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A C-glucosylflavone from leaves of Piper lhotzkyanum

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Abstract

The dichloromethane fraction obtained from the methanolic extract of leaves of *P. lhotzkyanum* afforded a C-glucosylflavone (kaplanin) together with sakuranetin, methyl 4-methoxydihydroferulate and a mixture of methyl ferulate and dihydroferulate derivatives. Spectrometric methods were used to elucidate the structures of these compounds. © 2000 Elsevier Science Ltd. All rights reserved.

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1. Introduction

The family Piperaceae principally comprises five genera (Piper, Peperomia, Lepianthes, Macropiper and Trianaeopiper) and about 1400 species with worldwide distribution (Mabberley, 1993). Species of the genus Piper and Peperomia are found in abundance in Brazil (700 and 600, respectively) (Joly, 1977), some of which are used in folk medicine (Duke, 1985; Corrêa, 1984). Phytochemical investigation of Piperaceae species has afforded a diverse range of flavonoids, especially those with an unsubstituted B-ring. Many dihydrochalcones have also been isolated, but C-glycosylflavonoids are rare (Parmar et al., 1997). Piper lhotzkyanum Kunth (syn. P. inversum C.DC., P. aromaticum C.DC., P. damazii C.DC., P. bennetianum C.DC., P. santa-barbaranum C.DC.) is a shrub 1–3 m high, commonly found in humid forests with low incidence of light, in Southeast Brazil (Guimarães et al., 1992). Previous phytochemical investigation of P. lhotzkyanum yielded (E) and (Z) derivatives of 4-hydroxy-3-(3',7'-dimethyl-1'-oxo)-2',6'-octadienylbenzoic acid, a mixture of sesquiterpenes and E-phytol, and a new chromene (lhotzchromene) from hexane extract of the leaves (Moreira et al., 1998). Column chromatography of the dichloromethane fraction obtained from the methanolic extract of the leaves afforded 5-hydroxy-7-methoxy-8-C-β-glucosylflavone (kaplanin, 1) as well as the known compounds methyl 4-methoxydihydroferulate (2), reported here for the first time in the family Piperaceae; sakuranetin (3), previously isolated from *P. aduncum* (Parmar et al., 1997); and a mixture of (2) with the C6-C3 derivatives 4-methoxydihydroferulic acid (4a), ethyl 4methoxydihydroferulate (4b) and methyl ferulate (4c). The isolated substances were identified using spectroscopic analysis (1D and 2D ¹H- and ¹³C-NMR, UV). The mixture was analysed by GC/MS and the substances were identified by comparison of their mass spectra with literature data, and by analysis of their mass fragmentation patterns (McLafferty and Stanffer, 1989).

2. Results and discussion

Compound **1** showed [M + H]⁺ at m/z 431.4210 in its HR-FABMS, corresponding to the molecular formula $C_{22}H_{22}O_9$. The UV spectrum of compound **1** showed signals at $\lambda_{\text{max}}(\log \varepsilon)$ 213.3 (2.6), 287.2 (1.8) and

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330.0 (0.3). A bathochromic shift (60 nm) of band I in the UV spectrum was observed upon addition of AlCl₃, indicating a 5-hydroxy-substituted flavonoid. NaOH (0.1 N) or fused NaOAc treatment had no effect on the UV spectrum, indicating the absence of hydroxyl group at position 7 (A ring) or 4' (B ring) (Mabry et al., 1970). The ¹H-NMR spectrum displayed signals for methoxyl protons at $\delta 3.95$ (3H, s), an olefinic proton at $\delta 6.80$ (1H, s), and a singlet and a multiplet due to aromatic protons at δ 6.48 (1H) and δ 7.6 – 8.2 (5H), respectively, which together suggested an unsubstituted B ring and a trisubstituted A ring. The presence of only one anomeric signal in the ¹H-NMR (4.98) and ¹³C-NMR (72.02) spectra (72.02) suggested 1 to be a monosaccharide derivative. The large H-1' and H-2' coupling constant (10 Hz) for the anomeric proton indicated a β-configuration. The ¹H–¹H COSY spectrum supported this configuration. The ¹H-NMR spectrum also showed signals for carbinolic protons at $\delta 3.0-4.98$ due to the sugar moiety (10H) and for a phenolic proton at δ 13.2 (5-OH). The ¹³C-NMR and HETCOR spectra of 1 showed signals at δ 95.85 and 105.97 due to carbons 6 and 8, indicating a C-substitution at carbon 8 of ring A. Also observed were signals at δ 62.9 and 82.5 arising from the glycosyl moiety. The ¹³C-NMR assignments

Table 1 ¹³C-NMR spectral data for compound 1^a, 5-hydroxy-7-methoxy-flavone (5)^b, vitexin-7-glucoside (6), isovitexin (7), orientin (8), isovientin (9), swertisin (10) and isoswertisin (11)^c

Position	1	5	6	7	8	9	10	11
2	165.0	163.5	164.2	163.9	164.0	163.6	164.0	164.5
3	105.5	105.4	102.4	102.9	102.5	102.8	103.1	102.5
4	183.6	182.1	182.3	182.1	181.9	181.8	182.3	182.4
4a	106.5	105.0	_d	103.5	104.0	103.3	104.1	105.9
5	163.2	161.3	160.7	160.8	160.0	160.2	160.9	161.0
6	95.9	98.2	98.5	109.0	97.9	108.5	109.8	95.1
7	165.2	165.4	162.6	163.3	162.2	163.0	164.2	163.5
8	105.9	92.8	107.3	93.7	104.4	93.6	90.7	104.8
8a	156.8	157.4	155.0	156.3	156.0	156.2	156.9	155.3
1'	132.3	130.6	121.2	121.1	122.0	121.4	121.1	121.6
2'	127.9	126.5	128.8	128.6	113.8	113.1	128.5	129.2
3'	129.9	129.2	115.8	116.0	145.5	145.4	116.0	116.0
4'	132.8	132.1	161.4	161.4	149.3	149.4	161.4	161.4
5'	129.9	129.2	115.8	116.0	115.6	116.0	116.0	116.0
6'	127.9	126.2	128.8	128.6	119.4	119.0	128.6	129.2
1"	72.0		73.9		73.3	72.9	72.9	
2"	74.5		71.4		70.6	70.3	70.3	
3"	80.3		78.5		78.4	78.6	79.3	
4"	72.5		69.9		70.5	70.0	71.4	
5"	82.5		81.4		81.7	81.3	81.5	
6"	62.9		61.5		61.4	61.2	62.0	
$O\underline{C}H_3$	56.9	56.7	_	_	_	_	-	

a $1 = acetone-d_6$.

of 1 (Table 1) were compared to those published for 5-hydroxy-7-methoxyflavone (5), vitexin-7-glucoside (6), isovitexin (7), orientin (8), isoorientin (9), swertisin (10) and isoswertisin (11), and confirmed this compound to be 5-hydroxy-7-methoxy-8-C-β-glucosylflavone (Davoust et al., 1980; Chopin et al., 1984). To the best of our knowledge, this is the first occurrence in the family Piperaceae of a C-glycosylflavonoid with an unsubstituted B-ring.

3. Experimental

3.1. General

UV spectra were recorded on a Shimadzu UV-1601 following the procedures for the identification of flavonoids (Mabry et al., 1970). Mass spectra were obtained on a VG Autospect operated at 70 eV. High resolution positive FAB-mass spectra were recorded using a glycerol

b $5 = CDCl_3$.

^c **6–11** = DMSO- d_6 .

^d Signal not reported.

matrix on a VG 7070E mass spectrometer. ¹H-NMR (200 MHz), ¹³C-NMR (50.2 MHz), APT, ¹H-¹H-COSY, and HETCOR spectra were recorded in acetone-*d*₆ or CDCl₃ on a Varian Gemini 200 spectrometer, using TMS as int. standard.

3.2. Separation

Silica gel (Merck); GC was performed using a Varian Star 3400 gas chromatograph, fused silica capillary column (DB-5, 30 m \times 0.20 mm), H₂ as carrier gas and temp. programming from 60 to 270°C (3°C/min); GC-MS was performed using a HP5890 SII gas chromatograph coupled to a VG Autospect mass spectrometer at 70 eV, fused silica capillary column (DB-1, 30 m \times 0.20 mm), He as carrier gas and temp. programming from 60 to 270°C (3°C/min).

3.3. Plant material

Leaves of *P. lhotzkyanum* were collected near Teresopolis, Rio de Janeiro, Brazil, in September 1996. Voucher samples (RB 323487) are deposited in the Herbarium of Rio de Janeiro Botanical Garden, Rio de Janeiro, Brazil.

3.4. Extraction and isolation

Dried and powdered leaves of *P. lhotzkyanum* (280 g) were extracted successively with hexane and methanol at room temperature. The MeOH extracts were evaporated to dryness under red. pres., then redissolved in H₂O/MeOH (7:3) and extracted successively with hexane, CH₂Cl₂, EtOAc and *n*-BuOH. The CH₂Cl₂ fraction (9.33 g) was fractionated on silica gel column eluted with a CH₂Cl₂–MeOH gradient. A fraction eluted with pure dichloromethane afforded a mixture of ferulates (3 mg) which were analysed and identified by GC/MS (Table 2). Compound 2 (10 mg) was eluted from the silica gel column with a mixture of CH₂Cl₂–MeOH 1%,

and purified over XAD-2 column eluted with MeOH. Compound **3** (15 mg) and compound **1** (28 mg) were eluted from a silica gel column with a mixture of CH₂Cl₂–MeOH (2.5 and 20%, respectively) and were purified by recrystallization from hexane and EtOAc.

3.4.1. Kaplanin 1

Yellow amorphous solid (CH₂Cl₂–MeOH 20%). HR-FABMS (positive mode) m/z: 431.4210 [M + H]⁺; C₂₂H₂₂O₉ + H requires, 431.4206; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε) MeOH: 213.3 (2.6), 287.2 (1.8), 330.0 (0.3); + AlCl₃: 204.8 (1.1), 220.4 (1.1), 281.0 (0.8), 326.6 (0.4), 391.6 (0.2); + NaOMe: 211.4 (3.1), 278.6 (1.8), 364.6 (0.4); ¹H-NMR spectral data (200 MHz, CDCl₃): δ 3.95 (3H, s, 7-OCH₃), 4.98 (1H, d, d = 10 Hz, H"-1; glucosyl), 3.0–4.2 (10H, d d + d

3.4.2. Compound **2**

Amorphous solid (CH₂Cl₂–MeOH 1%). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε) MeOH: 228.4 (0.3), 279.0 (0.1); EIMS m/z (rel. int.): 224 [M]⁺ (35), 164 (18), 151 (100), 91 (20), 77 (30) 65 (20); ¹H-NMR spectral data (200 MHz, CDCl₃): δ 2.60 (2H, t, J = 7 Hz, H-1'); 2.90 (2H, t, J = 7 Hz, H-2'); 3.68 (3H, s, O=C-O-CH₃); 3.84 (3H, s, O-CH₃); 6.72 (1H, d, J = 2 Hz, H-2); 6.74 (1H, dd, J = 2.8 Hz, H-6); 6.8 (1H, d, J = 8 Hz, H-5); 13 C-NMR spectral data (50.2 MHz, CDCl₃): δ 30.43 (C-1'), 35.82 (C-2'), 51.42 (O=C-O-CH₃), 55.65, 55.75 (2 × OCH₃), 111.19 (C-5), 111.53 (C-2), 119.93 (C-6), 132.99 (C-1), 147.36 (C-4), 148.74 (C-3), 173.21 (C=O).

3.4.3. Compound 3

White solid (CH₂Cl₂–MeOH 2.5%). MS, UV and ¹H- and ¹³C-NMR data were identical with published data (Liu et al., 1992).

Table 2 Identified substances in the mixture of C6–C3 derivatives

Substance	RT1a	RT2 ^b	(%)	MS
4a ^c	35.05	28.40	21.99	210 [M]+ (35), 150 (25), 137 (100), 77 (25), 65 (20)
2^{d}	36.77	29.40	60.86	224 [M] + (35), 164 (20), 151 (100), 91 (20), 77 (25), 65 (18)
4b ^e	38.95	31.78	5.89	238 [M] + (35), 164 (32), 151 (100), 91 (20), 77 (27), 65 (17)
$4c^{f}$	40.57	33.75	5.09	208 [M]+ (100), 177 (75), 145 (60), 133 (30), 117(31), 105 (28), 89 (55), 77 (56), 51 (77)

^a RT1 = RT in minutes from GC/MS.

 $^{^{\}rm b}$ RT2 = RT in minutes from GC.

^c **4a** = 4-Methoxydihydroferulic acid.

^d **2** = Methyl 4-methoxyferulate.

^e **4b** = Ethyl 4-methoxyferulate.

 $^{^{}f}$ 4c = Methyl ferulate.

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