



Phytochemistry 53 (2000) 991-995

www.elsevier.com/locate/phytochem

Chalconoids from Fissistigma bracteolatum

Trinh Phuong Lien^a, Andrea Porzel^b, Jürgen Schmidt^b, Tran Van Sung^a, Günter Adam^{b,*}

^aInstitute of Chemistry, National Centre for Natural Science and Technology of Vietnam, Nghia Do, Tu Liem, Hanoi, Viet Nam ^bInstitute of Plant Biochemistry, Weinberg 3, D-06120 Halle/Saale, Germany

Received 9 July 1999; received in revised form 2 November 1999

Dedicated to Prof. Gerhard Spiteller on the occasion of his 68th birthday.

Abstract

Phytochemical studies on the leaves of *Fissistigma bracteolatum* yielded besides the two known compounds 2-hydroxy-3,4,6-trimethoxychalcone (1) and 5,7,8-trimethoxyflav-3-ene (2), five new chalconoids 2-hydroxy-3,4,6-trimethoxychalcene (3), 2-hydroxy-3,4,6-trimethoxydihydrochalcone (4), 2'-hydroxy-3',4',6'-trimethoxydihydrochalcone (5), 2'-hydroxy-3',4',6'-trimethoxy- β '-methoxychalcane (6) and 2'-hydroxy-3',4',6'-trimethoxy- β '-ethoxychalcane (7). The structures of these compounds were determined by mass and NMR spectroscopic methods. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: Fissistigma bracteolatum; Annonaceae; Leaves; Chalconoids; 2-Hydroxy-3,4,6-trimethoxychalcene; 2-Hydroxy-3,4,6-trimethoxydihydrochalcone; 2'-Hydroxy-3',4',6'-trimethoxy-β'-methoxychalcane; 2'-Hydroxy-3',4',6'-trimethoxy-β'-methoxychalcane; 2'-Hydroxy-3',4',6'-trimethoxy-β'-ethoxychalcane

1. Introduction

The genus *Fissistigma* is a large tribe with ca. 70 species in the Annonaceae family (Leboeuf, Cave, Bhaumik, Mukherjee & Mukherjee, 1982). The decoctions of different plant, parts of several species have been used in southeast Asia as traditional medicines (Perry, 1980). Phytochemically, this tribe was reported to contain aporphine (Chang, Wu, Wu & Su, 1996), protoberberine (Chia, Chang, Li & Wu, 1998), phenanthrene alkaloids (Wu, Kao, Huang, Duh & Lu, 1990) and flavonoids (Alias, Awang, Hadi, Thoison, Sevenet & Pais, 1995) as their main components.

Fissistigma bracteolatum Chatt. is a creeper growing

from the leaves of *F. bracteolatum*.

The leaves of *F. bracteolatum* were extracted with *n*-hexane, ethyl acetate and *n*-butanol, successively. The *n*-hexane and ethyl acetate extracts were repeatedly

in the north of Vietnam (Pham Hoang Ho, 1993), the constituents of which have not yet been studied. In China this plant is applied externally on wounds to

stop bleeding or used to treat broken bones. In Viet-

nam it is used with other ingredients to treat infections

and also to enhance blood circulation (Vo Van Chi,

1997). In continuation of our search for new biologi-

cally active compounds from Vietnamese medicinal

plants (Thuy, Porzel, Ripperger, Sung & Adam, 1999)

we now report the structural elucidation of five new

chalconoids isolated besides two known compounds

E-mail address: gadam@ipb.uni-halle.de (G. Adam).

0031-9422/00/\$ - see front matter \odot 2000 Elsevier Science Ltd. All rights reserved. PII: S0031-9422(99)00570-1

^{2.} Results and discussion

^{*} Corresponding author. Tel.: +49-345-5582-216; fax: +49-345-5582-102

subjected to column chromatography on silica gel or reversed phase RP-8 using different solvent systems. Compounds 1, 2, 4, 6 and 7 were isolated from the *n*-hexane extract, compounds 3 and 5 from ethyl acetate extract.

Compounds 1 ($C_{18}H_{18}O_5$) and 2 ($C_{18}H_{18}O_4$) were identified as the known 2-hydroxy-3,4,6-trimethoxy-chalcone and 5,7,8-trimethoxyflav-3-ene, respectively, by comparison with spectroscopic data from the literature. Until now these two compounds were found only in the roots of *Uvaria dependens* (Annonaceae) (Nkunya, Waibel & Achenbach, 1993). Compound 1 belongs to the scarce group of retrochalcones.

Compound 3 had the elemental composition $C_{18}H_{20}O_4$, established by HRMS $(m/z\ 300.1380\ [M]^+)$. The ¹H NMR spectrum of 3 showed three aromatic methoxyl (δ 3.84, 3.79, 3.70), one methylene (δ 3.51), one phenolic hydroxyl (δ 7.94) and eight olefinic/aromatic protons. The EIMS peak at m/z 91 as well as the ¹H coupling pattern of a five spin-system of aromatic protons showed the presence of one monosubstituted phenyl ring. The absence of a carbonyl signal and the occurrence of a new methylene signal (δ 41.7) in the ¹³C NMR spectrum in comparison to 1 indicated that compound 3 had a methylene group at C-β' instead of a keto group. The HMBC spectrum showed the correlations between H₂- β' (δ 3.51) and C-1' (δ 142.3), and C-2'/6' (δ 129.3) confirming the unsubstituted ring to be connected at C- β '. The ¹³C signal at δ 122.7 was assigned to C-β due to its weak HMBC correlation with H-5 via ${}^4J_{\rm CH}$. The substitution pattern of the second aromatic ring was established by NOESY and HMBC spectra, especially by the HMBC correlations of OH-2 and C-1/C-2/C-3 and the NOESY cross peaks between H-5 (δ 6.25) and the two methoxyl signals at δ 3.84 (OMe-4) and δ 3.79 (OMe-6). From the spectral data the structure of the new compound 3 was determined as 2-hydroxy-3,4,6-trimethoxychalcene, which belongs to the retrochalcenes (Table 1).

The elemental composition of compound 4 was shown to be $C_{18}H_{20}O_5$ by HRMS (m/z 316.1306 $[M]^+$). The dominant peaks in the EIMS at m/z 105 (benzoyl ion) and 197 ($C_{10}H_{13}O_4$), resulting from α and a benzylic cleavage, respectively (Meksuriyen & Cordell, 1988), revealed the existence of a chalcone skeleton with an unsubstituted ring A and the ring B substituted with three methoxyl and one hydroxyl groups. Comparison of the ¹H, ¹³C NMR spectra of 4 with those of 1 suggested that compound 4 differed only by hydrogenation of the α,β -double bond ($\delta \alpha$: 3.14, m, 2H; δ β : 2.96, m, 2H). This was confirmed by the HMBC correlations of C- β' (δ 200.7) with H₂- α (δ 3.14) and H_2 - β (δ 2.96). The HMBC correlations of H- $2^{\prime}/6^{\prime}$ with C- β^{\prime} on the one hand and H_2 - β with C-2 and C-6 on the other hand confirmed the structure of **4** as 2-hydroxy-3,4,6-trimethoxydihydrochalcone, hitherto not yet described in the literature.

Compound 5 also displayed a molecular ion peak at m/z 316.1313, which was consistent with a formula of $C_{18}H_{20}O_5$. In analogy to 4 two intense fragments at m/z 211 (C₁₀H₁₁O₅) and 91 indicated a dihydrochalcone type, with all substituents located at ring A. Five aromatic protons at δ 7.28 (4H) and 7.19 (1H) were assigned to H-2/H-6, H-3/H-5 and H-4, respectively. Two methylene triplets at δ 3.33 and 2.96 with a coupling constant of 7.4 Hz were consistent with H_2 - α and H_2 - β , respectively. Furthermore, the hydroxyl signal at δ 13.7 indicated a chelat bonding, thus this group must be linked to C-2' of ring A. Finally, NOE difference spectra were used to determine the positions of the methoxyl groups in ring A. Irradiation of the methoxyl group at δ 3.68 enhanced the hydroxyl signal, hence this methoxyl group was attached to C-3'. Irradiation of the proton signal at δ 6.29 resulted in NOE enhancements of two methoxyl signals at δ 3.95 and 3.96. Consequently, this proton signal was assigned to position 5', the two methoxyl groups were connected to the positions 4' and 6'. The differentiation of OMe-4' and OMe-6' was possible by means of a weak NOE between the methoxy signal at δ 3.96 (OMe-6') and H_2 - α . These data led to the structure of the new compound 5 as 2'-hydroxy-3',4',6'-trimethoxvdihvdrochalcone.

Compound **6** (m/z 332.1639 [M]⁺, C₁₉H₂₄O₅) and compound **7** (m/z 346.1791 [M]⁺, C₂₀H₂₆O₅) also possessed the dihydrochalcone skeleton, which was deduced from EIMS as well as NMR spectra. Comparison of ¹H and ¹³C NMR data of **6** with those of **5** suggested that the β' -keto group in **5** was replaced by a methoxyl group in **6** (δ 3.31, 3H, s), which was confirmed by HMBC correlations of OCH₃- β' (δ 3.31) with C- β' (δ 78.5), of H- β' (δ 4.79) with C-1' (δ 107.5), C-2' (δ 151.2) and C-6' (δ 154.6).

Analysis of the spectroscopic data obtained for 7 indicated a close similarity to those of **6**. Compound 7 differed from **6** by the replacement of the methoxyl group at C- β ' by an ethoxyl group, which was proved by two additional signals at δ 3.49 (2H) and δ 1.19 (t, J=7.0 Hz, 3H) in the 1 H NMR spectrum as well as two signals at δ 65.6 (-O-CH $_{2}-$) and 15.4 (CH $_{3}$) in the 13 C NMR spectrum. Therefore, compound **6** was concluded to have the structure 2'-hydroxy-3',4',6'-trimethoxy- β '-methoxy-chalcane and compound 7 is 2'-hydroxy-3',4',6'-trimethoxy- β '-ethoxychalcane. Both compounds are new natural products.

Recently, potent antimitotic and cell growth inhibitory properties of several chalcones have been reported (Ducki et al., 1998). Comprehensive studies on the bioactivity of the described chalconoids are under way and will be published elsewhere.

3. Experimental

Mps. uncorr.; $[\alpha]_D^{20}$: JASCO DIP 1000 polarimeter; IR: Bruker IFS 28; UV: KONTRON UVIKON 940; EIMS (AMD 402, AMD Intectra GmbH): 70 eV (DIS), HR–EIMS (resolution ca. 5000); 1 H and 2D spectra were recorded on a Varian UNITY 500 spectrometer at 499.83 MHz. 13 C $\{^1$ H $\}$ and APT spectra were recorded on a Varian GEMINI 200–300 spectrometer at 75.5 MHz. Chemical shifts were referenced to internal TMS ($\delta = 0$, 1 H) and acetone- d_6 ($\delta = 29.8$, 13 C), respectively.

3.1. Plant material

Leaves and branches of *F. bracteolatum* Chatt. were collected in Hoa Binh province, Vietnam in August 1997. The species was identified by Mr. Ngo Van Trai, Institute of Materia Medica, Hanoi. A voucher specimen was deposited in the herbarium of this institute.

3.2. Extraction and isolation

The plant material (850 g) was dried at room temperature, ground and extracted three times for 12 h with 95% MeOH at room temperature. MeOH was evaporated in vacuo, and the aq. solution was extracted with *n*-hexane, followed by EtOAc and *n*-BuOH (each three times). The solvents were evaporated in vacuo. The *n*-hexane extract (11 g) was chromatographed over silica gel with *n*-hexane/EtOAc (8:2), increasing the amounts of EtOAc to 40%. Raw compounds 1, 2, 4, 6, 7 were obtained. The EtOAc extract (16 g) was fractionated with *n*-hexane/acetone (7:3), increasing the ratio of acetone to 100%, to afford the compounds 3 and 5.

3.2.1. 2-Hydroxy-3,4,6-trimethoxychalcone (1)

The compound was purified by CC [silica gel, n-hexane/EtOAc (6:4). Yellow needles from acetone. Mp 135–137°C. $R_{\rm f}$ 0.28 [silica gel, n-hexane/acetone (7:3)]. (see, Nkunya et al., 1993).

3.2.2. 5,7,8-Trimethoxyflav-3-ene (2)

The compound was purified by CC [silica gel, n-hexane/EtOAc (8:2) and n-hexane/ether (6:4)]. Oil. $R_{\rm f}$ 0.54 [silica gel, n-hexane/acetone (7:3)]. (see, Nkunya et al., 1993).

3.2.3. 2-Hydroxy-3,4,6-trimethoxychalcene (3)

The compound was purified by CC [silica gel, n-hexane/acetone (8:2) and RP8, acetonitril/H₂O (6:4)]. Oil. $R_{\rm f}$ 0.48 [silica gel, n-hexane/acetone (7:3)].

IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3513 (OH), 2938, 2841, 1724, 1613, 1506, 1465, 1425, 1347, 1267, 1114, 978. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 267 (3.91), 203 (4.54). EI–MS (70 eV) m/z (rel. int.): 300.1380 [M]⁺ (C₁₈H₂₀O₄, calcd. 300.1361) (100), 285 (11), 196 (30), 181 (20), 153 (21), 105 (19), 91 (70). ¹H NMR (300 MHz, (acetone- d_6) δ: 7.94 (1H, s, OH-2), 7.26 (4H, m, H-2'/H-6', H-3'/H-5'), 7.15 (1H, m, H-4'), 6.71 (2H, m, H-α, H-β), 6.25 (1H, s, H-5), 3.84 (3H, s, OMe-4), 3.79 (3H, s, OMe-6), 3.70 (3H, s, OMe-3), 3.51 (2H, m, H₂-β').

3.2.4. 2-Hydroxy-3,4,6-trimethoxydihydrochalcone (4)

The compound was purified by CC [silica gel, n-hexane/acetone (9:1) and RP8, MeOH/H₂O (8:2)]. Oil. $R_{\rm f}$ 0.35 [silica gel, n-hexane/acetone (7:3)].

IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3521 (OH), 2938, 2841, 1676 (C=O), 1618, 1598, 1511, 1465, 1427, 1349, 1241, 1159, 1112, 1039, 972, 872, 847. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 236 (4.16), 205 (4.67). EI–MS (70 eV) m/z (rel. int.): 316.1306 [M]⁺ (C₁₈H₂₀O₅, calcd. 316.1311) (100), 197 (66), 184 (40), 153 (14), 105 (26), 77 (18). ¹H NMR (300 MHz, acetone- d_6) δ : 8.04 (2H, d, 7.4 Hz, H-2'/H-6'), 7.85 (1H, s, OH-2), 7.61 (1H, tt, J = 7.4 and 1.3 Hz, H-4'), 7.51 (2H, tt, tt = 7.4 Hz, H-3'/H-5'), 6.25 (1H, tt)

Table 1 ¹³C NMR data of 3–7 (75 MHz, acetone-*d*₆)

3	4	5	6	7
107.4	109.5	142.7	142.9	142.9
149.8	149.5	129.3 ^a	129.2	129.2
131.3	131.6	129.2 ^a	129.1	129.1
152.4	152.2	126.7	126.4	126.5
89.5	89.6	129.2 ^a	129.1	129.1
155.2	154.9	129.3 ^a	129.2	129.2
122.7	19.4	31.3	32.7	32.6
131.1	39.1	46.7	37.4	37.6
41.7	200.7	206.1	78.5	76.9
142.3	137.9	106.6	107.5	108.1
129.3	128.9	159.6	151.2	151.4
129.1	129.4	131.6	132.2	132.4
126.6	133.6	159.9	153.9	153.9
129.1	129.4	88.3	89.7	89.6
129.3	128.9	159.9	154.6	154.2
60.9	60.9			
		60.3	60.6	60.5
56.1	56.2			
		56.4	56.2	56.3 ^a
56.1	56.1			
		56.4	56.2	56.2 ^a
			57.2	
				65.6
	107.4 149.8 131.3 152.4 89.5 155.2 122.7 131.1 41.7 142.3 129.3 129.1 126.6 129.1 129.3 60.9 56.1	107.4 109.5 149.8 149.5 131.3 131.6 152.4 152.2 89.5 89.6 155.2 154.9 122.7 19.4 131.1 39.1 41.7 200.7 142.3 137.9 129.3 128.9 129.1 129.4 129.2 129.4 129.3 128.9 60.9 60.9 56.1 56.2	107.4 109.5 142.7 149.8 149.5 129.3a 131.3 131.6 129.2a 152.4 152.2 126.7 89.5 89.6 129.2a 155.2 154.9 129.3a 122.7 19.4 31.3 131.1 39.1 46.7 41.7 200.7 206.1 142.3 137.9 106.6 129.3 128.9 159.6 129.1 129.4 131.6 126.6 133.6 159.9 129.1 129.4 88.3 129.3 128.9 159.9 60.9 60.9 60.9 60.3 56.1 56.2 56.4 56.1	107.4 109.5 142.7 142.9 149.8 149.5 129.3a 129.2 131.3 131.6 129.2a 129.1 152.4 152.2 126.7 126.4 89.5 89.6 129.2a 129.1 155.2 154.9 129.3a 129.2 122.7 19.4 31.3 32.7 131.1 39.1 46.7 37.4 41.7 200.7 206.1 78.5 142.3 137.9 106.6 107.5 129.3 128.9 159.6 151.2 129.1 129.4 131.6 132.2 126.6 133.6 159.9 153.9 129.1 129.4 88.3 89.7 129.3 128.9 159.9 154.6 60.9 60.9 60.9 60.3 60.6 56.1 56.2 56.4 56.2 56.1 56.1

^a Exchangeable.

H-5), 3.85 (3H, s, OMe-4), 3.78 (3H, s, OMe-6), 3.72 (3H, s, OMe-3), 3.14 (2H, m, H₂- α), 2.96 (2H, m, H₂- β).

3.2.5. 2'-Hydroxy-3',4',6'-trimethoxydihydrochalcone (5)

The compound was purified by CC [silica gel, n-hexane/acetone (8:2) and RP8, acetonitril/H₂O (7:3)]. Needles from acetone. Mp 101–103°C. $R_{\rm f}$ 0.38 [silica gel, n-hexane/acetone (7:3)].

IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3532 (*br.*, OH), 2935, 2851, 1619 (conj. C=O), 1598, 1497, 1439, 1416, 1358, 1287, 1149, 1126, 999, 977, 819. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 288 (4.17), 235 (sh.), 203 (sh.). EI–MS (70 eV) m/z (rel. int.): 316.1313 [M]⁺ (C₁₈H₂₀O₅, calcd. 316.1311) (57), 211 (100), 197 (11), 184 (24), 169 (11), 91 (30). ¹H NMR (300 MHz, acetone- d_6) δ: 13.69 (1H, s, OH-2'), 7.28 (4H, m, H-2/H-6, H-3/H-5), 7.19 (1H, m, H-4), 6.29 (1H, s, H-5'), 3.96 (3H, s, OMe-6'), 3.95 (3H, s, OMe-4'), 3.68 (3H, s, OMe-3'), 3.33 (2H, t, t = 7.4 Hz, H₂- α), 2.96 (2H, t, t = 7.4 Hz, H₂- β).

3.2.6. 2'-Hydroxy-3',4',6'-trimethoxy- β '-methoxychalcane ($\boldsymbol{6}$)

The compound was purified by CC [silica gel, n-hexane/EtOAc (8:2)] and then by prep. TLC [cyclohexane/acetone (6:4)]. Oil. $R_{\rm f}$ 0.46 [silica gel, n-hexane/acetone (7:3)]. [α] $_{\rm D}^{22.3}$ – 2.1° (MeOH, c 0.38).

IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3254 (OH), 2936, 2843, 1621, 1599,

1518, 1342, 1245, 1199, 1112, 1074, 864. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 277 (3.45), 223 (4.26). EI–MS (70 eV) m/z (rel. int.): 332.1639 [M]⁺ (C₁₉H₂₄O₅, calcd. 332.1624) (11), 300 (100), 285 (34), 227 (22), 196 (47), 91 (56), 77 (15). ¹H NMR (300 MHz, acetone- d_6) δ: 8.29 (1H, s, OH-2'), 7.27 (2H, dd, J = 7.8 and 7.2 Hz, H-3/H-5), 7.20 (2H, d, J = 7.8 Hz, H-2/H-6), 7.16 (1H, t, J = 7.2 Hz, H-4), 6.24 (1H, s, H-5'), 4.79 (1H, dd, J = 8.3 and 5.4 Hz, H-β'), 3.83 (3H, s, OMe-4'), 3.75 (3H, s, OMe-6'), 3.69 (3H, s, OMe-3'), 3.31 (3H, s, OMe-β'), 2.77 (1H, m, H_A-β), 2.62 (1H, m, H_B-β), 2.18 (1H, m, H_A-α), 2.00 (1H, m, H_B-α).

3.2.7. 2'-Hydroxy-3',4',6'-trimethoxy-β'-ethoxychalcane (7)

The compound was purified by CC [silica gel, *n*-hexane/EtOAc (8:2)] and then by prep. TLC [cyclohexane/acetone (1:1)]. Oil. $R_{\rm f}$ 0.45 [silica gel, *n*-hexane/acetone (7:3)]. [z]_D^{22.6} – 1.6° (MeOH, *c* 1.00). IR $\nu_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3254 (OH), 2937, 2842, 1622, 1599,

IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3254 (OH), 2937, 2842, 1622, 1599, 1519, 1399, 1341, 1245, 1199, 1112, 1074, 864, 820. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 276 (3.20), 222 (3.99). EI–MS (70 eV) m/z (rel. int.): 346.1791 [M]⁺ (C₂₀H₂₆O₅, calcd. 346.1780) (7), 300 (100), 285 (15), 196 (22), 181 (12), 91 (31). ¹H NMR (300 MHz, acetone- d_6) δ: 8.52 (1H, s, OH-2'), 7.27 (2H, dd, J = 7.8 and 7.2 Hz, H-3/H-5), 7.20 (2H, d, J = 7.8 Hz, H-2/H-6), 7.16 (1H, t, J = 7.2 Hz, H-4), 6.23 (1H, s, H-5'), 4.88 (1H, dd, J = 8.5 and 4.9 Hz, H-β'), 3.82 (3H, s, OMe-4'), 3.74 (3H, s, OMe-6'), 3.69 (3H, s, OMe-3'), 3.49 (2H, m, OCH₂CH₃-β'), 2.79 (1H, m, H_A-β), 2.63 (1H, m, H_B-β), 2.15 (1H, m, H_A-α), 2.02 (1H, m, H_B-α), 1.19 (3H, t, t = 7.0 Hz, OCH₂CH₃-β') (Table 1).

Acknowledgements

We thank the Volkswagenstiftung and the Bundesministerium für Bildung, Wissenschaft, Forschung und Technologie, Bonn, for financial support, Mr. Ngo Van Trai, Hanoi, for the identification of the plant material. One of us (T. P. L.) is indebted to the Deutscher Akademischer Austauschdienst (DAAD), Bonn, for a grant.

References

Alias, Y., Awang, K., Hadi, A. H. A., Thoison, O., Sevenet, T., & Pais, M. (1995). *Journal of Natural Products*, 58, 1160.

Chang, G. J., Wu, M. H., Wu, Y. C., & Su, M. J. (1996). Br. J. Pharmacol, 118, 1571.

Chia, Y. C., Chang, F. R., Li, C. M., & Wu, Y. C. (1998). Phytochemistry, 48, 367.

Ducki, S., Forrest, R., Hadfield, J. A., Kendall, A., Lawrence, N. J., McGown, A. T., & Rennison, D. (1998). Bioorganic and Medicinal Chemistry Letters, 8, 1051.

- Leboeuf, M., Cave, A., Bhaumik, P. K., Mukherjee, B., & Mukherjee, R. (1982). *Phytochemistry*, 21, 2783.
- Meksuriyen, D., & Cordell, G. A. (1988). *Journal of Natural Products*, 51, 1129.
- Nkunya, M. H. H., Waibel, R., & Achenbach, H. (1993). Phytochemistry, 34, 853.
- Perry, L. M. (1980). In *Medicinal plants of East and Southeast Asia* (p. 19). Cambridge: MIT Press.
- Pham, Hoang Ho (1993). Cay co Vietnam (An illustrated flora of Vietnam), vol. 2 (p. 329). Santa Ana, Canada: Mekong Printing.
- Thuy, T. T., Porzel, A., Ripperger, H., Sung, T. V., & Adam, G. (1999). *Phytochemistry*, 50, 903.
- Vo, Van Chi (1997). In *Tu Dien cay thuoc Vietnam (A dictionary of Vietnamese medicinal plant)* (p. 713). Ho Chi Minh City: Vietnam, Nha xuat ban Y hoc (Medicine Publication).
- Wu, Y. C., Kao, S. C., Huang, J. F., Duh, C. Y., & Lu, S. T. (1990). *Phytochemistry*, 29, 2387.