



# Flavonoids from *Goodyera schlechtendaliana*

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## Abstract

A flavonol glycoside, 3-[[6-*O*-(6-deoxy- $\alpha$ -L-mannopyranosyl)- $\beta$ -D-glucopyranosyl]oxi]-5,7-dihydroxy-8-[(4-hydroxy-3,5-dimethoxyphenyl)methyl]-2-(3,4-dihydroxyphenyl)-4*H*-1-benzopyran-4-one, trivially named goodyerin, was isolated from the whole plant of *Goodyera schlechtendaliana*, along with three known flavonoids, rutin, kaempferol-3-*O*-rutinoside and isorhamnetin-3-*O*-rutinoside. The structures were established by spectroscopic analysis. © 2000 Elsevier Science Ltd. All rights reserved.

**Keywords:** *Goodyera schlechtendaliana*; Orchidaceae; Flavonoids; Goodyerin

## 1. Introduction

*G. schlechtendaliana* Reichb (Orchidaceae) is a herbal drug used in Chinese folk medicine for fever, pain, snake-bites and lung disease since ancient time (Jiangsu New Medicinal College, 1986). Recently, it has also been used as a substitute for the precious crude drug, *Anoectochilus formosanus* Hayata (Kan, 1986; Lin & Namba, 1981a, 1981b; Du, Yoshizawa & Shoyama, 1998a). There have been no previous reports on the chemical constituents of *G. schlechtendaliana*. The present paper describes the isolation and structure elucidation of a novel flavonol glycoside, goodyerin (**1**), together with three known flavonol glycosides (**2**–**4**) from *G. schlechtendaliana*.

## 2. Results and discussion

Compound **1**, a yellow amorphous solid, gave posi-

tive results in Molish and Mg-HCl tests. Its UV spectrum in methanol showed characteristic absorption at 271 (band II), 320 (sh) and 360 nm (band I), indicating a 3-*O*-substituted flavonol skeleton (Markham, 1982), and analysis with the usual flavonoid shift reagents suggesting the presence of free hydroxyl groups at positions C-5, C-7, C-3' and C-4' (Markham, 1982). The monosaccharides obtained after complete acid hydrolysis were identified as D-glucose and L-rhamnose following GC analysis using standard samples (Hara, Okabe & Mihashi, 1987). Thus, **1** was deduced to be a 3-*O*-substituted flavonol glycoside.

The positive ion FABMS of **1** showed two quasimolecular ion peaks at  $m/z$  777 [ $M + H$ ]<sup>+</sup> and 799 [ $M + Na$ ]<sup>+</sup>, and high resolution MS analysis of quasimolecular ion peak [ $M + Na$ ]<sup>+</sup> revealed the molecular formula of **1** to be C<sub>36</sub>H<sub>40</sub>O<sub>19</sub>. Fragment ion peaks at  $m/z$  653 [ $M + Na - 146$  (deoxyhexose unit)]<sup>+</sup>, and 491 [653 – 162 (hexose unit)]<sup>+</sup>, confirmed the presence of rhamnosyl and glucosyl moieties in the molecule.

All the <sup>1</sup>H- and <sup>13</sup>C-NMR signals of **1** in CD<sub>3</sub>OD were assigned using <sup>1</sup>H–<sup>1</sup>H COSY, NOESY, DEPT, HMQC, and HMBC experiments (Table 1). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **1** showed the presence of a 3,8-disubstituted quercetin moiety, a glucose moiety, and a rhamnose moiety, in addition to a symmetric

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1,3,4,5-tetrasubstituted benzene ring system. In the  $^1\text{H}$ -NMR spectrum, a set of aromatic signals due to the 3',4'-disubstituted ring B were evident at  $\delta$  7.75 (1H, *dd*,  $J = 1.5, 0.7$  Hz, H-2'), 6.87 (1H, *dd*,  $J = 8.4, 0.7$  Hz, H-5'), and 7.58 (1H, *dd*,  $J = 8.4, 1.5$  Hz, H-6'). A signal at  $\delta$  6.31 (1H, *s*), which correlated to carbon at  $\delta$  99.5 in the HMQC spectrum, was characteristic of proton H-6 of ring A in the flavonoid skeleton (Markham, 1982). The H-6 signal was a singlet, suggesting that ring A of the flavonol has an 8-substituted skeleton. In the HMBC spectrum, the H-6 signal showed correlations with four quaternary carbons at  $\delta$  160.8 (C-5), 163.2 (C-7), 108.5 (C-8) and 105.8 (C-10), confirming the 8-substituted flavonol skeleton. Furthermore, a signal due to the methylene group at  $\delta$  4.04 (2H, *s*, H-7'') in  $^1\text{H}$ -NMR spectrum, correlated with the  $^{13}\text{C}$ -NMR signal at  $\delta$  29.1 (C-7'') in the HMQC spectrum. In addition, a chemically equivalent aromatic two-proton singlet at  $\delta$  6.58 (2H, *s*, H-2'' and H-6'') correlated with the  $^{13}\text{C}$ -NMR signal at  $\delta$  106.7 (C-2'' and C-6''), and two chemically equivalent methoxyl groups at  $\delta$  3.58 (3H  $\times$  2, *s*, 3''- and 5''-OMe) correlated with the  $^{13}\text{C}$ -NMR signal at  $\delta$  6.5 in the HMQC spectrum (Fig. 1).

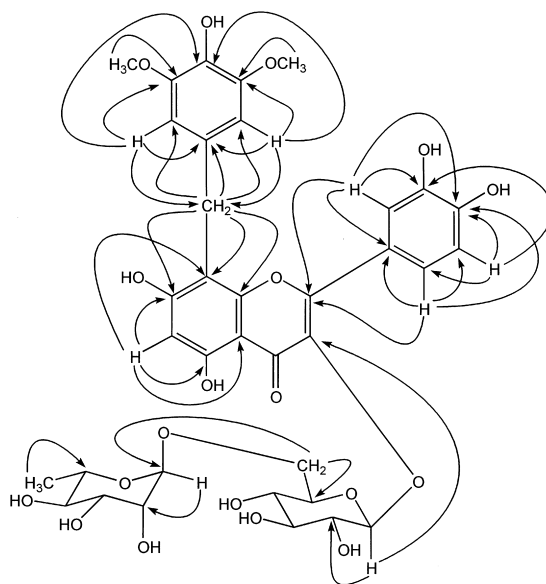


Fig. 1. Significant HMBC correlations observed for **1**.

Table 1  
 $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectral data of **1** in  $\text{CD}_3\text{OD}^a$

Attribution	$\delta_{\text{C}}$ (mult.)	$\delta_{\text{H}}$ (mult. $J_{\text{H-H}}$ )
2	159.3 ( <i>s</i> )	
3	135.6 ( <i>s</i> )	
4	179.7 ( <i>s</i> )	
5	160.8 ( <i>s</i> )	
6	99.5 ( <i>d</i> )	6.31 (1H, <i>s</i> )
7	163.2 ( <i>s</i> )	
8	108.5 ( <i>s</i> )	
9	155.8 ( <i>s</i> )	
10	105.8 ( <i>s</i> )	
1'	123.5 ( <i>s</i> )	
2'	117.9 ( <i>d</i> )	7.75 (1H, <i>dd</i> , 1.5, 0.7)
3'	146.0 ( <i>s</i> )	
4'	149.8 ( <i>s</i> )	
5'	116.1 ( <i>d</i> )	6.87 (1H, <i>dd</i> , 8.4, 0.7)
6'	123.7 ( <i>d</i> )	7.58 (1H, <i>dd</i> , 8.4, 1.5)
1''	133.2 ( <i>s</i> )	
2'', 6''	106.7 ( <i>d</i> )	6.58 (2H, <i>s</i> )
3'', 5''	149.0 ( <i>s</i> )	
4''	134.5 ( <i>s</i> )	
7''	29.1 ( <i>t</i> )	4.04 (2H, <i>s</i> )
3'', 5'' OMe	56.5 $\times$ 2 ( <i>q</i> )	3.58 (3H $\times$ 2, <i>s</i> )
Glc-1	104.9 ( <i>d</i> )	5.06 (1H, <i>d</i> , 7.3)
2	75.7 ( <i>d</i> )	3.44 (1H, <i>dd</i> , 9.2, 7.3)
3	78.2 ( <i>d</i> )	3.39 (1H, <i>dd</i> , 9.2, 9.2)
4	71.3 ( <i>d</i> )	3.25 (1H, <i>dd</i> , 9.2, 9.2)
5	77.2 ( <i>d</i> )	3.29 (1H, <i>m</i> )
6	68.5 ( <i>t</i> )	3.37 (1H, <i>dd</i> , 11.0, 5.5)
		3.77 (1H, <i>dd</i> , 11.0, 1.2)
Rha-1	102.4 ( <i>d</i> )	4.51 (1H, <i>d</i> , 1.0)
2	72.1 ( <i>d</i> )	3.63 (1H, <i>dd</i> , 3.3, 1.0)
3	72.3 ( <i>d</i> )	3.54 (1H, <i>dd</i> , 9.5, 3.3)
4	73.9 ( <i>d</i> )	3.27 (1H, <i>dd</i> , 9.5, 9.5)
5	69.7 ( <i>d</i> )	3.43 (1H, <i>m</i> )
6	17.9 ( <i>q</i> )	1.09 (3H, <i>d</i> , 6.2)

<sup>a</sup>  $^{13}\text{C}$  multiplicities from DEPT experiments.

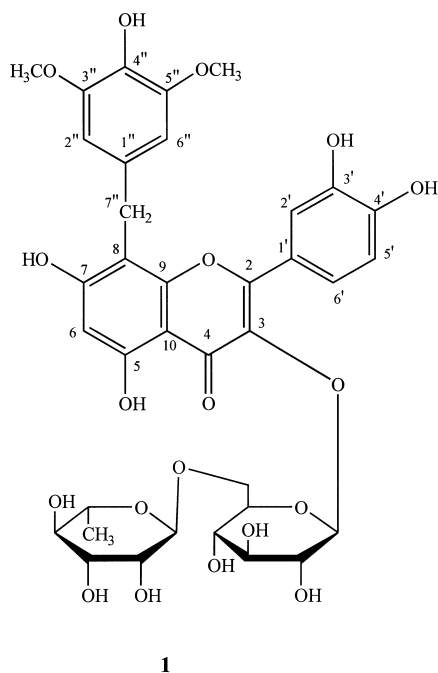
6'') correlated with the  $^{13}\text{C}$ -NMR signal at  $\delta$  106.7 (C-2'' and C-6''), and two chemically equivalent methoxyl groups at  $\delta$  3.58 (3H  $\times$  2, *s*, 3''- and 5''-OMe) correlated with the  $^{13}\text{C}$ -NMR signal at  $\delta$  6.5 in the HMQC spectrum (Fig. 1).

In the HMBC spectrum, long range correlations were observed from H-7'' to C-7, C-8, C-9, C-1'', C-2'' and C-6''. The same correlation of H-2'' and H-6'' was seen in C-1'', C-3'', C-4'', C-5'' and C-7''. The methoxyl group signals at  $\delta$  3.58 correlated to the carbons at  $\delta$  149.0 (C-3'' and C-5'') in the HMBC, and cross peaks between the methoxyl groups and H-2'' and H-6'' signals, and between H-7'' and H-2' signals were also observed in the NOESY spectra. On the basis of the above data, the symmetric 1,3,4,5-tetra-substituted benzene ring system was identified as a (4-hydroxy-3,5-dimethoxyphenyl)methyl group, attached to C-8 of the flavonol skeleton.

Two anomeric protons in the  $^1\text{H}$ -NMR spectrum of **1** were observed, and a signal at  $\delta$  5.06 (1H, *d*,  $J = 7.3$  Hz) was assigned as a glucosyl anomeric proton, suggesting that the glycosidic bond had a  $\beta$  linkage (Du, Kohinata, Kawasaki, Guo & Miyahara, 1998b). Another signal at  $\delta$  4.51 (1H, *d*,  $J = 1.0$  Hz) was assigned the rhamnosyl anomeric proton with an  $\alpha$  linkage (Du et al., 1998b). The sequence of sugar moieties was determined by NOESY spectroscopy, supported by a correlation between  $H_2$ -6 of the glucosyl group and the anomeric proton of rhamnosyl group. In the HMBC spectrum, the anomeric proton of the glucosyl group was correlated with C-3 ( $\delta$  135.6) of the aglycone. From these results, the structure of **1** was concluded to be 3-[[6-O-(6-deoxy- $\alpha$ -L-mannopyranosyl)- $\beta$ -D-glucopyranosyl]oxy]-5,7-dihydroxy-8-[(4-

hydroxy-3,5-dimethoxyphenyl)methyl]-2-(3,4-dihydroxyphenyl)-4H-1-benzopyran-4-one, and was trivially named goodyerin

The known flavonol glycosides were identified by UV, MS,  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra data in agreement with literature data (Markham, 1982; Agrawal, 1989) as rutin (**2**), kaempferol 3-*O*-rutinoside (**3**) and isorhamnetin 3-*O*-rutinoside (**4**).



### 3. Experimental

#### 3.1. General

Melting points were determined with a YAZAWA BY-2 apparatus and are uncorrected. Optical rotations were measured on a JASCO P-1010 digital polarimeter in MeOH at 26°C. UV spectra were recorded on a BECKMAN DU-70 spectrophotometer. The MS were recorded on a JEOL AX-500 instrument.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra were recorded on a JEOL GX-400 instrument (400 MHz for  $^1\text{H}$ -NMR and 100 MHz for  $^{13}\text{C}$ -NMR, respectively) using TMS as an internal standard.

#### 3.2. Plant material

The whole plants of *G. schlechtendaliana* were collected in Taiwan, in August 1997. A voucher specimen has been deposited at the Herbarium of Medicinal Plant Garden, Graduate School of Pharmaceutical Sciences, Kyushu University.

#### 3.3. Extraction and isolation

The air-dried whole plants of *G. schlechtendaliana* (1.5 kg) were powdered and percolated with MeOH at room temperature. Evaporation of the solvent from the extract under reduced pressure furnished the MeOH extract (667 g). The extract was suspended in  $\text{H}_2\text{O}$  and partitioned successively with *n*-hexane,  $\text{CHCl}_3$ , and *n*-BuOH. The *n*-BuOH soluble portion (58.2 g) was applied to a Diaion HP-20 column and eluted with  $\text{H}_2\text{O}$  and 20, 40, 50, 80 and 100% MeOH, successively. The 40% MeOH eluate (2.51 g) was chromatographed over silica gel and eluted with stepwise gradients of  $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$  (8:2:0.2 to 7:3:0.5) solvent system, then rechromatographed over a column of Polyamide C-200 (80% EtOH), and finally purified by silica gel column ( $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$ , 7:3:0.3) to give **2** (249 mg). The 50% MeOH eluate (5.49 g) was subjected to silica gel column chromatography and eluted with  $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$  (8:2:0.2 to 7:3:0.5) to give three fractions (frs. 1, 2 and 3). Fr. 3 (2.82 g) was further fractionated on a Sephadex LH-20 column eluted with MeOH, rechromatographed over silica gel eluted with  $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$  (8:2:0.2 to 7:3:0.5), and finally purified by reversed-phase HPLC (YMC-parck ODS-A) with 50% MeOH to yield of **3** (5 mg) and **4** (9 mg). The 60% MeOH eluate (545 mg) was chromatographed using Sephadex LH-20 column eluted with 80% MeOH, silica gel eluted with  $\text{CHCl}_3$ –MeOH– $\text{H}_2\text{O}$  (7:3:0.5), Polyamide C-200 (95% EtOH), then a silica gel column, and was finally crystallized from 80% EtOH to give **1** (282 mg).

#### 3.4. Goodyerin (**1**)

Yellow amorphous solid, mp: 180°C (80% EtOH),  $[\alpha]_D^{26} + 56.8^\circ$  (*c* 0.61, MeOH); UV  $\lambda_{\text{max}}$  (MeOH) nm: 271, 320 (*sh*), 360; + NaOMe: 281, 336 (*sh*), 415; +  $\text{AlCl}_3$ : 279, 370 (*sh*), 440; +  $\text{AlCl}_3/\text{HCl}$ : 279, 360 (*sh*), 408; + NaOAc: 278, 329 (*sh*), 398; + NaOAc/ $\text{H}_3\text{BO}_3$ : 268, 309 (*sh*), 381; positive ion HR-FABMS  $m/z$  799.2025  $[\text{M} + \text{Na}]^+$  (calculated for  $\text{C}_{36}\text{H}_{40}\text{O}_{19}\text{Na}$ : 799.2022); positive ion FABMS  $m/z$ : 815  $[\text{M} + \text{K}]^+$ , 799  $[\text{M} + \text{Na}]^+$ , 777  $[\text{M} + \text{H}]^+$ , 653  $[\text{M} + \text{Na} - 146 (\text{deoxyhexose unit})]^+$ , 491  $[653 - 162 (\text{hexose unit})]^+$ ;  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectral data ( $\text{CD}_3\text{OD}$ ): Table 1.

#### 3.5. Acid hydrolysis of **1**

A solution of **1** (2 mg) in 2N  $\text{H}_2\text{SO}_4$  (0.5 ml) was heated at 95°C for 0.5 h. After cooling, the reaction mixture was neutralized with  $\text{Ba}(\text{OH})_2$  and the insoluble portion was removed by filtration. The filtrate was evaporated under reduced pressure and the residue was separated on a Sep-Pack C18 cartridge column

eluted with H<sub>2</sub>O. The eluate was concentrated to give a syrup which exhibited two spots identical with those of authentic samples of D-glucose and L-rhamnose on Avicel SF TLC [BuOH–pyridine–H<sub>2</sub>O (6:2:3) upper layer + pyridine(1)]  $R_f$ : 0.38 (D-glucose), 0.63 (L-rhamnose). The syrup was converted to the trimethylsilyl ether of the thiazolidine derivative according to Hara et al., and was examined by GC (GL Sciences, OV-17, capillary column, 0.25 mm i.d.  $\times$  50 m; column temperature 220°C; carrier gas: He, 1.5 kg/cm<sup>2</sup>);  $R_t$  (min): 25.8 (D-glucose), 19.3 (L-rhamnose).

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