



Xanthenes from *Calophyllum teysmannii* var. *inophylloide*[☆]

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Received 3 January 2000; received in revised form 11 April 2000

Abstract

Further study of the wood of *Calophyllum teysmannii* Miq. var. *inophylloide* yielded xanthenes 7-hydroxy-1,2,8-trimethoxyxanthone, 6-hydroxy-1,2,5-trimethoxyxanthone, and 2-carbomethoxy-6-methoxyxanthone in addition to 3,8-dihydroxy-1,2,4-trimethoxyxanthone, 3-hydroxy-2,4-dimethoxyxanthone, 1,7-dihydroxy-3-methoxyxanthone (gentisin) and 2-hydroxyxanthone. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: *Calophyllum teysmannii* var. *inophylloide*; Guttiferae; Xanthenes; 7-Hydroxy-1,2,8-trimethoxyxanthone; 6-Hydroxy-1,2,5-trimethoxyxanthone; 2-Carbomethoxy-6-methoxyxanthone; 3,8-Dihydroxy-1,2,4-trimethoxyxanthone; 3-Hydroxy-2,4-dimethoxyxanthone; 1,7-Dihydroxy-3-methoxyxanthone; 2-Hydroxyxanthone

1. Introduction

Leaves, twigs, latex or stem bark of collections of *Calophyllum teysmannii* Miq. var. *inophylloide* (King) PF Stevens from Malaysia and Indonesia have yielded dipyrancoumarins which act as potent inhibitors of HIV-1 RT (Patil et al., 1993; Gustafson et al., 1994; Fuller et al., 1994; McKee et al., 1996, 1998; Cao et al., 1996, 1998a,b). However, there were no reports on constituents of the wood.

2. Results and discussion

We recently reported isolation and immunomodulatory activity of nine known xanthenes from the wood of a collection of *C. teysmannii* var. *inophylloide* from Narathiwat Province, Southern Thailand (Gonzalez et al., 1999) and subsequently described three new xanthenes and several

other constituents from the same collection (Kijjoa et al., 2000). Further study of the remaining fractions has now led to the isolation of three more new xanthenes, i.e. 7-hydroxy-1,2,8-trimethoxyxanthone (**1**), 6-hydroxy-1,2,5-trimethoxyxanthone (**2**), and 2-carbomethoxy-6-methoxyxanthone (**6a**). Previously known xanthenes from this fraction not previously found in *C. teysmannii* were 3,8-dihydroxy-1,2,4-trimethoxyxanthone (**3a**), 3-hydroxy-2,4-dimethoxyxanthone (**4**), 1,7-dihydroxy-3-methoxyxanthone (gentisin, **5**) (both listed in Peres and Nagem, 1997) and 2-hydroxyxanthone (**6b**, Sultanbawa 1980).

The ¹H NMR spectrum of xanthone **1**, C₁₆H₁₄O₆, exhibited signals of two sets of almost equivalent vicinally coupled aromatic protons and, in the ¹³C NMR spectrum (Table 1), three methoxys only one of which, at δ 57.28, was adjacent to an unsubstituted carbon on the xanthone ring system. In fact proton and carbon chemical shifts in one of the rings were essentially identical with those in ring A of 1,8-dimethoxy-2-hydroxyxanthone we previously reported from this species (Gonzalez et al., 1999) while shifts of protons and carbons in the second ring were identical with those in ring A of 1,2-methoxy-8-hydroxyxanthone (ibid). The

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HMBC correlations depicted in Fig. 1 confirmed the assignment as 7-hydroxy-1,2,8-trimethoxyxanthone.

In xanthone **2** with the same empirical formula $C_{16}H_{14}O_6$, chemical shifts of protons and carbons in ring A paralleled those in ring A of xanthone **1** (see Table 1). The situation in ring B was, however, reversed with the methoxyl group at $\delta 61.84$ again located next to a hydroxyl group, but now at C-4. Structure **2**, 6-hydroxy-1,2,5-trimethoxyxanthone, was again confirmed by the HMBC correlations shown in Fig. 1.

A third constituent was a dihydroxytrimethoxyxanthone **3a** or **3b**. One of the hydroxyls was hydrogen bonded to the carbonyl group at C-9 (1H NMR singlet at $\delta 13.08$) and attached to an otherwise unsubstituted aromatic ring as indicated by the 1H and ^{13}C chemical shifts in ring B and the HMBC correlations shown in Fig. 1. However this technique was unable to distinguish between formulas **3a** and **3b**. We opted for formula **3a**, 3,8-dihydroxy-1,2,4-trimethoxyxanthone, because of the pronounced bathochromic shift in the UV spectrum on addition of base characteristic of a phenolic hydroxyl *para* to a carbonyl group.

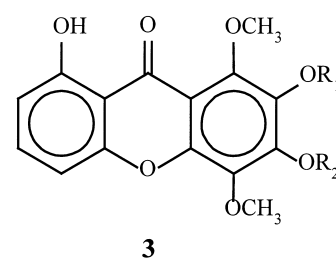
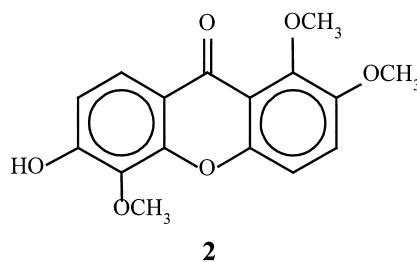
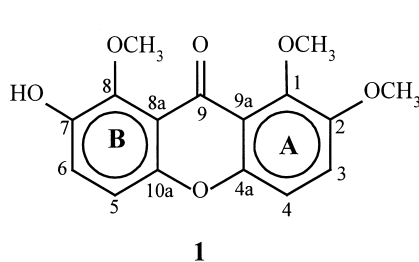
A xanthone assigned formula **3a** has been isolated previously from *Psorospermum febrigum* (Habib et al., 1987). Our ^{13}C NMR spectroscopy shifts in $CDCl_3$ (Table 1) differed only slightly from those reported earlier in DMSO; however, based on the HMBC correlations in Fig. 1 the shifts assigned to C-5 (111.7) and C-7 (104.0) by earlier workers probably need to be interchanged.

Xanthone **4** has been reported previously from two *Kielmeyera* (Gottlieb et al., 1970, 1971) and two *Hypericum* species (Luz Cardona et al., 1986, 1990); our ^{13}C NMR spectral data (Table 1) differ slightly from those reported previously (Castelão et al., 1977) and were verified by the HMBC correlations depicted in Fig. 1.

Xanthone **6a** appears to be new as a natural product although the literature contains two reports of its synthesis with little information provided on its properties (Eckstein et al., 1983; Jackson et al., 1993). Mass spectrum and 1H NMR data (see Section 3) indicated the presence of a carbomethoxy group in position 2 or 3 of ring A and a methoxy group in position 6 or 7 of ring B. The ^{13}C NMR spectrum (Table 1) supported these assignments while the HMBC correlations shown in Fig. 1 established the structure as 2-carbomethoxy-6-methoxyxanthone.

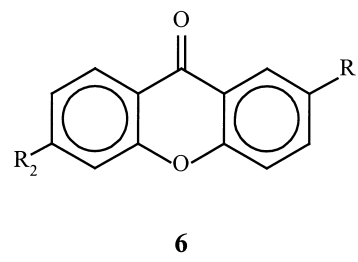
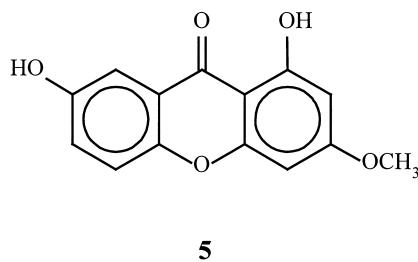
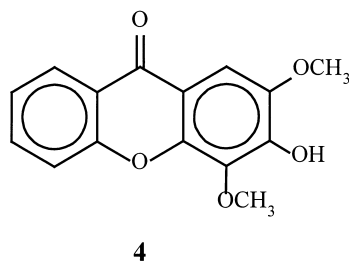
3. Experimental

Extraction and fractionation of the extract of the wood of *C. teysmannii* var. *inophylloide* from Narathiwat Province, Southern Thailand, were described earlier (Gonzalez et al., 1999, Kijjoa et al., 2000). Frs 49–73 (0.7 g) of the first chromatogram were combined and rechromatographed (Si gel 60), 100 ml subfrs being collected as follows: Subfrs 1–5 (petrol– $CHCl_3$, 7:3), 6–94 (petrol– $CHCl_3$, 1:1), 95–109



a - $R_1 = CH_3$, $R_2 = H$

b - $R_1 = H$, $R_2 = CH_3$



a - $R_1 = CO_2Me$, $R_2 = OMe$

b - $R_1 = OH$, $R_2 = H$

Table 1
¹³C NMR spectra of **1**, **2**, **3**, **4** and **6a** (75 MHz, CDCl₃)^a

C	1	2	3	4	6a
1	148.76	149.09	131.26	100.60	129.24
2	148.88	149.45	148.44	144.81	126.01
3	120.29	119.59	149.04	145.07	134.96
4	113.14	112.94	137.75	134.67	118.15
4a	150.73	150.88	147.47	146.17	158.79
5	112.62	133.24	106.41	117.98	100.44
6	121.80	153.75	136.06	134.19	165.43
7	145.03	112.12	110.86	123.96	113.74
8	144.34	122.77	162.06	126.60	128.46
8a	116.22	116.87	108.66	121.34	115.72
9	176.02	175.61	181.52	176.16	175.63
9a	117.34	116.93	108.74	114.57	121.62
10a	150.19	148.99	155.91	155.89	157.92
1-OMe	61.73	61.70	62.10 ^b		
2-OMe	57.28	57.12	61.80 ^b	56.53	
3-OMe					
4-OMe			61.80 ^b	61.73	
5-OMe		61.84			
6-OMe					55.93
8-OMe	62.54				
2-CO ₂ Me					165.99, 52.39

^a Assignments of multiples by DEPT and HETCOR.

^b Signals interchangeable.

(petrol–CHCl₃, 1:9), 160–170 (CHCl₃), 172–181 (CHCl₃–Me₂O, 4:1), 182–185 (CHCl₃–Me₂O, 3:2) and 186–197 (CHCl₃–Me₂O, 3:7). Purification of subfrs 12–16 by TLC (Si gel, CHCl₃–Me₂CO–HCO₂H, 80:20:1) gave **6a** (5 mg) and **3a** (15 mg). Purification of subfrs 17–31 by TLC (Si gel, toluene–EtOAc–HCO₂H, 65:35:1) gave 30 mg of 1,8-dimethoxy-2-hydroxyxanthone isolated previously from another fraction (Gonzalez et al., 1999), **1** (18 mg), **2** (16 mg), **4** (12 mg), an additional 12 mg of **6a**, **6b** (5 mg) and **5** (13 mg).

3.1. 7-Hydroxy-1,2,8-trimethoxyxanthone (**1**)

Brown solid, mp 73–75°C; HREIMS [M]⁺ *m/z* = 302.07901 (calculated for C₁₆H₁₄O₆ = 302.07905); EIMS *m/z* (rel. int.) 302 [M]⁺ (100), 287 [M–CH₃]⁺ (82.8), 272 (19.2), 256 (31.1), 244 (68.5), 227 (18.5), 201 (13.8), 149 (16.5); ¹H NMR (300 MHz, CDCl₃) δ 7.31 (*d*, *J* = 9.0 Hz, H-6), 7.30 (*d*, *J* = 9.2 Hz, H-3), 7.14 (*d*, *J* = 9.2 Hz, H-4), 7.10 (*d*, *J* = 9.2 Hz, H-5), 4.01 (*s*, 3H, 8-OMe), 3.99 (*s*, 3H, 1-OMe), 3.90 (*s*, 3H, 2-OMe); ¹³C NMR in Table 1; UV λ_{max}^{MeOH} nm (log ε) 386 (2.96), 265 (3.73), 242 (3.30); λ_{max}^{MeOH+NaOH} nm (log ε) 254 (3.85), 210 (4.57); λ_{max}^{MeOH+AlCl₃} no shift.

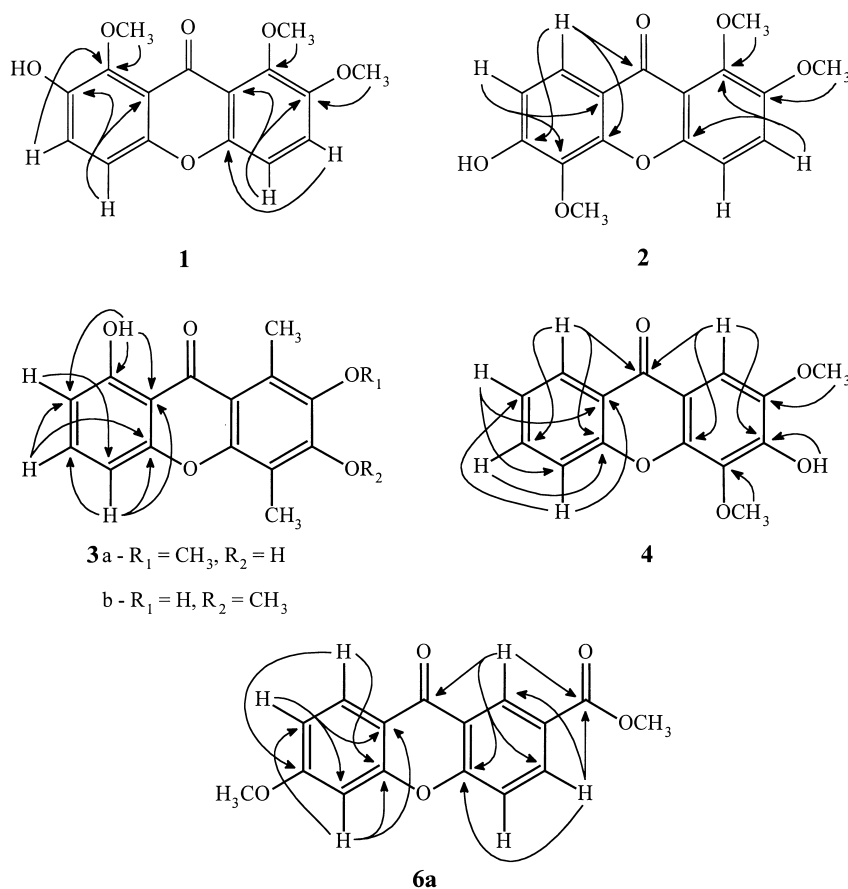


Fig. 1. HMBC of **1**, **2**, **3a**, **4** and **6a**.

3.2. 6-Hydroxy-1,2,5-trimethoxyxanthone (2)

Brown semisolid; HREIMS $[M]^+$ $m/z = 302.07907$ (calculated for $C_{16}H_{14}O_6 = 302.07904$); EIMS m/z (rel. int.) 302 $[M]^+$ (100) 287 $[M-CH_3]^+$ (82.8), 272 (10.7), 256 (31.1), 244 (68.5), 149 (16.5); 1H NMR (300 MHz, $CDCl_3$) δ 7.96 (d , $J = 8.9$ Hz, H-8), 7.32 (d , $J = 9.2$ Hz, H-3), 7.28 (d , $J = 9.2$ Hz, H-4), 6.94 (d , $J = 8.9$ Hz, H-7), 4.10 (s , 3H, 5-OMe), 3.99 (s , 3H, 1-OMe), 3.90 (s , 3H, 2-OMe); ^{13}C NMR in Table 1; UV λ_{max}^{MeOH} nm (log ϵ) 349 (3.50), 310 (3.80), 284 (3.68), 242 (4.38); $\lambda_{max}^{MeOH+NaOH}$ nm (log ϵ) 358 (4.10), 238 (4.04), 209 (4.98); $\lambda_{max}^{MeOH+AlCl_3}$ no shift.

3.3. 3,8-Dihydroxy-1,2,4-trimethoxyxanthone (3a)

Mp 170–172°C, lit 182–184°C (Habib et al., 1987) HREIMS $[M]^+$ $m/z = 318.07401$ (calculated for $C_{16}H_{14}O_7 = 318.07395$); EI MS m/z (rel. int.) 318 $[M]^+$ (78.5), 303 $[M-CH_3]^+$ (100), 288 (13.7), 275 (48.5), 260 (45.9), 149 (48.2); 1H NMR (300 MHz, $CDCl_3$) δ 13.08 (s , 8-OH), 7.52 (t , $J = 8.3$ Hz, H-6), 6.91 (d , $J = 8.1$ Hz, H-5), 6.78 (d , $J = 8.3$ Hz, H-7), 4.03, 4.01, 3.97 (each s and 3H, OMe); ^{13}C NMR in Table 1; UV λ_{max}^{MeOH} nm (log ϵ) 314 (2.94), 251 (3.29); UV $\lambda_{max}^{MeOH+NaOH}$ nm (log ϵ) 380 (3.00), 238 (2.88), 209 (3.58); $\lambda_{max}^{MeOH+NaOAc}$ nm (log ϵ) 368 (3.11), 209 (3.90); $\lambda_{max}^{MeOH+NaOAc+H_3BO_3}$ nm (log ϵ) 369 (2.97), 213 (3.93); $\lambda_{max}^{MeOH+AlCl_3}$ nm (log ϵ) 345 (2.93), 281 (3.18), 263 (2.87), 202 (3.62); $\lambda_{max}^{MeOH+AlCl_3+HCl}$ nm (log ϵ) 327 (2.90), 280 (3.18), 261 (3.27), 202 (3.63).

3.4. 2-Carbomethoxy-6-methoxyxanthone (6a)

Colorless crystals, mp 154–156°C; HREIMS $[M]^+$ $m/z = 284.06850$ (calculated for $C_{16}H_{12}O_5 = 284.06847$); EIMS m/z (rel. int.) 284 $[M]^+$ (80.4), 271 (7.5), 253 (100), 225 (25.5), 197 (11.9), 182 (16.8), 149 (10.7), 126 (16.5); 1H NMR (300 MHz, $CDCl_3$) δ 8.98 (d , $J = 2.1$ Hz, H-1), 8.24 (d , $J = 9.0$ Hz, H-8), 7.71 (dd , $J = 8.6, 1.2$ Hz, H-7), 7.48 (d , $J = 8.8$ Hz, H-3), 6.96 (dd , $J = 9, 2.5$ Hz, H-4), 6.89 (d , $J = 2.3$ Hz, H-5), ^{13}C NMR in Table 1; λ_{max}^{MeOH} nm (log ϵ) 303 (3.61), 244 (4.13). Addition of NaOH and $AlCl_3$ caused no shift.

Acknowledgements

Work in Portugal was supported by Fundação para a Ciência e Tecnologia (Unidade de I&D No. 226/94) FEDER and PRAXIS XXI. We thank the Aveiro University Mass Spectrometry Group for low resolution mass spectra and Dr Graham Eaton, Department of Chemistry, University of Leicester, UK, for providing HRMS.

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