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### Cyclopentene dialdehydes from Tabebuia impetiginosa

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#### Abstract

The isolation of two cyclopentene dialdehydes, 2-formyl-5-(4'-methoxybenzoyloxy)-3-methyl-2-cyclopentene-1-acetaldehyde and 2-formyl-5-(3',4'-dimethoxybenzoyloxy)-3-methyl-2-cyclopentene-1-acetaldehyde, from the bark of *Tabebuia impetiginosa* is reported. The structures were established by analysis of spectroscopic data. These compounds showed anti-inflammatory activity. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: Tabebuia impetiginosa; Bignoniaceae; Cyclopentene dialdehyde; Anti-inflammatory activity

#### 1. Introduction

The stem bark of the South American tree *Tabebuia impetiginosa* Mart. *ex* DC (Bignoniaceae), which is a source of furanonaphthoquinones, has been used in North and South America for many years as an anticancer, antifungal, antibacterial, and anti-inflammatory drug (Abbott, Hartwell, Leiter, Perdue & Schepartz, 1967; Hartwell, 1968; Zani, de Oliveira & de Oliveria, 1991). An investigation of the constituents of *T. impetiginosa* led us to isolate two new cyclopentene dialdehydes together with known furanonaphthoquinones and benzoic acid and benzaldehyde derivatives (Oliveira, Raslan, de Oliveira & Maia, 1993; Wagner, Kreher, Lotter, Hamburger & Cordell, 1989).

We used the nitro blue tetrazolium chloride (NBT) reduction system to determine the anti-inflammatory effects of two cyclopentene dialdehydes in activated human granular white blood cells (WBC, including neutrophils) (Christman, Holden & Blackwell, 1995). This paper describes the structure elucidation and anti-inflammatory activity of the new compounds.

#### 2. Results

The methanol extract of T. impetiginosa was dissolved in water and sequentially partitioned with n-hexane, chloroform, dichloroethane, and ethyl acetate. Silica gel column chromatography of the n-hexane and chloroform soluble layers yielded  $\beta$ -sitosterol, stigmasterol, 4'-methoxybenzyl 4-methoxybenzoate, 9-hydroxy-3-methylnaphtho[2,3-b]pyran-2,5,10-trione, (-)-3,4-dihydro-6,8-dihydroxy-3-methylisocoumarin, and seven known furanonaphthoquinones. The residue from the dichloroethane soluble layer was chromatographed on silica gel to afford compounds  $\mathbf{1}$  and  $\mathbf{2}$ .

Compound 1,  $C_{17}H_{18}O_5$ , showed IR absorptions due to an ester (1707 cm<sup>-1</sup>), a conjugated aldehyde (1664 cm<sup>-1</sup>), simple aldehyde (1712 cm<sup>-1</sup>), and an aromatic ring (1512, 1607 cm<sup>-1</sup>). Its <sup>1</sup>H-NMR spectrum indicated the presence of a para-substituted aromatic ring on the basis of chemical shift values of the AA'BB' system at  $\delta$  6.90 (d, J=9 Hz) and  $\delta$  7.95 (d, J=9 Hz). A methoxy signal was observed at  $\delta$  3.85 (s), along with two aldehyde protons at  $\delta$  9.79 (t, J=1.5 Hz) and  $\delta$  10.00 (s), and a methyl group at  $\delta$  2.22 (s). The multiplet signal at  $\delta$  2.70 and double double doublet at  $\delta$  2.87 (J=17, 5.5, 1.5 Hz) were assigned to the methylene protons adjacent to the simple alde-

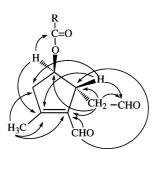
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Table 1 <sup>13</sup>C-NMR chemical shifts of compounds 1 and 2 (in CDCl<sub>3</sub>)

C	1	2
Me	14.36	14.38
1	45.95	45.95
2	137.29	137.26
3	160.60	160.56
4	45.88	45.95
5	76.87	77.10
CH 2 CHO	44.96	44.98
CH 2 CHO	200.87	200.80
СНО	187.44	187.45
Acyl 1'	122.23	122.32
2'	131.67	112.01
3'	113.64	148.68
4'	163.56	153.24
5'	113.64	110.22
6'	131.67	123.70
C=O	165.98	166.09
OMe	55.45	56.05
OMe		56.05

hyde (CH<sub>2</sub>CHO), and the doublet-like signal at  $\delta$  5.17 (J=7 Hz) is ascribable to H-5. The <sup>13</sup>C-NMR spectrum showed 17 carbon signals consisting of one methyl carbon ( $\delta$  14.36), one methoxy carbon ( $\delta$  55.45), two methylene carbons ( $\delta$  44.96 and 45.88), six methine carbons, four quaternary carbons, one carbonyl carbon ( $\delta$  165.98) and two aldehyde carbons ( $\delta$  187.44 and 200.87) (Table 1). NOEs were observed between H-1 at  $\delta$  3.57 and H-6 at  $\delta$  2.87, and between H-5 at  $\delta$  5.17 and H-6 at  $\delta$  2.70, showing that these protons are in a *cis* relationship with each other. The structure of 1 was determined by analysis of <sup>1</sup>H-, <sup>13</sup>C-NMR, HMQC, HMBC, NOESY (Fig. 1), and IR spectroscopic data as 2-formyl-5-(4'-methoxybenzoyloxy)-3-methyl-2-cyclopentene-1-acetaldehyde.

Compound 2, C<sub>18</sub>H<sub>20</sub>O<sub>6</sub>, showed ester, conjugated aldehyde, and aromatic ring absorptions in its IR spec-



relative conformation

1 R = 4'-methoxyphenyl 2 R = 3',4'-dimethoxyphenyl

Fig. 1. <sup>13</sup>C-<sup>1</sup>H long range correlations in the HMBC spectrum.

trum. The <sup>1</sup>H-NMR spectrum indicated the presence of a 3',4'-dimethoxybenzoyl moiety. Other spectral data were similar to those of 1. Thus, 2 was regarded as 2-formyl-5-(3',4'-dimethoxybenzoyloxy)-3-methyl-2-cyclopentene-1-acetaldehyde. Compounds 1 and 2 would be the hydrolysis products of the naturally-occurring iridoid glucoside (3) (Nakano, Maruyama, Murakami, Takaishi & Tomimatsu, 1993) (Davini, Iavarone & Trogolo, 1987; Watanabe, Takada, Matsuo & Nishimura, 1995; Drewes, Horn, Connolly & Bredenkamp, 1998).

Fig. 2 shows the inhibition curves of NBT reduction in the 12- $\sigma$ -tetradecanoylphorbol-13-acetate (TPA)-activated PMN by the samples. Compounds 1 and 2 showed potent anti-inflammatory activity. The 50 and 100% inhibitory concentrations (IC<sub>50</sub> and IC<sub>100</sub>) of 1 were 0.8 and 3.0 µg/ml, respectively. Compound 2 showed 1.05 µg/ml for IC<sub>50</sub> and 4.0 µg/ml for IC<sub>100</sub>.

R

- 1 R = 4'-methoxybenzoyl
- 2 R = 3',4'-dimethoxybenzoyl

#### 3. Experimental

#### 3.1. NMR

500 MHz for <sup>1</sup>H-NMR and 125 MHz for <sup>13</sup>C-NMR, CDCl<sub>3</sub>, TMS as int. standard; CC: silica gel (Mallinckrodt, AR) in amounts equivalent to 50 times of the extracts; PTLC: silica gel (Merck, 60F<sub>254</sub>; thickness, 0.5 mm); pre-packed column size B (Merck, LiChroprep Si 60).

#### 3.2. Isolation

The dried chips of the bark of *Tabebuia impetiginosa* (17 kg) (obtained from Santosflora, Sao Pâulo, Brazil) were extracted with hot MeOH. The MeOH extract was concentrated under reduced pressure; the residue (704 g) was dissolved in water and sequentially partitioned with *n*-hexane (residue 41.1 g), CHCl<sub>3</sub> (137.3 g), C<sub>2</sub>H<sub>4</sub>Cl<sub>2</sub> (35.1 g), and EtOAc (110.9 g). The C<sub>2</sub>H<sub>4</sub>Cl<sub>2</sub>-soluble portion (4 g) was chromatographed on silica gel (pre-packed column) using solvent systems of CHCl<sub>3</sub> (fraction 1: 500 ml, residue 48 mg; fr. 2: 500 ml, 103 mg; fr. 3: 1000 ml, 55 mg), CHCl<sub>3</sub>—MeOH (fr. 4: 1% MeOH 2000 ml, 176 mg; fr. 5: 3% MeOH 2000

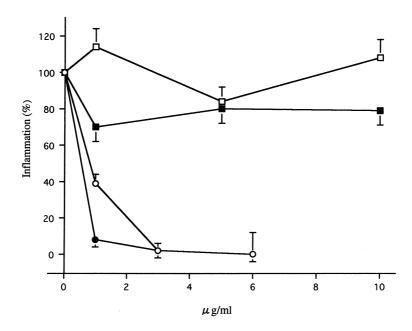


Fig. 2. Anti-inflammatory activities of 1 and 2 in the TPA-activated human PMN in comparison with alkylated benzoic acids.  $\bigcirc$  1;  $\bigcirc$  2;  $\square$  4-methoxybenzoic acid;  $\blacksquare$  3,4-dimethoxybenzoic acid.

ml, 127 mg). Fr. 2 was subjected to prepared TLC on silica gel using CHCl<sub>3</sub>–MeOH–acetone (120:2:1) to give the mixture of **1** and **2** (15 mg), a mixture of 5-and 8-hydroxy-2-(hydroxyethyl)naphtho[2,3-b]furan-4,9-dione (12 mg), 4-methoxy-, and 3,4-dimethoxyben-zoic acid. The mixture of **1** and **2** was subjected to prepared TLC with C<sub>6</sub>H<sub>6</sub>–Et<sub>2</sub>O (2:3) to afford **1** (Rf = 0.48) and **2** (Rf = 0.38) (2:1).

# 3.3. 2-Formyl-5-(4'-methoxybenzoyloxy)-3-methyl-2-cyclopentene-1-acetaldehyde (1)

Oil, IR  $v_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1712, 1707, 1664, 1607, 1512; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ :2.22 (3H, s, OMe), 2.70 (2H, m, H-4, CH<sub>2</sub>CHO), 2.87 (1H, ddd, J=17, 5.5 and 1.5 Hz, CH<sub>2</sub>CHO), 3.25 (1H, dd, J=20 and 7 Hz, H-4), 3.57 (1H, t-like, J=5.5 Hz, H-1), 3.85 (3H, s, OMe), 5.17 (1H, d-like, J=7 Hz, H-5), 6.90 (2H, d, J=9 Hz, H-3′, 5′), 7.95 (2H, d, J=9 Hz, H-2′,6′), 9.79 (1H, t, J=1.5 Hz, CH<sub>2</sub>CHO), 10.00 (1H, s, 2-CHO); MS m/z: 302.1140 (M<sup>+</sup>, calculated for C<sub>17</sub>H<sub>18</sub>O<sub>5</sub>, 302.1153).

## 3.4. 2-Formyl-5-(3',4'-dimethoxybenzoyloxy)-3-methyl-2-cyclopentene-1-acetaldehyde (2)

Oil, IR  $v_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1714, 1706, 1664, 1602, 1515;  $^{1}\text{H-NMR}$  (CDCl<sub>3</sub>)  $\delta$ :2.23 (3H, s, Me), 2.71 (2H, m, H-4, CH<sub>2</sub>CHO), 2.96 (1H, ddd, J=18.5, 5, and 1.5 Hz, CH<sub>2</sub>CHO), 3.27 (1H, dd, J=19.5 and 6.5 Hz, H-4), 3.58 (1H, t-like, J=5 Hz, H-1), 3.82, 3.91 (3H × 2, s, OMe), 5.18 (1H, dt, J = 6.5 and 2 Hz, H-5), 6.89 (1H, d, J = 8.5 Hz, H-5'), 7.50 (1H, d, J = 2 Hz, H-2'), 7.63 (1H, dd, J = 8.5 and 2 Hz, H-6'), 9.80 (1H, t, J = 1.5 Hz, CH<sub>2</sub>CHO), 10.00 (1H, t, 2-CHO); MS m/z: 332.1240 (M<sup>+</sup>, calculated for C<sub>18</sub>H<sub>20</sub>O<sub>6</sub>, 332.1258).

### 3.5. Anti-inflammatory assay with human WBC

Compounds 1 and 2 were tested. 4-methoxybenzoic acid and 3,4-dimethoxybenzoic acid that were also obtained from the  $C_2H_4Cl_2$ -soluble portion of T. *impetiginosa*, were used as controls.

Venous blood from healthy donors was harvested and the granular WBC fractions were isolated by the use of a Ficoll–Hypaque solution. The cells were suspended in an ice-cold phosphate buffered saline (PBS). The NBT medium consisted of 25  $\mu$ g/ml NBT, 9  $\mu$ M CaCl<sub>2</sub>, 5  $\mu$ M MgCl<sub>2</sub>, 0.1% glucose and 10 mM NaN<sub>3</sub> in PBS (Hirai, Moriguchi & Wang, 1991). The assay was performed by combining 1 × 10<sup>4</sup> WBC with 50 ng/ml TPA in an NBT medium for 20 min at 37°C. The diformazan pigments formed were extracted with DMSO, and absorbance at 560 nm was measured in a spectrophotometer. The inhibition rate was estimated by the modified method of Evans (Rice-Evans, Diplock & Symons, 1991).

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