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Cytotoxic caged-polyprenylated xanthonoids and a xanthone from *Garcinia cantleyana*

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Abstract

Phytochemical studies on the leaves and trunk bark of *Garcinia cantleyana* yielded five caged-xanthonoids including one tetra- and four tri-prenylated xanthones, cantleyanone A (1), 7-hydroxyforbesione (2) and cantleyanones B–D (4–6), as well as a simple xanthone, 4-(1,1-dimethylprop-2-enyl)-1,3,5,8-tetrahydroxyxanthone (3). Eight other known compounds, deoxygaudichaudione A, gaudichaudione H, friedelin, garbogiol, macranthol, glutin-5-en-3 β -ol, and a mixture of sitosterol and stigmasterol were also isolated. Their structures were elucidated by means of spectroscopic data and comparison of their NMR data with literature values. Significant cytotoxicity against MDA-MB-231, CaOV-3, MCF-7 and HeLa cancer cell-lines was demonstrated by cantleyanones B–D, 7-hydroxyforbesione, deoxygaudichaudione A and macranthol, with IC₅₀ values ranging from 0.22 to 17.17 μ g/ml. © 2007 Elsevier Ltd. All rights reserved.

Keywords: Garcinia cantleyana; Caged-polyprenylated xanthonoids; Cantleyanones

1. Introduction

In the course of our ongoing search for bioactive compounds from natural sources, we have screened eleven *Garcinia* species found in the Malaysia Peninsula for cytotoxic activity. Among the species showing strong activity was the MeOH extract of *Garcinia cantleyana* Whitmore. The species is endemic to Malaysia, found mostly in the upper mountain forest (Whitmore, 1983). It is a small tree, occasionally reaching 15 m tall and 90 cm girth. The inner bark is thick, opaque, and produces sulphur-yellow exudates. The crude chloroform extract of leaves and trunk barks of *G. cantleyana* were both found to be cytotoxic

against cancer cell-lines Fig. 1.

against MDA-MB-231, MCF-7, CaOV-3, and HeLa celllines with IC₅₀ values ranging from 2.33 to 8.33 μ g/ml. The leaf extract afforded the new tetra-prenylated caged-

xanthone cantleyanone A (1), 7-hydroxyforbesione (2),

anthone (3), in addition to the known compounds deoxy-

gaudichaudione A (Han et al., 2006), garbogiol (Iinuma

et al., 1998), macranthol (Kouno et al., 1991), and glutin-

5-en-3β-ol (Leong and Harrison, 1999). Meanwhile the

three new tri-prenylated caged-xanthones, cantleyanones

tion of compounds 1-6, and their cytotoxic properties

4-(1,1-dimethylprop-2-enyl)-1,3,5,8-tetrahydroxyx-

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B–D (4–6) were isolated from the trunk bark extract together with two known compounds; gaudichaudione H (Cao et al., 1998) and friedelin. Macranthol which also exhibited strong cytotoxic activity was isolated for the first time from *Garcinia*. Herein, we report the structure elucida-

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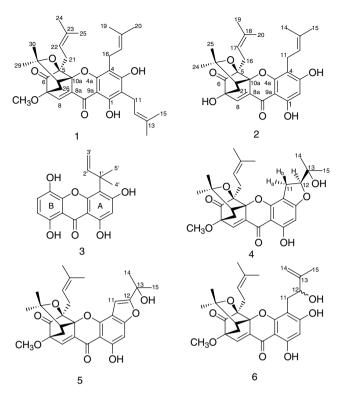


Fig. 1. Structures of cantleyanone A (1), 7-hydroxyforbesione (2), 4-(1,1-dimethylprop-2-enyl)-1,3,5,8-tetrahydroxyxanthone (3) cantleyanones B–D (4-6).

2. Results and discussions

Cantleyanone A (1) was isolated as yellow oil, and found to have the molecular formula C₃₄H₄₂O₇ by HRE-SIMS. Its IR spectrum exhibited absorption bands at 3449 (hydroxyl group), 1718 (unconjugated carbonyl group) and 1628 cm⁻¹ (ortho hydroxyl chelated carbonyl group). These data and the UV absorption band at λ_{max} 359 nm for a conjugated carbonyl chromophore were similar to those of deoxygaudichaudione A (Han et al., 2006) suggesting that 1 has a similar caged-polyprenylated structure. The ¹H and ¹³C NMR data were also comparable to those of deoxygaudichaudione A (Han et al., 2006), showing characteristic signals for the caged-prenylated and phloroglucinol halves of the structure. The only difference was the absence of the H-7 signal seen in the ¹H NMR spectrum of deoxygaudichaudone A, which has been replaced by a methoxyl singlet at $\delta_{\rm H}$ 3.66. The attachment of the methoxy group to the bridge-head carbon was apparent from HMBC correlation observed between its protons to C-7 ($\delta_{\rm C}$ 85.0). The olefinic proton H-8 ($\delta_{\rm H}$ 7.51) also appeared as a singlet since it no longer has a coupling partner on C-7. The ¹H and ¹³C NMR, as well as the HMBC data for cantleyanone A are summarized in Tables 1 and 2. A NOESY experiment supported the same relative stereochemistry for the tricyclic core of 1 as previously reported caged-xanthonoids (Cao et al., 1998; Rukachaisirikul et al., 2000; Xu et al., 2000). Cantleyanone A is thus assigned as structure 1.

Table 1 13 C NMR data in CDCl₃ of compounds 1–6 (δ in ppm)

Position	1	2	3	4	5	6
1	160.5	163.4	161.6	166.3	161.0	164.2
2	106.5	97.3	101.0	92.1	93.3	99.0
3	163.2	164.6	163.4	169.4	162.0	167.0
4	107.7	106.4	111.5	104.6	110.3	105.1
4a	156.3	158.2	155.5	155.4	153.6	158.6
5	84.4	84.3	136.3	84.2	84.3	84.3
6	202.8	203.9	123.3	201.5	201.3	201.9
7	85.0	79.5	110.2	84.9	85.0	85.0
8	134.2	137.9	153.7	134.5	135.3	134.6
8a	132.4	131.4	107.2	132.2	132.2	132.2
9	179.3	179.3	185.1	178.3	180.0	179.0
9a	101.1	101.4	103.4	101.3	102.5	101.2
10a	89.2	89.7	143.0	89.5	89.8	89.6
11	22.9	22.3		26.9	97.8	28.9
12	121.6	121.5		92.1	162.7	78.9
13	135.6	135.3		71.8	69.4	146.0
14	26.0	26.0		26.4	28.9	113.8
15	18.3	18.2		24.5	28.8	17.1
16	21.4	29.1		29.2	29.3	29.0
17	121.9	117.4		117.8	118.2	117.4
18	134.7	135.9		135.5	135.8	135.8
19	26.1	25.7		25.8	25.7	25.7
20	18.2	17.0		17.1	17.1	16.9
21	29.1	30.8		30.1	29.9	30.41
22	117.8	50.1		49.8	49.8	50.0
23	135.8	84.2		84.0	84.1	83.6
24	25.8	29.3		29.2	30.7	30.3
25	17.0	30.4		30.5	29.2	29.1
26	30.5					
27	49.9					
28	83.7					
29	29.3					
30	30.4					
1'			152.2			
2'			41.6			
3′			109.6			
4'			28.3			
5'			28.3			
7-O <u>C</u> H ₃	54.2			54.2	54.3	54.2

7-Hydroxyforbesione (2) with the molecular formula $C_{28}H_{32}O_7$ determined from HRESIMS, showed UV and IR absorptions similar to those of 1 suggesting that 2 also possessed similar structural features as 1 and deoxygaudichaudione A. However, a closer examination of the 1H and ^{13}C NMR spectra of 2 showed that, apart from the characteristic $-OC(Me)_2-CHCH_2-C-$ unit for the caged-prenylated moiety and the phloroglucinol-type aromatic ring, 2 has only two other free prenyl groups in its molecular structure. One of these was the prenyl group attached to C-5 (δ_C 84.3), as in 1, leaving the remaining prenyl group which must thus be attached on ring-A. The methoxy group at C-7 is also absent in this compound. However, the deshielded nature of C-7 (δ_C 79.5) indicated that this bridge-head carbon was hydroxylated.

The ¹H NMR spectrum of **2** exhibited an aromatic proton singlet at $\delta_{\rm H}$ 6.07 ($\delta_{\rm C}$ 97.3) which could be assigned as H-2 based on the ²J and ³J correlations observed from it to C-1 ($\delta_{\rm C}$ 163.4, ²J), C-3 ($\delta_{\rm C}$ 164.6, ²J), C4 ($\delta_{\rm C}$ 106.4, ³J) and

Table 2 ¹H and HMBC NMR data (CDCl₃) for compounds **1–6**

Position	$^{1}\mathrm{H}$	HMBC	$^{1}\mathrm{H}$	HMBC	$^{1}\mathrm{H}$	HMBC
	1	1	2	2	3	3
2			6.07 (1H, s)	C-1, 3, 4, 9a	6.28 s	C-1, 3, 4, 9a, 9
5					7.25 d (9)	C-5, 8, 10^a
7					6.71 d (9)	C-5, 8, $8a^a$
3	7.51 (1H, s)	C-6, 8^a , 9, 10^a	7.29 (1H, s)	C-6, 8^a , 9, 10^a , 21		
1	3.42 (2H, d, 7.0)	C-2, 12, 13	3.46 (2H, <i>m</i>)	C-4, 4a, 12, 13		
12	5.25 (1H, t, 7.0)	C-2, 11, 14	5.25 (1H, t, 6.5)	C-4, 11, 14, 15		
14	1.84 (3H, s)	C-12, 13, 15	1.76 (3H, s)	C-12, 13, 15		
15	1.78 (3H, s)	C-12, 13, 14	1.81 (3H, s)	C-12, 13, 14		
16	3.42 (2H, d, 7.0)	C-3, 4a, 4, 17	2.67 (2H, m)	C-5, 6, 10a, 17, 18		
17	5.25 (1H, t, 7.0)	C-16, 19	4.42 (1H, t, 7.0)	C-16, 19, 20		
19	1.75 (3H, s)	C-17, 18, 20	1.39 (3H, s)	C-17, 18, 20		
20	1.80 (3H, s)	C-17, 18, 19	1.07 (3H, s)	C-17, 18, 19		
21	2.63 (1H, m)	C-5, 6, 10a, 22	2.18 (1H, d, 13.0)	C-6, 7, 8, 10a, 22, 23		
			1.85 (1H, dd, 13.0, 9.5)			
22	4.44 (1H, m)	C-24	2.57 (1H, d, 9.5)	C-5, 7, 10a, 21, 23, 25		
24	1.39 (3H, s)	C-22, 23, 25	1.28 (3H, s)	C-22, 23, 25		
25	1.04 (3H, s)	C-22, 23, 24	1.70 (3H, s)	C-22, 23, 24		
26	2.38 (1H, d,12.5)	$C-6, 7, 8, 10^a, 27, 28$				
	1.64 (1H, m)					
27	2.57 (1H, d, 10.5)	C-5, 10a, 26				5.29 d (10.5)
29	1.32 (3H, s)	C-27, 28, 30				, ,
80	1.69 (3H, s)	C-27, 28				
2'					6.59 (1H, dd, 17.5, 10.5)	C-1', 4, 3', 4'/5'
3′					5.39 (1H, d, 17.5)	C-2', 1'
					5.29 (1H, d, 10.5)	C-1'
! /					1.73 (3H, s)	C-4, 1', 2', 5'
5′					1.73 (3H, s)	C-4, 1', 2', 4'
1-OH	12.89 (1H, s)	C-1, 2, 9a	12.55 (1H, s)	C-1, 2, 9a	12.32 (1H, s)	C-1, 2, 9a
8-ОН	6.53 (1H, s)	C-2, 3, 4				
8-OH					11.16 (1H, s)	C-7, 8, 8a
7-OCH ₃	3.66 (3H, s)	C-7				

(continued on next page)

Table 2 (continued)

Position	¹ H	HMBC	1 H	HMBC	1 H	HMBC
	4	4	5	5	6	6
2	6.07 (1H, s)	C-1, 3, 4, 9a	6.64 (1H, s)	C-1, 3, 4, 9a	6.19 (1H, s)	C-1, 3, 4
6						
7						
8	7.50 (1H, s)	C-6, 7, 8a, 9, 10a, 21	7.57 (1H, s)	C-6, 9, 10a, 21	7.52 (1H, <i>s</i>)	C-6, 7, 8a, 9, 10a
11	3.17 (1H, <i>dd</i> , 15.0, 9.0)	C-3, 4a, 4, 13	6.57 (1H, s)	C-4, 4a, 11	$3.19 (1H_a, d, 15.0)$	C-3, 4, 4a, 12
10	3.07 (1H, <i>dd</i> , 15.0, 9.0)	0.2 4 14 15			2.83 (1H _b , dd, 15.0, 10.0)	6 4 11 14 15
12	4.76 (1H, m)	C-3, 4, 14, 15	1.60 (211	G 12, 12, 15	4.38 (1H, <i>d</i> , 10.0)	C-4, 11, 14, 15
14	1.40 (3H, s)	C-12, 13, 15	1.69 (3H, s)	C-12, 13, 15	4.93 (1H, <i>br s</i>)	C-12, 13, 15
15	1.27 (3H, s)	C-12, 13, 14	1.69 (3H, s)	C-12, 13, 14	4.95 (1H, <i>br s</i>) 1.90 (3H, <i>s</i>)	C-12, 13, 14
16	2.60 (2H, <i>m</i>)	C-12, 13, 14 C-5, 6, 10a, 17, 18	2.64 (2H, <i>m</i>)	C-12, 13, 14 C-5, 6, 10a, 17, 18	2.50 (2H, <i>dd</i> , 14.0, 10.0)	C-12, 13, 14 C-5, 6, 10a, 17, 18
17	4.54 (1H, t, 7)	C-16, 19, 20	4.51 (1H, <i>m</i>)	C-16, 19, 20	4.45 (1H, <i>m</i>)	C-19, 20
1 /	4.34 (111, t, 7)	C-10, 19, 20	4.31 (111, <i>m</i>)	C-10, 19, 20	4.43 (111, m)	C-19, 20
19	1.42 (3H, s,)	C-17, 18, 20	1.35 (3H, s,)	C-17, 18, 20	1.39 (3H, s)	C-17, 18, 20
20	1.26 (3H, s)	C-17, 18,19	0.93 (3H, s)	C-17, 18, 19	1.04 (3H, s)	C-17, 18, 19
21	2.38 (1H, d, 12.5)	C-6, 7, 8, 10a, 22, 23	2.43 (1H, d, 13.0)	C-6, 7, 8, 10a, 22, 23	2.37 (1H _a , d, 13.5)	C-6, 7, 8, 10a, 22, 23
					1.66 (1H _b , dd, 13.5, 10.0)	
	1.64 (1H, dd, 12.5, 9.0)		1.64 (1H, dd, 10.0, 3.0)			
22	4.44 (1H, <i>m</i>)	C-5, 7, 10a, 21, 23, 25	2.60 (1H, d, 10.0)	C-5, 10a, 21, 23, 24, 25	2.56 (1H, d, 10.0)	C-5, 7, 10a, 21, 25
24	1.33 (3H, s)	C-22, 23, 25	1.78 (3H, s)	C-22, 23, 25	1.59 (3H, s)	C-22, 23, 25
25	1.67 (3H, s)	C-22, 23, 24	1.37 (3H, s)	C-22, 23, 24	1.30 (3H, s)	C-22, 23, 24
26						
27						
29						
30						
2'						
3'						
4'						
5'						
1-OH	12.72 (s)	C-1, 2, 9a	12.20 (s)	C-1, 2, 9a	12.59 (s)	C-1, 2, 9a
3-OH						
8-OH	2.65 (a)	C-7	2.67 (a)	C 7	2.65 (a)	C-7
OCH_3	3.65 (s)	C-/	3.67 (s)	C-7	3.65 (s)	C-/

C-9a ($\delta_{\rm C}$ 101.4, 3J). The remaining prenyl group was thus placed at C-4. The HMBC correlations observed between H₂-11 ($\delta_{\rm H}$ 3.46, m) with C-4 (2J) and C-4a ($\delta_{\rm C}$ 158.2, 3J) further support this assignment. Based on the above considerations, compound **2** was assigned as 7-hydroxyforbesione. The $^1{\rm H}$ and $^{13}{\rm C}$ NMR, as well as the HMBC data for **2** are summarized in Table 1 and 2. It is worth noting that this finding represents the first report of a 7-hydroxylated caged-xanthonoid.

The molecular formula for the new compound (3) was established as C₁₈H₁₆O₆ by HRESIMS. Its UV (257 and 343 nm) and IR (3435 cm⁻¹ and 1629 cm⁻¹) absorptions were typical of a xanthone derivative (Minami et al., 1996). Its ¹H NMR spectrum showed two sharp singlets at $\delta_{\rm H}$ 12.32 and 11.16 which suggested the presence of two chelated hydroxyl groups on C-1 and C-8, respectively. A pair of ortho-coupled AB-type aromatic proton signals at $\delta_{\rm H}$ 6.71 (d, J = 9.0 Hz, H-7), and 7.25 (d, J = 9.0 Hz, H-6) were also observed in addition to an aromatic oneproton singlet at $\delta_{\rm H}$ 6.28 (H-2), suggesting disubstituted and trisubstituted rings B and A, respectively, of the xanthone molecule. In the HMBC spectrum, cross-peaks for one of the chelated hydroxyl proton $\{\delta_{\rm H}\ 11.16\ (8\text{-OH})/\delta_{\rm C}\}$ 153.7 (C-8), 107.3 (C-8a), 110.2 (C-7)}, and the ortho-coupled pair of aromatic protons { $\delta_{\rm H}$ 6.71 (H-7)/ $\delta_{\rm C}$ 153.7 (C-8), 136.3 (C-5), 107.2 (C-8a) and 7.25 (H-6)/ δ_C 153.7 (C-8), 136.3 (C-5), 143.0 (C-10a)} were observed, confirming a disubstituted ring-B feature for 3. Similarly, HMBC cross-peaks observed for the other chelated hydroxyl proton $\{\delta_{\rm H}\ 12.32\ (1\text{-OH})/\delta_{\rm C}\ 101.0\ (\text{C-2}),\ 161.6\ (\text{C-1}),\ 103.4$ (C-9a)}, and the singlet aromatic proton $\{\delta_{\rm H} 6.28 \ (\text{H-2})/\text{M}\}$ $\delta_{\rm C}$ 111.5 (C-4), 163.4 (C-3), 161.6 (C-1), 103.4 (C-9a) established a trisubstituted ring-A feature, with hydroxyl groups located at C-1 and C-3, and a third substituent at C-4. The third substituent represented by the set of signals at $\delta_{\rm H}$ 1.73 (6H, s, H₃-4' and H₃-5'), $\delta_{\rm H}$ 5.39 (1H d, $J = 17.5 \text{ Hz}, \text{ H}_a - 3'), 5.29 \text{ (1H } d, J = 10.5 \text{ Hz}, \text{ H}_b - 3') \text{ and}$ $\delta_{\rm H}$ 6.59 (1H dd, J = 17.5, 10.5 Hz, H-2') in the ¹H NMR spectrum, was established to be a 1,1-dimethylprop-2-enyl group on the basis of 2D (¹H-¹H COSY, HSQC HMBC) spectral data. Compound 3 was therefore assigned as 4-(1,1-dimethylprop-2-enyl)-1,3,5,8tetrahydroxyxanthone.

Cantleyanones B, C and D (4, 5 and 6), obtained as yellow solids from the trunk bark extract, were all 7-methoxy caged-polyprenylated xanthones with a phloroglucinoltype aromatic ring-A. The three compounds showed UV and IR characteristics similar to 1 and 2 (UV absorption bands at $\lambda_{\rm max}$ 358–359 nm for a conjugated carbonyl chromophore, and IR absorption bands at 3423–3449 cm⁻¹ for hydroxyl, 1715–1718 cm⁻¹ for an unconjugated carbonyl, and 1636–1637 cm⁻¹ for a chelated *ortho*-hydroxyl carbonyl groups). The ¹H NMR and ¹³C NMR spectra (Tables 1 and 2) also showed the common signals for a caged-prenylated and phloroglucinol halves of the molecule. The C-7 ($\delta_{\rm C}$ 85.0) was methoxylated, and the phloroglucinol-type aromatic ring carries a single aromatic proton

on C-2 based on the observed HMBC cross-peaks from H-2 to C-1 (2J), C-3 (2J), C4 (3J) and C-9a (3J). The structures of **4**, **5** and **6** differ from each other by the modified prenyl side-chain on C-4.

Canthelyanone B (4) has a molecular formula C₂₉H₃₄O₈ as determined by HRESIMS. Analysis of the ¹H, ¹³C NMR, ¹H-¹H COSY, HSOC and HMBC data (Table 2) showed that the modified prenyl side-chain on C-4 for 4, represented by the set of signals at $\delta_{\rm H}$ 3.17 (dd, J=9.0and 15.0 Hz, H_a -11), 3.07 (dd, J = 9.0 and 15.0 Hz, H_{b} -11), 4.76 (m, H-12), 1.40 (s, H_{3} -14), and 1.27 (s, H_{3} -15), was in the form of a 2-(2-hydroxypropyl)dihydrofuran ring, similar to that found in mangostanin (Suksamrarn et al., 2006). The furan ring was further established to be fused to the C-3,4 position on ring-B based on the HMBC correlations H_2 -11/ δ_C 169.4 (C-3), 104.6 (C-4), 155.4 (C-4a), and H-12/169.4 (C-3), 104.6 (C-4). The relative stereochemistry at C-12 was deduced from a NOESY experiment. Whilst showing very strong NOEs with both Me-14 and Me-15, the oxymethine proton H-12 ($\delta_{\rm H}$ 4.76) showed very strong NOE interactions to only one of the C-11 methylene protons (δ_H 3.17, H_a -11). On the other hand, the other C-11 methylene proton at $\delta_{\rm H}$ 3.07 (H_b-11) was observed to show strong NOE interactions with both Me-14 and Me-15 as well as Me-25, which implied that they were on the same side of the molecule. This was further supported from the stronger NOE interactions observed between Me-25 and H_b-11 when compared to H_a -11. The structure of 4 was thus established and given the trivial name cantlevanone B.

The molecular formula of cantleyanone C (5) was determined to be $C_{29}H_{32}O_8$. The 1H and ^{13}C NMR spectra of 5 were very similar to cantleyanone E (4), the only difference being the absence of the C-11 methylene group. In its place, the 1H and ^{13}C NMR spectra of 5 exhibited an olefinic proton singlet at δ_H 6.57 (δ_C 97.8) and a very deshielded C-12 (δ_C 162.7), indicating the presence of a furan ring. Thus, 5 was assigned as the dehydrogenated product of cantleyanone E (4) and named cantleyanone C.

Cantleyanone D (6) has a molecular formula of C₂₉H₃₄O₈ as determined from its HRESIMS. The ¹H and ¹³C NMR spectra of **6** showed a different type of prenyl side-chain on C-4 when compared to 4 and 5. From the set of signals observed at $\delta_{\rm H}$ 2.83 (dd, J=15.0, 10.0 Hz, H_{b} -11), 3.19 (d, J = 15.0 Hz, H_{a} -11), 4.38 (d, J = 10.0 Hz, H-12), 4.93 and 4.95 (H_2 -14), 1.90 (s, H_3 -15), the prenyl side-chain was found to be in the form of a 2-hydroxy-3methylbut-3-enyl group. The assignment was supported by ¹H-¹H COSY and the HMBC correlations observed between the protons at $\delta_{\rm H}$ 2.83 and 3.19 (H_{a/b}-11) with C-4 ($\delta_{\rm C}$ 105.1), C-4a ($\delta_{\rm C}$ 158.6), C-3 ($\delta_{\rm C}$ 167.0), C-12 ($\delta_{\rm C}$ 78.9), C-13 ($\delta_{\rm C}$ 146.0), and between H-12 ($\delta_{\rm H}$ 4.38) with C-4, C-11 ($\delta_{\rm C}$ 105.1), C-14 ($\delta_{\rm C}$ 113.8) and C-15 ($\delta_{\rm C}$ 17.1). The stereochemistry at C-12 was left unresolved. The above spectral considerations allowed the assignment of cantleyanone D as the structure 6, which has not been reported before.

Table 3
Cytotoxic activities of the isolated natural products and CHCl₃ crude extract of *G. cantleyana*

Compounds	IC ₅₀ μg/ml ^a	IC ₅₀ μg/ml ^a					
	MDA-MB-231	MCF-7	CaOV-3	HeLa			
Cantleyanone A (1)	9.67 ± 1.53	24.33 ± 1.53	13.83 ± 1.26	20.67 ± 3.06			
7-Hydroxyforbesione (2)	2.17 ± 0.31	0.42 ± 0.08	0.28 ± 0.07	0.22 ± 0.03			
4-(1,1-Dimethylprop-2-enyl)-	25.50 ± 1.32	17.67 ± 1.53	27.50 ± 0.87	13.33 ± 0.58			
1,3,5,8-tetrahydroxyxanthone (3)							
Cantleyanone B (4)	6.23 ± 0.93	0.83 ± 0.07	0.28 ± 0.04	0.43 ± 0.10			
Cantleyanone C (5)	6.70 ± 0.36	1.48 ± 0.37	0.44 ± 0.06	0.48 ± 0.07			
Cantleyanone D (6)	17.17 ± 0.76	4.40 ± 0.46	3.47 ± 1.23	2.8 ± 0.3			
Deoxygaudichaudione A	0.44 ± 0.02	0.38 ± 0.06	0.44 ± 0.02	0.34 ± 0.03			
Garbogiol	>30	>30	>30	>30			
Macranthol	4.17 ± 0.35	3.70 ± 0.56	1.53 ± 0.57	2.53 ± 0.49			
G. cantleyana (chloroform extract)	2.71 ± 0.35	4.17 ± 0.25	4.63 ± 0.15	4.37 ± 0.15			

^a Results are expressed as IC₅₀ values (μg/ml) ±SD of three experiments performed in triplicate, CaOV-3, human ovarian cancer; HeLa, human cervical cancer; MDA-MB-231, human breast cancer (Re-); MCF-7, human breast cancer (Re+).

The isolated compounds were assayed for their cytotoxic activity on four cell-lines. The results are shown in Table 3. Compounds **2**, **4**–**6**, deoxygaudichaudione A and macranthol exhibited strong activity against the four cell-lines tested, with IC₅₀ values ranging from 0.22 to 7 μ g/ml, except for compound **6** which was weakly cytotoxic against MDA-MB-231 with an IC₅₀ value of 17.7 μ g/ml. Compounds **2**, **4** and **5** were also less active towards the MDA-MB-231 cell-line. Compounds **1** and **3** exhibited moderate to weak activity with IC₅₀ values ranging from 9.67 to 27.50 μ g/ml, respectively. Garbogiol showed weak activity against all tested cell-lines with IC₅₀ values more than 30 μ g/ml.

3. Conclusions

Similar to several other *Garcinia* species such as G. morella, G. hanburyi, G. gaudichaudii, G. bracteata, G. forbesii and G. scortechinii, G. cantleyana also elaborated caged-xanthones. The possible biosynthetic pathways of caged-xanthonoids have been discussed. From this investigation, the xanthonoids isolated from G. cantleyana seemed to be derived from mono-, tri- and tetra-prenylated xanthones. The prenyl moieties on the non-caged section of these xanthonoids retained their forms or are further oxidized and cyclized to form dihydrobenzofuran or benzofuran moieties. It is interesting to note the different attachment of the prenyl group in the mono-prenylated xanthone (3). The isolated xanthonoids are strongly cytotoxic except for cantleyanone A (1), 4-(1,1dimethylprop-2-enyl)-1,3,5,8-tetrahydroxyxanthone and garbogiol. Comparing the structures of the bioactive compounds 2, 4, 5 and 6 with that of 1 and 3, it is clear that the caged nature of ring-B as well as the presence of the peri-hydroxyl group on ring-A are important structural features for bioactivity. These are in agreement to previous findings (Cao et al., 1998; Mackeen et al., 2000). The C8-8a double bond is also thought to be

important for bioactivity (Zhang et al., 2004). However, as indicated by the weak cytotoxic activity exhibited by compound 1, prenylation on C-2, appears to reduce the cytotoxic activity of this class of compounds quite significantly.

4. Experimental section

4.1. General

Melting points were determined on a Fisher-Johns hot stage apparatus and are uncorrected. UV and IR spectra were recorded on CARY 100 Conc UV-Vis (Varian) and Perkin-Elmer RXI FTIR spectrometers, respectively. Optical rotation was measured on a JASCO DIP-370 digital polarimeter. Mass spectra were recorded on Polaris Q Mass Spectrometers (ThermoFinnigan San Jose CA), with ionization being induced by electron impact at 70 eV. HRESIMS were measured using Finnigan MAT95XL-T spectrometers. ¹H and ¹³C NMR spectra were recorded on Varian Unity INOVA 500 Spectrometer using CDCl₃ as solvent unless otherwise stated. Adsorbents used for vacuum liquid chromatography (VLC) and column chromatography (CC) were silica gel Merck 7749 and Silica gel Merck 7734 or 9385, respectively. Sephadex LH-20 was used for gel permeation chromatography (GPC). For analytical and preparative TLC, Merck TLC plates, Silica gel 60 F₂₅₄ and Merck PLC plates, Silica gel 60 F₂₅₄ (2 mm) were, respectively, employed.

4.2. Plant material

The plant materials were collected in July 2001, from the Cameron Highlands in Pahang, Malaysia, and were botanically identified by Mr. Shamsul Khamis. A voucher specimen (SK 58/01) has been deposited at the Laboratory of Natural Products, Institute of Bioscience, University Putra Malaysia.

4.3. Extraction and isolation

The air-dried, milled leaves of G. cantlevana were extracted in two batches. In the first batch, 425 g of the material was macerated in MeOH for 72 h and the extraction process repeated three times. The extracts were combined and the solvent removed under reduced pressure to give 89 g of the crude methanolic extract. A portion of the extract (42 g) was then triturated with chloroform (300 ml) by continuous shaking in a chloroform suspension for 10 min followed by filtration. The procedure was repeated four times $(4 \times 300 \text{ ml})$, and the collected filtrates combined and concentrated under reduced pressure to give 17 g of the chloroform soluble fraction. The fraction was further fractionated by VLC, eluting with hexane, CHCl₃, EtOAc and MeOH to afford seven fractions, (A1–A7). Fraction A4 (400 mg) was further chromatographed over silica gel, eluting with 100% dichloromethane to yield deoxygaudichaudione A (9 mg). Meanwhile, fraction A6 (2.6 g) was separated by a silica gel CC, eluting with hexane-EtOAc, 95:5 to give five sub-fractions, B1-B5. Sub-fraction B2 (400 mg) was further purified on silica gel column, eluting with 10% ethyl acetate-hexane, followed by repeated preparative TLC to afford 1 (6.9 mg) and 2 (16 mg). Separation of sub-fraction B3 (155 mg) was performed by CC over silica gel, eluting with 100% dichloromethane to afford garbogiol (8 mg). Fractions A2 (500 mg) and A3 (700 mg) were each re-crystallized to yield glutin-5-en-3β-ol (16 mg) and a mixture of sitosterol and stigmasterol (29 mg), respectively.

In the second batch of extraction, 256 g of the raw material was extracted three times with chloroform, and the combined extracts were concentrated under reduced pressure to give 15 g of the crude chloroform extract. The extract was then fractionated on a Sephadex LH-20 column, eluting with 100% MeOH, yielding four fractions, C1-C4. Further purification of fraction C2 (300 mg) on Sephadex LH-20, eluting with MeOH, afforded (11.6 mg). Finally, fraction C3 (5 g) was separated by CC, eluting with 1% MeOH in CHCl₃, to afford seven sub-fractions, D1-D7. Sub-fraction D2 (1.2 g) was further purified over Sephadex LH-20 column, eluting with MeOH to yield macranthol (7.9 mg).

The air-dried, milled trunk bark (1 kg) was macerated in chloroform for 72 h. The extraction process was repeated three times, the extracts combined and the solvent removed to yield 20 g of the crude chloroform extract. The extract was chromatographed over silica gel (250 g, 7734), eluting with 100% chloroform and finally flushed with methanol to afford seven major fractions (L-R). Further purification of fraction O (4 g) was performed by column chromatography over silica gel (110 g) with 4% MeOH in CHCl₃ as eluent. This afforded four combined sub-fractions which was further rechromatographed over silica gel (25 g, 9835) using 1% MeOH in CHCl₃ as eluent to afford 4 (8 mg), 5 (5.2 mg), and 6 (10 mg). Further separation of fraction Q (380 mg) by column chromatography over silica gel 9835 (22 g) was performed with 20% ethyl acetate in hexane as eluent to afford three combined sub-fractions Q1 (1-29), O2 (30–45) and O3 (46–72). Column chromatography of sub-fraction Q2 (60 mg) on silica gel using 2% ethyl acetate in dichloromethane as eluent, afforded three sub-fractions Q2.1 (1-13), Q2.2 (14-19) and Q2.3 (20-31). Sub-fraction O2.2 (20 mg) was repeatedly chromatographed over Sephadex LH-20, eluting with MeOH to give gaudichaudione H (12 mg).

4.3.1. Cantleyanone A (1)

Yellow oil; $[\alpha]_D^{25} - 125^{\circ}$ (c 0.02, CHCl₃); UV (CHCl₃) λ_{max} nm (log ϵ); 201 (3.40), 241 (3.47), 359 (3.65); IR (KBr) v_{max} cm⁻¹: 3449, 1718, 1628, 1054; for ¹H and ¹³C NMR spectroscopic data see Table 1; HRESIMS m/z: 561.2839 [M-H]^- (calcd. for $C_{34}H_{41}O_7$, 561.2931).

4.3.2. 7-Hydroxyforbesione (2)

Yellow amorphous solid, m.p. 188 °C; $[\alpha]_D^{25} - 135^\circ$ (c 0.02, CHCl₃); UV (CHCl₃) $\lambda_{\rm max}$ nm (log ε): 350 (3.50), 347 (3.49), 241 (3.43); IR (KBr) v_{max} cm⁻¹: 3431, 2929, 1730, 1636, 1426; For ¹H and ¹³C NMR spectroscopic data see Table 2; HRESIMS: m/z 479.2118 [M-H]⁻ (calcd. for $C_{28}H_{31}O_7$, 479.2149).

4.3.3. 4-(1,1-Dimethylprop-2-enyl)-1,3,5,8tetrahydroxyxanthone (3)

Yellow crystal, m.p. 182–184 °C; UV (CHCl₃) λ_{max} nm $(\log \varepsilon)$: 343 (3.69), 300 (3.63), 279 (3.60), 257 (3.56); IR (KBr) $v_{\text{max}} \text{ cm}^{-1}$: 3435, 2964, 1629, 1497, 1401, 1284, 1205, 1158; HRESIMS m/z: 327.0903 [M-H]⁻ (calcd. for $C_{18}H_{15}O_6$, 327.0947).

4.3.4. Cantleyanone B (*4*)

Yellow solid, m.p. $168 \,^{\circ}\text{C}$; $[\alpha]_D^{25} - 45^{\circ}$ (c 0.02, CHCl₃); UV (CHCl₃) λ_{max} nm (log ϵ): 359 (3.60), 348 (3.58), 334 (3.56), 263 (3.46); IR (KBr) $v_{\rm max}$ cm⁻¹: 3448, 1718, 1637, 1388, 1303 cm⁻¹; for ¹H and ¹³C NMR spectroscopic data see Table 2; EIMS m/z (rel. int.): 510 [M]⁺ (3), 482 (70), 467 (20), 413 (100), 385 (55), 353 (20), 251 (20), 245 (25), 203 (19); HRESIMS m/z: 509.2157 [M-H]⁻ (calcd. for C₂₉H₃₃O₈: 509.2253).

4.3.5. Cantleyanone C (5)

Yellow solid, m.p. 153 °C; $[\alpha]_D^{25} + 40^\circ$ (c 0.02, CHCl₃); UV (CHCl₃) λ_{max} nm (log ϵ): 318 (3.80), 261 (3.85), 241 (3.68); IR (KBr) v_{max} cm⁻¹: 3433, 1716, 1637, 1364 cm⁻¹; for ¹H and ¹³C NMR spectroscopic data see Table 2; ESIMS m/z (rel. int.): 507 [M-H]⁻ (100), 462 (25), 411 (86), 397 (90), 351 (24), 245 (88), and 217 (15); HRESIMS m/z: 507.2004 [M-H]⁻ (calc. for C₂₉H₃₁O₈: 507.2097).

4.3.6. Cantleyanone D (6) Yellow amorphous solid, m.p. 175 °C; $[\alpha]_D^{25} + 122^\circ$ (c 0.02, CHCl₃); UV (CHCl₃) λ_{max} nm (log ε): 358 (3.55), 333 (3.52); IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3423 (-OH), 1715 (C=O), 1636 (C=O), 1333 cm^{-1} ; ESIMS m/z (rel. int.): 509 [M-H]⁻ (84), 464

(35), 411 (100), 396 (35), 351 (83), 245 (23), and 165 (24); HRESIMS $[M-H]^-$ m/z 509.2163 (calc. for $C_{29}H_{33}O_8$: 509.2253); ¹H NMR: Table 1; ¹³C NMR: Table 2.

4.3.7. Cytotoxicity

Cytotoxic activities of new compounds (1–6) together with known compounds was measured against a panel of four human cancer cell lines as described previously (Mackeen et al., 2000).

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