

PHYTOCHEMISTRY

Phytochemistry 60 (2002) 375-379

www.elsevier.com/locate/phytochem

Flavonoids from Tephrosia aequilata

Paul K. Tarus, Alex K. Machocho, Caroline C. Lang'at-Thoruwa, Sumesh C. Chhabra*

Chemistry Department, Kenyatta University, PO Box 43844, Nairobi, Kenya

Received 29 May 2001; received in revised form 7 February 2002

Abstract

From the roots of the plant *Tephrosia aequilata* Baker, five flavonoids were isolated of which, 3,4:8,9-dimethylenedioxypterocarpan is reported for the first time. Its structure and those of the already known flavonoids were established by physical and spectroscopic analysis. Application of 2D NMR techniques was useful for complete characterization of the new pterocarpan as well as the other known flavonoids. © 2002 Elsevier Science Ltd. All rights reserved.

Keywords: Tephrosia aequilata; Papilionaceae; Roots; Parasitic protozoa; Antimicrobial activity; 3,4:8,9-Dimethylenedioxypterocarpan; Praecansone A; Praecansone B; Z-Praecansone A; Demethylpraecansone B

1. Introduction

Tephrosia (Papilionaceae) is a large genus of perennial woody shrubs, which are well distributed in the tropical and sub-tropical regions of the world (Gillet et al., 1971). Between 300 and 400 species are known (Willis, 1973), of which 35 occur in India, 30 are native to South America, 70 are found in South Africa and 50 in equatorial Africa of which 30 are found in Kenya (Chadra, 1976; Allen and Allen, 1981; Beentje, 1994).

Some of the species have been used in herbal remedies, insecticides and rat, fish and human poisons by the various indigenous people of Kenya (Gillet et al., 1971; Watt and Breyer-Brandwijk, 1962). Phytochemical studies have been carried out on the roots of some Tephrosia species. For example, the roots of T. emoroides A. Rich., yielded 4",5"-dihydro-5-methoxy-5"-isophenylfurano-[2",3",7,8]-flavanone which showed insect antifeedant activity against the larvae of stalk borer, Chillo partellus (Machocho et al., 1995). The roots of T. hildebrandtii Vatke yielded a pterocarpan, hildecarpin which exhibited antifeedant activity against the legume pod-borer, Maruca testulalis (Lwande et al., 1986). T. interrupta Engl. and T. linearis (Willd.) Pers. both yielded various rotenoids including deguelin and rotenone (Were, 1988). There has been no phytochemical

E-mail address: scchhabra@avu.org (S.C. Chhabra).

investigation of *T. aequilata*. The roots of *T. aequilata* are used to treat venereal diseases and the leaves to relieve abdominal pains (Kokwaro, 1993; Gillet et al., 1971).

In the present investigation, five flavonoids were isolated from the petrol (bp 40–60 °C) extract of the roots of *T. aequilata*. The pterocarpan, 3,4:8,9-dimethylenedioxypterocarpan (1) is reported for the first time. The β-oxygenated chalcones, praecansone A (2) and praecansone B (3) have been reported earlier from *T. praecans* Brummitt (Camele et al., 1980), *T. procumbens* (Venkataratnam et al., 1987) and *T. pumila* Lam. (Dagne et al., 1988; Yenesew et al., 1989). The *Z*-isomer of praecansone A (4) is also reported as a plant metabolite while demethylpraecansone A (5) was reported previously from *Lonchocarpus costaricensis* Pittier (Waterman and Mahmoud, 1985).

2. Results and discussion

The five flavonoids were isolated from the petrol (bp 40–60 °C) extract of the roots by a combination of chromatographic techniques followed by crystallization.

The ¹H NMR spectral data of **1** (Table 1) suggested a pterocarpan structure due to the splitting pattern of the protons of the heterocyclic ring B and the bridging protons of B and C rings (Màximo and Lourenço, 1998; Machocho et al., 1995; Pachler and Underwood, 1967). The shifts appeared at δ 5.43 d, 4.23 dd, 3.64 t and 3.45

^{*} Corresponding author. Tel.: +254-2-810901; fax: +254-2-811224

3

ddd, which were assigned to H-11a, H-6ax, H-6eq and H-6a, respectively. The spectrum exhibited four aromatic protons in which the two adjacent protons in positions 1 and 2 appeared at δ 6.96 d and 6.55 d, respectively, with J values of 8.1 Hz on the tetra-substituted ring A. The other two para-oriented signal at δ 6.66 s and 6.37 s were assigned to protons of positions 7 and 10, respectively, on ring D. The assignment of these sets of aromatic protons was supported by the ROESY spectrum, whereby H-1 showed spatial contours with H-11a and H-2, and H-7 with H-6ax, H-6eq and H-6a. These assignments were supported by 3-bond correlation in the HMBC spectrum. The two sets of aromatic protons suggested the placement of the two methylenedioxy groups at δ 5.84 (2d, J=1.5 Hz) and 5.93 s at positions 3,4 and 8,9, respectively. Additionally, the two pairs of methylenedioxy protons showed 3-bond correlation in the HMBC spectrum with their respective aromatic carbon atoms.

The 13 C NMR spectral data of 1 (Table 1) indicated 17 carbon atoms and was in agreement with the proposed structure. The HMQC and HMBC spectra confirmed structure 1 for the new pterocarpan. The HMQC spectrum provided the assignment of the protonated aromatic carbons as follows: δ 123.5 (C-1), 102.1 (C-2), 104.1 (C-7) and 93.3 (C-10). The protonated carbons of the heterocyclic rings were observed at δ 77.6 (C-11a), 66.0 (C-6) and 39.6 (C-6a), which compared closely with literature values (Màximo and Lourenço, 1998; Lwande et al., 1986). The methylenedioxy carbons appeared at δ 101.2 and 100.7. The quaternary aromatic carbons were assigned with the help of the HMBC spectrum. The

Table 1
The ¹H NMR, HMQC and HMBC spectral data for compound 1

Proton	¹ H NMR (<i>J</i> in Hz)	Correlated C-atom	
		HMQC	НМВС
1	6.96 d (8.1)	123.5 d	C-11a, C-4a, C-3
2	6.55 d (8.1)	102.1 d	C-11b, C-4
		143.0 s (C-3)	
		143.3 s (C-4)	
		166.8 s (C-4a)	
6 (eq)	4.23 dd	66.0 t	C-11a, C-6b, C-4a
	(10.8, 5.1)		
6 (ax)	3.64 t (10.8)	66.0 t	C-11a, C-6b, C-4a
6a	3.45 <i>ddd</i>	39.6 d	C-11b, C-10a
	(10.8, 6.9, 5.1)		
		117.0 s (C-6b)	
7	6.66 s	104.1 d	C-10a, C-9, C-6a
		148.3 s (C-8)	
		148.3 s (C-9)	
10	6.37 s	93.3 d	C-8, C-6b
		160.0 s (C-10a)	
11a	5.43 d (6.9)	77.6 d	C-10a, C-6b, C-4a
		114.8 s (C-11b)	
3,4-OCH ₂ O-	5.84 d (1.5)	100.7 t	C-3, C-4
8,9-OCH ₂ O-	5.93 s	101.2 t	C-9, C-8

The carbon multiplicities were determined by DEPT data.

mass spectrum of **1** showed an $[M]^+$ at m/z 312 for $C_{17}H_{12}O_6$ thus confirming the above deductions based on NMR data analysis.

The ¹H NMR and ¹³C NMR spectra of **2** are similar to those of praecansone A which had earlier been reported from T. praecans as praecansone A (Camele et al., 1980), and later the structure was revised by Dagne et al. (1988). Compounds 2 and 4 had different physical and chromatographic properties but had closely related UV and NMR spectral data. The ¹H NMR spectra of 2 and 4 were similar with minor variations. Each compound had three sets of methoxyl groups, two isolated protons, and unsubstituted benzene ring and a dimethyl chromene ring as the prenyl substitution. In the ¹H NMR of 2 the olefinic signal at δ 6.44 for H-8 (H- α) showed ROESY contours with H-2 or H-6 of the unsubstituted benzene ring and one of the methoxyl groups at δ 3.86, which was assigned to position 9. This implied that this particular methoxyl group was in the trans-orientation with respect to the hydrogen. One of the other methoxyl groups at δ 3.72 showed a spatial correlation with H-4" signal at δ 6.50 and this was assigned to position 6'. The remaining methoxyl group at δ 3.66 was assigned to position 2' based on the spatial correlation with the lone aromatic proton of position 3' at δ 6.19. The ¹³C NMR spectrum of **2** is identical to that of Praecansone A occurring as the E-isomer especially the 13 C NMR peak at δ 101.2 (C-8) as reported by Kiuchi et al. (1990). No spatial correlation in the ROESY spectra was observed between the olefinic proton at position 8 (H-α) with the methoxyl group at position 9 in 4. The 1 H NMR spectra of 4 showed the signal of H-8 (H-α) appearing at δ 6.03 which seems to have moved upfield compared to a similar proton in 2 while in the 13 C NMR, the C-8 signal has moved down field to δ 104.8. It therefore appears that 4 is the Z-isomer of praecansone A (2) when compared to the 1 H NMR and 13 C NMR data of the β-methoxychalcones (Kiuchi et al., 1990). The EIMS spectra of 2 and 4 had [M] $^{+}$ at m/z 380 for C₂₃H₂₄O₅. The base peaks at m/z 349 represented lose of methoxyl groups. A peak at m/z 365 in both compounds suggested loses of methyl groups from the molecular ions.

In vitro biological activity tests against parasitic protozoa were performed on the isolated flavonoids. Compounds **3** and **5** showed low activity against *Trypanosoma brucei rhodensiense* (strain STIB 900, stage trypomastigotes) with IC₅₀ 5.9 and 5.1 μg/ml, respectively, and *Trypanosoma cruzi* (strain Tulahuen C4, stage trypomastigotes) with IC₅₀ 7.6 and 6.0 μg/ml, respectively. The compounds exhibited no cytotoxicity towards L-6 cells and macrophages but showed considerable activity against *Leishmania donovani* (strain MHOM-ET-67, stage amastigotes) with at IC₅₀ values 17.2 and 9.0 μg/ml, respectively. Compounds **2** and **4** exhibited no activity against any of the parasitic protozoa or against *Plasmodium falciparum* (strain K1 and NF54, stages IEF).

Compounds 1–3 exhibited low activity against grampositive bacteria, Bacillus subtilis and Micrococcus lutea and 4 and 5 even less activity (≤ 8 mm). Compound 3 showed an inhibition zone of 11 and 13 mm against B. subtilis and M. lutea, respectively. The crude petrol (bp 40-60 °C) extract was inactive against the gram-negative bacteria, Escherichia coli and Pseudomonas aureginosa when tested at 100 µg/disk while 5 showed some activity (inhibition zone = 10 mm) against these bacteria. The other compounds showed slight or no activity against the gram-negative bacteria. The compounds were not active against the fungus, Aspergillus niger and the yeast, Saccharomyces cerevisiae. The antibacterial activities of these flavonoids were much lower than those observed for the standard antibiotics especially cotrimoxazole, streptomycin, kanamycin, gentamycin and chloramphenicol.

3. Experimental

3.1. General experimental procedures

Mps were uncorrected IR spectra: Perkin-Elmer 598 FTIR series spectrometer in KBr pellet. UV: Perkin-Elmer lambda 16 UV/vis spectrometer in MeOH. EIMS: Hewlett Packard 5989 A mass spectrometer at 70 eV with direct probe insert at 120–140 °C. NMR:

Varian VXR 500; CDCl₃ at 500 MHz for 1H NMR and 75 MHz for ^{13}C NMR with TMS as int. standard and the chemical shifts reported in δ (ppm) units relative to TMS signal and coupling constants (J) in Hz. Silica gel SDS chromagel 60 A CC (6–35 μ m) was used for VLC, and silica gel 60 F₂₅₄ (Macherey-Nagel) for analyt. (0.25 mm) and prep. (0.25 mm) TLC. Spots on chromatograms were detected under UV light (254 and 365 nm) and by spraying with 25% aqueous H₂SO₄.

3.2. Plant material

The roots of *T. aequilata* Baker were collected at the summit of Nzaui Hills in Makueni District, Kenya in December, 1999. The sample was authenticated by Mr. Simon Mathenge, Botany Department, University of Nairobi, Kenya. A voucher specimen (SM/PKT/02/99) has been deposited in the herbarium, Nairobi University, Nairobi.

3.3. Extraction and isolation

Air-dried roots (1.64 kg) were extracted with petrol (bp 40–60 °C). After evaporation of solvents, a yellow paste (11.3 g) was obtained and the dried extract was chromatographed on a silica gel by VLC and eluted with *n*-hexane–EtOAc mixtures from the ratio of 9:1 to 1:1 to obtain fractions A to D. When fraction A was repeatedly recrystalized in Me₂CO, 5 (30.0 mg) was obtained as yellow needle-like crystals. fraction B was rechromatographed on silica gel and eluted with petrol (bp 60–80 °C): EtOAc (9:1) which on recrystallization in Me₂CO yielded 1 (19.1 mg). Fraction C from the VLC column was twice subjected to silica gel prep. TLC eluting with CHCl₃:EtOAc (1:1) and CHCl₃ (100%), respectively, to yield 3 (11.7 mg). Fraction D was rechromatographed on silica gel by VLC with petrol (bp 60–80 °C):EtOAc (4:1) as eluant and was later subjected to silica gel prep. TLC with CHCl₃: EtOAc (1:1) as eluant to yield 2 (22.7 mg) and 4 (17.2 mg).

3.4. 3,4-Dimethylenedioxypterocarpan (1)

White crystals, mp 154–156 °C. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 2940, 1610, 1480, 1460, 1360, 1150, 1050, 1030, 1010, 930, 830. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 209, 239, 304. ¹H NMR spectral data (500 MHz, CDCl₃) and ¹³C NMR spectral data (75 MHz, CDCl₃): cf. Table 1. EIMS (probe) 70 eV, m/z (rel. int.): 312 [M]⁺ (100), 295 (13), 225 (6), 175 (16), 165 (10), 162 (26), 149 (21), 139 (11), 125 (8), 111 (13), 97 (18), 85 (17), 83 (20), 81 (15), 71 (28), 69 (31), 57 (46), 55 (31).

3.5. Praecansone A (2)

Yellow oil (17.9 mg), IR ν_{max}^{KBr} cm $^{-1}$: 2940, 1610, 1480, 1460, 1360, 1150, 1050, 1030, 1010, 930,830. λ_{max}^{MeOH} nm:

209, 239, 304. ¹H NMR spectral data (500 MHz, CDCl₃) and ¹³C NMR spectral data (75 MHz, CDCl₃): δ 7.83 (2H, *m*, H-2, H-6), 7.42 (1H, *m*, H-4), 7.37 (2H, *m*, H-3, H-5), 6.50 (1H, *d*, *J* = 10 Hz, H-4"), 6.44 (1H, *s*, H-8), 6.19 (1H, *s*, H-3'), 5.43 (1H, *d*, *J* = 10 Hz, H-3"), 3.86 (3H, *s*, 9-OMe), 3.72 (3H, *s*, 6'-OMe), 3.66 (3H, *s*, 2'-OMe), 1.42 (6H, *s*, 2"-Me₂). ¹³C xNMR spectral data (75 MHz, CDCl₃): 190.2 (C-7), 166.1 (C-9), 157.7 (C-4'), 154.5 (C-6'), 155.6 (C-2'), 139.7 (C-1), 131.6 (C-4), 128.0 (C-2, C-6), 127.7 (C-3, C-5), 127.0 (C-4"), 116.8 (C-3"), 111.9 (C-5'), 101.2 (C-8), 96.0 (C-3'), 76.5 (C-2"), 62.2 (2'-OMe), 56.2 (9-OMe), 55.8 (6'-OMe), 28.0 (2"-Me₂). EIMS (probe) 70 eV, *m/z* (rel. int.): 380 [M]⁺ (6), 365 (20), 349 (100), 335 (8), 319 (18), 245 (5), 217 (4), 167 (14), 105 (22), 77 (17).

3.6. Praecansone B (3)

Yellow oil (17.9 mg), IR ν_{max}^{KBr} cm⁻¹: 2980 and 3009, 1670, 1610, 1560, 1480, 1370, 1200, 1150, 1130. UV λ_{max}^{MeOH} nm: 274, 309. ¹H NMR spectral data (500 MHz, CDCl₃) and ¹³C NMR spectral data (75 MHz, CDCl₃): δ 7.90 (2H, m, H-2, H-6), 7.50 (1H, m, H-4), 7.40 (2H, m, H-3, H-5), 6.50 (1H, d, J=10 Hz, H-4"), 6.47 (1H, s, H-8), 6.23 (1H, s, H-3'), 5.50 (1H, d, J=10 Hz, H-3"), 3.78 (3H, s, 6'-OMe), 3.76 (3H, s, 2'-OMe), 1.43 (6H, s, 2"-Me₂). ¹³C NMR spectral data (75 MHz, CDCl₃): 188.0 (C-9), 182.0 (C-7), 158.2 (C-4'), 156.2 (C-2'), 155.0 (C-6'), 134.0 (C-1), 132.1 (C-4), 128.5 (C-2, C-6), 127.7 (C-4"), 127.0 (C-3, C-5), 116.5 (C-3"), 114.0 (C-5'), 100.5 (C-8), 96.2 (C-3'), 76.4 (C-2"), 63.2 (6'-OMe), 56.1 (2'-OMe), 28.0 (2"-Me₂). EIMS (probe) 70 eV, m/z (rel. int.): 366 [M]⁺ (16), 351 (100), 335 (95), 321 (4), 305 (11), 247 (16), 231 (9), 217 (17), 205 (52,), 190 (7), 175 (7), 160 (7), 105 (25), 91 (6), 77 (29), 69 (14), 51 (6).

3.7. cis-Praecansone A (4)

Yellow oil (17.9 mg), Found [M]⁺ 380.1537; $C_{23}H_{24}O_5$ Calc. for 380.1624. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1660. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 284. ¹H NMR spectral data (500 MHz, CDCl₃): δ 7.89 (2H, m, H-2, H-6), 7.43 (1H, m, H-4), 7.38 (2H, m, H-3, H-5), 6.50 (1H, d, J=10 Hz, H-4"), 6.22 (1H, s, H-3'), 6.03 (1H, s, H-8), 5.52 (1H, d, J = 10 Hz, H - 3''), 3.78 (3H, s, 9-OMe), 3.70 (3H, s, 6'-OMe), 3.65 (3H, s, 2'-OMe), 1.44 (6H, s, 2"-Me₂). ¹³C NMR spectral data (75 MHz, CDCl₃): 189.0 (C-7), 164.0 (C-9), 158.3 (C-4'), 156.0 (C-6'), 155.0 (C-2'), 140.0 (C-1), 131.5 (C-4), 128.1 (C-2, C-6), 127.8 (C-3, C-5), 127.6 (C-4"), 116.1 (C-3"), 110.0 (C-5'), 104.8 (C-8), 95.7 (C-3'), 76.4 (C-2"), 62.2 (2'-OMe), 57.3 (9-OMe), 55.9 (6'-OMe), 28.1 (2"-Me₂). EIMS (probe) 70 eV, m/z (rel. int.): 380 [M]⁺ (11), 365 (19), 349 (100), 335 (8), 319 (16), 245 (4), 217 (3), 167 (9), 105 (12), 77

3.8. Demethylpraecansone B (5)

Yellow needle-like crystals from *n*-hexane-EtOAc (9:1), (17.9 mg), mp 126-129 °C, lit. (Waterman and Mahmoud, 1985). Found [M]⁺ 352.1317; C₂₁H₂₀O₅ Calc. for 352.1311. $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1640, 1580, 1280. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 279, 369. ¹H NMR spectral data (500 MHz, CDCl₃): δ 7.86–7.91 (2H, m, H-2, H-6), 7.45–7.50 (3H, m, H-3, H-4, H-5), 7.32 (1H, s, H-8), 6.69 (1H, d, J=10Hz, H-4"), 5.95 (1H, s, H-3'), 5.47 (1H, d, J = 10 Hz, H-3"), 3.92 (3H, s, 2'-OMe), 1.46 (6H, s, 2"-Me₂). ¹³C NMR spectral data (75 MHz, CDCl₃): 194.0 (C-9), 175.8 (C-7), 162.0 (C-4'), 160.5 (C-2'), 159.9 (C-6'), 133.5 (C-1), 131.6 (C-4), 128.7 (C-2, C-6), 128.1 (C-3, C-5), 126.6 (C-4"), 116.1 (C-3"), 104.0 (C-3'), 103.0 (C-1'), 98.2 (C-8), 91.8 (C-5'), 76.4 (C-2"), 55.9 (2'-OMe), 28.4 $(2''-Me_2)$. EIMS (probe) 70 eV, m/z (rel. int.): 352 [M]⁺ (17), 337 (40), 319 (3), 232 (3), 217 (100), 202 (6), 191 (24), 105 (20), 77 (25), 69 (5), 51(4).

3.9. Parasitic assay

The assays to determine activity of the flavonoids for *T. brucei rhodensiense*, *T. cruzi* and *Leishmania donovani* were carried out according to the method of Räz et al. (1997) and Baltz et al. (1985). The antiplasmodium activity was determined as described by Ridley et al. (1996).

3.10. Antimicrobial assay

The bioassay for antimicrobial activity was carried out by agar diffusion assay method as described by Barry et al. (1979).

Acknowledgements

The authors are grateful to Dr. C. Codina and the staff at Department of Natural Products, Faculty of Pharmacy, University of Barcelona, Spain for running the NMR and MS spectra. P.K.T. thanks German Academic Exchange services (DAAD) for financial support for this research. Thanks are also due to Reto Brun of the Swiss Tropical Institute, Basel, Switzerland for carrying out the parasitic assays and finally to Mr. Simon Mathenge, Botany Department, University of Nairobi, Kenya for the authentication and collection of the plant material.

References

Allen, O.N., Allen, E.K., 1981. The Leguminoseae. The University of Wisconsin Press, Wisconsin, pp. 645–649.

- Baltz, T., Baltz, D., Goroud, C., Crocket, J., 1985. Cultivation in a semi-defined medium of animal infective forms of *Trypanosoma* brucei, T. equiperdum, T. evansi, T. rhodensiense, T. gambiense. EMBO Journal 4, 1273–1277.
- Barry, A.L., Coyle, M.B., Thornsberry, C., Gerlach, E.H., Hawkinson, R.W., 1979. Methods of measuring zones of inhibition with the, Bauer–Kirby disk susceptibility test. Journal of Clinical, Microbiology 10, 885–889.
- Beentje, H.J., 1994. Kenya trees shrubs and ianas. National Museums of Kenya, Nairobi, pp. 311–312.
- Camele, G., Monache, F.D., Monache, G.D., Marini-Betolo, G.B., 1980. Three new flavonoids from *Tephrosia praecans*. Phytochemistry 19, 707–709.
- Chadra, M. (Ed.), 1976. The Wealth of India. Publications and Information Directorate. CSIR, New Delhi, pp. 151–157.
- Dagne, E., Dinku, B., Gray, C.D., Waterman, P., 1988. Pumilaisoflavones A, B from the seedpods of *Tephrosia pumila*. Phytochemistry 27, 1503–1505.
- Gillet, J.B., Polhill, R.M., Verdcourt, B., 1971. Flora of tropical East Africa. The Government Printers, Nairobi, pp. 501.
- Kiuchi, F., Chen, X., Tsuda, Y.Z.-E., 1990. Isomerization of β-methoxychalcones: preferred existence of *E*-isomers in naturally occurring β-methoxychalcones. Chemical and Pharmaceutical Bulletin 38, 1862–1871.
- Kokwaro, J.O., 1993. Medicinal plants of East Africa. East African Literature Bureau, Nairobi, pp. 250–255.
- Lwande, W., Bentley, M.D., Hassanali, A., 1986. The structure of hildecarpin, an insect antifeedant 6a-hydroxypterocarpin from the roots of *Tephrosia hildebrandtii* Vatke. Insect, Science, Application 7, 501–503.
- Machocho, A.K., Lwande, W., Jondiko, J.I., Moreka, L.V.C., Hassanali, A., 1995. Three new flavonoids from the roots of *Tephrosia emoroides*, their antifeeding activity against the larvae of the spotted stalk borer *Chilo partellus* Swinhoe. Journal of Pharmacognosy 33, 222–227.
- Màximo, P., Lourenço, A., 1998. A pterocarpan from *Ulex parviflorus*. Phytochemistry 48, 359–362.
- Pachler, K.G.R., Underwood, W.G.E., 1967. NMR spectrum calculated data for the heterocyclic protons of (–)-pterocarpin. Tetrahedron 23, 1817.
- Räz, B., Iten, M., Grether-Bühler, Y., Kaminsky, R., Brun, R., 1997. The Alamar and Blue assay to determine drug sensitivity of African trypanosomes (*T. b. rhodensiense*, *T. b. gambiensiense*) in vitro. Acta Tropica 68, 139–147.
- Ridley, R.G., Hofheinz, W., Matile, H., Jacquet, C., Dorn, A., Masciadri, R., Jolidon, S., Richter, W.F., Guenzi, A., Girometta, M.A., Urwyler, H., Huber, W., Thaitong, S., Peters, W., 1996. Antimicrobial agents. Chemotherapy 40, 1846–1854.
- Venkataratnam, G., Rao, E.V., Vilain, C., 1987. Flavonoids of Tephrosia procumbens—revised structure for praecansone A conformation of Praecansone B. Journal of the Chemical Society Perkin Transactions 1, 2723–2727.
- Waterman, P., Mahmoud, E.N., 1985. Flavonoids from the seeds of Lonchocarpus costaricensis. Phytochemistry 24, 571–574.
- Watt, M.J., Breyer-Brandwijk, M.G., 1962. The Medicinal and Poisonous Plants of Southern and Eastern Africa, 1st ed. Livingstone, London, pp. 653–663.
- Were, O., 1988. Isolation and Chemical Characterization of Flavonoids from *Tephrosia interrupta* and *Tephrosia linearis*. MSc thesis, University of Nairobi, Nairobi.
- Willis, J.C., 1973. Dictionary of Flowering Plant and Ferns. University Press, Cambridge, 1135 pp.
- Yenesew, A., Dagne, E., Waterman, P.G., 1989. Flavonoids from the seed pods of *Tephrosia pumila*. Phytochemistry 28, 1291–1292.