



Triterpenoid glycosides and a triterpene from *Ilex brevicuspis*

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Abstract

Two saponins and a sapogenin were isolated from the leaves of *Ilex brevicuspis*. Their structures were established by means of spectroscopic methods as brevicuspisaponin 1 (3-*O*- α -L-arabinopyranosyl-20(*S*)-19 α ,24-dihydroxyursolic acid), brevicuspisaponin 2 (3-*O*- α -L-arabinopyranosyl-20(*S*)-19 α ,23,24-trihydroxyursolic acid) and 23-methylester of 20(*S*)-3 β ,19 α ,24-trihydroxyurs-12-en-23,28-dioic acid. © 2000 Elsevier Science Ltd. All rights reserved.

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1. Introduction

Ilex brevicuspis Reissek is a native tree in South Brazil and more particularly in the State of Rio Grande do Sul, Brazil, where it is popularly known as “caúna-da serra”, “congonha” or “orelha-de-mico”. *I. brevicuspis* is one of the species reported as an adulterant of *Ilex paraguariensis* St. Hil., the genuine maté product whose leaves are traditionally used to prepare the stimulating beverage named maté (Giberti, 1989). Continuing our systematic phytochemical study on *I. paraguariensis* and other South American *Ilex* species (Taketa & Schenkel, 1994; Gosmann, Guillaume, Taketa & Schenkel, 1995; Schenkel, Athayde, Giberti & Guillaume, 1995; Pires, Guillaume, Gosmann & Schenkel, 1997), we report here the isolation and structural elucidation of two new saponins and a new sapogenin isolated from the leaves of the title plant.

2. Results and discussion

Solvent partition and chromatographic procedures allowed the isolation of the main triterpenoids from the leaves of *I. brevicuspis*: brevicuspisaponin 1 (**1**), brevicuspisaponin 2 (**2**) and one sapogenin (**3**). Ursolic acid was also isolated and identified through comparison of its spectral data with that of the literature (Tkachev, Denisov, Gatilov, Bagryanskaya, Shevtsov & Rybalova, 1994).

Acid hydrolysis of **1** and **2** afforded, in both cases, only one sugar, identified as arabinose by co-TLC. The sugar configuration in each case was determined as α -L by means of GLC analysis (Mihashi, Hara & Okabe, 1986).

FAB MS (positive ion mode) of compound **1** displayed a quasi-molecular ion peak at m/z 621 [$M + H$]⁺ suggesting a molecular formula of C₃₅H₅₆O₉. In addition, the fragment at m/z 487 indicated the loss of an aldopentose moiety. The ¹³C-NMR spectrum of **1** showed 35 signals, whereas the DEPT spectrum revealed 6 methyl, 11 methylene, 10 methine and 8 quaternary carbon atoms (Table 1).

Careful comparison of ¹³C-NMR spectral data of **1**

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with that of ILA-1, isolated during a previous investigation (Schenkel et al., 1995), showed that **1** differs structurally from ILA-1 only by the absence of glucose signals. This assumption was confirmed by ^1H – ^1H COSY, ^1H – ^{13}C COSY, COLOC and ROESY experiments.

The most singularly important feature of this structure is the C-20 configuration, usually 20*R*, but in this case 20*S*. The axial configuration of CH_3 -30 (δ_{H} 1.11, *d*, *J* = 6 Hz) was deduced from its connectivities with $\text{H}\beta$ -18 (δ_{H} 3.26) and $\text{CH}_3\beta$ -29 (δ_{H} 1.42) in the ROESY experiment. Therefore, compound **1** was identified as 3-*O*- α -L-arabinopyranosyl-20(*S*)-19 α ,24-dihydroxyuronic acid, and named brevicuspisaponin 1. To the best

of our knowledge, this is the first report of the natural occurrence of this compound.

For compound **2**, the molecular formula $\text{C}_{35}\text{H}_{56}\text{O}_{10}$ was deduced based on the FAB MS, which displayed a quasi-molecular ion peak $[\text{M} + \text{H}]^+$ at *m/z* 637; the presence of 5 methyl, 12 methylene, 10 methine and 8 quaternary carbon atoms were also detected in the ^{13}C -NMR spectrum (Table 1). The sugar position was located at C-3 through ROESY correlation observed between the anomeric proton at δ_{H} 5.05 (*d*, *J* = 8 Hz; δ_{C} 104.4) and the aglycone signal at δ_{H} 4.47 ($\text{H}\alpha$ -3).

Careful comparisons of the ^{13}C -NMR spectrum signals for **2** and **1** revealed the presence in the spectrum of **2** of one additional hydroxymethyl

Table 1

^1H - (500 MHz) and ^{13}C -NMR (125 MHz) spectral data for compounds **1**, **2** (pyridine-*d*₅) and **3** (CDCl_3 : CD_3OD , 4 : 1)

C	DEPT	1		2		3	
		δ_{H} (mult, <i>J</i> = Hz)	δ_{C}	δ_{H} (mult, <i>J</i> = Hz)	δ_{C}	δ_{H} (mult, <i>J</i> = Hz)	δ_{C}
1	CH ₂	0.89; 1.53	37.7	1.03; 1.59	37.0	0.88; 1.45	38.1
2	CH ₂	2.00; 2.20	25.9	2.14; 2.31	24.9	1.52; 1.52	26.7
3	CH	3.51 (<i>dd</i> , 11.0, 5.0)	88.2	4.47 (<i>dd</i> , 13.0, 6.0)	80.5	3.89 (<i>dd</i> , 11.0, 5.0)	76.6
4	C	–	43.4	–	46.0	–	57.1
5	CH	0.95	55.4	1.88	46.3	1.19	51.7
6	CH ₂	1.35; 1.61	18.1	1.51; 1.90 (<i>t</i> , 13.0, $\text{H}\beta$)	17.2	1.18; 0.91	21.1
7	CH ₂	1.35; 1.50	32.9	1.37; 1.73 (<i>dt</i> , 10.0, 4.0, $\text{H}\alpha$)	31.9	0.99; 1.17	32.6
8	C	–	39.3	–	38.6	–	39.7
9	CH	1.78	46.7	1.88	45.8	1.42	47.4
10	C	–	35.9	–	34.9	–	36.3
11	CH ₂	1.78; 2.00	23.3	1.97; 2.05	22.6	1.65; 1.72	23.5
12	=CH	5.50 (<i>s</i>)	126.3	5.52 (<i>s</i>)	125.6	5.03 (<i>t</i> , 3.3)	127.4
13	=C	–	138.6	–	137.9	–	137.9
14	C	–	41.3	–	40.5	–	41.0
15	CH ₂	1.25; 2.26	28.4	1.22; 2.24	27.6	0.75; 1.47	28.0
16	CH ₂	2.06; 3.25 (<i>dt</i> , 13.0, 4.0, $\text{H}\alpha$)	26.2	2.05; 3.17 (<i>dt</i> , 13.0, 4.0, $\text{H}\alpha$)	25.4	1.30; 2.31 (<i>dt</i> , 13.0, 4.0, $\text{H}\alpha$)	25.8
17	C	–	47.1	–	46.4	–	47.0
18	CH	3.26 (<i>s</i>)	46.6	3.26 (<i>s</i>)	46.2	2.50 (<i>s</i>)	46.3
19	C	–	72.5	–	71.7	–	73.5
20	CH	2.00	42.2	2.00	41.5	1.48	41.3
21	CH ₂	1.30; 2.70 (<i>tt</i> , 13.0, 4.0, $\text{H}\alpha$)	24.0	1.27; 2.69 (<i>tt</i> , 10.0, 2.0, $\text{H}\alpha$)	23.3	0.92; 1.98 (<i>tt</i> , 13.0, 4.0, $\text{H}\alpha$)	23.7
22	CH ₂	1.95; 2.20	31.6	1.94; 2.24	30.8	1.28; 1.55	31.2
23	CH ₃	1.48 (<i>s</i>)	22.6	–	–	–	–
	O–CH ₂	–	–	4.33 (<i>d</i> , 13.0); 4.90 (<i>d</i> , 13.0)	59.4	–	–
	O=C	–	–	–	–	–	176.3
24	O–CH ₂	3.58 (<i>d</i> , 10.0); 4.35	62.5	3.89 (<i>d</i> , 13.0); 4.59 (<i>d</i> , 13.0)	61.2	3.78 (<i>d</i> , 12.0); 4.01 (<i>d</i> , 12.0)	60.8
25	CH ₃	0.81	14.5	0.91	13.9	0.68	15.7
26	CH ₃	1.05	16.3	1.08	14.5	0.48	16.0
27	CH ₃	1.73	23.5	1.68	22.7	1.00	23.8
28	O=C–O	–	179.9	–	179.1	–	181.0
29	CH ₃	1.42 (<i>s</i>)	29.0	1.40 (<i>s</i>)	28.2	0.90 (<i>s</i>)	29.2
30	CH ₃	1.11 (<i>d</i> , 6.0)	15.3	1.11 (<i>d</i> , 7.0)	15.6	0.73 (<i>d</i> , 7.0)	15.4
31	O–CH ₃	–	–	–	–	3.53	51.8
1'	–	4.73 (<i>d</i> , 7.0)	105.6	5.05 (<i>d</i> , 8.0)	104.4	–	–
2'	–	4.40 (<i>dd</i> , 8.0, 7.0)	72.0	4.43 (<i>t</i> , 8.0)	71.3	–	–
3'	–	4.20 (<i>dd</i> , 8.0, 3.0)	73.6	4.07 (<i>dd</i> , 8.0, 4.0)	73.0	–	–
4'	–	4.35	68.5	4.24	67.8	–	–
5'	–	3.85 (<i>d</i> , 10.0); 4.35	65.6	3.69 (<i>d</i> , 13.0); 4.31	65.1	–	–

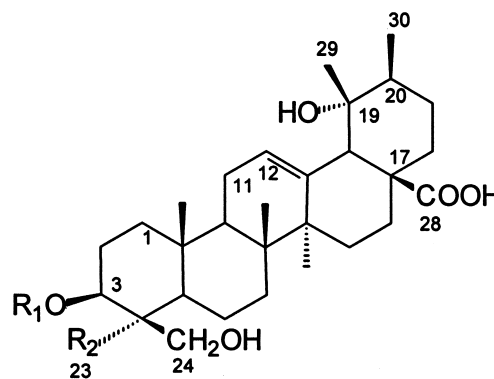
group (δ_C 59.4) that was located at C-23, in place of the methyl group at δ_C 22.6 (C-23) observed in **1**. The C-4 pattern could be established through the long range ^{13}C – ^1H COSY spectrum, which displayed cross peaks between the signals at δ_C 80.5 (C-3), δ_H 4.33 (H-23) and at δ_H 5.05 (H-1 ara). One- and two-dimensional NMR spectroscopic data (^1H – ^1H COSY, ^1H – ^{13}C COSY, COLOC, ROESY) confirmed **2** as 3-*O*- α -L-arabinopyranosyl-20(*S*)-19 α ,23,24-trihydroxyursolic acid, and named brevicuspisaponin **2**. This is the first report of its natural occurrence.

The FAB MS of compound **3** indicated the molecular formula $\text{C}_{31}\text{H}_{48}\text{O}_7$ through the positive molecular ion at m/z 533 $[\text{M} + \text{H}]^+$. Accordingly, the ^{13}C -NMR spectrum displayed 31 signals (Table 1) including, as determined from the DEPT spectrum, 6 methyl, 10 methylene, 6 methine and 9 quaternary carbon atoms. In the ^1H -NMR spectrum, the most salient features were the presence of a singlet at δ_H 2.50 (*s*, H β -18) suggesting a 19-hydroxy-ursene type skeleton and the presence of a singlet signal integrating for three protons at δ_H 3.53, and correlating in the ^{13}C – ^1H COSY spectrum with a carbon atom at δ_C 51.8, characteristic of a methoxy group. Protons from this latter group correlated, in the long range ^{13}C – ^1H COSY spectrum, with the carbon at δ_C 176.3, demonstrating the presence of a methyl ester group. In the same experiment, this carbonyl also correlated with protons at δ_H 3.78 and δ_H 4.01 (δ_C 60.8, CH_2O -24) demonstrating that the hydroxymethyl and methyl ester groups were both substituents of the same carbon. Finally, the hydroxymethyl proton at δ_H 3.78 also displayed a long range coupling with δ_C 76.6 (C-3).

The configuration of C-4 was deduced from the ROESY spectrum through the correlation between the β - CH_3 signal at C-25 (*s*, δ_H 0.68) and hydroxymethyl proton signals at δ_H 3.78 and at δ_H 4.01, confirming the β -position of this latter group at C-4.

Another important feature of the ROESY spectrum was the correlation between H β -18 (δ_H 2.50, *s*), CH_3 -29 (δ_H 0.90, *s*) and CH_3 -30 (δ_H 0.73, *d*, $J = 7.2$ Hz) indicating their β -orientation. This demonstrated the axial position of methyl group at C-30, and hence, the 20*S*-configuration. Chemical shifts of C-18 (δ_C 46.3) and C-22 (δ_C 31.2) are in agreement with the proposed configuration and can be explained in terms of a γ -effect induced by the 30 β -axial methyl group of **3**. Therefore, compound **3** was identified as 23-methylester of 20(*S*)-3 β ,19 α ,24-trihydroxyurs-12-en-23,28-dioic acid.

Very few 19 α -hydroxyursolic acid derivatives with the singular 20*S*-configuration have been isolated, most of them from *Ilex species* (Hidaka et al., 1987; Kakuno, Yoshikawa & Arihara, 1992), including South-American species (Schenkel et al., 1995).



	R ₁	R ₂
1	α -L-ara	CH_3
2	α -L-ara	CH_2OH
3	H	COOCH_3

3. Experimental

3.1. General

Melting points were obtained with a Kofler melting point apparatus and are uncorrected. IR spectra were recorded on a Perkin–Elmer 881 spectrophotometer. Optical rotations were measured on a Perkin–Elmer 241 polarimeter. EI MS spectra were recorded on a MS50 spectrometer and FAB MS spectra on a VG ZAB HS spectrometer. ^1H - and ^{13}C -NMR spectra were obtained on Bruker AM 400 and AMX 500 spectrometers. TLC was carried out on silica gel (Merck) GF₂₅₄ using CHCl_3 –MeOH (96 : 4) and CHCl_3 –EtOH– H_2O (8 : 4 : 0.5) as eluants for sapogenins and saponins, respectively, and EtOAc–MeOH–HOAc– H_2O (13 : 3 : 4 : 3) for sugars. Compounds were visualized using the vanillin–sulfuric acid reagent/100°/10 min.

3.2. Plant material

Leaves of *I. brevicuspis* Reissek were collected in Osório, State of Rio Grande do Sul, Brazil on April 1995. A herbarium specimen (leg. Coelho 163) is on deposit in the Herbarium of the Botany Department of the Federal University of Rio Grande do Sul (Herbarium ICN, Porto Alegre, Brazil).

3.3. Extraction and isolation

Air-dried leaves (589 g) were crushed and extracted with EtOH at room temperature (2×7 days). The ethanolic extract was evaporated to dryness under

reduced pressure and the residue (69 g) was suspended in water and extracted with chloroform. The chloroform and aqueous phases were evaporated to dryness to give the chloroform fraction (14 g) and the crude saponins fraction (19 g), respectively. Portions of these residues were repeatedly chromatographed on silica gel to give the pure compounds **1** (51 mg) and **2** (35 mg) from the saponin fraction, and **3** (58 mg) and **4** (13 mg) from the chloroform extract.

3.4. Acid hydrolysis

Compounds **1** and **2** were hydrolyzed on TLC plates in order to identify their sugar and aglycone moieties, as described by Kartnig and Wegschaidner (1972). Compounds **1** and **2** were also hydrolyzed in 5% aq. H₂SO₄ at 100°C for 1 h in preparation for GLC analysis after derivatization.

3.5. Sugar derivatization

The sugar obtained through acid hydrolysis of **1** and **2** was derivatized as described by Mihashi et al. (1986) using *L*-cysteine ethyl ester hydrochloride in the thiazolidine reaction.

3.6. Gas–liquid chromatography analysis of sugar derivatives from **1** and **2**

Analysis employed a WCOT SE-30 column (30 m, 0.25 mm i.d.), column temperature at 220°C and injection temperature at 252°C with H₂ as carrier gas at 2.2 ml min⁻¹. The trimethylsilyl ether of the arabinose thiazolidine derivative obtained from **1** and **2** gave a retention time of 3.50 min (same value as L-arabinose pattern; retention time of D-arabinose: 4.08 min).

3.7. Brevicuspisaponin 1 [3-*O*- α -L-arabinopyranosyl-20(*S*)-19 α ,24-dihydroxyursolic acid] (**1**)

White powder, mp 237–238°C; $[\alpha]_{589}^{18} + 36^\circ$ and $[\alpha]_{546}^{18} + 40^\circ$ (MeOH, *c* 1.346); IR ν_{\max}^{KBr} cm⁻¹: 3550–3050, 2966–2875, 1685, 1654, 1455, 1378, 1235, 1134, 865; FAB MS (positive-ion mode) *m/z*: 643.4 [M + Na]⁺, 621.4 [M + H]⁺, 487.4 [M-C₅H₉O₄]⁺; ¹H-NMR and ¹³C-NMR spectral data: Table 1.

3.8. Brevicuspisaponin 2 [3-*O*- α -L-arabinopyranosyl-20(*S*)-19 α ,23,24-trihydroxyursolic acid] (**2**)

White powder, mp 203–204°C; $[\alpha]_{589}^{16} + 29^\circ$, $[\alpha]_{578}^{16} + 29^\circ$, $[\alpha]_{546}^{16} + 34^\circ$, $[\alpha]_{436}^{16} + 57^\circ$ and $[\alpha]_{365}^{16} + 72^\circ$ (MeOH, *c* 0.64); IR ν_{\max}^{KBr} cm⁻¹: 3550–3100, 2924–2854, 1718, 1684, 1454, 1377, 1260–1215, 1140; FAB MS (positive ion mode) *m/z*: 659.3 [M + Na]⁺ and 637.3 [M + H]⁺; ¹H- and ¹³C-NMR spectral data: Table 1.

3.9. Compound **3** (23-methylester of 20(*S*)-3 β ,19 α ,24-trihydroxyurs-12-en-23,28-dioic acid)

Crystalline powder, mp 194–197°C; $[\alpha]_{589}^{20} + 68^\circ$, $[\alpha]_{578}^{20} + 71^\circ$, $[\alpha]_{546}^{20} + 82^\circ$, $[\alpha]_{436}^{20} + 145^\circ$, $[\alpha]_{365}^{20} + 227^\circ$ (CHCl₃, *c* 0.1); IR ν_{\max}^{KBr} cm⁻¹: 3434, 2936, 1710, 1456, 1382, 1236, 1066, 759; FAB MS (positive mode, matrix mNBA) *m/z* (rel. int.): 533.2 (40.3), 486.3 (9.3), 460.1 [3(mNBA + H) – H] (40.7), 391.2 (40), 307.0 [2mNBA – H] (100), 242.2 (5), 176.0 (13.5); EIMS *m/z*: 532, 514, 486, 470, 454, 437, 414, 269, 246, 232, 219, 201, 146. ¹H-NMR and ¹³C-NMR spectral data: Table 1.

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References

- Giberti, G. C. (1989). Los parientes silvestres de la yerba mate y el problema de su adulteración. *Dominguezia*, 7, 3.
- Gosmann, G., Guillaume, D., Taketa, A. T. C., & Schenkel, E. P. (1995). Triterpenoid saponins from *Ilex paraguariensis*. *J. Nat. Prod.*, 58, 438.
- Hidaka, K., Ito, M., Matsuda, Y., Kohda, H., Yamasaki, K., Yamahara, J., Chisaka, T., Kawakami, Y., Sato, T., & Kagei, K. (1987). New triterpene saponins from *Ilex pubescens*. *Chem. Pharm. Bull.*, 35, 524.
- Kakuno, T., Yoshikawa, K., & Arihara, S. (1992). Triterpenoid saponins from *Ilex crenata* fruit. *Phytochemistry*, 31, 3553.
- Kartnig, T., & Wegschaidner, O. (1972). Zur kenntnis der saponine aus *Herniaria glabra*. *Planta Med.*, 21, 144.
- Mihashi, K., Hara, S., & Okabe, H. (1986). Separation of aldose enantiomers by gas–liquid chromatography. *Chem. Pharm. Bull.*, 34, 1843.
- Pires, V. S., Guillaume, D., Gosmann, G., & Schenkel, E. P. (1997). Saponins from *Ilex dumosa*, an erva-maté (*Ilex paraguariensis*) adulterating plant. *J. Agric. Food Chem.*, 45, 1027.
- Schenkel, E. P., Athayde, M. L., Giberti, G. C., & Guillaume, D. (1995). A new saponin from *Ilex argentina*. *Acta Farm. Bonaerense*, 14, 5.
- Taketa, A. T. C., & Schenkel, E. P. (1994). Saponins from *Ilex pseudobuxus*. *Acta Farm. Bonaerense*, 13, 159.
- Tkachev, A. V., Denisov, A. Y., Gatilov, Y. V., Bagryanskaya, I. Y., Shevtsov, S. A., & Rybalova, T. V. (1994). Stereochemistry of hydrogen peroxide–acetic acid oxidation of ursolic acid and related compounds. *Tetrahedron*, 50, 11459.