



# Structure and synthesis of the first procassinidin dimers based on epicatechin, and gallo- and epigallo-catechin<sup>☆</sup>

Johan Coetzee<sup>a,1</sup>, Lulama Mciteka<sup>a</sup>, Elfranco Malan<sup>a,1</sup>, Daneel Ferreira<sup>b,\*</sup>

<sup>a</sup>Department of Chemistry, University of the Orange Free State, P.O. Box 339, Bloemfontein 9300, South Africa

<sup>b</sup>National Center for Natural Products Research, Research Institute of Pharmaceutical Sciences, School of Pharmacy, The University of Mississippi, University, MS 38677, USA

Received 10 September 1999; received in revised form 20 December 1999

## Abstract

The range of natural dimeric procassinidins is extended by identification of cassiaflavan-(4 $\alpha$   $\rightarrow$  8)-epicatechin, cassiaflavan-(4 $\alpha$   $\rightarrow$  8)-epigallocatechin, cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epicatechin, cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epigallocatechin, cassiaflavan-(4 $\beta$   $\rightarrow$  8)-gallo catechin, *ent*-cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epicatechin and cassiaflavan-(4 $\alpha$   $\rightarrow$  6)-epicatechin in the bark of *Cassia petersiana*. Their structures and absolute configuration were confirmed by synthesis. © 2000 Elsevier Science Ltd. All rights reserved.

**Keywords:** *Cassia petersiana*; Leguminosae; Flavanoids; Proanthocyanidins; Procassinidins; Synthesis

## 1. Introduction

The occurrence of flavan containing dimeric proanthocyanidins in nature is rare (Porter, 1988, 1994). Only four sources of these compounds were reported, viz cassiaflavan-epiafzelechin analogues from *Cassia fistula*, (Morimoto, Nonaka, Chen & Nishioka, 1988), (2*S*)-3',4',7-trihydroxyflavan-catechin analogues with lipase inhibitory activity from *Cassia nomame* (Hatano et al., 1997), (2*S*)-4',7,8-trihydroxyflavan-(4 $\beta$   $\rightarrow$  6)-epioritin-4 $\alpha$ -ol from *Acacia caffra* (Malan, Sireeparsad, Swinny & Ferreira, 1997) and butiniflavan-epicatechin and -epigallocatechin probutinidins from *Cassia petersiana* (Coetzee, Mciteka, Malan & Ferreira, 1999). We now report the structure and synthesis of the first procassinidins based on epicatechin and

gallo- and epigallo-catechin from the bark of *C. petersiana*, an aqueous extract which is used in traditional African medicine to treat fevers, gonorrhoea and skin infections (Palgrave, 1983).

## 2. Results and discussion

The bark of *C. petersiana* was extracted with acetone and yielded a complex mixture of phenolic monomers and dimeric proanthocyanidins. Catechin, epicatechin, gallo catechin, epigallocatechin and the three novel butiniflavan-epicatechin and -epigallocatechin probutinidins (Coetzee et al., 1999) were accompanied by seven dimeric procassinidins **1**, **3**, **5**, **7**, **9**, **11** and **13**. The mixture could be resolved by gel column chromatography and the compounds purified by TLC as their permethylaryl ether acetate derivatives **2**, **4**, **6**, **8**, **10**, **12** and **14**, respectively.

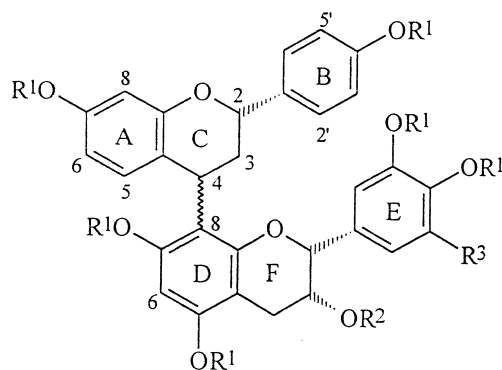
The structures and relative configuration of these procassinidin derivatives were determined by <sup>1</sup>H NMR (Tables 1 and 2). Owing to the adverse effects of rotational isomerism about the C-4(C)  $\rightarrow$  C-8(D) interflavanyl bond the spectra of compounds **6**, **8**, and **10**

<sup>☆</sup> Part 31 in the series 'Oligomeric Flavanoids'. Part 30 (Coetzee, Mciteka, Malan and Ferreira, *Phytochemistry*, 1999, 52, 737–743).

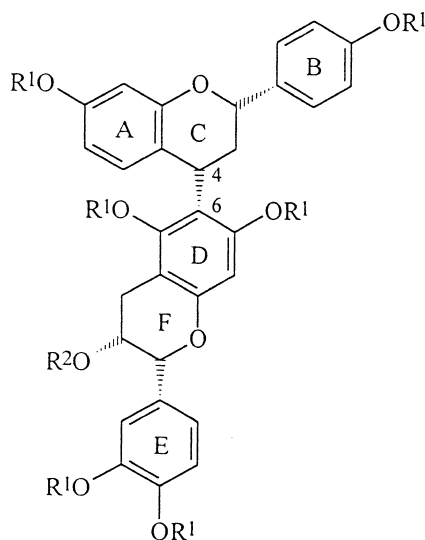
\* Corresponding author. Tel.: +1-91-601-232-1572; fax: +1-91-601-232-7062.

E-mail addresses: coetzeej@cem.nw.uovs.ac.za (J. Coetzee), dferreir@olemiss.edu (D. Ferreira).

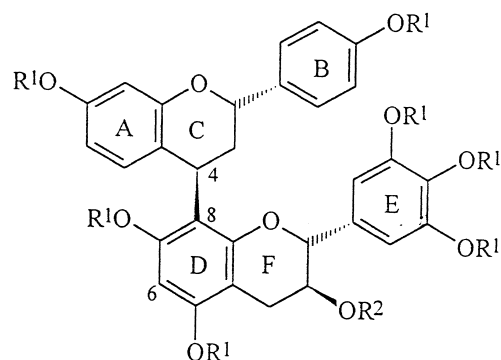
<sup>1</sup> Correspondence may also be addressed to Coetzee or Malan.



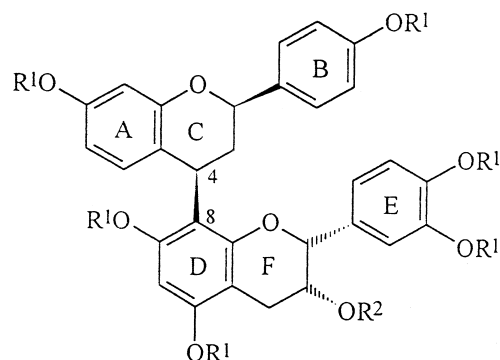
- 1  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=R^2=R^3=H$   
 2  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=Me, R^2=Ac, R^3=H$   
 3  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=R^2=H, R^3=OH$   
 4  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=Me, R^2=Ac, R^3=OMe$   
 5  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=R^2=R^3=H$   
 6  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=Me, R^2=Ac, R^3=H$   
 7  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=R^2=H, R^3=OH$   
 8  $\equiv$   $\equiv$   $\equiv$  ,  $R^1=Me, R^2=Ac, R^3=OMe$



- 13  $R^1=R^2=H$   
 14  $R^1=Me, R^2=Ac$



- 9  $R^1=R^2=H$   
 10  $R^1=Me, R^2=Ac$



- 11  $R^1=R^2=H$   
 12  $R^1=Me, R^2=Ac$

were broadened and had to be recorded at elevated temperatures (70°C,  $C_6D_6$ ) to obtain sharp sets of resonances. The spectra of compounds **2**, **4**, **12** and **14** showed two distinct rotamers at room temperature (20°C) for **2**, **12** (in  $CDCl_3$ ) and **14** ( $C_6D_6$ ), and at 40°C for **4** ( $C_6D_6$ ). The close proximity of the 4-H(C) resonances, however, precluded definition of the two rotamers since observation of NOE effects between these protons and 7-OMe(D) is a prerequisite for rotamer differentiation.

The  $^1H$  NMR spectral data (Tables 1 and 2) for compounds **2**, **4**, **6**, **8**, **10**, **12** and **14** showed similarly substituted ABC-units comprising AA'BB'- and ABX-systems for the aromatic protons and an AMNX-coupled pattern, reminiscent of the heterocyclic protons of a C-4 substituted flavan moiety (Hatano et al., 1997; Malan et al., 1997; Coetzee et al., 1999). The spectra of all the compounds additionally displayed a one-proton singlet in the high field aromatic region and an AMXY-system for protons in the heterocyclic

Table 1  
<sup>1</sup>H NMR peaks ( $\delta_H$ ) for compounds **2**, **4**, **6** and **8** at 300 MHz. Splitting patterns and *J*-values (Hz) are given in parentheses

Ring	H	<b>2</b> (CDCl <sub>3</sub> , 20°C)	<b>4</b> (C <sub>6</sub> D <sub>6</sub> , 40°C)	<b>6</b> (C <sub>6</sub> D <sub>6</sub> , 70°C)	<b>8</b> (C <sub>6</sub> D <sub>6</sub> , 70°C)
A	5	6.77, 6.66 <sup>a</sup> (d, 8.5)	7.18, 7.20 <sup>a</sup> (d, 8.5)	7.05 (d, 8.5)	7.04 (d, 8.5)
	6	6.40, 6.34 <sup>a</sup> (dd, 8.5, 2.5)	6.66, 6.73 <sup>a</sup> (dd, 8.5, 2.5)	6.56 (dd, 8.5, 2.5)	6.54 (dd, 8.5, 2.5)
B	8	6.49, 6.47 <sup>a</sup> (d, 2.5)	6.77, 6.91 <sup>a</sup> (d, 2.5)	6.73 (d, 2.5)	6.65 (d, 2.5)
	2', 6'	7.20, 7.45 <sup>a</sup> (d, 9.0)	7.27, 7.46 <sup>a</sup> (d, 9.0)	7.46 (d, 9.0)	7.46 (d, 9.0)
	3', 5'	6.82, 6.96 <sup>a</sup> (d, 9.0)	6.80, 6.93 <sup>a</sup> (d, 9.0)	6.85 (d, 9.0)	6.86 (d, 9.0)
C	2	5.17, 5.10 <sup>a</sup> (dd, 12.0, 2.0)	5.27, 5.16 <sup>a</sup> (dd, 11.5, 2.0)	5.78 (dd, 7.0, 3.5)	5.78 (dd, 7.0, 3.5)
	3	2.00, 2.20 <sup>a</sup> (ddd, 13.0, 2.0, 5.5)	2.28, 2.41 <sup>a</sup> (ddd, 13.0, 5.0, 2.5)	2.55 (ddd, 13.5, 7.0, 6.0)	2.56 (ddd, 13.5, 7.0, 6.0)
	3	2.72, 2.78 <sup>a</sup> (m)	3.32, 3.13 <sup>a</sup> (m)	2.97 (ddd, 13.5, 7.0, 3.5)	2.94 (ddd, 13.5, 7.0, 3.5)
D	4	4.95, 5.01 <sup>a</sup> (dd, 12.0, 5.5)	5.43, 5.51 <sup>a</sup> (dd, 12.0, 5.0)	5.07 (dd, 7.0, 6.0)	5.06 (t, 6.0)
	6	6.25, 6.15 <sup>a</sup> (s)	6.18, 6.06 <sup>a</sup> (s)	6.11 (s)	6.12 (s)
	8	—	—	—	—
E	2'	6.53, 7.01 <sup>a</sup> (d, 2.0)	6.46, 6.90 <sup>a</sup> (s)	7.06 (d, 2.0)	6.68 (s)
	5'	6.65, 6.88 <sup>a</sup> (d, 8.5)	—	6.75 (d, 8.5)	—
F	6'	6.20, 6.99 <sup>a</sup> (dd, 8.5, 2.0)	6.46, 6.90 <sup>a</sup> (s)	6.83 (dd, 8.5, 2.0)	6.68 (s)
	2	4.89, 5.13 <sup>a</sup> (br.s)	4.65, 5.02 <sup>a</sup> (br.s)	4.68 (br.s)	4.60 (br.s)
	3	5.31, 5.56 <sup>a</sup> (m)	5.51, 5.78 <sup>a</sup> (m)	5.62 (m)	5.59 (m)
	4	2.94, 2.98 <sup>a</sup> (m)	3.06, 3.19 <sup>a</sup> (dd, 18.0, 5.0)	3.06 (dd, 18.0, 5.0)	3.06 (dd, 18.0, 5.0)
	4	2.94, 2.98 <sup>a</sup> (m)	3.31, 3.46 <sup>a</sup> (dd, 18.0, 2.5)	3.32 (dd, 18.0, 2.5)	3.33 (d, 18.0, 2.5)
	OMe	3.51 <sup>a</sup> , 3.75 <sup>a</sup> , 3.76, 3.78, 3.79, 3.84 <sup>a</sup> , 3.85 <sup>a</sup> (×2), 3.87, 3.89, 3.90 <sup>a</sup> , 3.91 (all s)	3.27 <sup>a</sup> , 3.36, 3.43 <sup>a</sup> , 3.48 <sup>a</sup> , 3.49 <sup>a</sup> , 3.50, 3.55, 3.61 <sup>a</sup> (×2), 3.62, 3.70 (×2), 3.92, 3.96 <sup>a</sup> (s)	3.41, 3.46, 3.49, 3.53, 3.60, 3.71 (all s)	3.43, 3.47, 3.49, 3.53, 3.68 (×2), 3.94 (s)
	OAc	1.79, 1.91 <sup>a</sup> (s)	1.51, 1.53 <sup>a</sup> (s)	1.63 (s)	1.64 (s)

<sup>a</sup> Signals of minor rotamer.

Table 2  
<sup>1</sup>H NMR peaks ( $\delta_H$ ) for compounds **10**, **12** and **14** at 300 MHz. Splitting patterns and *J*-values (Hz) are given in parentheses

Ring	H	<b>10</b> (C <sub>6</sub> D <sub>6</sub> , 70°C)	<b>12</b> (CDCl <sub>3</sub> , 20°C)	<b>14</b> (C <sub>6</sub> D <sub>6</sub> , 20°C)
A	5	7.02 ( <i>dd</i> , 8.5)	6.61, 6.81 <sup>a</sup> ( <i>d</i> , 8.5)	7.43 ( <i>d</i> , 8.5)
	6	6.56 ( <i>dd</i> , 8.5, 2.5)	6.38, 6.23 <sup>a</sup> ( <i>dd</i> , 8.5, 2.5)	6.70 ( <i>dd</i> , 8.5, 2.5)
	8	6.63 ( <i>d</i> , 2.5)	6.50, 6.15 <sup>a</sup> ( <i>d</i> , 2.5)	6.92 ( <i>d</i> , 2.5)
B	2', 6'	7.41 ( <i>d</i> , 9.0)	7.41, 7.43 <sup>a</sup> ( <i>d</i> , 9.0)	7.30 ( <i>d</i> , 9.0)
	3', 5'	6.85 ( <i>d</i> , 9.0)	6.92, 6.94 <sup>a</sup> ( <i>d</i> , 9.0)	6.77 ( <i>d</i> , 9.0)
C	2	5.67 ( <i>dd</i> , 8.0, 3.5)	5.14, 5.08 <sup>a</sup> ( <i>dd</i> , 12.0, 2.0)	5.30 ( <i>dd</i> , 12.0, 2.0)
	3	2.56 ( <i>ddd</i> , 13.5, 7.0, 6.5)	2.22, 2.06 <sup>a</sup> ( <i>ddd</i> , 13.0, 5.5, 2.0)	2.44 ( <i>dd</i> , 13.0, 5.5, 2.0)
	3	2.80 ( <i>ddd</i> , 13.5, 6.5, 3.5)	2.77, 2.81 <sup>a</sup> ( <i>m</i> )	3.23 ( <i>m</i> )
D	4	5.01 ( <i>t</i> , 6.0)	4.92, 5.00 <sup>a</sup> ( <i>dd</i> , 12.0, 5.5)	4.87 ( <i>dd</i> , 12.0, 5.5)
	6	6.12 ( <i>s</i> )	6.25, 6.13 <sup>a</sup> ( <i>s</i> )	—
	8	—	—	6.38 ( <i>s</i> )
E	2'	6.67 ( <i>s</i> )	6.71, 6.99 <sup>a</sup> ( <i>d</i> , 2.0)	6.64 ( <i>d</i> , 2.0)
	5'	—	6.77, 6.87 <sup>a</sup> ( <i>d</i> , 8.5)	6.78 ( <i>d</i> , 8.5)
F	6'	6.67 ( <i>s</i> )	6.69, 7.01 <sup>a</sup> ( <i>dd</i> , 8.5, 2.0)	6.46 ( <i>dd</i> , 8.5, 2.0)
	2	4.73 ( <i>d</i> , 7.5)	0.44, 5.12 <sup>a</sup> ( <i>br.s</i> )	4.58 ( <i>br.s</i> )
	3	5.58 ( <i>m</i> )	5.40, 5.49 <sup>a</sup> ( <i>m</i> )	5.45 ( <i>m</i> )
	4	2.99 ( <i>dd</i> , 17.0, 7.5)	2.87, 3.04 <sup>a</sup> ( <i>dd</i> , 18.0, 2.0)	2.94 ( <i>dd</i> , 18.0, 4.5)
	4	3.40 ( <i>dd</i> , 17.0, 6.0)	2.98, 3.04 <sup>a</sup> ( <i>dd</i> , 18.0, 5.0)	3.39 ( <i>m</i> )
	OMe	3.45, 3.46, 3.52, 3.53, 3.63 (×2), 3.91 (all <i>s</i> )	3.54 <sup>a</sup> , 3.56, 3.76 <sup>a</sup> , 3.77 <sup>a</sup> , 3.78, 3.83 <sup>a</sup> , 3.84 <sup>a</sup> , 3.85, 3.87, 3.88, 3.90 <sup>a</sup> , 3.91 (all <i>s</i> )	3.31, 3.39, 3.40, 3.43, 3.49, 3.66 (all <i>s</i> )
OAc		1.64 ( <i>s</i> )	1.88, 1.94 <sup>a</sup> ( <i>s</i> )	2.00 ( <i>s</i> )

<sup>a</sup> Signals of minor rotamer.

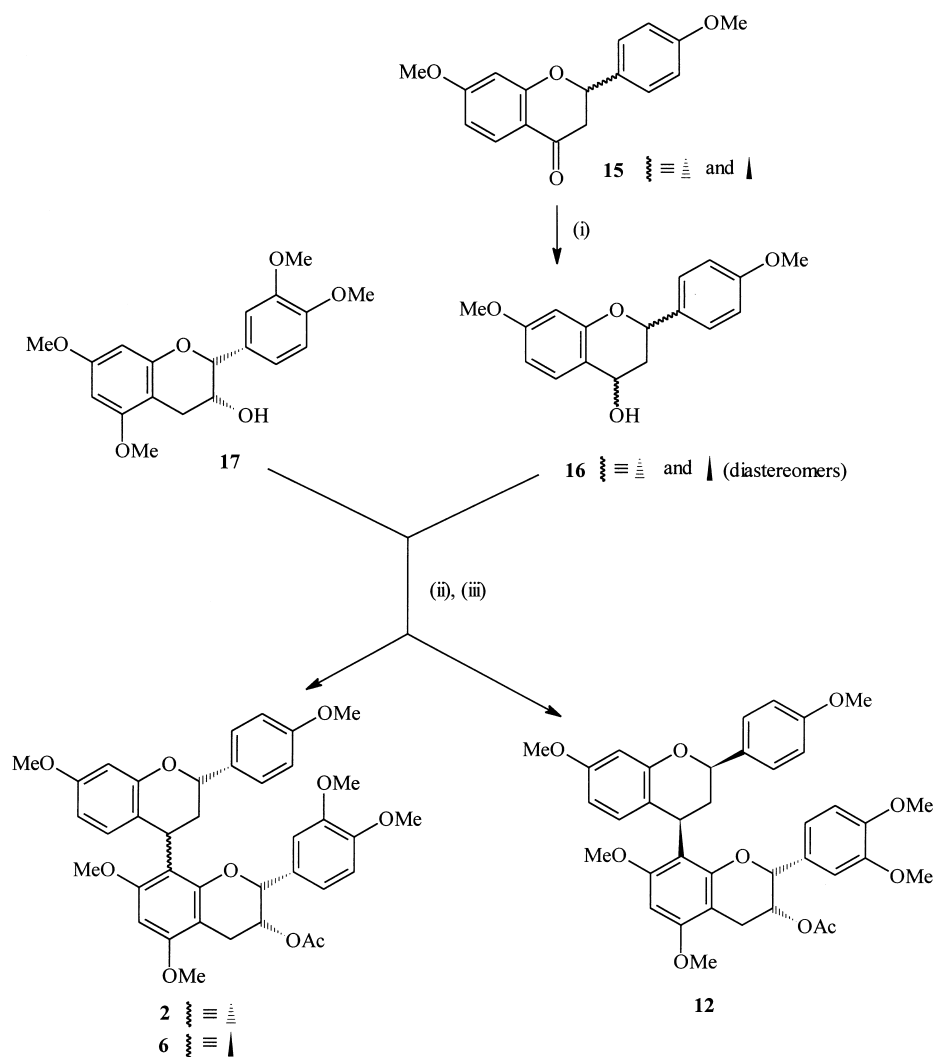
Table 3  
<sup>13</sup>C NMR peaks ( $\delta_c$ ) for compounds **2**, **4**, **6**, **8**, **10** and **12**

Ring	C	<b>2</b> (CDCl <sub>3</sub> , 20°C)	<b>4</b> (C <sub>6</sub> D <sub>6</sub> , 40°C)	<b>6</b> (C <sub>6</sub> D <sub>6</sub> , 70°C)	<b>8</b> (C <sub>6</sub> D <sub>6</sub> , 70°C)	<b>10</b> (C <sub>6</sub> D <sub>6</sub> , 70°C)	<b>12</b> (CDCl <sub>3</sub> , 20°C)
A	5	128.7	128.7	128.7	128.7	128.9	128.1
	6	107.3	107.7	107.7	107.8	107.7	107.9
	8	101.5	101.9	101.9	102.0	101.9	101.5
	2', 6'	127.9	127.5	127.2	127.2	127.8	127.9
B	3', 5'	114.2	114.4	114.4	114.4	114.4	114.2
	2	79.4	78.7	75.8	75.8	75.6	78.8
C	3	35.0	36.5	36.2	36.3	36.7	35.8
	4	32.0	29.1	28.9	29.1	29.2	32.0
D	6	88.6	89.3	90.3	90.3	90.3	88.7
	2'	109.6	105.8	104.3	105.7	105.8	109.9
E	5'	110.8	—	106.7	—	—	110.7
	6'	119.4	105.8	114.2	105.7	105.8	119.2
F	2	77.3	77.8	78.1	78.3	79.7	77.6
	3	68.0	68.9	68.1	68.1	70.0	67.9
	4	26.6	27.0	27.1	27.0	26.0	26.7
	OCH <sub>3</sub>	55.7, 55.8, 56.1, 56.3, 56.4, 56.9	55.0 (×2), 55.2, 56.1, 56.4 (×2), 60.5	55.0, 55.3, 56.2, 56.4, 56.7, 56.9	55.0 (×2), 55.1, 56.1, 56.4 (×2), 60.5	54.9, 55.0, 55.2, 56.1, 56.3 (×2), 60.5	55.5, 55.7 (×2), 56.2, 56.3, 56.8
	CH <sub>3</sub> COO—	170.5	169.5	169.4	169.4	169.5	170.9

region. When taken in conjunction with the presence of an aromatic ABX-system for derivatives **2**, **6**, **12** and **14**, an aromatic  $A_2$ -spin system for derivatives **4**, **8** and **10** and the relevant number of aromatic *O*-methyl and aliphatic *O*-acetyl resonances, the data collectively indicated 5,7,3',4'-tetrahydroxyflavan-3-ol DEF units for compounds **1**, **5**, **11** and **13**, and 5,7,3',4',5'-penta-hydroxyflavan-3-ol units for **3**, **7** and **9**. The dimeric structures were supported by FAB-MS data which indicated molecular ions  $m/z$  656 ( $C_{38}H_{40}O_{10}$ ) and 686 ( $C_{39}H_{42}O_{11}$ ), for derivatives **2**, **6**, **12** and **14**, and **4**, **8** and **10**, respectively.

COSY experiments indicated coupling of 2-H(C) with 2',6'-H(B) and 2-H(F) with 2'- and 6'-H(E) for derivatives **2**, **4**, **6**, **8**, **10**, **12** and **14**. The same experiment revealed  $^4J_{HH}$  coupling between 4-H(C) and 5-H(A) which defined the ABC- and DEF-units for all the compounds. Phase sensitive NOESY experiments of derivatives **2**, **4**, **12** and **14** showed association

between 2- and 4-H(C), indicating a 2,4-*cis* relative configuration of the C-ring. The conspicuous absence of any NOE association between 2- and 4-H(C) in compounds **6**, **8** and **10** strongly suggested 2,4-*trans* relative configuration of their C-rings. These allocations of relative configuration were corroborated by the multiplicity and coupling constants of 2- and 4-H(C) for the 2,4-*cis* configuration, e.g. **2**, where both 2- and 4-H(C) resonate as double doublets ( $^3J_{2,3} = 2.0$ , 12.0 Hz;  $^3J_{3,4} = 5.5$ , 12.0 Hz). Analogues with a 2,4-*trans* relative configuration, e.g. **6**, show a double doublet for both 2- ( $^3J_{2,3} = 3.5$ , ca. 7.0 Hz) and 4-H(C) ( $^3J_{3,4} = 6.0$ , 7.0 Hz) and in some instances a triplet for 4-H(C) ( $^3J_{3,4} = 6.0$  Hz) (Coetzee et al., 1999). The chemical shifts of the C-2 (C-ring) resonances in the  $^{13}C$  NMR spectra of derivatives **2**, **4**, **6**, **8**, **10** and **12** fully supported these relative configurations. Those compounds with a 2,4-*trans* configuration (**6**, **8** and **10**) possess shielded 2-C(C) signals (ca. 4 ppm) com-



Scheme 1. Reagents and conditions: (i)  $NaBH_4$ , EtOH; (ii) **17** then,  $TiCl_4/CH_2Cl_2$ ; (iii)  $Ac_2O$ , pyridine.

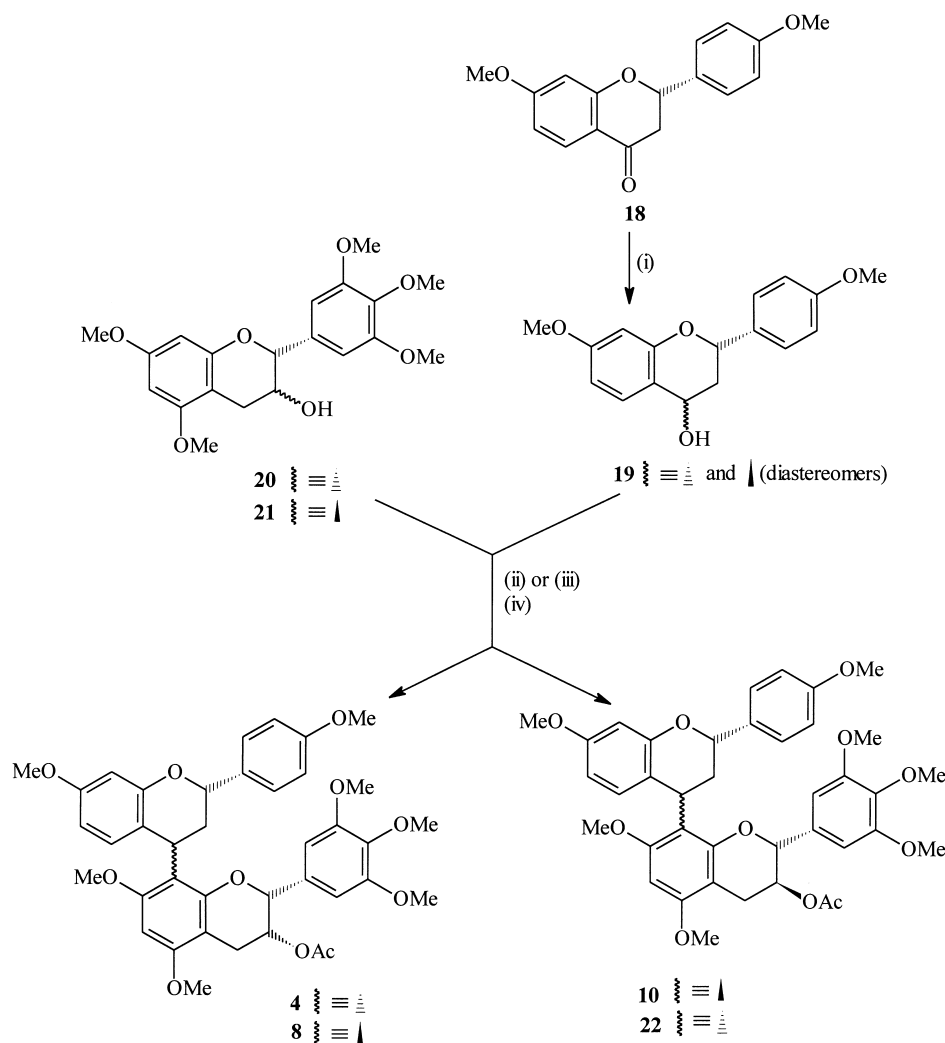
pared to the chemical shifts of these carbons in derivatives with a 2,4-*cis* configuration (Table 3) as a result of the  $\gamma$ -*gauche* effect (Fletcher, Porter, Haslam & Gupta, 1977) operating in the former group of compounds. The same shielding phenomena are also evident from the results of Hatano et al. (1997). The chemical shifts of the hydrogen-bearing carbons in compounds **2**, **4**, **6**, **8**, **10** and **12** are collated in Table 3. Sample size precluded the collection of  $^{13}\text{C}$  NMR data for the 4,6-linked analogue **14**.

Bonding positions to the D-ring were established for all the derivatives by the familiar NOE associations exhibited by 'residual' D-ring protons with methoxyl(s) i.e. the D-ring singlet showing strong association with the 5- and 7-methoxy groups of the D-ring in derivatives **2**, **4**, **6**, **8**, **10** and **12** and selectively with 7-OMe in **14** (Young, Brandt, Young, Ferreira & Roux, 1986).

The CD spectra of compounds **2**, **4** and **14** exhibited high amplitude negative Cotton effects at ca. 240 nm,

indicating a 4 $\alpha$ -flavanyl C-ring substituent (De Angelis & Wildman, 1969; Van der Westhuizen, Ferreira & Roux, 1981). Derivatives **6**, **8**, **10** and **12** showed positive Cotton effects near 240 nm which is reminiscent of a 4 $\beta$  C-ring substituent. When taken in conjunction with the relative 2,4-*cis* configuration of derivatives **2**, **4**, **12** and **14**, the negative Cotton effects for **2**, **4** and **14** indicated their 2*S*,4*S* absolute configuration and the positive Cotton effect for **12** its 2*R*,4*R* absolute stereochemistry. The positive Cotton effects similarly supported the 2*S*,4*R* absolute configuration of derivatives **6**, **8** and **10** with their 2,4-*trans* relative configurations.

Derivatives **2**, **4**, **6**, **8**, **12** and **14** showed 2-H(F) as a broad singlet [ $^3J_{2,3(\text{F})}$  ca 1.0 Hz] while **10** exhibited a coupling constant of 7.5 Hz. This indicated a 2,3-*cis* relative configuration of the F-rings (Steinberg, Burger, Malan, Cronjé, Young & Ferreira, 1990) of the former group and 2,3-*trans* for derivative **10**. Thus, we opted to synthesize these derivatives in order to estab-



Scheme 2. Reagents and conditions: (i) NaBH<sub>4</sub>, EtOH; (ii) **20** then, TiCl<sub>4</sub>/CH<sub>2</sub>Cl<sub>2</sub>; (iii) **21** then, TiCl<sub>4</sub>/CH<sub>2</sub>Cl<sub>2</sub>; (iv) Ac<sub>2</sub>O/pyridine.

lish the absolute configuration (Nonaka, Miwa & Nishioka, 1982) and constitution of the natural products **1**, **3**, **5**, **7**, **9**, **11** and **13**.

Racemic 4',7-dimethoxyflavanone **15** was reduced by sodium borohydride (Hatano et al., 1997; Malan et al., 1997) to give the diastereomeric mixture of the flavan-4-ol **16** in 96% yield (Scheme 1). Treatment of this mixture with optically pure tetra-*O*-methylepicatechin **17** in dichloromethane containing titanium tetrachloride (TiCl<sub>4</sub>) as Lewis acid (Kawamoto, Nakatsubo & Murakami, 1991) afforded a mixture of compounds, which were purified by PLC to give three procassinidin-type dimers in fair yields. Acetylation yielded the permethylaryl ether acetates **2**, **6** and **12** which had identical <sup>1</sup>H NMR and CD spectral data when compared to derivatives obtained from the natural compounds **1**, **5** and **11**. We could not detect any of the expected permethyl aryl ether of the (4 $\alpha$   $\rightarrow$  6)-analogue **13**, an observation which is supported by previous reports of absence (Hatano et al., 1997; Coetzee et al., 1999) or formation in low yields (Young, Cronjé, Botes, Ferreira & Roux, 1985) of the (4  $\rightarrow$  6)-linked dimers. Compounds **1**, **5** and **11** were hence unequivocally identified as cassiaflavan-(4 $\alpha$   $\rightarrow$  8)-epicatechin, cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epicatechin, and *ent*-cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epicatechin, respectively, and **13** tentatively as cassiaflavan-(4 $\alpha$   $\rightarrow$  6)-epicatechin.

In contrast to the above and our previous syntheses of dimeric proanthocyanidins with flavan constituent units (Malan et al., 1997; Coetzee et al., 1999) where racemic flavanones were utilized as source of the flavan molecular backbone, compounds **4** and **8** were synthesized from optically pure (2*S*)-di-*O*-methylliquiritigenin **18** and penta-*O*-methylepigallocatechin **20** (Scheme 2). Initial NaBH<sub>4</sub>-reduction of **18** gave the intermediate 4',7-dimethoxyflavan-4-ol **19** (diastereomers, quantitative yield) which was subsequently treated with penta-*O*-methylepigallocatechin **20** in dichloromethane containing TiCl<sub>4</sub>. The <sup>1</sup>H NMR and CD spectra of the *O*-acetyl derivatives of the synthetic dimer derivatives **4** and **8** corresponded with those of the same derivatives of the procassinidins from *C. petersiana*. The structures and absolute stereochemistry of the natural products **3** and **7** could thus be defined as cassiaflavan-(4 $\alpha$   $\rightarrow$  8)-epigallocatechin and cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epigallocatechin, respectively.

A similar procedure (Scheme 2) was employed to synthesize **10** by using (2*S*)-4',7-di-*O*-methylliquiritigenin **18** and penta-*O*-methylgalloocatechin **21**. Subsequent acetylation afforded the cassiaflavan-(4 $\beta$   $\rightarrow$  8)-galloocatechin and cassiaflavan-(4 $\alpha$   $\rightarrow$  8)-galloocatechin derivatives **10** and **22**. The lower yields of **4** and **22** compared to those of **8** and **10** indicated that the stereochemical course of coupling is directed by the C-2 *eq.* aryl group.

The *ent*-cassiaflavan-(4 $\beta$   $\rightarrow$  8)-epicatechin **11** is the

first compound in the class of flavan  $\rightarrow$  flavan-3-ol proanthocyanidins possessing a flavan top unit with 2*R* absolute configuration, hence presumably also indicating the involvement of a 2*R*-flavanone in their biosynthesis (Stich & Forkmann, 1988).

### 3. Experimental

<sup>1</sup>H NMR spectra were recorded at 300 MHz for solutions in CDCl<sub>3</sub> or C<sub>6</sub>D<sub>6</sub>, with TMS as int. standard. FAB-MS were recorded on a VG 70-70E instrument with a VG 11-250J data system and an iontech saddle-field FAB gun. CD data were obtained in MeOH. TLC was performed on precoated Merck plastic sheets (silica gel 60 PF<sub>254</sub> 0.25 mm) and the plates were sprayed with H<sub>2</sub>SO<sub>4</sub>-HCHO (40:1, vol/vol) after development. Prep. TLC plates, Kieselgel PF<sub>254</sub> (1.0 mm) were air dried and used without prior activation. Compounds were recovered from the absorbent with Me<sub>2</sub>CO. CC was on Sephadex LH-20 in EtOH. Methylations were performed with an excess of CH<sub>2</sub>N<sub>2</sub> in MeOH-Et<sub>2</sub>O over a period of 48 h at -15°C, while acetylations were in Ac<sub>2</sub>O-pyridine at ambient temperature. Evaporations were done under reduced pressure at ambient temperatures in a rotary evaporator, and freeze drying of aqueous solutions on a Virtis 12SL freezemobile.

#### 3.1. General procedure for the synthesis of procassinidin derivatives

The reduction of flavanones **15** and **18** was done according to standard procedures (Hatano et al., 1997; Malan et al., 1997). To a dry solution of 4',7-dimethoxyflavan-4-ol **16** (90.0 mg) in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) was added the permethylaryl ether **17** of epicatechin (296 mg) and TiCl<sub>4</sub> (0.04 ml, 1.2–1.4 equivalent). The mixture was stirred at 0°C under N<sub>2</sub> for 60 min and the temperature was allowed to rise to 40°C for a further 6 h. An excess of cold H<sub>2</sub>O (40 ml) was added and the mixture was extracted with Et<sub>2</sub>O (3  $\times$  20 ml). After drying (Na<sub>2</sub>SO<sub>4</sub>) the Et<sub>2</sub>O was removed under vacuum and the mixture was resolved by prep. TLC in hexane-C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO (5:2:3) to give three bands at *R*<sub>f</sub> 0.53, 0.40 and 0.33. After acetylation of the *R*<sub>f</sub> 0.33 band (42 mg) it was further separated in hexane-C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO (5:4:1) to give four bands at *R*<sub>f</sub> 0.57 (5.5 mg), 0.37 (5.7 mg), 0.29 (8.2 mg) and 0.22 (14.1 mg). The *R*<sub>f</sub> 0.57 band yielded starting material **17**. The *R*<sub>f</sub> 0.37 band yielded 4',7-di-*O*-methyl-*ent*-cassiaflavan-(4 $\beta$   $\rightarrow$  8)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetyl-epicatechin **12** as a light-brown amorphous solid. The *R*<sub>f</sub> 0.29 band gave 4',7-di-*O*-methylcassiaflavan-(4 $\beta$   $\rightarrow$  8)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetylepicatechin **6** and



the  $R_f$  0.22 band 4',7-di-*O*-methylcassiaflavan-(4 $\alpha$   $\rightarrow$  8)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetylepicatechin **2**.

(2*S*)-4',7-dimethoxyflavan-4-ol **19** (90.0 mg) was dissolved in dry  $\text{CH}_2\text{Cl}_2$  (20 ml), the  $\text{TiCl}_4$  (0.04 ml) added to give a dark purple solution after which the epigallocatechin or galocatechin permethylaryl ethers **20/21** (290 mg) was added during separate experiments. The mixture was treated and worked up as above, and the residue was separated by prep. TLC in  $\text{C}_6\text{H}_6$ – $\text{Me}_2\text{CO}$  (8:2) to give two bands at  $R_f$  0.39 (40 mg) and 0.26 (17.3 mg). The  $R_f$  0.39 band yielded starting material **20**. Acetylation of the  $R_f$  0.26 band followed by prep. TLC in hexane– $\text{C}_6\text{H}_6$ – $\text{Me}_2\text{CO}$  (5:4:1,  $\times 4$ ) gave two bands which comprised 4',7-di-*O*-methylcassiaflavan-(4 $\beta$   $\rightarrow$  8)-epigallocatechin penta-*O*-methylether acetate **8** and 4',7-di-*O*-methylcassiaflavan-(4 $\alpha$   $\rightarrow$  8)-epigallocatechin-penta-*O*-methylether acetate **4** at  $R_f$  0.29 (4.0 mg) and 0.39 (10 mg), respectively. Derivatives **4** and **8** showed identical NMR and CD spectral data as the same derivatives of the corresponding natural products **3** and **7**. The yield of compound **10** and very likely also that of **22** during a similar procedure using the galocatechin permethylaryl ether **21** were low and after initial prep. TLC separation in hexane– $\text{C}_6\text{H}_6$ – $\text{Me}_2\text{CO}$  (5:2:3) only 6 mg of an inseparable mixture of **10** and **22** was obtained. However, it was possible to identify compound **10** by using the  $^1\text{H}$  NMR spectral data of the natural compound to fingerprint the synthetic mixture.

### 3.2. Isolation of phenolic compounds

Milled bark (6.3 kg) of *C. petersiana* was repeatedly extracted with  $\text{Me}_2\text{CO}$  ( $3 \times 7.5$  l) for 48 h periods at 25°C. The  $\text{Me}_2\text{CO}$  was removed under vacuum at 35°C and the residue dissolved in  $\text{H}_2\text{O}$  and freeze dried to give a brown powder (370 g).

The extract (two batches of 25 g each) was subjected to CC on Sephadex LH-20 in EtOH (6  $\times$  180 cm) column, 0.5 ml/min flow rate, 32 min fractions) to give the following fractions: B<sub>1</sub> (tubes 225–264, 82 mg), B<sub>2</sub> (265–279, 62 mg), B<sub>3</sub> (280–285, 10 mg), B<sub>4</sub> (286–319, 70 mg), B<sub>5</sub> (320–354, 110 mg), B<sub>6</sub> (355–364, 29 mg), B<sub>7</sub> (365–399, 260 mg) and B<sub>8</sub> (400–414, 117 mg). A portion (200 mg) of fraction B<sub>8</sub> was methylated and the mixture was separated by prep. TLC in  $\text{C}_6\text{H}_6$ – $\text{Me}_2\text{CO}$  (9:1,  $\times 3$ ) to give four bands at  $R_f$  0.64 (B<sub>8,1</sub>, 33 mg), 0.40 (B<sub>8,2</sub>, 24.5 mg), 0.37 (B<sub>8,3</sub>, 12.5 mg) and 0.13 (B<sub>8,4</sub>, 15.6 mg). All four fractions were separately acetylated and purified by prep. TLC in  $\text{C}_6\text{H}_6$ – $\text{Me}_2\text{CO}$  (24:1,  $\times 3$ ) to yield the following compounds.

4',7-Di-*O*-methylcassiaflavan-(4 $\alpha$   $\rightarrow$  8)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetylepicatechin **2** ( $R_f$  0.64, 15 mg) from B<sub>8,1</sub> as a *white amorphous solid*. (Found:  $\text{M}^+$ , 656.2623.  $\text{C}_{38}\text{H}_{40}\text{O}_{10}$  requires  $\text{M}^+$ , 656.2620;  $\delta_{\text{H}}$  (Table 1);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 20°C):  $\delta$  (major rotamer)

21.2 [ $\text{CH}_3\text{COO}-$ ], 26.6 [4-C(F)], 32.0 [4-C(C)], 35.0 [3-C(C)], 55.7 [ $-\text{OCH}_3$ ], 55.8 [ $-\text{OCH}_3$ ], 56.1 [ $-\text{OCH}_3$ ], 56.3 [ $-\text{OCH}_3$ ], 56.4 [ $-\text{OCH}_3$ ], 56.9 [ $-\text{OCH}_3$ ], 68.0 [3-C(F)], 77.3 [2-C(F)], 79.4 [2-C(C)], 88.6 [6-C(D)], 101.5 [8-C(A)], 107.3 [6-C(A)], 109.6 [2'-C(E)], 110.8 [5'-C(E)], 114.2 ( $\times 2$ ) [3',5'-C(B)], 119.1, 119.4 [6'-C(E)], 119.5, 127.9 ( $\times 2$ ) [2',6'-C(B)], 128.2, 128.5, 128.7 [5-C(A)], 130.6, 134.4, 148.8, 156.3 [ $\text{Ar-OCH}_3$ ], 157.7 [ $\text{Ar-OCH}_3$ ], 158.0 [ $\text{Ar-OCH}_3$ ], 158.6 [ $\text{Ar-OCH}_3$ ], 158.9 [ $\text{Ar-OCH}_3$ ], 159.6 [ $\text{Ar-OCH}_3$ ], 170.5 [ $\text{CH}_3\text{COO}-$ ]; CD [ $\theta$ ]<sub>276.1</sub> 2704, [ $\theta$ ]<sub>243.9</sub> –16710, [ $\theta$ ]<sub>234.7</sub> 1016.

4',7-Di-*O*-methyl-ent-cassiaflavan-(4 $\beta$   $\rightarrow$  8)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetylepicatechin **12** ( $R_f$  0.61, 4.5 mg) from B<sub>8,1</sub> as a *white-yellowish amorphous solid*. (Found:  $\text{M}^+$ , 656.2621.  $\text{C}_{38}\text{H}_{40}\text{O}_{10}$  requires  $\text{M}^+$ , 656.2620.  $\delta_{\text{H}}$  (Table 2);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 20°C):  $\delta$  (major rotamer) 21.5 [ $\text{CH}_3\text{COO}-$ ], 26.7 [4-C(F)], 32.0 [4-C(C)], 35.8 [3-C(C)], 55.5 [ $-\text{OCH}_3$ ], 55.7 [ $-\text{OCH}_3$ ], 55.7 [ $-\text{OCH}_3$ ], 56.2 [ $-\text{OCH}_3$ ], 56.3 [ $-\text{OCH}_3$ ], 56.8 [ $-\text{OCH}_3$ ], 67.9 [3-C(F)], 77.6 [2-C(F)], 78.8 [2-C(C)], 88.7 [6-C(D)], 101.5 [8-C(A)], 107.9 [6-C(A)], 109.9 [2'-C(E)], 110.7 [5'-C(E)], 114.2 ( $\times 2$ ) [3',5'-C(B)], 119.0, 119.2 [6'-C(E)], 119.5, 127.86, 127.9 ( $\times 2$ ) [2',6'-C(B)], 128.1 [5-C(A)], 128.5, 130.4, 134.4, 148.7, 156.3 [ $\text{Ar-OCH}_3$ ], 157.5 [ $\text{Ar-OCH}_3$ ], 157.8 [ $\text{Ar-OCH}_3$ ], 158.6 [ $\text{Ar-OCH}_3$ ], 158.7 [ $\text{Ar-OCH}_3$ ], 159.6 [ $\text{Ar-OCH}_3$ ], 170.9 [ $\text{CH}_3\text{COO}-$ ]; CD [ $\theta$ ]<sub>286.5</sub> 4178, [ $\theta$ ]<sub>273.3</sub> –4770, [ $\theta$ ]<sub>246.5</sub> 6630.

4',7-Di-*O*-methylcassiaflavan-(4 $\alpha$   $\rightarrow$  6)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetylepicatechin **14** ( $R_f$  0.36, 8.3 mg) from B<sub>8,2</sub> as a *white amorphous solid*. (Found:  $\text{M}^+$ , 656.2620.  $\text{C}_{38}\text{H}_{40}\text{O}_{10}$  requires  $\text{M}^+$ , 656.2620.  $\delta_{\text{H}}$  (Table 2); CD [ $\theta$ ]<sub>286.8</sub> 4591, [ $\theta$ ]<sub>244.1</sub> –15230, [ $\theta$ ]<sub>230.3</sub> 3321.

4',7-Di-*O*-methylcassiaflavan-(4 $\beta$   $\rightarrow$  8)-3',4',5,7-tetra-*O*-methyl-3-*O*-acetylepicatechin **6** ( $R_f$  0.47, 2.5 mg) from B<sub>8,3</sub> as a *white amorphous solid*. (Found:  $\text{M}^+$ , 656.2621.  $\text{C}_{38}\text{H}_{40}\text{O}_{10}$  requires  $\text{M}^+$ , 656.2620.  $\delta_{\text{H}}$  (Table 1);  $^{13}\text{C}$  NMR ( $\text{C}_6\text{D}_6$ , 70°C):  $\delta$  20.5 [ $\text{CH}_3\text{COO}-$ ], 27.1 [4-C(F)], 28.9 [4-C(C)], 36.2 [3-C(C)], 55.0 [ $-\text{OCH}_3$ ], 55.3 [ $-\text{OCH}_3$ ], 56.2 [ $-\text{OCH}_3$ ], 56.4 [ $-\text{OCH}_3$ ], 56.7 [ $-\text{OCH}_3$ ], 56.9 [ $-\text{OCH}_3$ ], 68.1 [3-C(F)], 75.8 [2-C(C)], 78.1 [2-C(F)], 90.3 [6-C(D)], 101.9 [8-C(A)], 102.6, 104.3 [2'-C(E)], 106.7 [5'-C(E)], 107.7 [6-C(A)], 111.4, 114.2 [6'-C(E)], 114.4 ( $\times 2$ ) [3',5'-C(B)], 118.4, 119.4, 127.2 ( $\times 2$ ) [2',6'-C(B)], 128.7 [5-C(A)], 133.4, 136.8, 140.2, 154.6 [ $\text{Ar-OCH}_3$ ], 156.6 [ $\text{Ar-OCH}_3$ ], 157.9 [ $\text{Ar-OCH}_3$ ], 158.4 [ $\text{Ar-OCH}_3$ ], 159.8 [ $\text{Ar-OCH}_3$ ], 159.8 [ $\text{Ar-OCH}_3$ ], 169.4 [ $\text{CH}_3\text{COO}-$ ]; CD [ $\theta$ ]<sub>275.8</sub> –7354, [ $\theta$ ]<sub>244.0</sub> 15780, [ $\theta$ ]<sub>230.6</sub> 1358.

Fraction B<sub>8,4</sub> ( $R_f$  0.29, 8.5 mg) comprised the heptamethyl ether diacetate of fisetinidol-(4 $\beta$   $\rightarrow$  8)-epicatechin (Steynberg et al., 1990).

A portion of fraction B<sub>7</sub> (200 mg) was methylated and the mixture was separated by PLC in  $\text{C}_6\text{H}_6$ – $\text{Me}_2\text{CO}$  (8:2), to give a band at  $R_f$  0.67 (47 mg) which was subsequently acetylated. Further separation in

hexane–Me<sub>2</sub>CO–EtOAc (12:5:3, vol/vol) afforded two bands at *R<sub>f</sub>* 0.51 (5.2 mg) and 0.45 (6.0 mg) comprising compounds **10** and **8**, respectively. 4',7-Di-*O*-methylcassiaflavan-(4β → 8)-3',4',5',5,7-penta-*O*-methyl-3-*O*-acetylalloocatechin **10** was obtained as *colorless amorphous flakes*. (Found: M<sup>+</sup>, 686.2723. C<sub>39</sub>H<sub>42</sub>O<sub>11</sub> requires M, 686.2724; δ<sub>H</sub> (Table 2); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 70°C): δ 20.5 [CH<sub>3</sub>COO–], 26.0 [4-C(F)], 29.2 [4-C(C)], 36.7 [3-C(C)], 54.9 [–OCH<sub>3</sub>], 55.0 [–OC<sub>2</sub>H<sub>3</sub>], 55.2 [–OCH<sub>3</sub>], 56.1 [–OCH<sub>3</sub>], 56.3 (×2) [–OCH<sub>3</sub>], 60.5 [–OCH<sub>3</sub>], 70.0 [3-C(F)], 75.6 [2-C(C)], 79.7 [2-C(F)], 90.3 [6-C(D)], 101.9 [8-C(A)], 103.7, 105.8 (×2) [2',6'-C(E)], 107.7 [6-C(A)], 110.2, 114.4 (×2) [3',5'-C(B)], 114.9, 119.4, 127.8 (×2) [2',6'-C(B)], 128.9 [5-C(A)], 133.4, 135.4, 140.1, 154.1 [Ar-OCH<sub>3</sub>], 154.2 [Ar-OCH<sub>3</sub>], 156.5 [Ar-OCH<sub>3</sub>], 157.9 [Ar-OCH<sub>3</sub>], 158.5 [Ar-OCH<sub>3</sub>], 159.7 [Ar-OCH<sub>3</sub>], 160.0 [Ar-OCH<sub>3</sub>], 169.5 [CH<sub>3</sub>COO–]; CD [θ]<sub>275.8</sub> –4174, [θ]<sub>238.9</sub> 14440 and [θ]<sub>227.2</sub> –313. 4',7-Di-*O*-methylcassiaflavan-(4β → 8)-3',4',5',5,7-penta-*O*-methyl-3-*O*-acetylalloocatechin **8** was isolated as *colorless amorphous platelets*. (Found: M<sup>+</sup>, 686.2722. C<sub>39</sub>H<sub>42</sub>O<sub>11</sub> requires M, 686.2724; δ<sub>H</sub> (Table 1); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 70°C): (20.5 [CH<sub>3</sub>COO–], 27.0 [4-C(F)], 29.1 [4-C(C)], 36.3 [3-C(C)], 55.0 (×2) [–OCH<sub>3</sub>], 55.1 [–OCH<sub>3</sub>], 56.1 [–OCH<sub>3</sub>], 56.4 (×2) [–OC<sub>2</sub>H<sub>3</sub>], 60.5 [–OCH<sub>3</sub>], 68.1 [3-C(F)], 75.8 [2-C(C)], 78.3 [2-C(F)], 90.3 [6-C(D)], 102.0 [8-C(A)], 103.6, 105.7 (×2) [2',6'-C(E)], 107.8 [6-C(A)], 110.2, 114.4 (×2) [3',5'-C(B)], 114.8, 119.3, 127.2 (×2) [2',6'-C(B)], 128.7 [5-C(A)], 133.4, 135.3, 140.1, 154.1 [Ar-OCH<sub>3</sub>], 154.6 [Ar-OCH<sub>3</sub>], 156.4 [Ar-OCH<sub>3</sub>], 157.9 [Ar-OCH<sub>3</sub>], 158.4 [Ar-OCH<sub>3</sub>], 159.7 [Ar-OCH<sub>3</sub>], 159.8 [Ar-OCH<sub>3</sub>], 169.4 [CH<sub>3</sub>COO–]; CD [θ]<sub>275.0</sub> –1254, [θ]<sub>248.1</sub> 12180 and [θ]<sub>239.7</sub> –804.

A portion (200 mg) of fraction C<sub>4</sub> (Coetzee et al., 1999) was methylated and subjected to the same procedure of separation and purification as for compounds **8** and **10**. The band at *R<sub>f</sub>* 0.64 (22 mg) obtained after the first separation was acetylated and purified to yield 4',7-Di-*O*-methylcassiaflavan-(4α → 8)-3',4',5',5,7-penta-*O*-methyl-3-*O*-acetylalloocatechin **4** (*R<sub>f</sub>* 0.54, 6.0 mg) as a *white amorphous solid*. (Found: M<sup>+</sup>, 686.2724. C<sub>39</sub>H<sub>42</sub>O<sub>11</sub> requires M, 686.2724; δ<sub>H</sub> (Table 1); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 40°C): δ (major rotamer) 20.4 [CH<sub>3</sub>COO–], 27.0 [4-C(F)], 29.1 [4-C(C)], 36.5 [3-C(C)], 55.0 (×2) [–OCH<sub>3</sub>], 55.2 [–OCH<sub>3</sub>], 56.1 [–OC<sub>2</sub>H<sub>3</sub>], 56.4 (×2) [–OCH<sub>3</sub>], 60.5 [–OCH<sub>3</sub>], 68.9 [3-C(F)], 77.8 [2-C(F)], 78.7 [2-C(C)], 89.3 [6-C(D)], 101.9 [8-C(A)], 103.7, 105.8 (×2) [2',6'-C(E)], 107.7 [6-C(A)], 110.2, 114.4 (×2) [3',5'-C(B)], 114.8, 119.4, 127.5 (×2) [2',6'-C(B)], 128.7 [5-C(A)], 133.4, 135.3, 140.0, 154.1

[Ar-OCH<sub>3</sub>], 154.3 [Ar-OCH<sub>3</sub>], 156.4 [Ar-OCH<sub>3</sub>], 157.8 [Ar-OCH<sub>3</sub>], 158.5 [Ar-OCH<sub>3</sub>], 159.7 [Ar-OCH<sub>3</sub>], 159.8 [Ar-OCH<sub>3</sub>], 169.5 [CH<sub>3</sub>COO–]; CD [θ]<sub>275.5</sub> 2535, [θ]<sub>243.7</sub> –11860 and [θ]<sub>222.9</sub> 2205.

## Acknowledgements

Financial support by the Foundation for Research Development, Pretoria and the 'Sentrale Navorsingsfonds' of the UOFS, is gratefully acknowledged. This work was supported in part by the United States Department of Agriculture, Agricultural Research Service Specific Cooperative Agreement No. 58-6408-7-012. We thank Mrs. L Davies from Skukuza who identified *Cassia petersiana* after it was collected in the vicinity of Hazyview, Mpumalanga. A voucher specimen is being kept in the Chemistry Department at UOFS.

## References

- Coetzee, J., Mciteka, L., Malan, E., & Ferreira, D. (1999). *Phytochemistry*, 52, 737–743.
- De Angelis, G. G., & Wildman, W. C. (1969). *Tetrahedron*, 25(20), 5099–6012.
- Fletcher, A. C., Porter, L. J., Haslam, E., & Gupta, R. K. (1977). *J. Chem. Soc., Perkin Trans. 1*, 1628–1637.
- Hatano, T., Yamashita, A., Hashimoto, T., Ito, H., Kubo, N., Yoshiyama, M., Shimura, S., Ito, Y., Okuda, T., & Yoshida, T. (1997). *Phytochemistry*, 46(5), 893–900.
- Kawamoto, H., Nakatsubo, F., & Murakami, K. (1991). *Mokuzai Gakkaishi*, 37(5), 488–493.
- Malan, E., Sireeparsad, A., Swinny, E., & Ferreira, D. (1997). *Phytochemistry*, 44(3), 529–531.
- Morimoto, S., Nonaka, G., Chen, R., & Nishioka, I. (1988). *Chem. Pharm. Bull.*, 36(1), 39–47.
- Nonaka, G., Miwa, N., & Nishioka, I. (1982). *Phytochemistry*, 21(2), 429–432.
- Palgrave, K. C. (1983). In E. J. Moll, *Trees of southern Africa* (p. 288). Cape Town: C. Struik Publishers.
- Porter, L. J. (1988). In J. B. Harborne, *The flavonoids — advances in research since 1980* (pp. 21–62). London: Chapman & Hall.
- Porter, L. J. (1994). In J. B. Harborne, *The flavonoids — advances in research since 1986* (pp. 23–55). London: Chapman & Hall.
- Steynberg, J. P., Burger, J. F. W., Malan, J. C. S., Cronjé, A., Young, D. A., & Ferreira, D. (1990). *Phytochemistry*, 29(1), 275–277.
- Stich, K., & Forkmann, G. (1988). *Phytochemistry*, 27(3), 785–789.
- Van der Westhuizen, J. H., Ferreira, D., & Roux, D. G. (1981). *J. Chem. Soc., Perkin Trans. 1*, 1220–1226.
- Young, D. A., Cronjé, A., Botes, A. L., Ferreira, D., & Roux, D. G. (1985). *J. Chem. Soc., Perkin Trans. 1*, 2521–2527.
- Young, D. A., Brandt, E. V., Young, E., Ferreira, D., & Roux, D. G. (1986). *J. Chem. Soc., Perkin Trans. 1*, 1737–1749.