

## Triterpenoids with antimicrobial activity from *Drypetes inaequalis*

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### ARTICLE INFO

#### Article history:

Received 26 March 2007

Received in revised form 10 December 2008

Available online 13 February 2009

#### Keywords:

*Drypetes inaequalis*

Euphorbiaceae

Stems

Fruit

Triterpenoid esters

Saponins

Antimicrobial activity

### ABSTRACT

The air-dried stems and ripe fruit of *Drypetes inaequalis* Hutch. (Euphorbiaceae) were studied. Four triterpene derivatives, characterized as lup-20(29)-en-3 $\beta$ ,6 $\alpha$ -diol, 3 $\beta$ -acetoxylup-20(29)-en-6 $\alpha$ -ol, 3 $\beta$ -caffeoyloxylup-20(29)-en-6 $\alpha$ -ol and 28- $\beta$ -D-glucopyranosyl-30-methyl 3 $\beta$ -hydroxyolean-12-en-28,30-dioate along with 10 known compounds were isolated from the whole stems. One triterpene, characterized as 3 $\alpha$ -hydroxyfriedelan-25-al along with six known compounds were isolated from the ripe fruit. Their structures were established on the basis of spectroscopic analysis and chemical evidence. The triterpenes were tested for antimicrobial activity against some Gram-positive and Gram-negative bacteria, and two of them appeared to be modestly active.

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### 1. Introduction

*Drypetes inaequalis* Hutch. (Euphorbiaceae) is a forest shrub growing in the Centre and East provinces of Cameroon. Therapeutic applications of the *Drypetes* plants in West and Central Africa concern the treatment of sinusitis, swellings, boils, gonorrhoea and dysentery (Dalziel, 1937; Irvine, 1961; Bouquet and Debray, 1974; Walker et al., 1961). In our previous study on the *Drypetes* genus, we have reported on the anti-inflammatory and analgesic actions of the crude extract and compounds isolated from *D. molunduana* (Wandji et al., 2000; Chungag-Anye et al., 2001, 2002), the phenolic constituents from *D. armoracia* (Wandji et al., 2003) and the antileishmanial furanosesquiterpenes and triterpenoids from *D. chevalieri* (Wansi et al., 2007). As a continuation of our search for compounds with biological activities from the *Drypetes* species, we studied the whole stems and the ripe fruit of *D. inaequalis*. From the whole stems, we isolated four new and 10 known compounds, and from the ripe fruit we obtained one new and six known compounds. The known compounds from both parts of the plant were identified as serjanic acid (5) (Javasinghe et al.,

1993), oleanolic acid (6) (Mahato and Kundu, 1994), hederagenin (7) (Mahato and Kundu, 1994), queretaroic acid (8) (Agrawal and Jain, 1992), serragenic acid (9) (Agrawal and Jain, 1992), 28- $\beta$ -D-glucopyranosyl 3 $\beta$ -hydroxyolean-12-en-28-oate (10) (Srivastava and Jain, 1989), friedelin (12) (Li et al., 2006), 3,7-dioxofriedelane (13) (Mahato and Kundu, 1994), 3 $\alpha$ -friedelanol (14) (Salazar et al., 2000), 3-oxofriedelan-25-al (15) (Anjaneyulu and Narayana, 1980), stigmaterol (16) (Wandji et al., 2000), 3 $\beta$ -D-glucopyranosylstigmaterol (17) (Wandji et al., 2000), sitosterol (18) (Wandji et al., 2003) and 3 $\beta$ -D-glucopyranosylsitosterol (19) (Wandji et al., 2003). The structures of the five new compounds have been determined as lup-20(29)-en-3 $\beta$ ,6 $\alpha$ -diol (1), 3 $\beta$ -acetoxylup-20(29)-en-6 $\alpha$ -ol (2), 3 $\beta$ -caffeoyloxylup-20(29)-en-6 $\alpha$ -ol (3), 28- $\beta$ -D-glucopyranosyl-30-methyl 3 $\beta$ -hydroxyolean-12-en-28,30-dioate (4) and 3 $\alpha$ -hydroxyfriedelan-25-al (11), on the basis of spectroscopic analysis and chemical evidence. In the present paper, the isolation, structural determination and antimicrobial activity of the new compounds will be described.

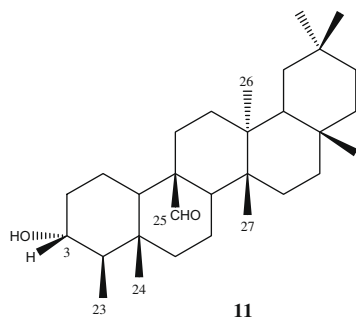
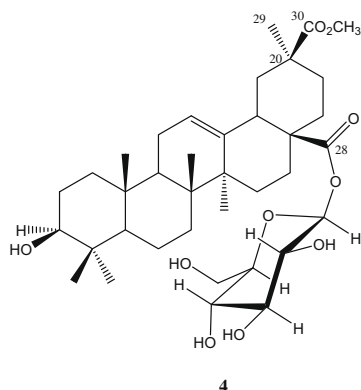
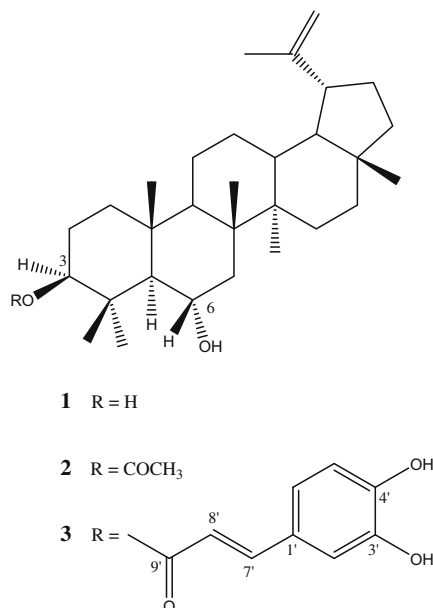
### 2. Results and discussion

The whole stems and the ripe fruit of *D. inaequalis* were sun-dried, ground into a powder form, macerated with a mixture of solvents and chromatographed on silica gel to afford 19 compounds

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**1–19.** Compounds **5–10** and **12–19** were identified as known compounds by comparison of their  $^{13}\text{C}$  NMR data and other physical properties with reported values. Compounds **1–4** and **11** were characterized as five new triterpene derivatives.



Compound **1** was obtained as a colourless amorphous solid. The pseudo molecular ion peaks at  $m/z$  443  $[\text{M}+\text{H}]^+$  and 460  $[\text{M}+\text{NH}_4]^+$  in its  $\text{Cl}/\text{NH}_3$  MS, and the HR TOF MS ES<sup>+</sup> at  $m/z$  442.3825 suggested its molecular formula to be  $\text{C}_{30}\text{H}_{50}\text{O}_2$ . The IR spectrum indicated the presence of hydroxyl ( $3400\text{ cm}^{-1}$ ) and olefinic ( $1660\text{ cm}^{-1}$ ) groups. The  $^{13}\text{C}$  NMR and DEPT spectra of **1** showed 30 carbon signals including seven methyls, nine methylenes, seven methines (two of which are oxygenated ( $\delta_{\text{H}}$  3.20 and 4.10)) and six quaternary

**Table 1**

$^{13}\text{C}$  NMR data for compounds **1**<sup>a</sup>, **2**<sup>a</sup>, **3**<sup>b</sup>, **4**<sup>c</sup>, **11**<sup>c</sup> and **15**<sup>c</sup> (100 MHz)<sup>d</sup>.

N° C	1	2	3	4	11	15
1	38.5	38.1	38.3	39.2	21.0	22.4
2	27.0	23.3	23.3	28.1	37.0	40.9
3	78.7	80.8	81.2	78.8	71.2	211.7
4	39.1	38.0	38.2	39.5	53.3	59.0
5	60.6	60.6	60.5	55.9	38.3	41.6
6	68.8	68.5	67.6	19.0	41.9	40.2
7	46.7	46.7	45.8	33.4	18.0	17.2
8	42.1	42.0	41.9	39.8	52.2	52.5
9	49.9	49.8	49.9	48.2	51.5	51.4
10	39.3	39.1	39.1	37.6	58.2	57.0
11	20.8	20.8	20.5	24.4	28.2	28.8
12	25.0	24.9	25.2	123.1	31.2	31.6
13	37.6	37.5	37.9	144.3	39.2	39.4
14	43.0	43.0	42.9	41.4	38.2	38.4
15	27.4	27.4	27.5	28.3	31.4	31.7
16	35.5	35.5	35.4	24.0	35.7	35.1
17	42.9	42.8	42.8	46.4	30.0	30.1
18	48.3	48.2	48.2	43.1	42.8	42.7
19	47.9	47.9	47.9	42.2	35.5	34.9
20	150.8	150.7	150.6	44.2	28.1	28.1
21	29.8	29.7	29.6	30.7	32.8	31.9
22	39.9	39.9	39.8	33.8	38.9	39.0
23	30.9	30.6	30.1	29.3	10.6	7.2
24	15.5	16.5	16.5	17.1	14.7	15.9
25	17.1	17.2	16.0	16.2	204.7	204.9
26	17.5	17.5	16.8	17.7	19.6	19.7
27	14.5	14.5	13.8	26.5	18.5	18.4
28	18.0	18.0	17.2	175.8	31.9	31.7
29	109.4	109.4	109.0	28.8	35.1	35.1
30	19.3	19.2	18.3	177.3	31.6	31.6
		Ac	Caffeoyl	Glc		
1'			126.5	95.1		
2'			113.9	73.4		
3'			145.6	77.7		
4'			148.4	70.6		
5'			115.3	77.4		
6'			121.7	61.7		
7'			145.5			
8'			114.4			
9'			168.1			
3-OCOCH <sub>3</sub>		21.2				
30-CO-OCH <sub>3</sub>		171.0		52.7		

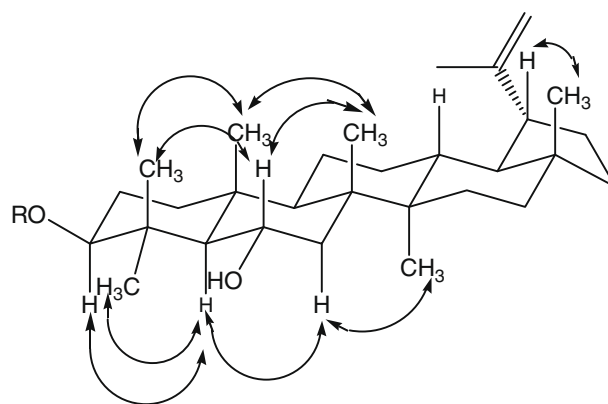
<sup>a</sup>  $\text{CDCl}_3$ .

<sup>b</sup>  $\text{CD}_3\text{OD}$ .

<sup>c</sup>  $\text{C}_5\text{D}_5\text{N}$ .

<sup>d</sup> Assignments were made on the basis of DEPT,  $^1\text{H}$ – $^1\text{H}$  COSY, NOESY, HMQC and HMBC experiments.

carbons (Table 1). The signals exhibited at  $\delta_{\text{C}}$  109.4 and 150.8 confirmed **1** to be a lupane-type triterpene bearing two hydroxyl



**Fig. 1.** Important NOESY correlations in compounds **1**; **2** and **3**.

groups. The EIMS of **1** exhibited key peaks at  $m/z$  218, 205, 203 and 189, indicating that the two hydroxyl groups are located on rings A and B. The determination of the positions of both hydroxyl groups is based on the HMBC and NOESY experiments. The HMBC spectrum of **1** showed correlations between the oxymethine proton signal at  $\delta_H$  3.20 and the carbon signals at  $\delta_C$  27.0 (C-2), 39.1 (C-4), 60.6 (C-5), 30.9 (C-23), and 15.5 (C-24), confirming the location of one hydroxyl group at C-3. Interactions were also exhibited between the second oxymethine proton signal at  $\delta_H$  4.10 and the carbon signals at  $\delta_C$  39.1 (C-4), 60.6 (C-5), 46.7 (C-7), 42.1 (C-8), and 39.3 (C-10) suggesting that the second hydroxyl group was connected to C-6. The NOESY (Fig. 1) cross-peaks exhibited between both protons H-3 $\alpha$  and H-5 ( $\delta$  0.78), in addition to the coupling constant of H-3 $\alpha$  ( $J$  = 10.7, 5.6 Hz) confirmed the  $\beta$ -orientation of the 3-hydroxyl group; the NOESY spectrum also showed correlations between the proton signal at  $\delta_H$  4.10 (H-6 $\beta$ ) and the methyl proton signals at  $\delta_H$  0.98 (CH<sub>3</sub>-24), 0.85 (CH<sub>3</sub>-25) and 1.08 (CH<sub>3</sub>-26), confirming the  $\alpha$ -equatorial orientation of the 6-hydroxyl group. The configuration of C-6 was also supported by the coupling constant of H-6 $\beta$  ( $J$  = 9.6, 4.3 Hz) and the  $^{13}\text{C}$  NMR spectrum of **1** which exhibited a key and characteristic signal at  $\delta_C$  60.6 for carbon C-5 as deduced from the HMQC spectrum; this value was 4 ppm higher than that of some reported data for C-5 (56.6 ppm) in similar lupane-type triterpenes with a 6 $\beta$ -OH group (Núñez et al., 2005). Therefore, the structure of compound **1** was established as, lup-20(29)-ene-3 $\beta$ ,6 $\alpha$ -diol. The same 3 $\beta$ ,6 $\alpha$ -structure was postulated before for a metabolite isolated from *Periploca aphylla* (Ghulam et al., 2000). However, the configuration of C-6 subsequently corresponded to lup-20(29)-en-3 $\beta$ ,6 $\beta$ -diol (Núñez et al., 2005).

Compound **2** was obtained as a colourless amorphous solid, and was deduced to have the molecular formula C<sub>32</sub>H<sub>52</sub>O<sub>3</sub> on the basis of CI/NH<sub>3</sub> MS ( $m/z$  485 [M+H]<sup>+</sup>, 502 [M + NH<sub>4</sub>]<sup>+</sup>, and the HR TOF MS ES<sup>+</sup> ( $m/z$  484.3934). The IR spectrum of **2** showed signals assigned to hydroxyl (3400 cm<sup>-1</sup>), ester (1740 cm<sup>-1</sup>) and olefinic (1668 cm<sup>-1</sup>) groups. On comparison, the  $^1\text{H}$  NMR spectra of compounds **2** and **1** were almost identical, apart from the change of chemical shift of H-3 from  $\delta_H$  3.20 in **1** to  $\delta_H$  4.42 in **2**, and the presence of one additional acetyl proton singlet at  $\delta_H$  2.04 in **2**. The second oxymethine proton at  $\delta_H$  4.10 in **1** did not change in **2** ( $\delta_H$  4.03). In the  $^{13}\text{C}$  NMR (Table 1) the carbon C-3 signal,  $\delta_C$  78.7 in **1**, changed to  $\delta_C$  80.8 in **2**. Thus, compound **2** was deduced to be the 3-monoacetylated derivative of **1**. On the basis of the  $^1\text{H}$ - $^1\text{H}$  COSY and NOESY (Fig. 1) spectra, the stereochemistry of the two oxymethine carbons C-3 and C-6 in both compounds **2** and **1** were confirmed to be the same. Accordingly, the structure of **2** was established as 3 $\beta$ -acetoxylup-20(29)-en-6 $\alpha$ -ol.

Compound **3** was obtained as colourless crystals. The pseudo molecular ion peaks at  $m/z$  605 [M+H]<sup>+</sup> and 622 [M + NH<sub>4</sub>]<sup>+</sup> in its CI/NH<sub>3</sub> MS, and the HR TOF MS ES<sup>+</sup> at  $m/z$  604.4105 suggested its molecular formula to be C<sub>39</sub>H<sub>56</sub>O<sub>5</sub>. The IR spectrum showed absorption bands assigned to hydroxyl (3500, 3310 cm<sup>-1</sup>), ester (1680 cm<sup>-1</sup>), and aromatic (1600 cm<sup>-1</sup>) groups. The  $^1\text{H}$  NMR spectral data of **3** showed two sets of signals: the first set of signals were analysed as an (*E*)-caffeoyl moiety [ $\delta_H$  6.22 (H-7'), 7.50 (H-8'), and three aromatic protons at  $\delta_H$  6.75 (H-5'), 7.01 (H-2') and 6.92 (H-6')]; the second set of signals in **3** were almost similar to those of **1**, except few modifications observed on the oxymethine protons,  $\delta_H$  3.20 in **1** and  $\delta_H$  4.50 in **3**. The second oxymethine proton at  $\delta_H$  4.10 in **1** did not change in **3** ( $\delta_H$  4.00). The  $^{13}\text{C}$  NMR (Table 1) confirmed the presence of an (*E*)-caffeoyl moiety in **3** ( $\delta_C$  113.9, 114.4, 115.5, 121.7, 126.5, 145.5, 145.6, 148.4 and 168.1). The carbon C-3 signal at  $\delta_C$  78.7 in **1** changed to  $\delta_C$  81.2 in **3**. Thus, compound **3** was deduced to be the 3-(*E*)-caffeoyl derivative of **1**. On the basis of the  $^1\text{H}$ - $^1\text{H}$  COSY and NOESY (Fig. 1) spectra, the stereochemistry of the two oxymethine carbons (C-3 and C-6) in both compounds **3** and **1** were confirmed to be the same.

Therefore, the structure of **3** was established as 3 $\beta$ -caffeoyloxy-lup-20(29)-en-6 $\alpha$ -ol.

The molecular formula of compound **4** was deduced as C<sub>37</sub>H<sub>58</sub>O<sub>10</sub> from the FAB MS and  $^{13}\text{C}$  NMR data. The positive FAB MS of **4** revealed a quasi-molecular ion at  $m/z$  669.5 [M + Li]<sup>+</sup>. The  $^1\text{H}$  NMR of **4** showed the presence of six tertiary methyl singlets at  $\delta_H$  0.80–1.19 (each, 3H, s), a doublet of doublets at  $\delta_H$  2.73 (1H,  $J$  = 3.65, 14.35 Hz, H-18) and a triplet at  $\delta_H$  5.36 (1H,  $J$  = 3.42 Hz, H-12). These signals and the  $^{13}\text{C}$  NMR signals (Table 1) at  $\delta_C$  123.1 and 144.3 were in agreement with reported data of olean-12-ene type triterpenes. The  $^{13}\text{C}$  NMR and DEPT spectra of **4** showed an oxymethine carbon signal at  $\delta_C$  78.8 and two C=O groups at  $\delta_C$  175.8 and 177.3. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR signals at  $\delta_H$  3.73 (3H, s) and  $\delta_C$  52.7 indicated that one of them was present as a carbomethoxyl group. Moreover, the  $^{13}\text{C}$  NMR resonances of the carbons C-12, C-13, C-14, C-17 and C-20 were identical with spectral values of compounds having carboxyl functions at C-17 and C-20 (Hassanean and Mohamed, 1998). The HMBC spectrum showed key correlations between the proton H-18 ( $\delta_H$  2.73) and carbons C-28 ( $\delta_C$  177.3), C-20 ( $\delta_C$  44.2), and between the proton H-1' ( $\delta_H$  5.37) and carbon C-28 ( $\delta_C$  177.3), confirming the position of connectivity of the  $\beta$ -glucopyranosyl ester to be to the 28-carboxyl group. Consequently, the carbomethoxyl was deduced to be connected at C-20, and its position was established to be C-30, as deduced from the  $^{13}\text{C}$  NMR spectrum of **4** which exhibited resonance for C-29 methyl at  $\delta_C$  28.8 in agreement literature (Hassanean and Mohamed, 1998). In addition, alkaline saponification of **4** gave glucose and the corresponding sapogenin which was identical to compound **5**, isolated from the same plant and identified as serjanic acid (**5**) (Javasinghe et al., 1993). Accordingly, compound **4** was elucidated as, 28- $\beta$ -D-glucopyranosyl-30-methyl 3 $\beta$ -hydroxyolean-12-en-28,30-dioate.

Compound **11** was obtained as colourless amorphous powder. Its molecular formula C<sub>30</sub>H<sub>50</sub>O<sub>2</sub> was established on the basis of the HR TOF MS ES<sup>+</sup>,  $m/z$  442.3821, the CI/NH<sub>3</sub> MS,  $m/z$  443 [M+H]<sup>+</sup>, 460 [M+NH<sub>4</sub>]<sup>+</sup> and NMR data. The  $^{13}\text{C}$  NMR (Table 1),  $^1\text{H}$  NMR, HMQC, HMBC and DEPT spectra suggested a friedelin-type triterpene skeleton containing one oxymethine ( $\delta_C$  71.2 and  $\delta_H$  3.49) and one aldehyde function ( $\delta_C$  204.4 and  $\delta_H$  10.19). From the GC-SM, the fragment ion at  $m/z$  205 suggested the absence of oxygen function on rings D and E. Also, the fragments at  $m/z$  125 and 315 resulting from the cleavage of ring B suggested the location of one oxygen function on ring A. The HMBC spectrum showed correlations between the proton signal ( $\delta_H$  3.49) and carbons C-4 ( $\delta_C$  53.3), C-5 ( $\delta_C$  38.3) and C-23 ( $\delta_C$  10.6), and between the aldehyde proton signal at  $\delta_H$  10.19 and the carbons C-8 ( $\delta_C$  52.2), C-9 ( $\delta_C$  51.5), C-10 ( $\delta_C$  58.2) and C-11 ( $\delta_C$  28.2). These data confirmed the position of the hydroxyl group at C-3 and the aldehyde function (C-25) at C-9. The NOESY spectrum of **11** (Fig. 2) showed interactions between the aldehyde proton ( $\delta_H$  10.19) and the methyl groups CH<sub>3</sub>-24 ( $\delta_H$  0.65) and CH<sub>3</sub>-26 ( $\delta_H$  0.95). The NOESY cross-peaks from H-3 ( $\delta_H$  3.49) to CH<sub>3</sub>-23 ( $\delta_H$  1.01) and CH<sub>3</sub>-24 ( $\delta_H$  0.65), in addition to the coupling constant of H-3 ( $J$  = 10.0,

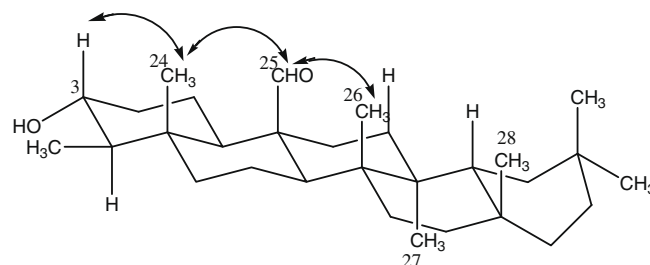


Fig. 2. Important NOESY correlations in compound **11**.

**Table 2**  
Antimicrobial activities of compounds **1**, **2**, **3**, **4** (each conc. 200 mg/l in DMSO).

Micro-organisms used	Inhibition zone diameter (mm)				
	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	Gentamicin (control)
<i>Staphylococcus aureus</i> Gram-(+)	15			11	34
<i>Escherichia coli</i> Gram-(–)				14	35
<i>Salmonella typhi</i> Gram-(–)				13	42
<i>Shigella dysenteriae</i> Gram-(–)					30
<i>Klebsiella pneumoniae</i> Gram-(–)					40
<i>Pseudomonas aeruginosa</i> Gram-(–)					43

4.0 Hz) confirmed the  $\beta$ -axial orientation of H-3 and consequently the  $\alpha$ -equatorial orientation of the hydroxyl group in agreement with reported data (Salazar et al., 2000). Therefore, the structure of **11** was established as 3 $\alpha$ -hydroxyfriedelan-25-al.

From the antimicrobial test results (Table 2), it appears that compound **1** exhibits antimicrobial activity against *Staphylococcus aureus*. Compounds **2** and **3** which are the 3-acylated derivatives of compound **1** showed no inhibitory activity on the six bacterial strains. Compound **4** reveals antimicrobial activity against *S. aureus*, *Escherichia coli* and *Salmonella typhi*. The activities of both compounds **1** and **4** were lower in comparison to that of gentamicin which was used as control.

### 3. Experimental

#### 3.1. General

MPs were determined using a Kofler microhot stage apparatus. IR spectroscopy was performed on a Perkin–Elmer 257 spectrometer. Specific rotations were measured on a Perkin–Elmer 241 polarimeter. MS were registered on a Micromass Q-ToF instrument, on a Nermag R10-10C spectrometer and a HP-5973 Mass Selective Detector. NMR experiments were performed on a Varian Gemini 400 MHz instrument and a Bruker AC 400 spectrometer, the residual solvent signal was taken as reference in each case (CDCl<sub>3</sub>, CD<sub>3</sub>OD and C<sub>5</sub>D<sub>5</sub>N). Si gel 60 (240–400 mesh) was used for CC at normal pressure while Si gel 60 H (5–40  $\mu$ m) and Si gel 60 C (20–40  $\mu$ m) were used for CC under compressed air (300 mbar). Pre-coated Si gel 60 F<sub>254</sub> aluminium plates were used for TLC.

#### 3.2. Plant material

The whole stems and the ripe mature fruit of *D. inaequalis* Hutch. (Euphorbiaceae) were collected from Eloundem (Centre province of Cameroon) in August 2004. The herbarium specimen documenting the collection has been deposited in the National Herbarium, Yaoundé, Cameroon (Ref 4981/SRFK).

#### 3.3. Extraction and isolation

The whole stems and the ripe fruit of *D. inaequalis* were sun-dried and ground separately into a powder form. The ground stems (10.0 kg) were macerated at room temperature with a mixture of CH<sub>2</sub>Cl<sub>2</sub>–MeOH (1:1) (3  $\times$  25 l) for 9 days. The solvents were evaporated under reduced pressure to yield the total crude extract (390.0 g). Part of the extract (250.0 g) was subjected to CC over Si gel [60 (240–400 mesh), 800 g]. A total of 75 fractions (400 ml each) were eluted with hexane, CH<sub>2</sub>Cl<sub>2</sub> and MeOH in increasing polarity. TLC permitted the combination the resulting fractions into 8 groups of fractions coded A, B, C, D, E, F, G and H, obtained as follow: A (5.0 g) [Fr. 1–5 (hexane–CH<sub>2</sub>Cl<sub>2</sub> 100:0 to 75:25)]; B (30 g) [Fr. 6–12 (hexane–CH<sub>2</sub>Cl<sub>2</sub> 70:30 to 50:50)]; C (35 g) [Fr.

13–20 (hexane–CH<sub>2</sub>Cl<sub>2</sub> 45:55 to 25:75)]; D (27 g) [Fr. 21–27 (hexane–CH<sub>2</sub>Cl<sub>2</sub> 20:80 to 0:100)]; E (25 g) [Fr. 28–35 (CH<sub>2</sub>Cl<sub>2</sub>–MeOH 100:0 to 95:5)]; F (30 g) [Fr. 36–45 (CH<sub>2</sub>Cl<sub>2</sub>–MeOH 93:7 to 90:10)]; G (36 g) [Fr. 46–57 (CH<sub>2</sub>Cl<sub>2</sub>–MeOH 88:12 to 80:20)] and H (50 g) [Fr. 58–75 (CH<sub>2</sub>Cl<sub>2</sub>–MeOH 75:25 to 50:50)]. Further CC over Si gel 60 C (20–40  $\mu$ m) of group C fractions using hexane–CH<sub>2</sub>Cl<sub>2</sub> (50:50) yielded compounds **2** (30 mg), **12** (40 mg), **13** (30 mg) and **16** (50 mg). Further CC over Si gel 60 C (20–40  $\mu$ m) of group D fractions using hexane–CH<sub>2</sub>Cl<sub>2</sub> (75:25) afforded compounds **1** (22 mg), **5** (15 mg) and **6** (70 mg). Further CC over Si gel 60 H (5–40  $\mu$ m) of group E fractions by using CH<sub>2</sub>Cl<sub>2</sub> (100%) yielded compounds **7** (20 mg), **8** (12 mg) and **9** (10 mg). Further CC over Si gel 60 H (5–40  $\mu$ m) of group F fractions using CH<sub>2</sub>Cl<sub>2</sub>–MeOH (95:5) yielded compounds **3** (20 mg), **4** (45 mg), **10** (15 mg) and **17** (20 mg). The ground fruit (2.0 kg) was macerated at room temperature with a mixture of EtOAc–MeOH (3:1), (3  $\times$  4 l) for 6 days. The solvents were evaporated under reduced pressure to yield the total crude extract (157.0 g) which was subjected to CC over Si gel [60 (230–400 mesh), 500 g]. A total of 144 fractions (200 ml each) were eluted with hexane, EtOAc and MeOH in increasing polarity. TLC permitted the combination the resulting fractions into five series of fractions coded A, B, C, D and E, obtained as follows: series A (10.0 g) [Fr. 1–10 (hexane–EtOAc, 100:0 to 75:25)]; series B (15 g) [Fr. 11–19 (hexane–EtOAc 70:30 to 50:50)]; series C (10 g) [Fr. 20–50 (hexane–EtOAc 45:55 to 25:75)]; series D (50 g) [Fr. 51–102 (hexane–EtOAc 20:80 to 0:100)]; series E (55 g) [Fr. 103–144 (EtOAc–MeOH 100:0 to 50:50)]. Further CC over Si gel 60 C (20–40  $\mu$ m) of series A using hexane–EtOAc in increasing polarity yielded compounds **12** (50 mg) and **14** (17 mg). Repeated CC of series B over Si gel 60 C (20–40  $\mu$ m) using hexane–EtOAc in increasing polarity afforded compounds **11** (7 mg) and **15** (18 mg). Further CC of series C over Si gel 60 C (20–40  $\mu$ m) using hexane–EtOAc in increasing polarity yielded compound **18** (60 mg). Repeated CC over Si gel 60 C (20–40  $\mu$ m) of series D using EtOAc–MeOH in increasing polarity afforded compounds **10** (65 mg) and **19** (200 mg).

##### 3.3.1. lup-20(29)-ene-3 $\beta$ ,6 $\alpha$ -diol (**1**)

Colourless amorphous solid; [ $\alpha$ ]<sub>D</sub><sup>20</sup> = +24.5° (CHCl<sub>3</sub>, c 0.60); IR (KBr)  $\nu_{\max}$  cm<sup>–1</sup>: 3400, 3030, 1660, 1260, 1180, 890; <sup>1</sup>H NMR spectral data (400 MHz, CDCl<sub>3</sub>):  $\delta$  3.20 (1H, dd, *J* = 5.6, 10.7 Hz, H-3), 4.10 (1H, dt, *J* = 4.3, 9.6 Hz, H-6), 4.58 (1H, d, *J* = 2.3 Hz, H-29), 4.68 (1H, d, *J* = 2.3 Hz, H-29), 2.35 (1H, m, H-19), 1.68 (3H, s, CH<sub>3</sub>-30), 1.32 (3H, s, CH<sub>3</sub>-23), 1.08 (3H, s, CH<sub>3</sub>-26), 0.98 (3H, s, CH<sub>3</sub>-24), 0.96 (3H, s, CH<sub>3</sub>-27), 0.85 (3H, s, CH<sub>3</sub>-25), 0.78 (m, H-5), 0.75 (3H, s, CH<sub>3</sub>-28), 1.40 and 1.67 (m, H-7); <sup>13</sup>C NMR spectral data (100 MHz, CDCl<sub>3</sub>), see Table 1; Cl/NH<sub>3</sub> MS, *m/z*: 443 [M+H]<sup>+</sup>, 460 [M+NH<sub>4</sub>]<sup>+</sup>; EIMS (70 eV) *m/z* 442 [M]<sup>+</sup> (10), 424 [M–H<sub>2</sub>O]<sup>+</sup> (20), 409 [M–H<sub>2</sub>O–CH<sub>3</sub>]<sup>+</sup> (5), 406 [M–2H<sub>2</sub>O]<sup>+</sup> (3), 236 (6), 218 (20), 205 (40), 203 (12), 189 (50); HR TOF MS ES<sup>+</sup> (calcd. for C<sub>30</sub>H<sub>50</sub>O<sub>2</sub> 442.3811, found 442.3825).

##### 3.3.2. 3 $\beta$ -acetoxylup-20(29)-en-6 $\alpha$ -ol (**2**)

Colourless amorphous solid; [ $\alpha$ ]<sub>D</sub><sup>20</sup> = +18.5° (CHCl<sub>3</sub>, c 0.80); IR (KBr)  $\nu_{\max}$  cm<sup>–1</sup>: 3400, 3000, 1740, 1668, 1255, 880; <sup>1</sup>H NMR spectral data (400 MHz, CDCl<sub>3</sub>):  $\delta$  4.42 (1H, dd, *J* = 5.5, 10.5 Hz, H-3), 4.03 (1H, dt, *J* = 4.2, 9.5 Hz, H-6), 4.56 (1H, d, *J* = 2.3 Hz, H-29), 4.68 (1H, d, *J* = 2.3 Hz, H-29), 2.35 (1H, m, H-19), 1.68 (3H, s, CH<sub>3</sub>-30), 1.16 (3H, s, CH<sub>3</sub>-23), 1.10 (3H, s, CH<sub>3</sub>-26), 1.04 (3H, s, CH<sub>3</sub>-24), 0.95 (3H, s, CH<sub>3</sub>-27), 0.91 (3H, s, CH<sub>3</sub>-25), 0.78 (3H, s, CH<sub>3</sub>-28), 2.04 (3H, s, COCH<sub>3</sub>-3); <sup>13</sup>C NMR spectral data (100 MHz, CDCl<sub>3</sub>), see Table 1; Cl/NH<sub>3</sub> MS, *m/z* 485 [M+H]<sup>+</sup>, 502 [M+NH<sub>4</sub>]<sup>+</sup>; EIMS (70 eV) *m/z* 484 [M]<sup>+</sup> (7), 466 (8), 426 (10), 408 (5), 383 (5), 218 (100), 205 (40), 203 (12), 189 (12); HR TOF MS ES<sup>+</sup> (calcd. for C<sub>32</sub>H<sub>52</sub>O<sub>3</sub> 484.3916, found 484.3934).



### 3.3.3. 3 $\beta$ -caffeoyloxylyp-20(29)-en-6 $\alpha$ -ol (**3**)

Colourless crystals; mp 272.2–273.7 °C;  $[\alpha]_D^{20} = +105.4^\circ$  (CHCl<sub>3</sub>, c 0.08); IR (KBr)  $\nu_{\max}$  cm<sup>-1</sup>: 3500, 3310, 2930, 2860, 1680, 1600, 1255, 1180, 880; UV (CH<sub>3</sub>OH)  $\lambda_{\max}$  nm log( $\epsilon$ ): 303.6 (0.504), 327.6 (0.636); <sup>1</sup>H NMR spectral data (400 MHz, CD<sub>3</sub>OD): triterpene moiety:  $\delta$  4.50 (1H, dd,  $J$  = 5.70, 10.50 Hz, H-3), 4.00 (1H, (1H, dt,  $J$  = 4.10, 9.50 Hz, H-6), 4.55 (1H, d,  $J$  = 2.30 Hz, H-29), 4.68 (1H, d,  $J$  = 2.30 Hz, H-29), 2.39 (1H, m, H-19), 1.68 (3H, s, CH<sub>3</sub>-30), 1.17 (3H, s, CH<sub>3</sub>-23), 1.14 (3H, s, CH<sub>3</sub>-26), 0.96 (3H, s, CH<sub>3</sub>-24), 1.02 (3H, s, CH<sub>3</sub>-27), 1.12 (3H, s, CH<sub>3</sub>-25), 0.82 (3H, s, CH<sub>3</sub>-28); caffeoyl moiety:  $\delta$  7.01 (1H, d,  $J$  = 1.83 Hz, H-2'), 6.75 (1H, d,  $J$  = 8.06 Hz, H-5'), 6.92 (1H, dd,  $J$  = 2.00, 8.24 Hz, H-6'), 7.50 (1H, d,  $J$  = 15.74 Hz, H-7'), 6.22 (1H, d,  $J$  = 15.74 Hz, H-8'); <sup>13</sup>C NMR spectral data (100 MHz, CD<sub>3</sub>OD), see Table 1; CI/NH<sub>3</sub> MS,  $m/z$ : 605 [M+H]<sup>+</sup>, 622 [M+NH<sub>4</sub>]<sup>+</sup>; EIMS (70 eV)  $m/z$  604 [M]<sup>+</sup> (25), 239(10), 221 (15), 203 (65), 163 (100); HR TOF MS ES<sup>+</sup> (calcd. for C<sub>39</sub>H<sub>56</sub>O<sub>5</sub> 604.4128, found 604.4105).

### 3.3.4. 28- $\beta$ -D-glucopyranosyl-30-methyl 3 $\beta$ -hydroxyolean-12-en-28,30-dioate (**4**)

White crystals from CH<sub>2</sub>Cl<sub>2</sub>; mp 230–232 °C;  $[\alpha]_D^{20} = +45.5^\circ$  (MeOH, c 0.75); IR (KBr)  $\nu_{\max}$  cm<sup>-1</sup>: 3450, 1720 (COOR), 1710 (COOCH<sub>3</sub>); <sup>1</sup>H NMR spectral data (400 MHz, C<sub>5</sub>D<sub>5</sub>N)  $\delta$ : 5.37 (1H, d,  $J$  = 8.00 Hz, H-1'), 5.36 (1H, t,  $J$  = 3.42 Hz, H-12), 3.73 (3H, s, COOCH<sub>3</sub>), 3.19 (1H, t,  $J$  = 8.00 Hz, H-3), 2.73 (1H, dd,  $J$  = 3.65, 14.35 Hz, H-18), 1.19 (3H, s, CH<sub>3</sub>), 1.17 (3H, s, CH<sub>3</sub>), 1.00 (3H, s, CH<sub>3</sub>), 0.98 (3H, s, CH<sub>3</sub>), 0.82 (3H, s, CH<sub>3</sub>), 0.80 (3H, s, CH<sub>3</sub>); <sup>13</sup>C NMR spectral data (100 MHz, C<sub>5</sub>D<sub>5</sub>N) data, see Table 1. CI/NH<sub>3</sub> MS,  $m/z$  663 [M+H]<sup>+</sup>, 680 [M+NH<sub>4</sub>]<sup>+</sup>; FAB/NBA + Li: ion mode FAB<sup>+</sup>,  $m/z$ : 669.5 [M+Li]<sup>+</sup>, 630, 625, 581.6, 580.6, 460.3, 397.5, 307.0, 292.1, 291.1, 290.0, 154.1. Alkaline hydrolysis of compound (**4**): The glycoside **4** (15 mg) was refluxed with 5% KOH for about 5 h. After completion, the reaction mixture was neutralized with diluted H<sub>2</sub>SO<sub>4</sub> and extracted with *n*-BuOH. Work up of the *n*-BuOH soluble portion yielded glucose identified by <sup>1</sup>H and <sup>13</sup>C NMR to the available authentic compound. The purification of the aqueous fraction afforded the aglycone, identical to serjanic acid (**5**), also isolated from the same plant.

### 3.3.5. Serjanic acid (3 $\beta$ -hydroxyolean-12-en-28,30-dioic acid 30-methyl ester) (**5**)

White crystals from CH<sub>2</sub>Cl<sub>2</sub>; mp 280–281 °C; CI/NH<sub>3</sub> MS,  $m/z$ : 501 [M+H]<sup>+</sup>, 518 [M+NH<sub>4</sub>]<sup>+</sup>; EI MS (probe) 70 eV,  $m/z$ : 500 [M]<sup>+</sup> (15), 454 (15.4), 292 (64.6), 247 (41.5), 246 (100.0), 233 (15.4), 232 (17.7), 207 (33.8), 187 (93.8), 186 (46.2), 173 (23.1), 159 (21.5).

### 3.3.6. 3 $\alpha$ -hydroxyfriedelane-25-al (**11**)

Colourless amorphous powder;  $[\alpha]_D^{20} = +14.5^\circ$  (CHCl<sub>3</sub>, c 0.60); IR (KBr)  $\nu_{\max}$ : 3400, 3030, 1720, 1260, 1180, 890 cm<sup>-1</sup>; <sup>1</sup>H NMR spectral data (400 MHz, CDCl<sub>3</sub>):  $\delta$  10.19 (1H, s, H-25), 3.49 (1H, dt,  $J$  = 10.0, 4.0 Hz, H-3ax), 2.16 (qd,  $J$  = 13.0, 3.0, H-2eq), 1.19 (3H, s, CH<sub>3</sub>-28); 1.07 (3H, s, CH<sub>3</sub>-30); 1.01 (3H, d,  $J$  = 6.0 Hz, CH<sub>3</sub>-23); 0.95 (3H, s, CH<sub>3</sub>-26); 0.94 (3H, s, CH<sub>3</sub>-29); 0.93 (3H, s, CH<sub>3</sub>-27); 0.65 (3H, s, CH<sub>3</sub>-24); <sup>13</sup>C NMR spectral data (100 MHz, CDCl<sub>3</sub>), see Table 1. GC–SM  $m/z$ : [M]<sup>+</sup> 442, 315, 205, 125. CI/NH<sub>3</sub> MS,  $m/z$ : 443 [M+H]<sup>+</sup>, 460 [M+NH<sub>4</sub>]<sup>+</sup>.

## 3.4. Antimicrobial activity

### 3.4.1. Microbial strains

A total of six micro-organisms belonging to one Gram-(+) bacterial species (*S. aureus*) and five Gram-(−) bacteria (*E. coli*, *S. typhi*, *Shigella dysenteriae*, *Klebsiella pneumoniae* and *Pseudomonas aeruginosa*) were clinically isolated from patients in the “Centre Pasteur de Yaoundé” Cameroon. They were maintained on agar slants at

4 °C in the Laboratory of the Applied Microbiology and Molecular Pharmacology (Faculty of Science, Yaounde).

### 3.4.2. Antimicrobial assays

Antimicrobial activity was evaluated using the agar diffusion method, according to the NCCLS (2002) protocol with slight modifications. Briefly, sterile cylinders of 6 mm were used to make wells inside Mueller-Hinton agar plates. The plates were inoculated with  $2 \times 10^4$  l of the test micro-organisms equivalent to  $5 \times 10^5$  CFU/ml. All the compounds were dissolved in DMSO or in heated sterilized distilled water at a concentration 200 mg/l. Wells were filled with  $15 \times 10^{-5}$  l of solution of each test compound, the positive control drug (gentamicin) and the negative control DMSO, and allowed to diffuse for 45 min at 4 °C. The plates were incubated at 37 °C for 24 h. The sensitivity was recorded by measuring the clear zone of growth inhibition around the wells (mm diameter). Each set was tested in triplicate.

## Acknowledgements

One of the authors (J. WANDJI) is grateful for a grant (F/2624-3F) from the International Foundation for Science (Sweden), and for the sponsorship of “Université Paris Descartes, France” during his multiple research visits in France.

## References

- Agrawal, P.K., Jain, D.C., 1992. <sup>13</sup>C NMR Spectroscopy of oleanane triterpenoids. Progress in NMR Spectroscopy 24, 1–90.
- Anjaneyulu, A.S.R., Narayana, R.M., 1980. Phytochemistry 19, 1163.
- Bouquet, A., Debray, L., 1974. Plantes Médicinales de la Côte-d'Ivoire. Travaux et documents de l'ORSTOM 32, 82–87.
- Chungag-Anye, N.B., Njamen, D., Dongmo, A.B., Wandji, J., Nguetefack, T.B., Wansi, J.D., Kamanyi, A., Fomum, Z.T., 2001. Anti-inflammatory and analgesic properties of the stem extract of *Drypetes molunduana* Pax and Hoffm. (Euphorbiaceae) in rats. Pharmaceutical and Pharmacological Letters 11, 61–63.
- Chungag-Anye, N.B., Njamen, D., Dongmo, A.B., Wandji, J., Fomum, Z.T., Nguetefack, T.B., Kamanyi, A., 2002. Anti-inflammatory and analgesic effects of *Drypetes molunduana* A, a sesquiterpene lactone from *Drypetes molunduana*. Pharmaceutical Biology 41 (1), 26–30.
- Dalziel, J.M., 1937. The Useful Plants of West Tropical Africa. The Crown Agents for the Colonies. Westminster, London, pp. 140–141.
- Ghulam, M., Erum, A., Saeed, A., Itrat, A., Habib, A., Abdul, M., Syed, S-u-H., Choudhary, M.I., 2000. Lupene-type triterpenes from *Periploca aphylla*. Journal of Natural Products 63, 881–883.
- Hassanean, H.A., Mohamed, M.H., 1998. Novel oleanene saponins from *Taverniera aegyptiaca* Boiss. Pharmazie 53, 195–199.
- Irvine, R.F., 1961. Woody Plant of Ghana. Oxford University Press, London, pp. 223–226.
- Javasinghe, U.L.B., Wannigama, G.P., MacLeod, J.K., 1993. Saponins of *Diploclisia glaucescens*. Natural Product Letters 2, 249–253.
- Li, L., Huang, X., Sattler, I., Fu, H., Grabley, S., Lin, W., 2006. Structure elucidation of a new friedelane triterpene from the mangrove plant *Hibiscus tiliaceus*. Magnetic Resonance in Chemistry 44, 624–628.
- Mahato, S.B., Kundu, A.P., 1994. <sup>13</sup>C NMR spectra of pentacyclic triterpenoids. A compilation and some salient features. Phytochemistry 37, 1517–1575.
- National Committee for Clinical Laboratory Standards (NCCLS), 2002. Methods for Dilution Antimicrobial Susceptibility Test for Bacteria that Grow Aerobically, third ed. Approved standard NCCLS document M100-812, Wayne, PA.
- Núñez, M.J., Carolina, P., Reyes, Ignacio, A., Jiménez, Laila, M., Bazzocchi, Isabel L., 2005. Lupane triterpenoids from *Maytenus* Species. Journal of Natural Products 68, 1018–1021.
- Salazar, G.C., Silva, G.D.F., Duarte, L.P., Vieira Filho, S.A., Lula, I.S., 2000. Two epimeric friedelane triterpenes isolated from *Matenus truncate* Reiss: <sup>1</sup>H and <sup>13</sup>C chemical shift assignments. Magnetic Resonance in Chemistry 38, 977–980.
- Srivastava, S.K., Jain, D.C., 1989. Triterpenoid saponins from plants of *Araliaceae*. Phytochemistry 28, 644–647.
- Walker, A.R., Sillans, R., Trochain, J.L., 1961. Les plantes utiles du Gabon. In: Paul Lechevalier (Ed.), 12-Rue de Tournon Paris VI, pp. 165–166.
- Wandji, J., Wansi, J.D., Fuendjie, V., Dagne, E., Mulholland, A.D., Tillequin, F., et al., 2000. Sesquiterpene lactone and friedelane derivative from *Drypetes molunduana*. Phytochemistry 54, 811–815.
- Wandji, J., Tillequin, F., Mulholland, A.D., Temgoua, D.A., Wansi, J.D., Seguin, E., Fomum, Z.T., 2003. Phenolic constituents from *Drypetes armoracia*. Phytochemistry 63, 453–456.
- Wansi, J.D., Wandji, J., Lallemand, M.-C., Chiozem, D.D., Samreen, Iqbal, M.C., Tillequin, F., Fomum, Z.T., 2007. Antileishmanial furanosesquiterpene and triterpenoids from *Drypetes chevalieri* Beille (Euphorbiaceae). BLACPM 6 (1), 5–10.