazuyoshi Kawazoe*, Aki Yutani, Yoshihisa Takaishi

Faculty of Pharmaceutical Sciences, University of Tokushima, 1-78, Sho-machi, Tokushima, 770-8505, Japan Received 28 January 1999; received in revised form 1 June 1999

Abstract

Three arylnaphthalene norlignans were isolated from the root of *Vitex rotundifolia*. Their structures were established from spectroscopic evidence as 1-(3,4-dimethoxyphenyl)-7-hydroxy-8- methoxynaphthalene-3-carbaldehyde, 1-(3,4-dimethoxyphenyl)-2,7-dihydroxy-8-methoxynaphthalene-3-carbaldehyde and 2,7-dihydroxy-1,9,10-trimethoxy-7*H*-benzo[*c*] fluorene-6-carbaldehyde, respectively. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Vitex rotundifolia; Verbenaceae; Arylnaphthalene; Lignan; Vitrofolal

1. Introduction

Vitex rotundifolia (Verbenaceae) is a common plant growing on East Asian beaches, and its seed is not only used in Japan for a cold remedy, i.e., treatment of headaches, but also used as raw material for Chinese traditional medicine. From this plant, some diterpenoids and sesquiterpenoids have been obtained (Asaka, Kamikawa & Kubota, 1973; Tada & Yasuda, 1984), and interestingly, some insecticidal substances are present in its leaf tissue as well (Watanabe, Takasa, Matsuo & Nishimura, 1995). In this paper, we describe the charactersiation of three norlignans, vitrofolals A, B and C, in roots of V. rotundifolia.

2. Results and discussion

A methanol extract obtained from the roots of V. rotundifolia was partitioned between EtOAc and water, with the resulting EtOAc soluble fraction subjected to chromatography to obtain each component (see Section 3).

Vitrofolal A (1) had a strong UV absorption (269, 224 nm) indicator of a naphthalene chromophore,

0031-9422/99/\$ - see front matter \odot 1999 Elsevier Science Ltd. All rights reserved. PII: S0031-9422(99)00405-7

whereas in the IR spectrum, both hydroxy and carbonyl groups were detected at 3600–2800 and 1687 cm⁻¹, respectively. From its EIMS spectrum, a formula of C₂₀H₁₈O₅ was deduced. The ¹H-NMR spectrum indicated eight deshielded protons, including a formyl group and three methoxyl groups (Table 1); thus, a naphthalene bearing an aryl ring was proposed for the skeleton of 1. All C-H correlations were revealed by analysis of a HSQC spectrum. In the ¹H-¹H COSY, three correlation series were shown, i.e., I: δ 7.83 (H-2) and 8.27 (H-4), II: δ 7.83 (H-5) and 7.39 (H-6), and III: δ 7.03 (H-2'), 6.93 (H-5') and 7.06 (H-6'). These groups were assigned to three aromatic rings and their position was established by the coupling constants. From the NOESY, the nOe correlations between H-2 and both H-2' and H-6', and between H-4 and both H-5 and an aldehyde proton were observed. Thus, an aromatic ring containing series III and the series I and II form a 1-arylnaphthalene skeleton with a formyl group at C-3. The HMBC spectrum showing correlations between the aldehyde proton and C-2 and C-4, H-6 and C-4a, C-7 and C-8, H-2' and C-1 and C-4', supported these findings, and assigned three methoxyl groups to C-8, C-3' and C-4'. Consequently, we were able to assign all proton and carbon signals, and 1 was defined as 1-(3,4-dimethoxyphenyl)-7-hydroxy-8-methoxynaphthalene-3-carbaldehyde.

The ¹H-NMR spectral pattern of vitrofolal B (2)

^{*} Corresponding author.

Table 1 ¹H-NMR spectral data for 1^a, 2^a and 3^b

	1	2		3
2	7.83, 1H, d , $J = 1.5$ Hz	=	3	7.40, 1H, d , $J = 8.8$ Hz
4	8.27, 1H, d , $J = 1.5$ Hz	8.11, 1H, s	4	7.83, 1H, d , $J = 8.8$ Hz
5	7.83, 1H, d , $J = 8.8 \text{ Hz}$	7.68, 1H, d , $J = 8.8$ Hz	5	8.21, 1H, s
6	7.39, 1H, d , $J = 8.8 \text{ Hz}$	7.18, 1H, d , $J = 8.8$ Hz	7	5.85, 1H, d , $J = 3.0$ Hz
2'	7.03, 1H, d , $J = 1.9$ Hz	6.99, 1H, brs	8	7.34, 1H, <i>s</i>
5′	6.93, 1H, d , $J = 8.2 \text{ Hz}$	ca. 7.0, 1H, m	11	8.32, 1H, s
6′	7.06, 1H, dd , $J = 1.9$, 8.2 Hz	ca. 7.0, 1H, m	1-OMe	3.69, 3H, s
2-OH	=	10.71, 1H, s	2-OH	6.52, 1H, s
3-СНО	10.12, 1H, s	10.05, 1H, s	6-CHO	10.14, 1H, s
7-OH	6.32, 1H, s	6.25, 1H, s	7-OH	5.06, 1H, d , $J = 3.0$ Hz
8-OMe	3.12, 3H, s	3.13, 3H, s	9-OMe	4.03, 3H, s
3'-OMe	3.90, 3H, s	3.89, 3H, s	10-OMe	4.03, 3H, s
4'-OMe	3.97, 3H, s	3.97, 3H, s		

^a Spectra were measured in CDCl₃.

was very similar to that of vitrofolal A (1), except that it had one D_2O exchangeable singlet signal at low-field (δ 10.71) which resulted in a singlet aromatic methine signal (δ 8.11) appearing instead of two doublet signals coupled with each other.

Furthermore, the formula $C_{20}H_{18}O_6$ suggested that **2** is an oxidized compound of **1**. A proton at δ 10.71 should contribute to forming a hydrogen bond because it appeared at low-field as a sharp singlet signal. This indicated that the hydroxyl group was adjacent to the formyl group. In the nOe experiment, a correlation between H-4 and H-5 (δ 8.11 and 7.68) was observed. From these facts, it was revealed that **2** has a hydroxy group at C-2, and its structure was deduced as 1-(3,4-dimethoxyphenyl)-2,7-dihydroxy-8-methoxynaphthalene-3-carbaldehyde.

Vitrofolal C (3), $C_{21}H_{18}O_6$, displayed analogous spectral data to vitrofolal A (1). One doublet methine proton (δ 5.85, J=3.0 Hz) coupled with the hydroxy proton showed long-range correlations with carbon signals at δ c 135.8 (C-7a), 132.6 (C-11a), and 139.0 (C-11b) in the HMBC spectrum. Furthermore, in the NOESY spectrum, correlations were observed between H-7 and singlet aromatic proton H-8, and between H-

11 and a methoxy group attached to C-1. From these facts, it was revealed that two aromatic rings and the methine carbon make a fluorene ring system. Except for near C-7, the correlation patterns of nOe were close to those of vitrofolal A (1). Consequently, vitrofolal C (3) was deduced to be 2,7-dihydroxy-1,9,10-trimethoxy-7*H*-benzo[*c*] fluorene-6-carbaldehyde.

Each of the compounds has an arylnaphthalene skeleton that has peviously been only occasionally reported (Rischmann, Mues, Geiger, Laas & Eicher, 1989; Feliciano, Corral, Gordaliza & Castro, 1991). From the viewpoint of their biosynthesis, vitrofolal A (1), B (2) and C (3) should be formed via coniferyl alcohol coupling and subsequent metabolism as described elsewhere (Lewis & Davin, 1999).

3. Experimental

3.1. General

 1 H- (400 MHz) and 13 C-NMR (100 MHz) in CDCl₃ and pyridine- d_5 with TMS as internal standard. Column chromatograhpy used silica gel 60 (230-400 mesh, Merck). Silica gel HPLC and GPC (Gel

^b Spectra were measured in pyridine-d₅.

Table 2 ¹³C-NMR spectral data for **1**^a, **2**^a and **3**^b

	1	2		3
1	137.3	120.0	1	140.8
2	127.2	154.6	2	150.9
3	131.5	128.7	3	118.3
4	134.5	138.5	4	129.1
4a	129.9	124.6	4a	130.6
5	128.4	128.7	5	138.2
6	118.3	116.2	6	129.0
7	150.2	151.5	6a	144.5
8	141.0	140.1	7	74.3
8a	128.8	131.6	7a	135.8
1'	134.6	119.7	8	109.8
2'	113.7	114.7	9	149.3
3′	147.7	148.0	10	149.7
4'	148.4	148.3	11	107.9
5'	109.9	110.1	11a	132.6
6′	122.0	123.2	11b	139.0
3-CHO	191.9	196.2	11c	125.7
8-OMe	61.6	61.6	1-OMe	63.0
3'-OMe	56.0	56.0	6-CHO	195.0
4'-OMe	56.1	56.2	9-OMe	56.4
			10-OMe	56.3

^a Spectra were measured in CDCl₃.

Permeation Chromatography) employed SIL-06 (250×20 mm, YMC) and H- 2001 (1000×20 mm, Shodex), respectively.

3.2. Plant material

Roots of *V. rotundifolia* were collected in 1997, at Tosa-domari beach in Tokushima Prefecture, Japan. The identification of the plant material was comfirmed by Dr. K. Murakami, and a voucher specimen is deposited in the herbarium in Faculty of Pharmaceutical Sciences, University of Tokushima.

3.3. Extraction and fractionation

The MeOH extract from 12.1 kg of air-dried material was partitioned between EtOAc and H₂O to obtain an EtOAc-soluble portion (106 g). This was applied to a silica gel column which was eluted with CHCl₃–MeOH (4:1) to obtain frs. 1–9. Fr. 2 (19.9 g) was purified by silica gel HPLC (*n*-hexane–EtOAc, 4:1) and GPC (CHCl₃) to obtain vitrofolal A (1, 10.6 mg) and vitrofolal C (3, 18.2 mg). Fr. 3 (11.2 g) was purified by GPC to obtain vitrofolal B (2, 3.7 mg).

3.4. *Vitrofolal A* (1)

Amorphous yellow powder, HR-EI MS: m/z 338.1158 [M]⁺, calcd. for C₂₀H₁₈O₅, 338.1155. IR $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 3600–2800, 1687, 1608, 1511, 1467, 1250, 1157, 1084, 1024. UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 329 (4.0), 269 (4.5), 224 (4.5), 208 (4.6). ¹H-NMR spectral data (Table 1), ¹³C-NMR spectral data (Table 2).

3.5. *Vitrofolal B* (2)

Amorphous yellow powder, HR-EI MS: m/z 354.1100 [M]⁺, calcd. for C₂₀H₁₈O₆, 354.1103. IR $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 3600–2800, 1656, 1517, 1385, 1252, 1026. UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 338 (3.8), 283 (4.2), 206 (4.3). ¹H-NMR spectral data (Table 1), ¹³C-NMR spectral data (Table 2).

3.6. *Vitrofolal C* (**3**)

Amorphous red powder, $[\alpha]_{\rm D}$ + 3.7° (CHCl₃, c 0.53), HR-EI MS: m/z 366.1096 [M]⁺, calcd. for C₂₁H₁₈O₆, 366.1104. IR $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 3600–2800, 1612, 1499, 1470, 1297. UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 365 (4.0), 237 (4.5). ¹H-NMR spectral data (Table 1), ¹³C-NMR spectral data (Table 2).

Acknowledgements

The authors thank Dr. K. Murakami for identification of the plant material.

References

Asaka, Y., Kamikawa, T., & Kubota, T. (1973). Chemistry Letters, 1973, 937.

Feliciano, A. S., Corral, J. M. M. D., Gordaliza, M., & Castro, A. (1991). *Phytochemistry*, 30, 3483.

Lewis, N.G., & Davin, L.B. (1999). Comprehensive natural products chemistry. In D.H.R. Borton Jr., K. Nakanishi, K. & O. Meth-Cohn (Eds.), Vol. 1, pp. 639–712.

Rischmann, M., Mues, R., Geiger, H., Laas, H. J., & Eicher, T. (1989). *Phytochemistry*, 28, 867.

Tada, H., & Yasuda, F. (1984). Heterocycles, 22, 2203.

Watanabe, K., Takasa, Y., Matsuo, N., & Nishimura, H. (1995). Biosci. Biotech. Biochem, 59, 1979.

^b Spectra were measured in pyridine-d₅.