



Squamocin-O₁ and squamocin-O₂, new adjacent bis-tetrahydrofuran acetogenins from the seeds of *Annona squamosa*

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Abstract

Two bis-tetrahydrofuran acetogenins, squamocin-O₁ (**1**) and squamocin-O₂ (**2**), were isolated from a MeOH extract of seeds of *Annona squamosa* L. Their structures were determined by spectral means including precursor-ion scanning mass spectral analysis for their aminor derivatives. The configurations at the oxymethine chiral centers were assigned as 12*R*,15*R*,16*R*,19*R*,20*R*,23*R*,24*S*,28*S*,36*S* for **1** and 12*S*,15*R*,16*R*,19*R*,20*R*,23*R*,24*S*, 28*S*,36*S* for **2**, based on ¹H NMR analysis of their Mosher's ester derivatives and CD data.

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1. Introduction

Annonaceous tetrahydrofuran acetogenins have attracted much interest due to their broad range of biological activities (Alali et al., 1999). *Annona squamosa* L. (Annonaceae) is well known for its edible tropical fruits and as custard apple, and its seeds are reported to have insecticidal and abortifacient properties (Chopra et al., 1956). In a continuation of previous studies, two adjacent bis-tetrahydrofuran acetogenins named squamocin-O₁ (**1**) and squamocin-O₂ (**2**) (Fig. 1) were isolated from a fraction of the MeOH extract. Previously, the isolation and structure elucidation of more than twenty acetogenins from the seeds of *Annona squamosa* L., among which squamocin (**3**) and squamostatin-A were two major constituents (Fujimoto et al., 1988, 1990, 1994; Sahai et al., 1994; Araya et al., 1994a,b).

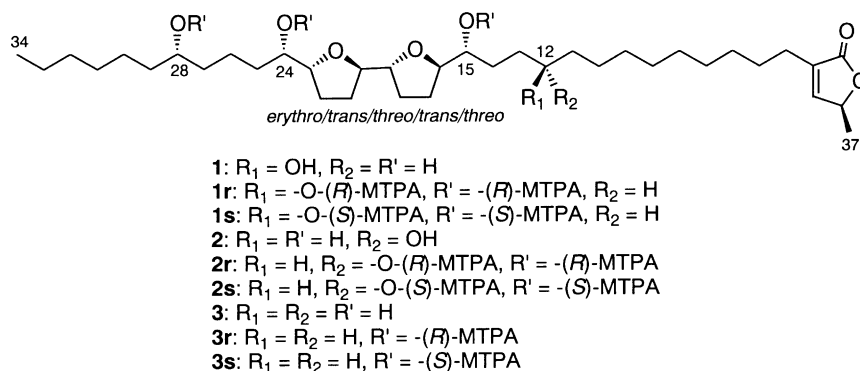
2. Results and discussion

Compounds **1** and **2** were initially obtained as a mixture (planar structure of this mixture was briefly reported: Araya et al., 1994c) by reversed phase HPLC from a fraction which was more mobile than squamostatin-A. The separation of the mixture was achieved by reversed-phase HPLC with MeOH–CH₃CN–H₂O–iPrOH (120:40:30:1) as an eluting solvent, to afford the more mobile squamocin-O₁ (**1**) and the less mobile squamocin-O₂ (**2**).

Compounds **1** and **2** showed UV (210 nm) and IR (1750 cm⁻¹) absorptions typical of α,β-unsaturated-γ-lactone moiety of annonaceous acetogenins. The same molecular formula, C₃₇H₆₆O₈, was assigned to **1** and **2** on the basis of HR-FAB-MS data. The ¹H and ¹³C NMR spectra of compounds **1** and **2** resembled those of **3**, but had additional oxymethine signal (δ_H 3.60/δ_C 71.5 for **1** and 3.58/71.7 for **2**). The spectral data suggested that compounds **1** and **2** were mono-hydroxylated analogs of **3**. The EI-MS spectra indicated that compounds **1** and **2** belong to an adjacent bis-tetrahydrofuran family and the bis-tetrahydrofuran moiety is located

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Fig. 1. Structures of squamocin- O_1 (1), squamocin- O_2 (2) and squamocin (3).

from C-15 to C-24 (Fig. 2). Further, the fragmentation pattern suggested that the extra hydroxyl group should be located along the methylene chain between the lactone and bis-tetrahydrofuran moieties, although routine mass analysis failed to assign the position of the hydroxyl group.

We previously developed a precursor ion-scanning method for mass spectral analysis of acetogenin aminal derivatives (Hirayama et al., 1993; Araya et al., 1994c). Application of this method was found to be useful in establishing the position of the hydroxyl group. The *N,N*-dimethylethylenediamine derivative of **1** (**1a**, for the structure, see Fig. 3) clearly showed ions at m/z 293 and 323 due to the fission C11–C12 and C12–C13 followed by dehydration, respectively, in the precursor-ion spectrum from m/z 72 ion $[\text{CH}_2=\text{CHN}^+\text{H}(\text{CH}_3)_2]$ (Fig. 3). Thus, the hydroxyl group was unequivocally assigned to the C-12 position. The spectrum also confirmed the positions of the bis-tetrahydrofuran moiety and C-28 hydroxyl group. The precursor ion-scanning spectrum of the *N,N*-dimethylethylenediamine derivative of **2** was essentially same as that of **1a**, confirming that **1** and **2** have the same planar structure.

The configurations at the oxymethine centers were established as follows. The stereochemistry around the bis-tetrahydrofuran moiety of **1** and **2** was readily determined to be *threo/trans/threo/trans/erythro* (from C-15 to C-24) by comparing their ^1H and ^{13}C NMR spectral data (Table 1) with those of **3** and stereochemically defined bis-tetrahydrofuran acetogenins (Sahai et al., 1994). The absolute configuration at C-36 was established to be *S*, typical for annonaceous acetogenins, on the basis of negative Cotton effects at 239

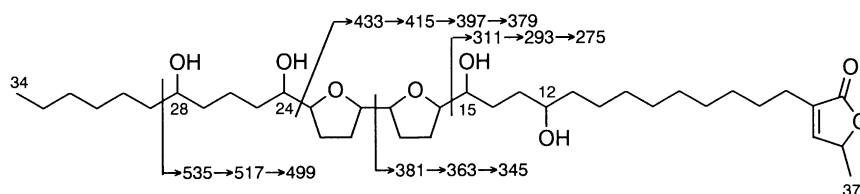
Table 1
 ^{13}C NMR spectral data for squamocin- O_1 (1) and squamocin- O_2 (2)

C	1	2
1	173.9	173.9
2	134.3	134.3
3	25.1	25.1
4	27.3	27.4
5–9	a	a
10	25.8	25.8
11	37.5	37.7
12	71.5	71.7
13	33.5	34.3
14	a	a
15	74.3	74.6
16	83.1	83.2
17	28.4	28.4
18	28.9	29.0
19	82.2 ^b	82.2 ^c
20	82.5 ^b	82.5 ^c
21	28.9	29.0
22	24.8	24.8
23	82.8	82.9
24	71.3	71.2
25	32.4	32.5
26	22.1	22.2
27	37.3	37.4
28	71.7	71.9
29	37.5	37.7
30	25.6	25.7
31	a	a
32	31.8	31.9
33	22.6	22.6
34	14.0	14.1
35	148.9	148.8
36	77.4	77.4
37	19.2	19.2

^a The signals appeared in the region of δ 29.0–30.0.

^b Assignments may be interchanged within the column.

^c Assignments may be interchanged within the column.

Fig. 2. EI-MS fragmentation pattern of squamocin- O_1 (1) and squamocin- O_2 (2).

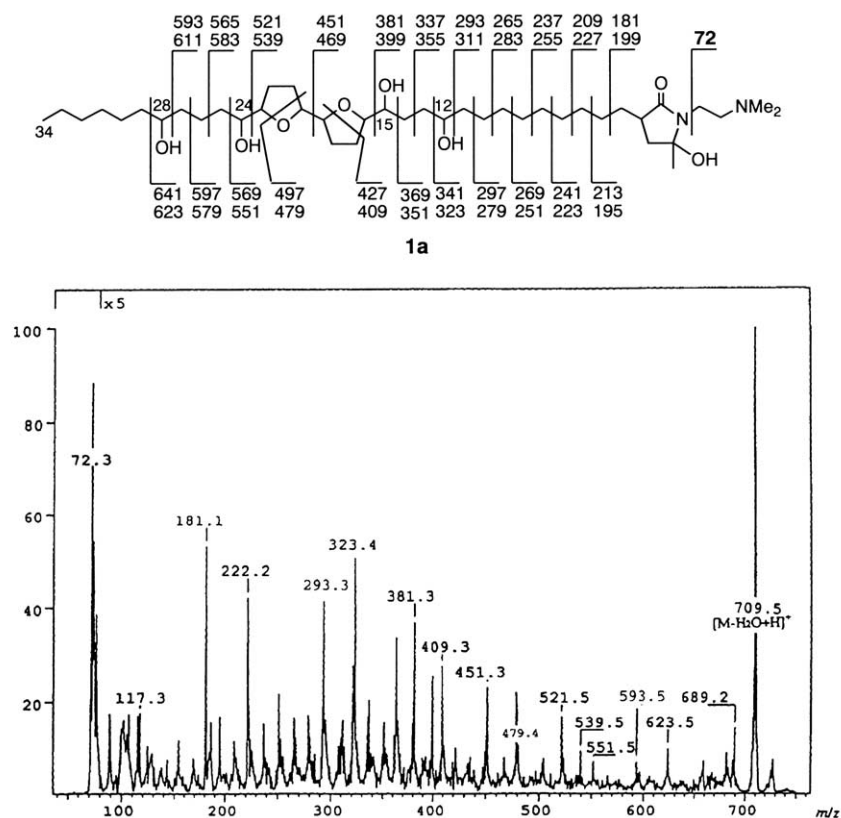


Fig. 3. Precursor ion-scanning spectrum of *N,N*-dimethylethylenediamine derivative (**1a**) of squamocin- O_1 (**1**). The accompanied m/z values are due to dehydration ions.

Table 2
 ^1H NMR spectral data for (*R*)- and (*S*)-MTPA esters (**1rs–3rs**)

No.	1r	1s	2r	2s	3r	3s
4	2.27 (t^a)	2.26 (t^a)	2.27 (t^a)	2.26 (t^a)	2.26 (t^a)	2.26 (t^a)
12	4.94 (<i>m</i>)	5.00 (<i>m</i>)	5.02 (<i>m</i>)	5.09 (<i>m</i>)	—	—
15	4.98 (<i>m</i>)	5.00 (<i>m</i>)	4.95 (<i>m</i>)	5.09 (<i>m</i>)	5.02 (<i>m</i>)	5.06 (<i>q</i>)
16	3.96 (<i>q</i>)	3.94 (<i>m</i>)	3.89 (<i>m</i>)	3.99 (<i>m</i>)	3.99 (<i>m</i>)	4.03 (<i>q</i>)
19	3.63 (<i>m</i>)	3.77 (<i>m</i>)	3.62 (<i>m</i>)	3.78 (<i>m</i>)	3.65 (<i>m</i>)	3.79 (<i>m</i>)
20	3.80 (<i>m</i>)	3.77 (<i>m</i>)	3.78 (<i>m</i>)	3.78 (<i>m</i>)	3.83 (<i>m</i>)	3.79 (<i>m</i>)
23	3.86 (<i>m</i>)	3.94 (<i>m</i>)	3.86 (<i>m</i>)	3.95 (<i>m</i>)	3.87 (<i>m</i>)	3.96 (<i>m</i>)
24	5.14 (<i>q</i> -like ^b)	5.20 (<i>q</i> -like ^b)	5.14 (<i>m</i>)	5.20 (<i>m</i>)	5.14 (<i>m</i>)	5.20 (<i>q</i> ^b)
28	5.02 (<i>m</i>)	5.00 (<i>m</i>)	5.02 (<i>m</i>)	4.99 (<i>m</i>)	5.02 (<i>m</i>)	4.99 (<i>m</i>)
34	0.874 (t^c)	0.859 (t^c)	0.874 (t^c)	0.860 (t^c)	0.874 (t^c)	0.860 (t^c)
35	6.98 (<i>brs</i>)	6.98 (<i>brs</i>)	6.98 (<i>brs</i>)	6.98 (<i>brs</i>)	6.98 (<i>brs</i>)	6.98 (<i>brs</i>)
36	4.99 (<i>m</i>)	4.99 (<i>m</i>)	4.99 (<i>m</i>)	4.99 (<i>m</i>)	4.99 (<i>m</i>)	4.99 (<i>m</i>)
37	1.40 (d^d)	1.40 (d^d)	1.39 (d^d)	1.39 (d^d)	1.41 (d^d)	1.40 (d^d)
MeO	3.48	3.49	3.51	3.50	3.51	3.51
	3.51	3.51	3.51	3.50	3.53	3.53
	3.53	3.53	3.53	3.52	3.61	3.55
	3.60	3.53	3.57	3.53		

^a $J=6.9$ Hz.

^b $J=6.6$ Hz.

^c $J=7.2$ Hz.

^d $J=6.4$ Hz.

nm in their CD spectra. The C-28 configuration of **1** and **2** was deduced from the ^1H chemical shifts of the terminal methyl group (C-34) in their (*R*)- and (*S*)-tetra-MTPA ester derivatives (**1r/1s** and **2r/2s**) (Table 2) (Nishioka et al., 1994). The (*R*)-MTPA esters **1r**, **2r** and **3r** all displayed signals due to 34- H_3 at δ 0.874, while the (*S*)-MTPA esters **1s**, **2s** and **3s** consistently exhibited the corresponding signals at δ 0.860. Thus, 28*S* configuration was assigned to compounds **1** and **2**, since compound **3** is known to have 28*S*.

Absolute configuration of the bis-tetrahydrofuran moiety of **1** was also determined to be 15*R*,16*R*,19*R*,20*R*,23*R*,24*S*, the same as previously reported for **3** (Sahai et al., 1994). The alternative 15*S*,16*S*,19*S*,20*S*,23*S*,24*R* configuration was ruled out, since the chemical shifts of 28-H, 24-H and 23-H of **1r**s and **2r**s were essentially identical to those of the respective squamocin derivatives **3r** and **3s**. The whole data discussed above revealed that compounds **1** and **2** have 15*R*,16*R*,19*R*,20*R*,23*R*,24*S*,28*S*,36*S* configuration, but are epimeric at the C-12 position. The chemical shifts of H-15 and H-16 of **1r** and **2r** were not in accord with those of **3r**, as expected from the occurrence of the additional MTPA group at C-12. Shi et al. proposed a novel application of Mosher's ester method for the determination of the absolute stereochemistry of epimeric carbinol centers (Shi et al., 1997). This method was successfully applied to compounds **1** and **2**. The

chemical shifts for pertinent ^1H signals of the MTPA esters, **1r** and **2r**, are listed in Table 3. The Table also includes the corresponding data [(*R*)-MTPA esters (**4r** and **5r**)] of structurally related acetogenins, 12-hydroxybullatacins A (**4**) and B (**5**) (Shi et al., 1997). The sign and magnitude of the values of $\Delta\delta_{\text{H}}$ (**1r–2r**) are in excellent agreement with those of $\Delta\delta_{\text{H}}$ (**4r–5r**). These studies established that compound **1** has 12*R* configuration whereas compound **2** has 12*S*. The assignments were further corroborated by comparing the ^{13}C NMR data of **1**, **2**, **4** and **5**: $\delta_{\text{C-12}}$ and $\delta_{\text{C-15}}$; 71.5, 74.3 for **1**; 71.7, 74.6 for **2**; 71.5, 74.2 for **4** (12*R*) (Shi et al., 1997); 71.8, 74.4 for **5** (12*S*) (Shi et al., 1997). The stereochemical structure of **1** (12*R*,15*R*,16*R*,19*R*,20*R*,23*R*,24*S*,28*S*,36*S*) and **2** (12*S*,15*R*,16*R*,19*R*,20*R*,23*R*,24*S*,28*S*,36*S*) are shown in Fig. 1.

Squamocin-F (**6**) (Fig. 4), isolated by our group from the seeds of *A. squamosa*, is a related 12-hydroxylated acetogenin, and the configuration at C-12 of **6** remained to be elucidated (Sahai et al., 1994). With a set of accurate ^{13}C NMR spectral data for compounds **1** and **2**, the C-12 configuration of **6** was determined as *R* (therefore the absolute configuration 12*R*,15*R*,16*R*,19*R*,20*R*,23*R*,24*S*,36*S* could be assigned), since the ^{13}C NMR values near the stereogenic center [37.6 (C-11), 71.7 (C-12), 33.5 (C-15)] are in good agreement with those of squamocin-O₁, but not with those of squamocin-O₂. It should be noted that the C-15 chemical shifts show diagnostic difference between the C-12 epimers (δ 33.5 for **1** vs 34.3 for **2**, see Table 1).

Salzmanin, recently isolated from *A. salzmanii* roots, has the same planar structure as compounds **1** and **2**. This closely related bis-tetrahydrofuran acetogenin is reported to have 12*R**,15*R**,16*R**,19*R**,20*S**,23*R**,24*S**,28*S** (*implies relative configuration) configuration (*threo/trans/erythro/cis/erythro*) (Queiroz et al., 1999).

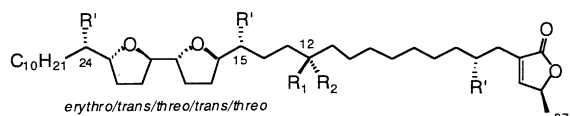
The brine shrimp toxicity (Alkofahi et al., 1989) of **1** and **2** was much less than that of **3** (LD₅₀ values: 1.0 ppm for **1**, 1.0 ppm for **2**, and 0.07 ppm for **3**). The cytotoxic activity of squamocin-O₁ (**1**) and squamocin-O₂ (**2**) against human K562 leukemia and HLE hepatoma cells were also investigated. Compounds **1** and **2** displayed much lower activity on the cancer cell lines, compared with squamocin (**1**: K562, IC₅₀ = 4.0×10^{-4} $\mu\text{g/ml}$; HLE, IC₅₀ = 3.7×10^{-3} $\mu\text{g/ml}$; **2**: K562, IC₅₀ = 4.3×10^{-4} $\mu\text{g/ml}$; HLE, IC₅₀ = 3.5×10^{-3} $\mu\text{g/ml}$; **3**: K562, IC₅₀ = 2.4×10^{-5} $\mu\text{g/ml}$; HLE, IC₅₀ = 5.0×10^{-5} $\mu\text{g/ml}$).

Table 3

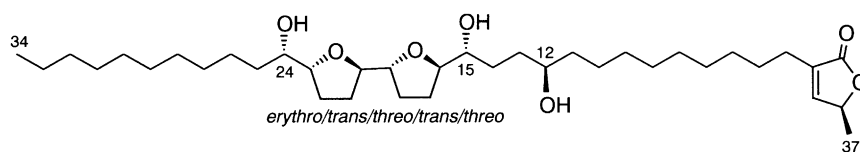
Difference in ^1H chemical shifts (ppm) between MTPA esters, **1r**, **2r**, **4r** and **5r**

Esters	H-12	H-15	H-16	H-19	H-20
1r	4.94	4.98	3.96	3.63	3.80
2r	5.02	4.95	3.89	3.62	3.78
$\Delta\delta_{\text{H}}$ (1r–2r)	–	+0.03	+0.07	+0.01	+0.02
4r ^a	4.93	4.97	3.96	3.63	3.80
5r ^a	5.00	4.95	3.89	3.62	3.79
$\Delta\delta_{\text{H}}$ (4r–5r) ^a	–	+0.02	+0.07	+0.01	+0.01

^a Adopted from Shi et al., 1997.



4r: $\text{R}_1 = \text{R}' = -\text{O}-(R)\text{-MTPA}$, $\text{R}_2 = \text{H}$
5r: $\text{R}_1 = \text{H}$, $\text{R}_2 = \text{R}' = -\text{O}-(R)\text{-MTPA}$

Fig. 4. Structure of squamocin-F (**6**).

3. Experimental

3.1. General

EI- (70 eV) and FAB-MS spectra were obtained with a JEOL JMS-AX505HA spectrometer. ^1H and ^{13}C NMR spectra were recorded on a JEOL GSX-500 spectrometer or LAMBDA-400 spectrometer in CDCl_3 with tetramethylsilane as an internal standard. IR spectra were determined on a JASCO IR-810 spectrometer. UV spectra were obtained on a Shimadzu UV-200 spectrometer. The method and derivatization for the precursor ion scanning were detailed in our previous paper (Hirayama et al., 1993). HPLC was performed on a Shimadzu LC-6A apparatus equipped with a SPD-6A UV detector (220 nm).

3.2. Isolation

In the previous HPLC separation of the MeOH extract of *A. squamosa* seeds (1 kg) (Fujimoto et al., 1994), a more mobile fraction than squamostatin-A was collected and stored. This fraction showed a broad peak about 12.5 min (14.0 min for squamostatin-A) when analyzed by HPLC (column: Shimadzu Shim-Pack CLC-ODS (25 cm \times 10 mm i.d.); solvent: MeOH– H_2O 10:1; flow rate 1.0 ml/min). The separation of the peak afforded a mixture of compounds **1** and **2** (46 mg). This was further separated by HPLC (column: STR PREP-ODS (25 cm \times 10 mm i.d.); solvent, MeOH– CH_3CN – H_2O –iPrOH (120:40:30:1); flow rate 6.0 ml/min; typical retention times, 39 min for **1**, 43 min for **2**) to furnish compounds **1** and **2**. (*R*)- and (*S*)-tetra-MTPA esters were prepared from compounds **1** and **2** (1 mg each) as described previously (Sahai et al., 1994).

3.3. Squamocin-*O*₁ (**1**)

White wax (20 mg), $[\alpha]_{\text{D}}^{25} + 17.7^\circ$ ($c = 0.6$, MeOH), CD (MeOH) $\Delta\epsilon$ (nm) -0.45 (239), UV λ_{max} (MeOH) nm (log ϵ): 210 (3.8). IR ν_{max} (CHCl_3) cm^{-1} : 3690, 3585, 3460, 1750. EI-MS m/z : 620, 602, 517, 505, 415, 397, 379, 363, 345, 293, 275, 97. HR-FAB-MS m/z : 639.4792 $[\text{M} + \text{H}]^+$ (calc. for $\text{C}_{37}\text{H}_{67}\text{O}_8$, 639.4835). ^1H NMR δ : 0.88 (3H, *t*, $J = 6.4$ Hz, H-34), 1.41 (3H, *d*, $J = 6.4$ Hz, H-37), 2.26 (2H, *t*, $J = 7.8$ Hz, H-3), 3.45 (1H, *br t*, $J = 7.8$ Hz, H-15), 3.60 (2H, *m*, H-12, H-28), 3.76–3.96 (5H, *m*, H-16, -19, -20, -23, -24), 5.00 (1H, *qq*, $J = 6.8$, 1.9 Hz, H-36), 6.99 (1H, *s*, H-35). ^{13}C NMR spectral data: Table 1. ^1H NMR spectral data of (*R*)- and (*S*)-MTPA esters: Table 2.

3.4. Squamocin-*O*₂ (**2**)

White wax (9 mg), $[\alpha]_{\text{D}}^{25} + 17.4^\circ$ ($c = 1.0$, MeOH), CD (MeOH) $\Delta\epsilon$ (nm) -0.45 (239), UV λ_{max} (MeOH) nm

(log ϵ): 210 (3.8). IR ν_{max} (CHCl_3) cm^{-1} : 3690, 3585, 3460, 1750. The EI-MS spectrum was essentially identical to that of **1**. HR-FAB-MS m/z : 639.4781 $[\text{M} + \text{H}]^+$ (calc. for $\text{C}_{37}\text{H}_{67}\text{O}_8$, 639.4835). ^1H NMR δ : 0.88 (3H, *t*, $J = 6.9$ Hz, H-34), 1.41 (3H, *d*, $J = 6.4$ Hz, H-37), 2.26 (2H, *t*, $J = 7.8$ Hz, H-3), 3.45 (1H, *br t*, $J = 7.8$ Hz, H-15), 3.58 (2H, *m*, H-12, H-28), 3.76–3.96 (5H, *m*, H-16, -19, -20, -23, -24), 5.00 (1H, *qq*, $J = 7.0$, 1.8 Hz, H-36), 6.99 (1H, *s*, H-35). ^{13}C NMR data: Table 1. ^1H NMR spectral data of (*R*)- and (*S*)-MTPA esters: Table 2.

3.5. Biological assays

Brine shrimp test was performed according to a published method (Alkofahi et al., 1989). Cytotoxic activities against K562 and HLE cells were determined according to the procedure described in our previous paper (Yoshida et al., 2001). The IC_{50} values showed in the text are the average of triplicate assays.

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